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6 **Development of high performance lightweight concrete using** 7 **ultra high performance cementitious composite and different** 8 **lightweight aggregates**

9
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16 **Abstract:** To reduce the dead load of concrete structures, this study developed a high performance
17 lightweight aggregate concrete (HPLAC) by combining the use of ultra high performance
18 cementitious composite (UHPC) and different types of aluminosilicate lightweight aggregates
19 (LWAs). The physicochemical properties of two types of LWAs (i.e. expanded clay and expanded
20 shale) influencing the HPLAC were elaborated and compared. The composition distribution and
21 micromechanical properties in the interfacial regions of paste and LWAs were revealed by
22 elemental mapping and nanoindentation. The results showed that the incorporation of the clay
23 LWAs or shale LWAs in the HPLAC led to similar density and thermal conductivity values, while
24 the use of the shale LWAs induced a lower water absorption and higher strength of HPLAC as
25 compared to the clay LWAs due to the fine pore structure and higher pozzolanic activity of the
26 former. The internal curing effect provided by the pre-wetted shale LWAs was more efficient in
27 enhancing the hydration of binder, and the Al dissolution from the shale LWAs further densified
28 the interfacial bonding to form a dense rim surrounding LWAs, resulting in improved
29 micromechanical properties at the interface. The X-ray CT results indicated that the adoption of
30 UHPC was beneficial to preventing segregation of the LWAs and steel fibers in the HPLAC. By
31 virtue of the physical-chemical interactions of LWAs, the synergetic use of UHPC and pre-wetted
32 shale LWAs was able to produce an HPLAC with high structural efficiency, good thermal

33 insulation, low autogenous shrinkage and permeability.

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35 **Keywords:** High performance lightweight concrete; Ultra high performance cementitious
36 composite; Durability; Pozzolanic reactivity; Micromechanical properties

37

38 **1 Introduction**

39 Modular integrated construction (MiC) is perceived as an innovative construction concept by
40 adopting two construction phases: factory assembly and on-site installation [1]. The development
41 of the MiC in recent years seems to provide a solution for addressing the shortage of manpower,
42 increased costs of construction, and declining productivity in the construction industry.
43 Furthermore, the adoption of the MiC can increase the quality of construction, enhance site safety
44 and reduce environmental impacts, thus the MiC is getting more popular around the world [2, 3].
45 However, conventional concrete materials similar to those used for on-site construction are
46 normally used in the MiC [4], which poses challenges for requiring stronger hoist cranes and
47 special logistics. Therefore, it is of significance to develop a high performance lightweight
48 concrete for enhancing the lifting capacity and reducing the transportation cost of the MiC
49 construction.

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51 Lightweight aggregates (LWAs) are generally used to produce high strength lightweight concrete
52 for structural applications. Due to the presence of pores in the LWAs, the weight of high-strength
53 lightweight aggregate concrete (HSLAC) is 20~40% lower than conventional concrete, with
54 density values of less than 2000 kg/m³ [5] and 28 day compressive strength higher than 40 MPa
55 [6]. To prepare high performance HSLAC, three strategies have been used: use of good-quality
56 binder matrix, selection of proper LWAs, and addition of fibers.

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58 For improving the quality of the binder matrix, supplementary cementitious materials (SCMs) and
59 a low water-to-binder (*w/b*) ratio were adopted [7, 8]. Some researchers [9, 10] prepared HSLAC
60 with the incorporation of silica fume, and found that the HSLAC had similar resistance to the
61 chloride ion penetration and bond strength as compared to normal weight concrete. These findings
62 are in agreement with the studies of Thomas and co-workers [11, 12], which indicated that the
63 additions of SCMs (e.g. silica fume, fly ash, and blast furnace slag) were effective in decreasing

64 penetration of chloride ions in 25-year-old lightweight concrete compared to concrete containing
65 only Portland cement. Nadesan and Dinakar [13] utilized silica fume and metakaolin to improve
66 the mechanical and durability properties of HSLAC, and the benefits were more pronounced at a
67 lower w/b ratio of 0.25. The study of Chen and Liu [14] indicated that the combined use of fly ash,
68 blast furnace slag, and silica fume in HSLAC could prevent bleeding and up-floating of LWAs. To
69 reduce the dead weight and increase the strength, Kılıç et al. [15] developed an HSLAC
70 incorporating silica fume and fly ash. Similarly, Wilson and Malhotra [16] found that HSLAC
71 prepared with silica fume had a satisfactory resistance to cyclic freezing and thawing. In addition,
72 Seabrook and Wilson [17] pointed out that the optimum silica fume content was in the range of 5%
73 to 10% by the weight of cement. Atmaca et al. [18] added 3% nano-silica into HSLAC to mitigate
74 the negative effects of LWAs. As a result, a remarkable increase in mechanical properties and
75 reductions in sorptivity and gas permeability were achieved. Except using the SCMs, by reducing
76 the w/b ratio, Moreno et al. [19] prepared a more durable HSLAC (higher strength and lower ion
77 permeability) in comparison to normal lightweight aggregate concrete with a higher w/b ratio. Real
78 and his colleagues [20, 21] also reported that the dense HSLAC with lower w/b ratio had lower
79 permeability regardless of the type of LWAs.

