

1 **Ambient volatile organic compounds at a receptor site in the**
2 **Pearl River Delta region: variations, source apportionment and**
3 **effects on ozone formation**

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23 **Highlights**

- 24 • Ambient VOCs were measured by an online GC-MS/FID system.
- 25 • Solvent usage, LPG usage and vehicle exhaust are main VOC sources.
- 26 • The formation of O₃ is anthropogenic VOC-limited with dominant contributions
27 from aromatics.
- 28 • Frequent high-O₃ episodes revealed the importance of regional transport in
29 photochemical pollution.

30

31 **Abstract**

32 We present the continuously measurements of volatile organic compounds (VOCs)
33 at a receptor site (Wan Qing Sha, WQS) in the Pearl River Delta (PRD) region from
34 September to November of 2017. The average mixing ratios of total VOCs (TVOCs)
35 was 36.4 ± 27.9 ppbv with the dominant contribution from alkanes (55.6%), followed
36 by aromatics (33.3%). The diurnal variation of TVOCs showed a strong photochemical
37 consumption during daytime, which is related to the formation of ozone (O_3). Five VOC
38 sources were resolved by the positive matrix factorization (PMF) model, including
39 solvent usage (28.6%), liquid petroleum gas (LPG) usage (24.4%), vehicle exhaust
40 (21.0%), industrial emissions (13.2%) and gasoline evaporation (12.9%). The regional
41 transport air masses from the upwind cities of south China can result in the elevated
42 concentrations of TVOCs. Lower ratios of TVOCs/ NO_x (1.59 ± 1.12) suggested that
43 the O_3 formation regime at WQS site was VOC-limited, which also confirmed by a
44 photochemical box model with the master chemical mechanism (PBM-MCM).
45 Furthermore, the observation on high- O_3 episode days revealed that frequent O_3
46 outbreaks at WQS were mainly caused by the regional transport of anthropogenic VOCs
47 especially for aromatics and the subsequent photochemical reactions. This study
48 provides valuable information for policymakers to propose the effective control
49 strategies on photochemical pollution in a regional perspective.

50

51 **Keywords:** VOCs; PRD region; Source apportionment; O_3 ; Regional transport

52 **1. Introduction**

53 Volatile organic compounds (VOCs) play important roles in atmospheric
54 environmental chemistry because they can affect regional air quality and human health
55 and climate (Guo et al., 2017; Seinfeld and Pandis, 2006). VOCs can react with nitrogen
56 oxides ($\text{NO}_x = \text{NO} + \text{NO}_2$) in the presence of sunlight leading to the production of
57 tropospheric ozone (O_3) and thus trigger photochemical pollution (Atkinson, 2000).
58 VOCs such as aromatic compounds and biogenic species are significant precursors for
59 the formation of secondary organic aerosols (SOA) (Hallquist et al., 2009).

60 The sources of VOCs include biogenic and anthropogenic emissions. Biogenic
61 VOCs mainly emitted from terrestrial plants, marine phytoplankton and other natural
62 sources, which are regarded as the largest contributor of VOCs globally (Guenther et
63 al., 2012). In urban and rural areas, anthropogenic VOCs can be dominant (Guo et al.,
64 2017). Compared to biogenic VOCs, the sources of anthropogenic VOCs are often
65 complex including vehicular exhaust, fuel evaporation, industrial emissions and
66 biomass burning (Guo et al., 2017; Liu et al., 2008a). The emissions of anthropogenic
67 VOCs depend largely on the energy consumption and industries. In many Chinese
68 urban areas, vehicular emission is widely reported as one of the major sources of
69 anthropogenic VOCs (Cai et al., 2010; Guo et al., 2011; Li et al., 2016; Lyu et al., 2016),
70 while industrial emissions are the largest contributor of anthropogenic VOCs in highly
71 industrialized regions (An et al., 2014; Mo et al., 2017). Urban and industrial emissions
72 can contribute to the VOCs in downwind rural regions through long-range or regional
73 transport (He et al., 2019; Russo et al., 2010b; Song et al., 2020; Tang et al., 2009).

74 It is well known that the photochemical formation of O₃ is non-linearly related to
75 the precursors of VOCs and NO_x. The formation of O₃ is usually found to be VOC-
76 limited in highly-polluted urban areas. While in the downwind rural regions, the O₃-
77 VOC-NO_x relationship is not always constant with the changes of meteorological
78 conditions such as tropical cyclones, continental anticyclones and mesoscale
79 circulations (e.g., sea-land breezes) (Ding et al., 2004; Guo et al., 2009; Wang et al.,
80 2017; Zeren et al., 2019). The concentrations and photochemical reactivity of the VOC
81 species are key factors in terms of the impacts on O₃ formation. Ozone formation
82 potential (OFP) is widely used as a parameter to evaluate the combination effects by
83 multiplying VOC concentration and its corresponding maximum incremental reactivity
84 (MIR) factor (Carter, 1994). In addition, the observation-based model (OBM) coupled
85 with Master Chemical Mechanisms (MCM) is developed as a powerful tool to
86 investigate the O₃-VOC-NO_x relationship in recent years (He et al., 2019; Ling and Guo,
87 2014; Wang et al., 2017; Zeren et al., 2019). It can simulate the O₃ photochemical
88 production and destruction based on observed ambient concentrations of O₃ and its
89 precursors. He et al., (2019) used a photochemical box model incorporating the master
90 chemistry mechanism (PBM-MCM) to investigate the photochemical formation of O₃
91 at a downwind rural site (Heshan) in the Pearl River Delta (PRD) during autumn 2014,
92 and found that O₃ formation was VOC-limited. Wang et al., (2017) applied the PBM-
93 MCM investigated the O₃ photochemistry at a downwind coastal site over the South
94 China Sea (SCS). They revealed that the O₃ formation changed from transition (VOC
95 and NO_x-limited jointly) to VOC-limited regime under the influence of sea-land breezes

96 from non-O₃ episode days to O₃ episode days.

97 As one of the most developed city clusters in China, the PRD region is densely
98 populated and rapidly developed with the pollution of O₃ and particulate matter {Lai,
99 2016 #174}. Over the last decade, a series of field campaigns and modeling studies
100 were performed in this region to understand the characteristics of VOCs including
101 abundances, composition, temporal variations, source contributions and chemical
102 reactivity, as well as the role in photochemical pollutant production (Barletta et al., 2008;
103 Guo et al., 2011; Guo et al., 2009; Ling et al., 2011; Liu et al., 2008b; Louie et al., 2013;
104 Yuan et al., 2013; Zeng et al., 2019; Zhang et al., 2013; Zou et al., 2015). Although
105 significant advances have obtained to understand the photochemical pollution
106 formation processes and to implement the control strategies, the PRD region is still
107 suffered from the frequent outbreaks of high-O₃ episodes. Even in the downwind areas
108 of PRD with sparse anthropogenic emissions, high-O₃ episodes were often observed as
109 well (He et al., 2019; Wang et al., 2017; Zeren et al., 2019). Here, we presented the
110 results of online VOCs observation at a receptor site in the PRD region in autumn 2017.
111 We intend (1) to characterize the concentration levels and diurnal variations of VOCs;
112 (2) to understand the contribution of VOC sources; and (3) to assess the impacts of
113 VOCs on O₃ formation with a focus on photochemical pollution periods.