80

81 For producing high strength lightweight concrete, the type of LWAs used also plays a vital role.
82 Several kinds of LWAs have been adopted in the production of HSLAC, including cold bonded
83 and sintered fly ash LWAs [13, 18, 22, 23], expanded clay and slate [24-27], oil-palm-boiler clinker
84 and oil palm shell [28-30], and other LWAs (e.g. scoria, tuff, expanded shale) [12, 15, 16, 31].
85 Kockal and Ozturan [23] produced a high strength sintered fly ash LWA (crushing strength of 12
86 MPa) and compared it with cold bonded sintered fly ash LWA (crushing strength of 3.7 MPa) in
87 HSLAC. Their study showed that the HSLAC incorporating high strength LWAs had better
88 mechanical properties (compressive and splitting tensile strengths, modulus of elasticity) than the
89 HSLAC using low strength LWAs. In terms of durability performance, Real and Bogas [20] found
90 that the permeability of HSLAC decreased when the open porosity of LWAs was reduced and the
91 permeability of HSLAC with denser aggregates could be similar to that of normal weight concrete.
92 They also reported a higher carbonation resistance of HSLAC prepared with the lower porosity
93 LWAs [32]. Normally, the use of fine LWAs induced a higher compressive strength of lightweight
94 concrete as compared to coarse LWAs [33, 34]. Moreover, the larger amount of fine LWAs could
95 attain a higher compressive strength of HSLAC [35]. Due to the presence of lower stiffness of

96 LWAs, lightweight concrete usually exhibits lower flexural/tensile properties [36, 37] and higher
97 long-term shrinkage [38]. Polypropylene and steel fibers have thus been added to enhance the
98 flexural toughness and splitting tensile strength of HSLAC [39-42]. Compared to the
99 polypropylene fibers, the addition of steel fibers was more effective in increasing the flexural
100 strength and toughness [39, 42], whereas the latter increased the density considerably and reduce
101 the slump of concrete [40, 41]. Thus, a proper volume fraction of steel fibers should be taken into
102 account in the design of HSLAC. Since the LWAs have a lower modulus than the natural
103 aggregates, other advantages of using fibers in HSLAC include the increase of modulus of
104 elasticity [41], enhancement of ductility [42], and reduction of shrinkage [43] thanks to the crack
105 arresting ability of the fibers.

106
107 Given the LWAs are the weakest component in the HSLAC, the durability and mechanical
108 properties of the lightweight concrete would be mainly governed by the qualities of cement mortar
109 forming a protective coating around the LWAs [44, 45]. Thus, how to improve the quality of the
110 mortar matrix is the key point in the production of HSLAC. In this study, an ultra high performance
111 cementitious composite (UHPC) was used as a binder to prepare the HSLAC to leverage its very
112 high strength and superior durability. Previously, some studies [46-50] have incorporated pre-
113 wetted LWAs into the low *w/b* concrete, but the intentions of these investigations were to enhance
114 the strength [46, 49] or reduce the shrinkage [47, 48, 50] of high performance concrete, the weight
115 of concrete was thus not reduced significantly (e.g. the density of the LWC was about 2,300 kg/m³
116 in ref [46]). Therefore, this study aimed to produce high quality lightweight concrete combining
117 1) UHPC as a paste, and 2) a large volume of LWAs aiming to improve the performance of
118 lightweight concrete and reduce the dead load of concrete.

119
120 Aforementioned, with the use of SCMs, good quality LWAs and fibers, the mechanical properties
121 and durability of lightweight concrete can be enhanced considerably. Until now, few studies have
122 provided a comprehensive perspective (i.e. an integration of the above three approaches) to
123 produce a high-performance lightweight concrete. This work integrated the use of a UHPC paste,
124 steel fibers and two typical types of LWAs (expanded clay and expanded shale) to produce a
125 UHPC-based HPLAC. In particular, the roles of the physicochemical properties of LWAs in the
126 UHPC matrix were emphatically investigated and compared.

127

128 **2 Materials and methods**

129 **2.1 Materials**

130 A Type I ordinary Portland cement (OPC) sourced from Hong Kong Green Island Cement
 131 Company was used as the main cementitious material. Silica fume (SF) and ground granulated
 132 blast furnace slag (GGBS) were purchased from Mainland China. Metakaolin (MK) was supplied
 133 by a local company (Nano and Advanced Materials Institute). The chemical compositions of the
 134 cementitious materials were analyzed by X-ray fluorescence technique using a Rigaku
 135 Supermini200 spectrometer. Table 1 lists the composition results, which show that the SF was
 136 rich in SiO₂, the MK was rich in SiO₂ and Al₂O₃, and the GGBS was rich in SiO₂ and CaO. The
 137 particle size distributions of cementitious materials (except SF) were measured by a laser
 138 diffraction analyzer (Malvern Instrument), while the size of SF was tested by a Zetasizer (Malvern,
 139 Nano-ZS90) due to its extremely fine particle. The particle size distributions are depicted in Fig.
 140 1 and the average size and density of cementitious materials are shown in Table 1.