114

115 **2. Experimental**

116 **2.1. Sampling site**

117 A field measurement was carried out at the site of Wan Qing Sha (WQS, 22°43'N,

118 113°33'E) from 1 September to 30 November 2017 (**Fig. S1**). WQS is a coastal site
119 located at the Pearl River Estuary (PRE) connecting the land areas of PRD and the SCS.
120 This site is located in the downwind area of urban Guangzhou (~50 km southeast to the
121 city center of Guangzhou) and surrounded by several fast-developing cities (Dongguan,
122 Shenzhen and Zhongshan, ~25-40 km). During autumn and winter seasons, this site can
123 be regarded as a receptor site and frequently influenced by the air masses carrying large
124 amounts of atmospheric pollutants from the upwind areas (Guo et al., 2011; Ling and
125 Guo, 2014). It is an ideal site to study the characteristics of regional pollution in the
126 PRD region.

127

128 **2.2. Sampling and chemical analysis**

129 Ambient VOCs were sampled and analyzed continuously by an online gas
130 chromatography-mass spectrometry/flame ionization detector system (GC-MS/FID,
131 Tianhong Instrument (TH-300B), Wuhan) with 1-h time resolution. The sampling tube
132 inlet was 1.5 m above the rooftop (~15 m), and the outlet was connected to a PFA-made
133 manifold with a by-pass pump drawing air at a rate of 15 L/min. Ambient air was
134 continuously drawn through a PFA tube with an inner diameter of 7.6 cm. The air was
135 then pre-concentrated by passing through a cold trap maintaining $-80\text{ }^{\circ}\text{C}$ for the
136 removal of water and carbon dioxide (CO_2), and then trapped at $-150\text{ }^{\circ}\text{C}$ with an empty
137 capillary column. After pre-concentration, the VOCs were desorbed by heating up to
138 $100\text{ }^{\circ}\text{C}$ rapidly and introduced into the GC-MS/FID system for analysis. The
139 Photochemical Assessment Monitoring Station (PAMS) gas was used for the daily

140 calibration of GC-MS/FID system at 22:00 to 0:00 local time. The coefficients (R^2) of
141 calibration curves were higher than 0.99 for each VOC species presented in this study.
142 The method detection limits (MDLs) for VOC species ranged from 0.003-0.047 ppbv.

143 The trace gases including O_3 , nitric oxide (NO), nitrogen dioxide (NO_2), carbon
144 monoxide (CO), and sulfur dioxide (SO_2) were simultaneously monitored at this
145 sampling site. Briefly, O_3 was monitored with a commercial UV photometric analyzer
146 (Thermo 49i) with the lower detection limit of 1.0 ppbv. NO and NO_x were detected by
147 a chemiluminescence analyzer (Thermo 42i) with the detection limit of 0.5 ppbv. CO
148 was measured by a nondispersive infrared analyzer (Thermo 48i). SO_2 was measured
149 by a pulsed UV fluorescence (Thermo 43i) with the detection limit of 0.5 ppbv. The
150 meteorological parameters (i.e., temperature, relative humidity, wind direction, wind
151 speed, and pressure) were recorded by a portable weather station (Model WXT520,
152 Vaisala, Finland).

153

154 **2.3. Models**

155 **2.3.1 Air mass back-trajectories**

156 To trace the potential source regions of atmospheric pollutants, air mass back-
157 trajectories are computed by using the Hybrid Single Particle Lagrange Integrated
158 Trajectory (HYSPLIT) model (Stein et al., 2015). In this study, three-day air mass back-
159 trajectories for each day were ran using HYSPLIT model with the arrival height of 200
160 m at every 6 hours for WQS site. Furthermore, we performed the air mass cluster
161 analysis to categorize the back-trajectories into three groups for source origin analysis.

162

163 **2.3.2 Positive matrix factorization (PMF) model**

164 In this study, US EPA PMF 5.0 model was used for the source apportionment of
165 VOCs (Norris, 2014). The PMF model is a multivariate factor analysis tool that
166 decomposes a matrix of sample data into two matrices of factor contributions and factor
167 profiles, which can be interpreted to explore the source types and contributions at a
168 receptor site (Paatero, 1997; Paatero and Tapper, 1994). The decomposition of sample
169 data matrix can be simplified as follows:

$$170 \quad x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (1)$$

171 where x_{ij} is the j th species concentration measured in the i th sample, g_{ik} is the species
172 contribution of the k th source to the i th sample, f_{kj} is the j th species fraction from the
173 k th source, e_{ij} is the residual for each species and p is the total number of independent
174 sources. Based on the least squares method, the number of sources and VOC source
175 profiles of each species in the individual source can be obtained from PMF model
176 calculation. The detailed operation of PMF model is given in supplement (**Text S1**).

177

178 **2.3.3 Photochemical box model (PBM)**

179 A photochemical box model incorporating the Master Chemical Mechanism (PBM-
180 MCM) was used to simulate the photochemical formation of O₃. This model has been
181 successfully applied to investigate the O₃ formation and its relationships with
182 precursors (VOCs and NO_x) in previous studies (He et al., 2019; Lam et al., 2013; Ling
183 and Guo, 2014; Lyu et al., 2016; Wang et al., 2017; Zeng et al., 2019). The sensitivity

184 of O₃ formation to its precursors was assessed using the relative incremental reactivity
185 (RIR), which can be calculated using the following equation (Carter and Atkinson
186 1989):

$$187 \quad RIR^S(X) = \frac{[P_{O_3-NO}^S(X) - P_{O_3-NO}^S(X-\Delta X)] / P_{O_3-NO}^S(X)}{\Delta S(X) / S(X)} \quad (1)$$

188 where $P_{O_3-NO}^S(X)$ and $P_{O_3-NO}^S(X-\Delta X)$ represent the O₃ formation potential in a base
189 run with original concentrations, and in a run with a hypothetical change (ΔX) (10% in
190 this study) in source/species X. In both runs, net O₃ production with titration by NO is
191 considered. $S(X)$ means the measured concentration of X, and ΔX is the concentration
192 change of X caused by the hypothetical change $\Delta S(X)$. According to equation (1), a
193 higher positive RIR value of a given precursor indicates a higher probability that O₃
194 production will be more effectively reduced by cutting down emissions of this precursor.
195 A more detailed description of PBM-MCM is shown in supplement (**Text S2**).

196

197 **3. Results and discussion**

198 **3.1. Data overview**

199 The time series of air pollutant concentrations including VOCs, O₃, NO_x, CO as
200 well as meteorological parameters during the campaign are shown **Fig. 1**. During the
201 whole measurement period, the ambient temperature ranged from 13.2 to 36.4 °C with
202 an average of 25.7 ± 4.6 °C. The relative humidity (RH) ranged from 24.0 % to 87.0 %
203 with an average of 63.0 ± 13.2 %. The wind speed varied between 0.3 m/s to 5.4 m/s
204 (average: 1.8 ± 0.8 m/s) with the prevailing wind direction of northeasterlies. The
205 average concentrations of NO_x and CO were 26.2 ± 19.3 ppbv and 497 ± 158 ppbv,

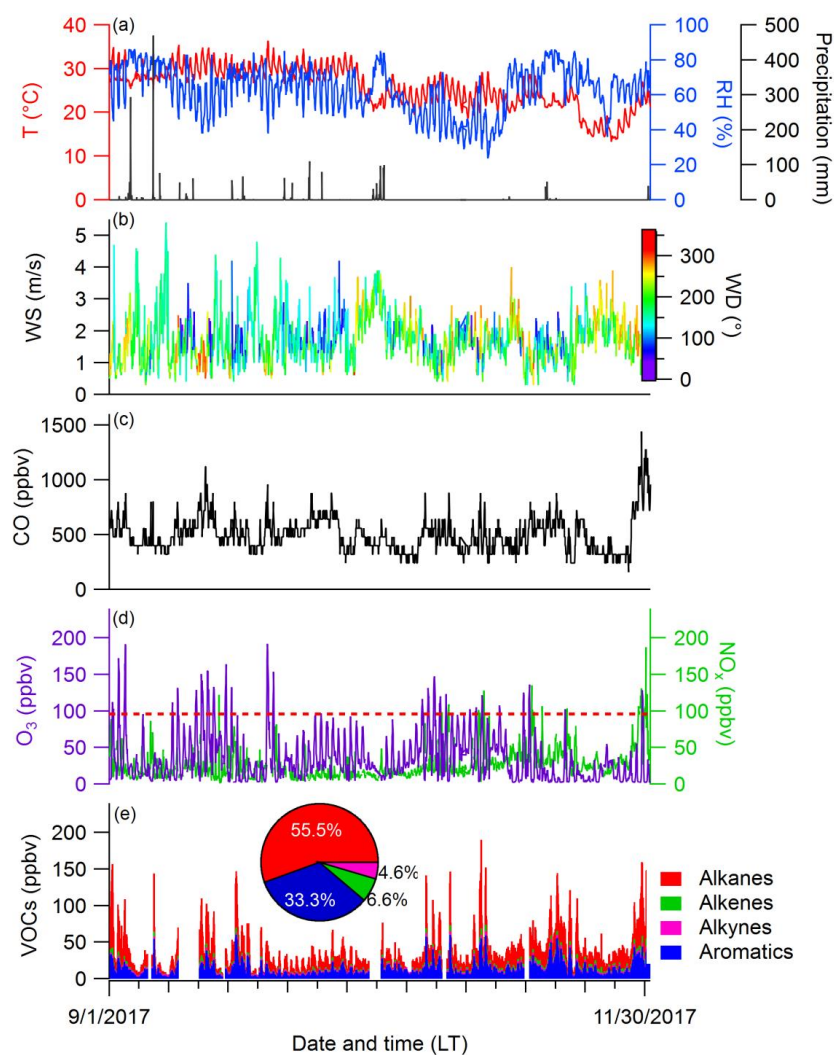
206 respectively. The average concentrations of O₃ ranged from 2.33 to 192 ppbv with an
207 average of 34.9 ± 32.2 ppbv. According to the National Ambient Air Quality Standard
208 II of China, high O₃ episodes are defined as the days with maximum hourly average
209 mixing ratio of O₃ exceeding ~ 100 ppbv. It can be seen clearly in **Fig. 1** that high-O₃
210 days occurred during the sampling period. Specifically, 26 high-O₃ days were observed
211 during 1-3, 11-12, 15-21, 27-28 September, 23-27, 31 October, 1-2, 5, 9-10, 16, 29
212 November (**Table S1**). During most of the high O₃ days, we also observed high
213 concentrations of VOCs, suggesting that the O₃ formation is probably VOC-limited at
214 WQS site. We will further discuss the effects of VOCs on O₃ formation in **Section 3.5**
215 to illustrate the causes of these photochemical pollution episodes.

216 The mixing ratios of total VOCs (TVOCs) ranged from 4.51 to 190 ppbv with an
217 average of 36.4 ± 27.9 ppbv (**Fig. 1**). Among the measured VOCs, the group of alkanes
218 accounts for the largest fraction of TVOCs (55.6%), followed by aromatics (33.3 %),
219 alkenes (6.6 %) and alkyne (4.6 %). The dominance of alkanes followed by aromatics
220 is consistent with previous results widely reported in Chinese cities (Cai et al., 2010;
221 Liu et al., 2008b; Louie et al., 2013). The top 10 VOC species were ethane, propane, *n*-
222 butane, *i*-butane, *i*-pentane, *n*-pentane, ethyne, toluene, ethylbenzene and *m,p*-xylene,
223 totally accounting for 78.3% of TVOCs. Generally, ethane and ethyne are typical
224 species emitted from incomplete combustion of fossil fuel and biomass/biofuel (Guo et
225 al., 2007b; Lai et al., 2011; Tang et al., 2009). Their average mixing ratios were $2.40 \pm$
226 1.40 and 1.69 ± 1.10 ppbv, respectively. Propane, *i*-butane and *n*-butane are the major
227 components of liquid petroleum gas (LPG), which are widely regarded as tracers of

228 LPG usage (Blake and Rowland, 1995; Ling and Guo, 2014). The average mixing ratios
229 were 4.63 ± 3.51 , 2.22 ± 1.94 and 3.86 ± 3.74 ppbv for propane, *i*-butane and *n*-butane,
230 respectively. The average concentration of two pentane isomers, *i*-pentane and *n*-
231 pentane were 2.31 ± 3.05 and 1.67 ± 2.10 ppbv respectively, which were mainly emitted
232 from gasoline evaporation and/or gasoline-fueled vehicle exhaust (Liu et al., 2008a; Liu
233 et al., 2008b). In addition, high levels of aromatic hydrocarbons were observed with
234 average mixing ratios of 5.96 ± 4.90 ppbv for toluene, 2.11 ± 1.15 ppbv for
235 ethylbenzene and 2.75 ± 2.00 ppbv for *m,p*-xylene, respectively. Toluene, ethylbenzene
236 and *m,p*-xylene (TEX) are common species emitted from traffic-related and industrial
237 solvent usage in southern China (Zhang et al., 2012; Zhang et al., 2013). Overall, high
238 concentrations of alkanes and aromatics suggest the important contributions of traffic
239 emissions and industrial solvent usage to VOCs.

240 We also made a comparison of major VOC species (**Table S1**) with the offline
241 measurement results at the same site from October to December 2007 (Ling et al., 2011).
242 The mixing ratios of LPG tracers (propane, *i*-butane and *n*-butane) were significantly
243 higher in 2017 than those observed in 2007. In contrast, we found lower concentrations
244 of fossil fuel combustion-related species including ethane, ethylene, ethyne, *i*-pentane
245 and benzene in 2017 compared to those in 2007. This could be attributable to the
246 implementations of switching from fossil fuel/biofuel to LPG as vehicle fuel in the PRD
247 region in recent years (Zhang et al., 2018). Besides, the mixing ratios of TEX were
248 lower in 2017 than those in 2007. It suggests the effective reduction of VOCs emitted
249 from industrial emissions in the PRD region. It should be noted that the concentration

250 and composition of VOCs could be also influenced by other factors such as the
251 sampling period, meteorological conditions, analytical method and atmospheric
252 reactions besides source emissions.



253
254 **Fig. 1** Time series of air pollutants and meteorological conditions at WQS site from
255 September to November 2017: (a) temperature (T), relative humidity (RH) and
256 precipitation; (b) wind speed (WS) colored by wind direction (WD);(c) carbon
257 monoxide (CO); (d) ozone (O₃) and nitrogen oxide (NO_x); the red dash line in (d) shows
258 a threshold (O₃ > 100 ppbv) for the definition of high-O₃ days. (e) mixing ratios of
259 VOCs
260 The pie chart shows the relative abundance of VOCs during the whole measurement
261 period.