141

142 Table 1. Chemical compositions and physical properties of cementitious materials and aggregates.

Compositions (wt. %)	OPC	SF	MK	GGBS	CLWA	SLWA	RS
SiO ₂	19.0	95.7	53.00	34.78	60.4	62.0	90.4
Al ₂ O ₃	5.68	0.45	43.60	14.22	19.8	19.8	4.71
CaO	65.8	0.73	0.12	38.38	2.57	0.76	0.66
MgO	0.97	0.71	0.10	7.32	1.23	/	0.26
K ₂ O	0.66	1.74	1.68	0.77	2.89	4.58	2.06
P ₂ O ₅	0.16	0.09	0.13	/	0.37	0.23	0.14
Fe ₂ O ₃	3.12	0.07	0.70	0.27	11.2	7.16	0.88
Na ₂ O	/	/	/	/	/	1.29	0.66
SO ₃	4.03	0.27	0.23	3.12	0.15	0.13	0.10
Physical properties							
Specific gravity	3.15	2.69	2.11	2.56	2.63	2.52	2.53
Average particle size (μm)	22.68	0.44	6.21	15.64	/	/	/

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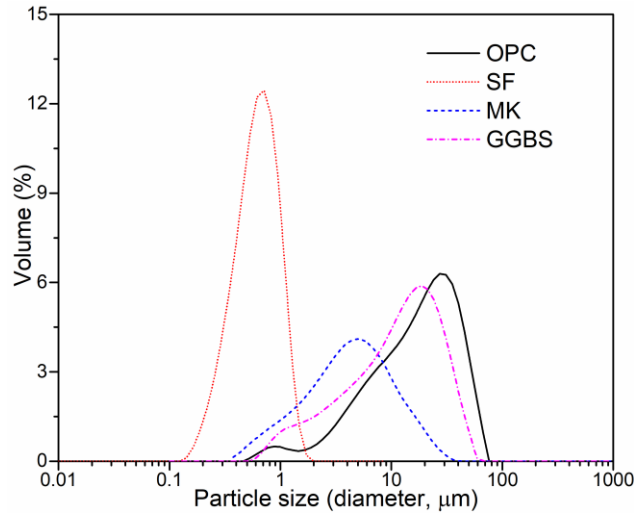


Fig. 1. Particle size distributions of cementitious materials.

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145 Aggregates used in this study included river sand (RS), expanded clay lightweight aggregate
 146 (CLWA), and expanded shale lightweight aggregate (SLWA). The CLWA was made from fresh
 147 clay, and produced through pelletization and sintering in the form of rounded pellets. The angular
 148 shaped SLWA was crushed from a big bulk of expanded shale. Thus, the surface of the SLWA was
 149 very rugged and the surface pores were open (see Fig. 2). The particle size of these aggregates was
 150 ranged between 2.36-5 mm as finer LWAs usually have a higher density. The compositions of
 151 aggregates are presented in Table 1, which indicate that SiO_2 and Al_2O_3 were the main components
 152 in the CLWA and SLWA (devoid of calcium). In order to know the crushing resistance strength of
 153 the two types of LWAs, a number of the CLWA pellets and SLWA cubes (cubic particles cut from
 154 bulk shale) with a size/diameter of around 5 mm were tested. The crushing resistance of the CLWA
 155 pellet was calculated based on a method reported in previous studies [51, 52]. The average strength
 156 results are shown in Table 2, which also lists other physical properties of the LWAs. The total and
 157 open porosities of the two aggregates were measured by submerging them in water under vacuum
 158 for 24 h [53]. It is found that the SLWA had a far higher strength than the CLWA despite they had
 159 similar porosities. The water absorption and apparent density values of the two LWAs after soaking
 160 in water for 24 h were similar (under ambient condition). The pre-wetted density of the SLWA was
 161 slightly higher than that of CLWA, whereas the dry bulk density of the former was a little lower.

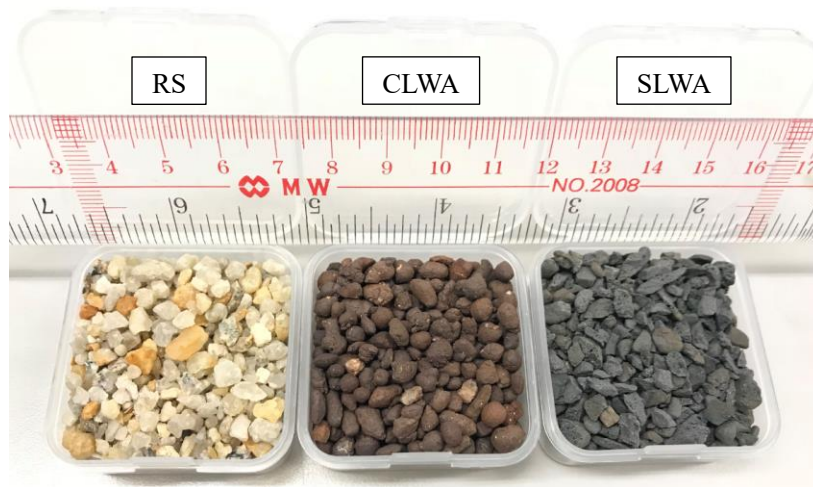


Fig. 2. Appearances of RS, CLWA and SLWA used in this study.

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163 Table 2. Physical properties of LWAs.