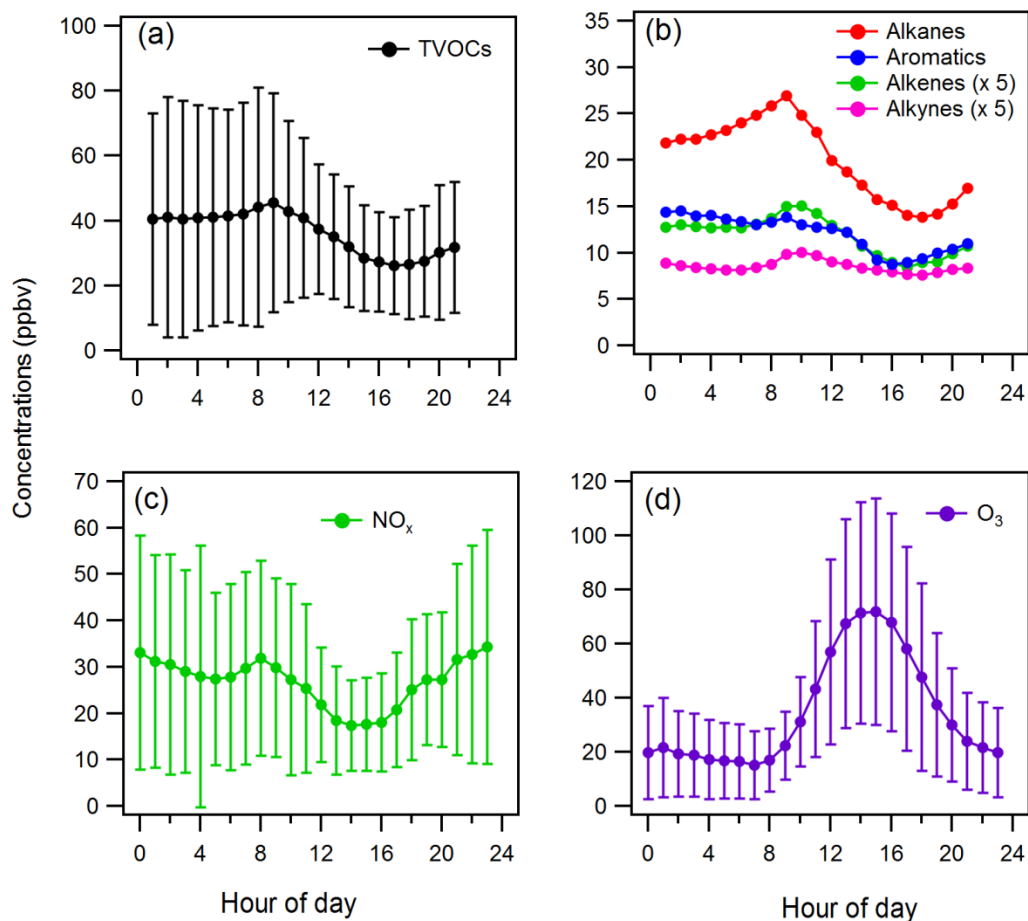
262 3. 2. Diurnal variations of VOCs, NO_x and O₃

263 **Fig. 2** shows the diurnal variations of TVOCs, VOC groups, NO_x, and O₃ during
264 the sampling period. The diurnal variations of VOCs at 22:00 to 0:00 local time (LT)
265 were missing due to the daily calibration. In general, meteorological conditions play an
266 important factor influencing the diurnal variations of atmospheric pollutants. In this
267 study, lower wind speeds were observed during nighttime than daytime, indicating a
268 more stagnant condition in favor of air pollutant accumulation during nighttime (**Fig.**
269 **S2**). In addition to meteorology, emission sources and chemical reactions can also
270 influence the diurnal variations of air pollutants. The diurnal variation of TVOC
271 concentration was characterized with a peak at 8:00 to 9:00 LT in the morning. This
272 peak of TVOCs can be attributed to the strong emissions of traffics in the morning rush
273 hours. Subsequently, the concentration of TVOCs decreased quickly during the daytime
274 (10:00 to 18:00 LT), which is related to the strong photochemical consumption of
275 TVOCs. After 19:00 LT, the concentration of TVOCs increased and then stayed
276 constantly during nighttime. It is likely caused by the increase of traffic emissions
277 during evening rush hours and the shrink of atmospheric boundary layer during
278 nighttime. The groups of alkanes, aromatics and alkenes showed fast photochemical
279 consumptions during daytime (10:00 to 18:00 LT). The diurnal variations of some
280 representative VOC species were shown in **Fig. S3**. C₂ species (ethane, ethylene,
281 ethyne), propane, *i/n*-butanes, *n*-pentane and benzene exhibited the highest
282 concentrations in the morning rush hours (8:00-9:00 LT), which are related to traffic
283 emissions. However, there were no obvious morning peaks observed for toluene,

284 ethylbenzene and xylene, indicating that they mainly came from other sources such as
285 solvent usage rather than traffic emissions. After the morning rush hours, the
286 concentrations of ethene, propane, *i/n*-butanes, *n*-pentane, the so call BTEX (benzene,
287 toluene, ethylbenzene and xylene) and styrene showed significant decreases during
288 daytime, whereas they showed higher concentrations but with relatively flat trends
289 during nighttime. Ethene and BTEX are highly reactive species with OH radicals, and
290 thus their fast decreases were due to strong photochemical reactions during daytime.
291 However, the constant trends of reactive VOC species during nighttime could be due to
292 reduced oxidation capacity and increased stagnant condition of the atmosphere.

293 Similarly, the NO_x concentration also peaked at 8:00-9:00 LT, following a
294 decreasing trend during daytime. This suggests the strong traffic emissions in the
295 morning rush hours, followed by the fast-photochemical consumption of both TVOCs
296 and NO_x. The diurnal variation of O₃ showed an opposite trend with TVOCs and NO_x
297 during daytime with the highest concentration during the afternoon (13:00-16:00 LT).
298 This indicates that the consumptions of VOCs and NO_x result in the photochemical
299 formation of O₃ during daytime. During nighttime, the reduction of O₃ was caused by
300 the titration of NO ($\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$). We further investigated the diurnal variation
301 of TVOCs/NO_x ratios, which have been widely used to determine the O₃ formation
302 regime (Li et al., 2019; Liu et al., 2016; Seinfeld, 1989; Zou et al., 2015). Generally,
303 VOC-limited regimes occur when TVOCs/NO_x ratios are lower than 8, while NO_x-
304 limited regimes occur when TVOCs/NO_x ratios are larger than 8 (Li et al., 2019; Liu et
305 al., 2016; Zou et al., 2015). In this study, the average ratio of TVOCs/NO_x was $1.59 \pm$

306 1.12 during the whole measurement period. Furthermore, high concentrations of O₃
 307 were often observed with TVOCs/NO_x ratios of 1~6 from midday to afternoon (**Fig.**
 308 **S4**). These results suggest that O₃ formation is VOC-limited at WQS site.



309
 310 **Fig. 2** Diurnal variations of TVOCs, VOCs groups, NO_x and O₃ at WQS site from
 311 September to November 2017: (a) TVOCs, (b) VOC groups, (c) NO_x and (d) O₃
 312 The values of VOCs at 22:00-0:00 were excluded due to daily calibration.

313

314 3.3 Source identification of VOCs

315 The ratios of specific VOCs are useful indicators for investigating the sources of
 316 VOCs and the aging of air masses (Gilman et al., 2013; Guo et al., 2007a; Rossabi and
 317 Helmig, 2018). As *i*-pentane and *n*-pentane have similar sources and reactivity, their

318 ratio is widely used to reveal the VOC sources (Gilman et al., 2013; Rossabi and Helmig,
319 2018; Swarthout et al., 2013). In general, lower *i*-pentane/*n*-pentane ratios were
320 observed in the areas highly-influenced by tropical forest fires (0.43-0.57) (Andreae
321 and Merlet, 2001; Rossabi and Helmig, 2018), oil and natural gas operations (0.81-1.1)
322 (Gilman et al., 2013; Swarthout et al., 2013). However, relatively high values of *i*-
323 pentane/*n*-pentane ratios were reported for gasoline fuel evaporation (1.8-4.6), and in
324 urban areas or tunnels with strong vehicle emissions (2.2-3.8) (Gentner et al.; Jobson et
325 al., 2004; Russo et al., 2010a). In this study, *i*-pentane and *n*-pentane correlated well
326 ($R^2 = 0.88$, **Fig. S5**), indicating that they originated from the common sources. The
327 average ratio of *i*-pentane/*n*-pentane was 1.45 ± 0.45 , which is close to the values for
328 gasoline evaporation. Moreover, the average ratio of *i*-pentane/*n*-pentane was lower
329 than that (4.16) observed at WQS site in 2007 (Ling et al., 2011). This implies the
330 effective reduction of VOCs from gasoline-fueled vehicular emissions in the PRD
331 region over the last decade.

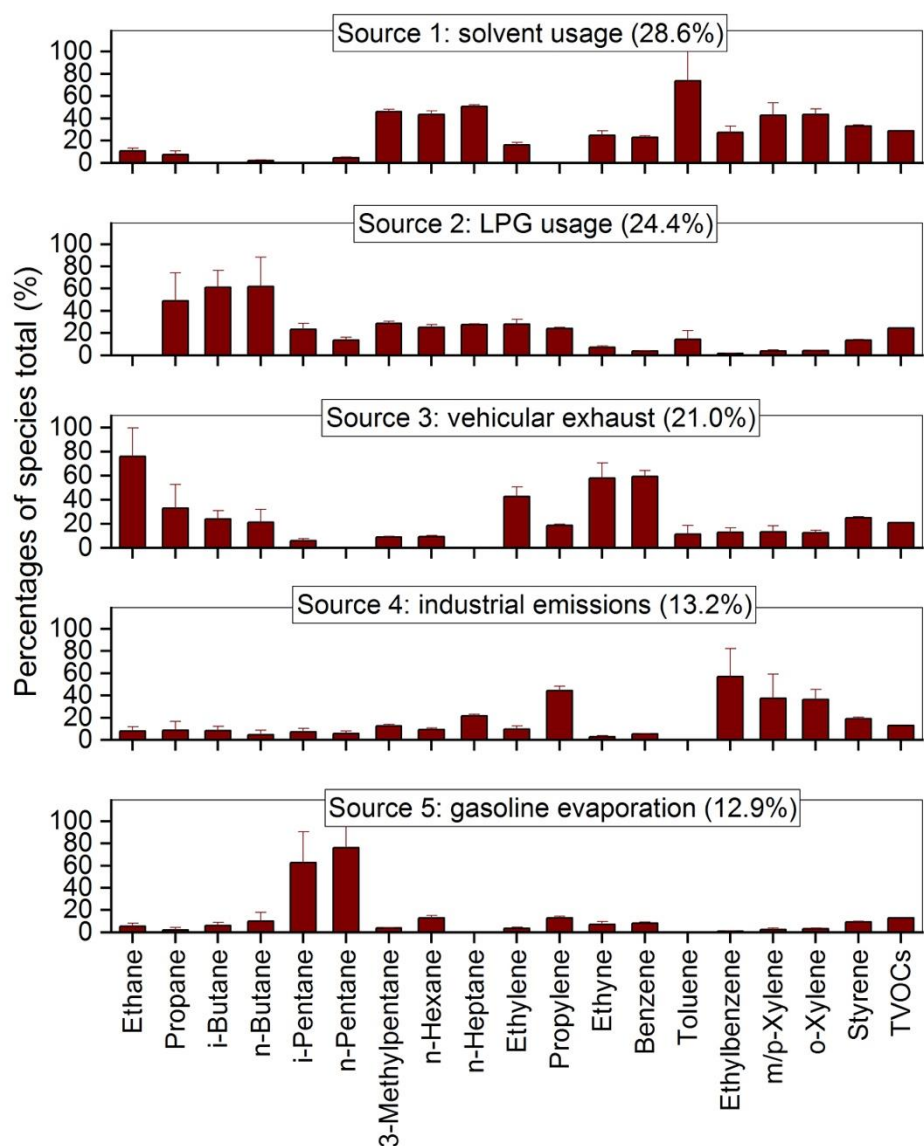
332 Benzene is usually emitted from vehicle exhaust in the urban and rural areas (Li et
333 al., 2017), while toluene mainly originated from the evaporation of industrial solvents
334 used in paint, printing and coating (Yuan et al., 2010; Zhang et al., 2013). The ratio of
335 toluene to benzene (T/B) is often used to identify the non-traffic source contributions
336 to VOCs (Barletta et al., 2008; Elbir et al., 2007). When the ratio of T/B ratio is lower
337 than 2, it indicates the strong influences from combustion sources including vehicle
338 emissions and/or biomass burning. When the ratio of T/B is greater than > 2 indicate
339 the major contributions from non-traffic sources especially paint solvent usage (Barletta

340 et al., 2008; Elbir et al., 2007). In this study, benzene showed a weak correlation with
341 toluene ($R^2 = 0.36$, **Fig. S4**), indicating that they came from different sources.
342 Furthermore, the ratios of T/B ranged from 1.1 to 33.0 with an average of 7.3 ± 5.2 ,
343 suggesting the important influence of solvent usage on VOCs at WQS.

344 The PMF model was used to resolve the source profiles and contributions of VOCs
345 in this study. Five-factor was chosen as the optimal solution of PMF model results and
346 the factor profile are shown in **Fig. 3**. Factor 1 is dominated by TEX, *n*-hexane, 3-
347 methylpnene and *n*-heptane. TEX are the most common VOC species emitted from
348 solvent usage in the PRD region (Zhang et al., 2012; Zhang et al., 2013). *n*-Hexane is
349 used in the solvents for cooking oil extraction and as one of the cleaning reagents for
350 shoe-making, furniture and textile. It can also be regarded as a tracer for household
351 solvent usage (Guo et al., 2011). 3-methylpnene and *n*-heptane are also associated with
352 the solvent usage in painting and varnishing (Guo et al., 2011; Ling and Guo, 2014).
353 Therefore, factor 1 can be assigned to solvent usage. Factor 2 is characterized with high
354 loadings of propane, *i*-butane and *n*-butane that are typical tracers of LPG usage (Blake
355 and Rowland, 1995). As mentioned earlier, LPG has been widely used as the fuel for
356 vehicles in the PRD regions in recent years (Zhang et al., 2018). This factor also has
357 relative high percentages of ethene, propylene, *i*-pentane, which are likely associated
358 with LPG usage. Therefore, we attributed to factor 2 to LPG usage. Factor 3 is identified
359 with high abundances of ethane, ethylene, ethyne and benzene, which are typical tracers
360 of incomplete combustion including fossil fuel and biomass burning (Guo et al., 2007b;
361 Lai et al., 2011; Tang et al., 2009). In terms of the diurnal variations of ethane, propane,

362 ethylene, ethyne and benzene showing significant peaks in morning rush hours, it is
363 reasonable to expect that they come from traffic emissions rather than biomass burning.
364 Therefore, factor 3 is assigned to vehicle exhaust. Compared to factor 1, factor 4 is
365 characterized with high loadings of ethylbenzene, *m/p*-xylene, *o*-xylene, and lower
366 percentages of benzene and toluene. In addition to solvent usage, other industrial
367 processes can also emit considerable amounts of aromatics in the PRD region (Zhang
368 et al., 2012; Zhang et al., 2013). For example, xylenes are widely used in shoemaking,
369 printing, packaging, toy and textiles industry in the PRD region (Guo et al., 2011; Ling
370 et al., 2011). Therefore, we assigned factor 4 to industrial emissions. Factor 5 is
371 dominated by high percentages of *i*-pentane and *n*-pentane, which were mainly emitted
372 from gasoline evaporation (Guo et al., 2011; Ling et al., 2011). Therefore, factor 5 was
373 regarded as gasoline evaporation.