Physical properties	CLWA	SLWA
Crushed resistance (N/mm ²)	0.76	4.58
Open porosity (%)	20.5	18.7
Total porosity (%)	68.0	63.9
Water absorption (24 h soaking in water, %)	13.4	14.2
Apparent density (24 h soaking in water, kg/m ³)	956	1065
Loose bulk density (Dry, kg/m ³)	539	525

164

165 2.2 Mix proportions and research framework

166 The HPLAC mixtures were comprised of two parts, one was a UHPC paste and the other was
 167 LWAs. The mix proportions of HPLAC mixtures are listed in Table 3. The pozzolans with
 168 different particle sizes were combined to prepare a high performance paste. For the reference
 169 mixture (100R), the natural RS was adopted as an aggregate and then compared with the LWAs.
 170 The aggregate-to-binder ratio was set as 1:1 by mass. A *w/b* ratio of 0.17 was used for the
 171 reference mixture. Since the LWAs would absorb water during mixing, they were pre-soaked in
 172 water for 24 h [54], and then the LWAs were used to replace 50% and 100% RS by volume. Since
 173 part of water would be released from pre-soaked LWAs due to the centrifugal force during the
 174 high-speed mixing [55], the *w/b* of HPLACs was reduced slightly (Table 3) to maintain a similar

175 flow value with the reference mixture (180±5 mm determined by flow table [56]). To further
 176 reduce the weight of the mixture, more pre-wetted LWAs (150% and 200% volume of RS based
 177 on the reference mixture) were introduced into the mixture. For all the mixtures, the dosage of a
 178 polycarboxylate ether (PCE) superplasticizer (BASF, SKY 8588) was 3% of the binder by mass.
 179 The water content in the superplasticizer was 80%. Short straight steel fibers coated by copper
 180 (diameter of 0.22 mm and a length of 13 mm) were employed to reinforce the samples. As the
 181 addition of the steel fibers with very high density would increase the weight of concrete largely,
 182 the dosage of the fibers was controlled at 1.5% by the paste volume. The number in the codified
 183 mix refers to the percentage of aggregates. The ‘C’ and ‘S’ denote the CLWA and SLWA,
 184 respectively. For example, the 100C means that the mixture was prepared with 100% CLWA as
 185 aggregates.

186

187 Table 3. Mix proportions of HPLACs prepared with SLWA and CLWA (kg/m³).

Mix	Cementitious materials				Fiber	PCE	Aggregates			Water	w/b
	OPC	SF	MK	GGBS			RS	CLWA	SLWA		
100R	827	124	41	83	61	32	1075	/	/	157	0.17
50C	827	124	41	83	61	32	537	203	/	146	0.16
100C	827	124	41	83	61	32	/	406	/	135	0.15
150C	687	103	34	69	51	27	/	506	/	113	0.15
200C	584	88	29	58	43	23	/	574	/	96	0.15
50S	827	124	41	83	61	32	537	/	226	146	0.16
100S	827	124	41	83	61	32	/	/	452	135	0.15
150S	687	103	34	69	51	27	/	/	564	113	0.15
200S	584	88	29	58	43	23	/	/	639	96	0.15

188

189 For the preparation of HPLACs, the cementitious materials were dry-mixed for 3 minutes in a
 190 laboratory mechanical mixer. Then, water and superplasticizer were added followed by further
 191 mixing for 6 minutes. After the mixture was turned into a paste, the RS/pre-soaked LWAs were
 192 poured into the mixer and the whole mixture was continued mixing for another 2 minutes. Then,
 193 the steel fibers were added followed by 1 minute of additional mixing to obtain a homogenous
 194 fresh mixture. After casting into steel molds, a laboratory vibrating table was used to compact the
 195 mixtures. Plastic sheets were used to cover the surface of mixtures to avoid moisture loss. 24 h
 196 later, the specimens were demoulded and cured in a water tank (25±2 °C) for different periods
 197 before testing.

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2.3 Methods

2.3.1 Chapelle test

The Chapelle test is used for assessing the reactivity of pozzolanic materials in cement-based systems [57, 58]. In this study, the pozzolanic activity of the LWAs was evaluated by this method. One gram of the powdered aggregate (RS, CLWA, and SLWA) was allowed to react with one gram of Ca(OH)_2 in 200 mL distilled water. The prepared solution was placed in sealed glass bottles and heated to 80 °C and kept for 16 hours. After cooling down to room temperature, 20 grams of sucrose were added to dissolve the remaining unreacted Ca(OH)_2 , with the aid of magnetic stirring for 20 mins to make sure the complete complexion of the Ca ions. Afterward, the suspensions were filtered through 2.5 μm filter papers using vacuum filtration. 0.1 N HCl and methyl orange indicator were employed to titrate the filtered solution. Thus, the remaining and reacted Ca(OH)_2 amounts could be calculated based on the consumed HCl dosage. The reactivity of the aggregates was then obtained as the consumed Ca(OH)_2 amount by the aggregates divided by the original Ca(OH)_2 amount. After drying the solid residues (obtained from filtration) at 105 °C, a thermogravimetric (TG, Rigaku Thermo Plus EVO2) analysis was conducted to obtain the bound water of the residues [59].

2.3.2 Physical properties

The compressive strength of the samples was measured at 7, 28, and 90 days of curing following ASTM C109 [60]. Samples with the size of 40×40×40 mm³ were prepared for measurements. A compression machine with a maximum force of 3000 KN was used for testing. The loading rate for the test was set as 0.6 MPa/s. Three specimens were measured for obtaining an average value. The wet density of the samples was also tested at different curing times. The volume of the samples was measured by a water displacement method based on BS EN 12390-7:2019 [61]. The mass of the samples was weighted in the saturated surface dry (SSD) condition. Besides, the water absorption of the 90 days samples was measured according to ASTM C642 [62]. Before recording the dry mass, the samples were dried in a ventilated oven at 105 °C until constant mass was achieved.

228 2.3.3 Thermal conductivity

229 The thermal insulation property of HPLACs was assessed by a thermal conductivity test. Samples
230 with a dimension of $40 \times 40 \times 40 \text{ mm}^3$ were used for testing after curing at 28 and 90 days. A thermal
231 conductivity meter equipped with a Hot Disk Kapton sensor was employed for the measurements.
232 The measurement range of the device was from 40 down to 0.03 W/m/K. The temperature during
233 the measurement was 21 °C. At least five samples from each mixture were tested to obtain an
234 average value.

235

236 2.3.4 Isothermal calorimetry

237 The isothermal calorimetry was tested to monitor the hydration kinetic of HPLACs prepared with
238 pre-soaked LWAs. The hydration heat of samples containing different amounts of the SLWA was
239 recorded and the cases of 100R and 100C were also tested to make comparisons with the SLWA-
240 incorporated samples. The proportions of the samples complied with Table 3. The fresh mixtures
241 located in insulated containers were transferred to an isothermal calorimeter (Calmetrix I-CAL)
242 for measurements. The heat of hydration was monitored up to 168 h at a constant temperature of
243 20 °C.

244

245 2.3.5 Durability related properties

246 The ion permeability of 100R and 100S was determined to evaluate the concrete's resistance to
247 chloride-ion penetration according to ASTM C1202 [63]. This rapid electricity flux method has
248 been widely used for characterizing the permeability of high strength lightweight concrete [9, 19],
249 although it is not without limitations [64]. The 28 days cured samples were used for testing. Slice
250 samples of 50 mm thickness were cut from cylinder concrete samples. Before testing, the samples
251 were vacuumed for 3 h, followed by saturated in water for 18 h. A voltage of 60 V was exerted on
252 a cell with one side filled with 3% NaCl solution and the other side 0.3 M NaOH solution. The
253 total charge passed after 6 h was recorded as an indication of chloride ion penetrability.

254

255 The autogenous shrinkage of 100R and 100S was measured using the corrugated tube method
256 according to ASTM C1698 [65]. The fresh mixtures were filled into polyethylene tubes (420 mm
257 in length and 29 mm in outer diameter) held vertically while the vibrating table was switched on.
258 The measurement of autogenous deformation was started after setting and continuously monitored
259 for 14 days under the sealed condition at 23 °C. The data were recorded every minute during the

260 testing period.

261

262 2.3.6 Microstructure tests

263 X-ray diffraction (XRD, Rigaku SmartLab-Advance) equipped with a 9 kW rotating anode X-ray
264 source ($\lambda \sim 1.54 \text{ \AA}$) was employed to identify the mineralogy of the aggregates (RS, CLWA, and
265 SLWA). A high quality semiconductor detector (HyPix-3000) was used to scan the powdered
266 samples. The scan ranged from 10° to 70° in this measurement with a scan speed of $5^\circ/\text{min}$ and a
267 0.01° step size. To quantify the crystal phases in the aggregates, 20 wt% of corundum was added
268 to the samples as an internal standard. Rietveld refinement was conducted to analyze the data by
269 using Topas 4.2 software.

270

271 A mercury intrusion porosimeter (MIP, Micromeritics AutoPore IV 9500) was adopted to analyze
272 the pore structure of LWAs. Pores with sizes from $360 \mu\text{m}$ down to 5 nm could be measured via
273 the intrusion of mercury into the connected pores under an intrusion pressure of up to 227 MPa .
274 The pore shapes were assumed to be cylindrical and the contact angle of the mercury on the solid
275 surface was taken as 140° [66]. Despite the debate using MIP for measuring porosity due to the
276 irregularity of pore geometries and ink-bottle effects [67], the information obtained is still useful
277 for qualitative analysis and comparison [53]. The surface of the raw LWA particles and their
278 interior structure were observed using a scanning electron microscope (SEM, Tescan VEGA3).
279 The backscattered electron (BSE) imaging mode was performed on the carbon-coated polished
280 cross section of LWAs to explore the internal pore connectivity in the LWAs. The pores in the
281 LWAs were filled with epoxy and then the samples were polished before BSE observation. For
282 the concrete samples, the cube samples after 90 days curing were cut into slices and then
283 embedded in epoxy and polished by the different fineness of papers (180 grits , $9 \mu\text{m}$, $3 \mu\text{m}$, 0.05
284 μm). After obtaining a smooth surface, the samples were cleaned in ethanol in an ultrasonic bath
285 for 5 mins to remove the residual suspensions on the surface. Before the BSE testing, the samples
286 were dried in a 40°C vacuum oven for 3 days and coated with carbon for conduction. Energy-
287 dispersive X-ray spectroscopy (EDX) was conducted to analyze the elemental distribution of the
288 matrix. The above SEM/BSE observations were carried out under 20 kV voltage.