374 Overall, solvent usage (28.6%) was the largest contributor to VOCs at WQS site,
375 followed by LPG usage (24.4%) and vehicle exhaust (21.0%). We also compared the
376 source contributions of VOCs with the previous study at WQS site in 2007 (Guo et al.,
377 2011). Although lower contribution of solvent usage in 2017 than that in 2007 (46%)
378 suggests the effective control in recent years, it is still an important source of VOCs at
379 WQS site. The contribution of LPG was higher in 2017 than that (9%) in 2007,
380 indicating the influence of LPG usage in the region over the last decade. In addition,
381 the contribution of vehicle exhaust in 2017 was slightly lower than that (26%) in 2007.



382

383

Fig. 3 Source profiles of VOCs identified by PMF model and the relative contributions of each source to the measured VOCs at WSQ site.

384

385

386 3.4 Atmospheric processing of VOCs

387 The ratios of VOCs with different photochemical lifetimes can be used to evaluate

388 the atmospheric processes including photochemical aging and atmospheric transport

389 (Guo et al., 2007b; Louie et al., 2013; Song et al., 2018). Ethyne and CO share common

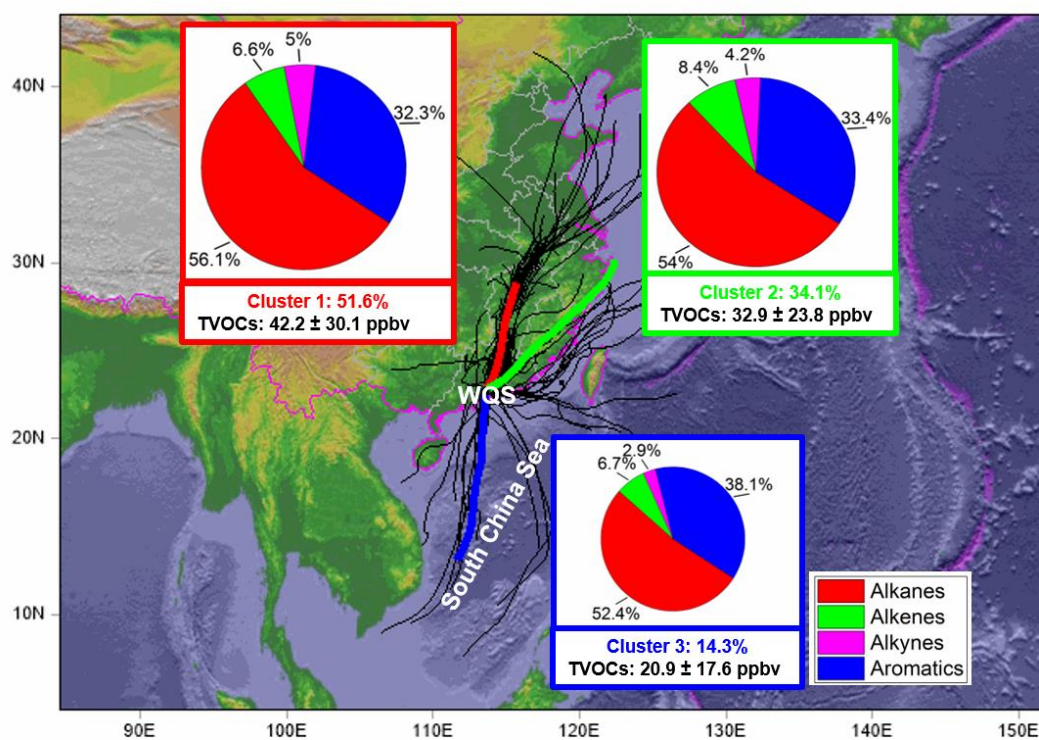
390 sources but have different photochemical lifetime with approximately 2 weeks for

391 ethyne and 2 months for CO. The ratio of ethyne/CO (E/C) have been widely used as a
392 measure of photochemical aging. In the PRD region, high ratios of E/C (5.6-7.5
393 pptv/ppbv) were reported for fresh combustion sources (Guo et al., 2007b; Louie et al.,
394 2013). In this study, the ratios of E/C ranged from 0.28 to 12.4 with an average of 3.49
395 ± 1.48 , suggesting that the air masses at WQS site have experienced a certain extent of
396 photochemical aging.

397 The sources of ethylbenzene and *m/p*-xylene have similar sources such as solvent
398 usage, while they have different reactivities with OH radical. Therefore, the ratio of
399 *m/p*-xylene/ethylbenzene (X/E) can be used to assess the photochemical aging
400 processes. In this study, a good correlation ($R^2 = 0.87$, **Fig. S5**) was observed between
401 ethylbenzene and *m/p*-xylene, confirming that they came from the common sources.
402 Generally, the ratio of X/E is 3 for fresh emissions in urban environment (Monod et al.,
403 2001), while when the ratio lower than 3 indicates the occurrence of photochemical
404 aging. In this study, the ratios of X/E ranged from 0.68 to 2.23 with an average of 1.25
405 ± 0.30 , suggesting the importance of photochemical aging and regional air mass
406 transport.

407 To investigate the atmospheric transport pathways, three-day air mass back-
408 trajectories were calculated using HYSPLIT model and further grouped into three
409 clusters (**Fig. 4**). The first cluster (Cluster-1) originated from the upwind areas of south
410 China (Jiangxi province) and upwind PRD cities (e.g., Dongguan and Huizhou),
411 accounting for 51.6% of total back-trajectories. In this cluster, the highest
412 concentrations of TVOCs (43.2 ± 30.1 ppbv) were observed, pointing to the influence

413 of upwind areas of south China. The second cluster (Cluster-2) accounted for 34.1% of
 414 total back-trajectories, which shows the air masses transported along the southeast
 415 coastal regions of China (Fujian and Zhejiang provinces). In the Cluster-2, the average
 416 mixing ratio of TVOCs was 32.9 ± 23.8 ppbv. In addition, the third cluster (Cluster-3)
 417 originated from the SCS, accounting for 14.3% of total air mass back-trajectories, in
 418 which lower level of TVOCs were observed (20.9 ± 17.6 ppbv). Although the
 419 compositions of measured VOCs showed no significant changes in different clusters,
 420 the predominance of alkanes and aromatics suggested the importance of traffic and
 421 industrial emissions by regional transport.



422
 423 **Fig. 4** Concentrations and composition of TVOCs in three air mass clusters
 424 (red: cluster 1; green: cluster 2, blue: cluster 3).