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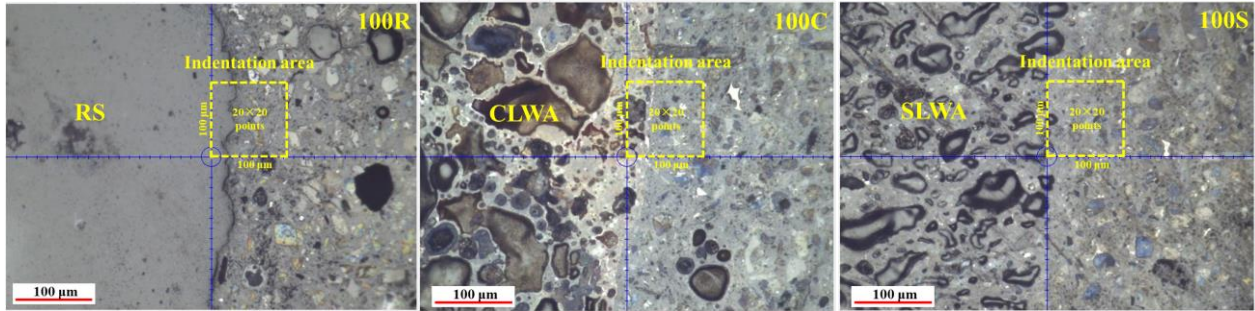


Fig. 3. Indentation areas in 100R, 100C and 100S.

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291 A Bruker's Hysitron TI Premier nanoindenter was employed to measure the micromechanical
 292 properties of the interfacial zone between the aggregate and the paste. A square indentation grid of
 293 400 points (20×20) was programmed for the indentation (see Fig. 3) and the interval distance
 294 between the grids was $5 \mu\text{m}$. A trapezoid loading program was used for the measurement. The
 295 indentation force was loaded up to a maximum value of $2000 \mu\text{N}$ at a rate of $400 \mu\text{N/s}$ using a
 296 diamond tip with Berkovich triangular pyramidal. Then, the maximum force was held for 2 s,
 297 followed by unloading within 5 s. The hardness and elastic modulus of the indented points were
 298 obtained according to the depth-load curves [68, 69]. The preparation of samples (90 days of curing)
 299 for the nanoindentation test was similar to that of the SEM-BSE testing. However, to achieve a
 300 very smooth surface, the samples were further polished by fine grits ($9 \mu\text{m}$, $3 \mu\text{m}$, $0.05 \mu\text{m}$) for 30
 301 mins, respectively.

302

303 The pore structure in the paste of HPLACs was evaluated by the Brunauer-Emmett-Teller (BET)
 304 test (the MIP method would involve the information of pores in LWAs). A porosimetry instrument
 305 with a two-station degas system (Micromeritics ASAP 2020) was used for nitrogen adsorption and
 306 desorption. Small fragments ($75\text{-}315 \mu\text{m}$) crushed from the 90 days cured samples were used in
 307 this test. The steel fibers in the samples were removed by a magnet. The vacuum degassing
 308 temperature was set at $60 \text{ }^\circ\text{C}$ to remove the remaining water from the smallest pores, followed by
 309 an analysis procedure under liquid nitrogen conditions. Pore information in the range of $2\text{-}100 \text{ nm}$
 310 could be obtained from this measurement. Since the HPLACs contained LWAs, the actual pore
 311 information of the paste matrix was calculated by subtracting the pores in the LWAs in the samples.

312

313 A high resolution X-ray computed tomography (μCT , YXLON FF35) was employed to observe
 314 the distributions of LWAs and fibers in the HPLACs in a non-destructive way. $40 \times 40 \times 40 \text{ mm}^3$
 315 cube samples were used for CT scanning and the internal voids in the LWAs (about 5 mm in

316 diameter) were also detected. The X-ray tube was operated at 200 kV voltage and 65 μ A current
317 for the concrete samples and 100 kV voltage and 60 μ A current for the LWA particles. For both
318 samples, 1800 projections of two-dimensional (2D) images were captured in a 360° rotation with
319 an integration time of 1.67 s. 0.1 mm tin and 0.1 mm copper were taken as filters for concrete and
320 LWAs samples, respectively. The three-dimensional (3D) volumetric structures were generated via
321 real time reconstruction software (CERA). Then, the 2D and 3D CT images were visualized and
322 analyzed by VGStudio MAX.

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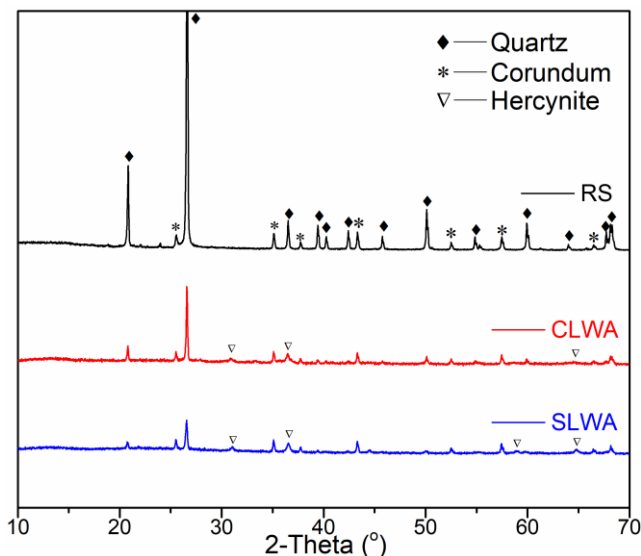
324 **3 Experimental results**

325 **3.1 Physicochemical properties of LWAs**

326 3.1.1 Mineralogy of LWAs

327 The XRD spectra of the different aggregates are displayed in Fig. 4. As seen, the distinct peaks
328 of quartz were identified in all aggregates. However, the intensity of the quartz signal was
329 strongest in the RS, and lowest in the shale-based SLWA. To some extent, this implies the highest
330 content of quartz in the RS and the lowest content of quartz in the SLWA. The results are
331 consistent with the Q-XRD analyses shown in Table 4, which presents that the order of quartz
332 content was RS (sand)>CLWA (clay)>SLWA (shale). On the other hand, the amount of
333 amorphous phases in the aggregates followed the sequence: SLWA>CLWA>RS. From the
334 diffractograms, hercynite (FeAl_2O_4 , PDF#34-0192) was also observed as crystal phase in the
335 LWAs and its content was slightly higher in the CLWA than that in the SLWA (see Table 4). Zhang
336 and Gjørsv also compared the mineral compositions of various LWAs (including Liapor, Leca, and
337 Lytag), they found different minerals in the various LWAs due to the completely different raw
338 materials and manufacturing procedures [70]. In this study, although the two types of LWAs were
339 produced with different raw materials (clay and shale), they still had similar chemical and mineral
340 compositions. However, the fraction of each mineral in the two types of LWAs was different,
341 which might induce different chemical reactivities. As indicated by previous studies [70, 71], the
342 LWAs possessed some pozzolanic reactivity due to the high temperature manufacturing process
343 (typically in the range of 1000-1200 °C [72]). The ions (e.g. Al and Si) released from the LWAs
344 would react with $\text{Ca}(\text{OH})_2$ to form additional hydrates [71, 72]. Hence, the higher content of
345 amorphous phases revealed that the LWAs would likely involve in the pozzolanic reaction with

346 the surrounding paste.



347 Fig. 4. XRD patterns of the RS, SLWA and CLWA (with 20 wt.% corundum).

348 Table 4. Phase contents in the aggregates analyzed by Q-XRD (wt%).

Aggregates	RS	CLWA	SLWA
Quartz	89.9	31.8	15.0
Hercynite	-	9.7	8.3
Amorphous	10.1	58.5	76.7

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350 3.1.2 Pore structure of LWAs

351 The pore structures of the LWAs measured by MIP are presented in Fig. 5. The pore size of the
 352 SLWA was much smaller than that of the CLWA. Most of the pores in the CLWA were larger than
 353 1 μm , while the pores in the SLWA were mainly smaller than 1 μm . The MIP results indicate that
 354 the SLWA had a much finer pore structure than the CLWA. As indicated in ref. [53], the density
 355 and water absorption process of LWAs were related to their pore structures. Pores larger than 1 μm
 356 in the LWAs would be saturated rapidly but the absorbed water was desorbed easily, while water
 357 in smaller pores was sorbed and desorbed more slowly [55]. Therefore, it is anticipated that the
 358 internal curing effect provided by the SLWA would be adequate due to the steady and gradual
 359 water desorption. In any case, however, a large number of pores in the LWAs could provide spaces
 360 to accommodate water for internal curing. Moreover, the fine and uniform pore structure in the
 361 SLWA might account for its higher crushing strength as compared to that of the CLWA (Table 2).

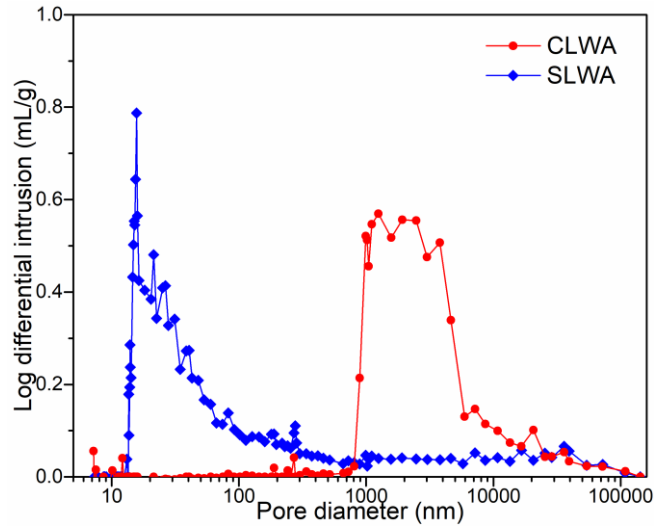


Fig. 5. Pore size distributions of SLWA and CLWA.