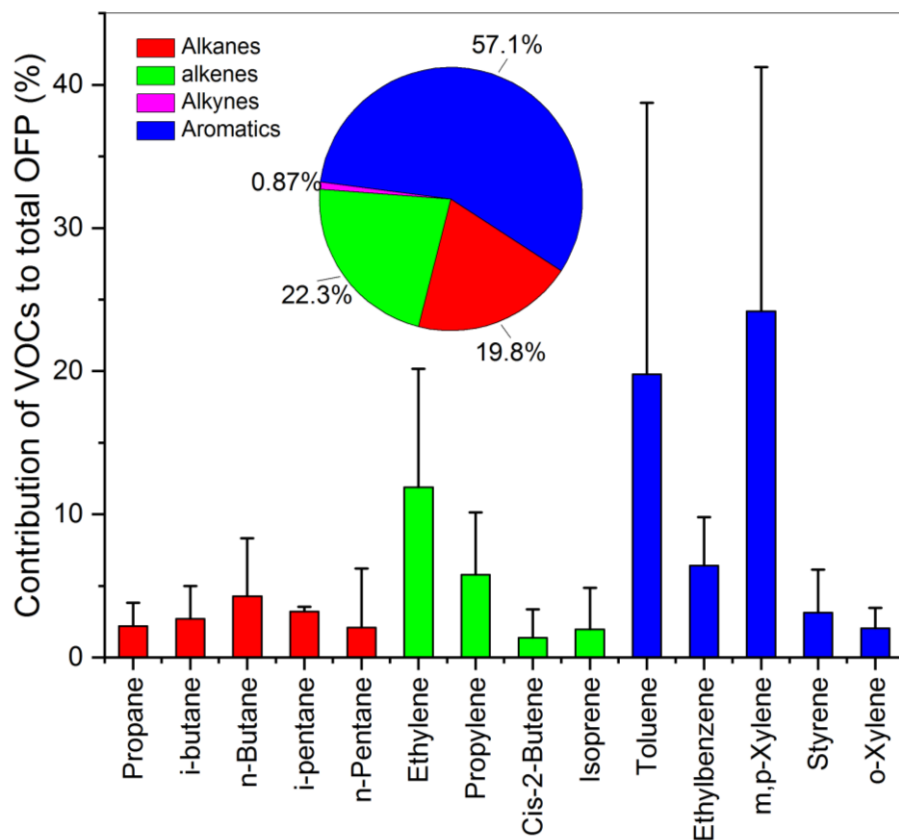
426 3.5. Effects of VOCs on O₃ formation

427 In order to estimate the potential effects of VOCs on O₃ formation, we calculated

428 firstly the OFP of VOC species as the following equation:

$$429 \quad OFP_i = VOC_i \times MIR_i$$

430 where OFP_i and VOC_i are the OFP and the mixing ratios of VOC species i , and the MIR
431 values are obtained from the study by Carter (1994). **Fig. 5** shows the contributions of
432 different VOC groups and top 15 VOC species to the total OFP during the whole
433 measurement period. Aromatics were the largest contributor, accounting for 57.1% of
434 total OFP, followed by alkenes (22.3%) and alkanes (19.8%). As for individual VOC
435 species, five abundant aromatics including toluene, ethylbenzene, *m/p*-xylene, styrene
436 and *o*-xylene totally accounted for 55.6% of total OFP. Since toluene, ethylbenzene,
437 *m/p*-xylene, styrene and *o*-xylene mainly contributed by the emissions of solvent usage,
438 their high contributions to total OFP suggest that the reduction of regional industrial
439 emissions could be the key to control photochemical O_3 pollution in the PRD region.
440 In addition, ethene, propylene, propane and *i/n*-butanes also showed large contributions
441 to total OFP, which are emitted from LPG usage. Even though LPG is commonly
442 regarded as a 'cleaner' vehicle fuel, its usage can emit VOCs contributing to O_3
443 formation significantly. Therefore, LPG usage might also play important role in the
444 formation of photochemical pollution.



445

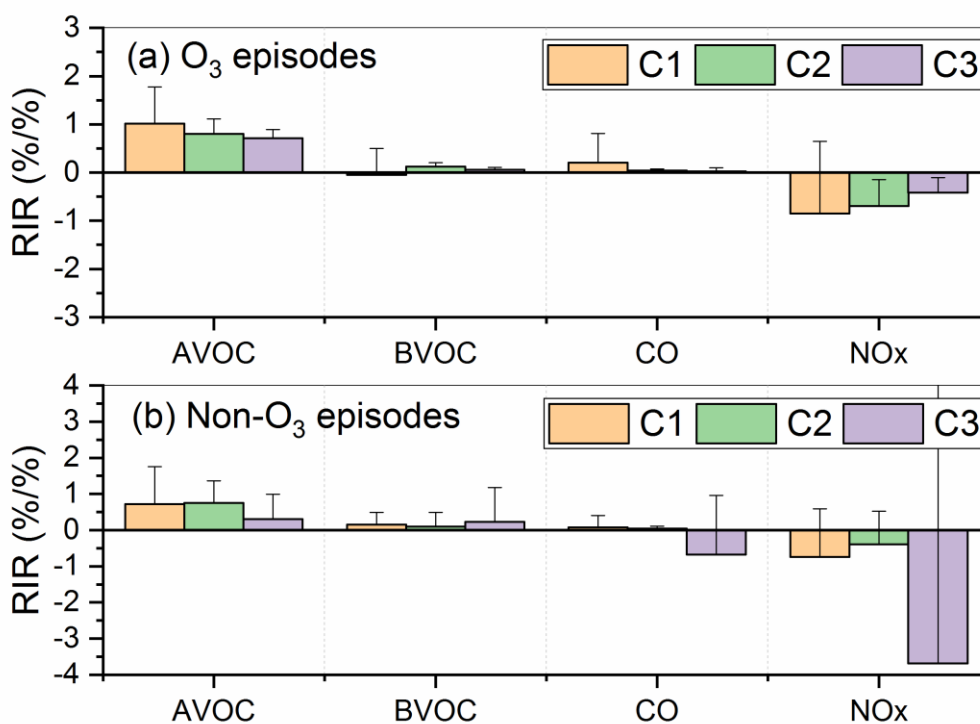
446 **Fig. 5** Contributions of major VOCs to total ozone formation potential (OFP) at WQS
 447 site. The pie chart shows the relative contribution of each VOC group.

448

449 Furthermore, we simulated the O₃ formation processes by PBM-MCM and
 450 calculated the RIR values of precursors to investigate their effects of on O₃ production.

451 **Fig. 6** shows the RIR of anthropogenic VOCs (AVOC), biogenic VOCs (BVOC), CO
 452 and NO_x to investigate the sensitivity of O₃ to the changes of the precursors. As
 453 mentioned in Section 3.1, we totally characterized 26 high-O₃ days during the entire
 454 sampling period. Therefore, we classified the whole measurement period into O₃-
 455 episode and non-O₃ episode days. Meanwhile, in order to better understand the
 456 influence of regional transport in these photochemical pollution events, we further
 457 classified these O₃-episode and non-O₃ episode days based on the influence of different

458 types of source origins as discussed above. On both O₃-episode and non-O₃-episode
 459 days, the RIR values of AVOC were always positive, indicating that the O₃ formation
 460 regime is AVOC-limited, which is in accordance with the discussion based on the
 461 TVOCs/NO_x ratios. In contrast, the RIR values of NO_x were always negative,
 462 suggesting that reducing NO_x levels would increase the O₃ production.



463
 464 **Fig. 6** Average RIR values of O₃ precursors during O₃ episodes and non-O₃ episodes.
 465 The precursors include anthropogenic VOCs (AVOC), biogenic VOCs (BVOC), CO,
 466 NO_x. C1, C2 and C3 represents the different influence of source origins.

467
 468
 469
 470

471 **Table 1** Summary of air pollutant concentrations and typical diagnostic ratios of
 472 VOCs during 26 high-O₃ days in three types of air mass clusters.

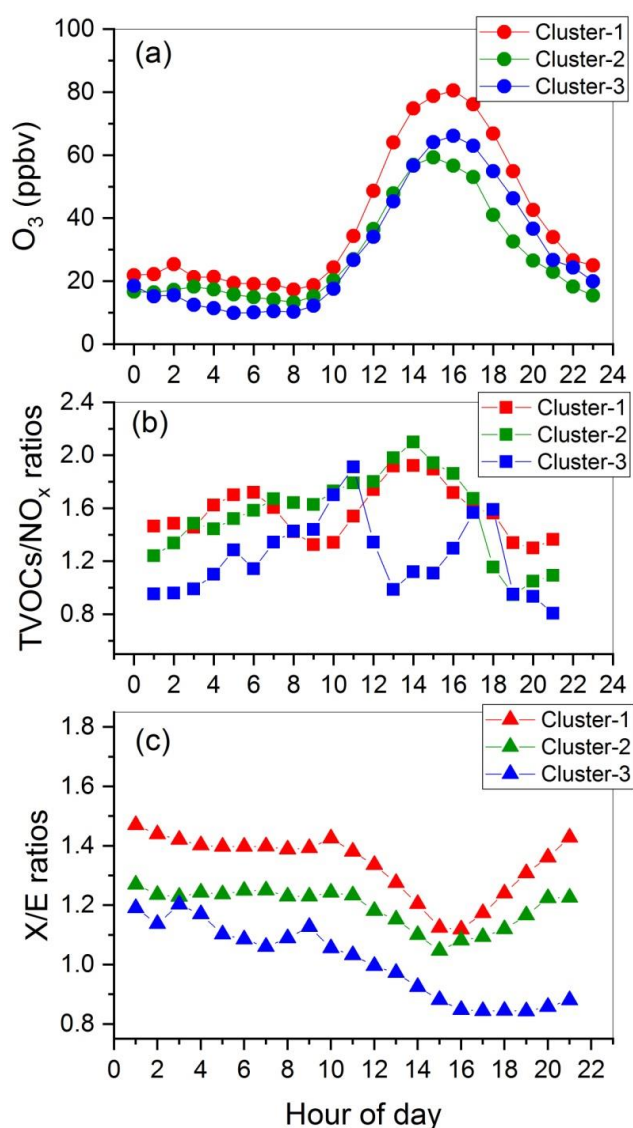
Clusters	Cluster-1 (<i>n</i> = 18)			Cluster-2 (<i>n</i> = 4)			Cluster-3 (<i>n</i> =4)		
Air mass origins	Southeast China/PRD			Southeast China coast			South China Sea		
	Mean	S.d ^a	Median	Mean	S.d	Median	Mean	S.d	Median
VOCs (ppbv)									
Alkanes	32.4	23.2	22.6	20.6	12.6	17.6	19.1	13.1	15.9
Alkenes	3.3	2.2	2.5	2.2	1.3	1.9	2.0	1.0	0.7
Alkyne	2.4	0.9	2.3	1.6	0.9	1.3	1.1	0.7	0.9
Aromatics	16.8	10.0	14.2	11.5	6.1	10.0	10.4	5.3	9.3
TVOCs	54.9	34.6	41.3	35.9	20.5	30.6	32.6	18.9	26.4
Air pollutants (ppbv)									
NO _x	35.4	22.1	29.0	27.2	16.0	24.7	21.4	10.5	18.8
O ₃	56.0	41.0	46.7	49.1	42.2	37.3	53.8	52.1	32.0
CO	572	168	560	535	124	560	561	126	560
Diagnostic ratios									
Ethyne/CO	4.24	1.13	4.42	2.75	1.33	2.33	1.83	0.88	1.61
X/E	1.31	0.28	1.24	1.07	0.31	0.98	1.03	0.26	0.93

473
 474 During all O₃-episode days, in addition to AVOC, CO also showed positive RIR values,
 475 indicating that reducing the levels of CO would lead to the decrease of O₃ formation as
 476 well. Given the different origins of air mass clusters, the RIR value of AVOC and CO
 477 were higher in Cluster-1 than those in other two clusters (Cluster-2 and Cluster-3). This
 478 suggests that the formation of O₃ was mainly caused by regional transport of
 479 anthropogenic emissions from the upwind PRD urban regions. In particular, the
 480 occurrence frequency on high-O₃ days in Cluster-1 (*n* =18) was higher than that in
 481 Cluster-2 (*n* = 4) and Cluster-3 (*n* =4). Therefore, we suggested that the regional
 482 reduction of anthropogenic VOCs emissions would be effective for controlling the O₃
 483 formation in the downwind PRD region. The variations of VOC and NO_x, CO and O₃
 484 concentrations as well as the diagnostic ratios of E/C and X/E during O₃-episode days
 485 in different clusters were given in **Table 1**. In Cluster-1, higher concentrations of

486 TVOCs, NO_x, O₃ and CO were observed compared to those in another two clusters.
487 Furthermore, the average ratios of E/C and X/E were higher in Cluster-1 than those in
488 other two clusters (**Table 1**). The results indicate that more fresh emissions of
489 anthropogenic pollutants transported regionally from the upwind PRD urban regions.

490 Besides, we also investigated the diurnal variations of O₃, TVOCs/NO_x and X/E
491 ratios during O₃-episodes in different clusters (**Fig. 7**). In Cluster-1, the concentrations
492 of O₃ were elevated with the increase of TVOCs/NO_x ratios from 10:00 to 16:00 LT
493 during daytime. Correspondingly, the ratios of X/E showed fast decrease after 10:00 to
494 16:00 LT. The results suggest that the subsequent photochemical reactions lead to the
495 formation of O₃ after the regional transport of anthropogenic emissions from upwind
496 urban regions to WQS site. This result is in line with the recent findings that the
497 downwind regions (PRE) can be act as a sink area of O₃ in the PRD region resulting
498 from the transported air pollutants and intensive photochemical reactions (Zeren et al.,
499 2019). Compared to Cluster-1, Cluster-2 had lower concentrations of O₃ during daytime
500 (10:00-16:00) with the corresponding weak decrease of X/E ratios, indicating less
501 photochemical reactions when WQS site was influenced by the air masses from
502 Southeast coastal regions. However, in Cluster-3, we found that TVOCs/NO_x ratios
503 showed fast decrease from 10: 00 to 12:00 with continuously increase of O₃. This is
504 likely related to the shift of sea-land breezes during daytime. As shown **Fig. 7c**, a
505 slightly increase of X/E ratios were observed in morning hours (7:00-9:00 LT), which
506 is likely related to fresh emissions from the land air. In contrast, the ratios of X/E
507 showed decrease continuously from 10:00 to 16:00, suggesting the photochemical

508 aging under the influence of SCS air. However, the limited number of high-O₃ days
509 under the influence of air masses from the SCS cannot sufficiently demonstrate the
510 effects of sea-land breezes on O₃ formation in this study. More long-term field
511 measurements and modeling studies are necessary to investigate the role of sea-land
512 breezes in photochemical pollution in the PRD region.



513
514 **Fig. 7** Diurnal variations of O₃ concentrations, TVOCs/NO_x ratios and *m,p*-
515 xylene/ethylbenzene (X/E) ratios during high-O₃ days with the influence of three air
516 mass clusters.

517 **4. Conclusions**

518 In this study, VOCs were continuously measured at a receptor site in the PRD region
519 from September to November 2017. The results showed that the dominant VOC species
520 were alkanes and aromatics. The source apportionment results suggest solvent usage,
521 LPG usage, vehicle exhaust were the major contributors of VOCs. The air mass back-
522 trajectories showed that the upwind south China was the major source region of VOCs.
523 The ratios of VOCs/NO_x and the result of PBM-MCMs indicated that the
524 photochemical O₃ formation was anthropogenic VOC-limited at WQS site. Moreover,
525 toluene, ethylbenzene, xylene, ethylene, propylene and *i/n*-butanes were the major
526 contributors of total OFP. Finally, O₃ formation during the high-O₃ days were further
527 investigated by PBM model simulation and diagnostic ratios of VOCs. The results
528 suggest that frequent high-O₃ outbreaks were caused by the regionally transported
529 pollutants followed by strong photochemical reactions. In future, more studies on O₃-
530 VOC-NO_x relationship at downwind region are still needed, which is crucial for further
531 improvement of regional photochemical pollution control strategies in the PRD region.

532

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