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363 In order to closely examine the structure of the two lightweight aggregates, SEM-BSE was
 364 employed to observe the surface and internal pores. The appearance of the CLWA surface shown
 365 in Fig. 6a was full of cavities, but the cavities seemed to be non-connected. From the cross-section
 366 of the CLWA, an outer shell of about 500 μm thick can be identified in Fig. 6b. This relatively
 367 denser shell covered the internal pores, whose sizes were much larger than the pores in the shell.
 368 These behaviors are in line with the observation of another study [53], which showed that the outer
 369 shell of expanded clay aggregates was much less porous than the interior. Fig. 6c shows the pores
 370 of the SLWA. As the SLWA was crushed from the big bulk of expanded shale, the surface was very
 371 rugged and the pores were opened. A close examination in Fig. 6d displays that the surface pores
 372 with different sizes were separated by the shale matrix. Zhang and Gjrv [73] found that the
 373 penetration of cement paste into LWAs depended on the surface of the LWAs. The cement paste
 374 did not penetrate through the dense shell, while it could penetrate into the opening pores of surface.
 375 Hence, for the SLWA, the bond was expected to be better due to an improved mechanical
 376 interlocking between the aggregates and the cement paste.

377

378 BSE images of the polished cross-section in Fig. 7 show the interior pores in the CLWA and SLWA.
 379 For the CLWA (Fig. 7a), a clear shell in the outer part of the aggregate was observed. In contrast,
 380 the interior structure in the CLWA was observed more porous. As shown in Fig. 7b with greater
 381 details, many interior pores were well-interconnected or separated by a thin wall. This
 382 characteristic of pore structure was prone to water absorption and storage. However, the large voids

383 inside the CLWA would make the aggregate weak and friable [53]. Figs. 7c and 7d present the
 384 interior structure of the SLWA. The cross-section images demonstrate the open-cell pores on the
 385 surface of the particle. In comparison with the CLWA, the interior pores in the SLWA were more
 386 isolated, and the walls between the pores were thicker. The refinement of the pore structure might
 387 be responsible for the higher strength of the SLWA compared to the CLWA (see Table 2).
 388 Additionally, the open pores of the SLWA could increase the bonding area with the paste and
 389 reinforce the interfacial pozzolanic reaction, in contrast, the densely sintered shells on the surface
 390 of the CLWA would lower chemical reactivity. The surface structure of SLWA was similar to that
 391 of Lytag (sintered fly ash) without a dense shell, which facilitated to form a more dense and
 392 homogeneous ITZ [74]. For the LWA with a dense shell, some crystalline hydration products, such
 393 as $\text{Ca}(\text{OH})_2$ and ettringite, would accumulate at the dense outer layer to make the ITZ more porous
 394 [74]. As a consequence, although the two types of LWAs had similar porosities (Table 2), they had
 395 different pore structures and geometries, both of them affecting the overall strength of HPLACs
 396 by processes which took place in the ITZ.

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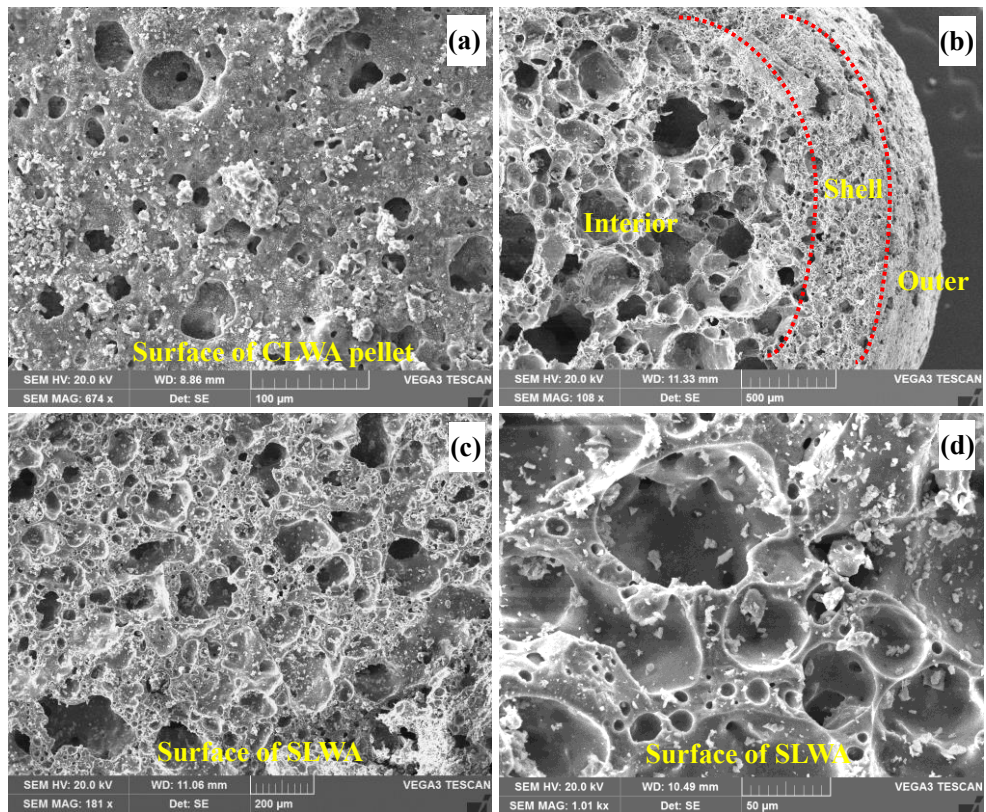


Fig. 6. SEM of pores in the CLWA and SLWA, (a) surface of CLWA; (b) internal pores of CLWA; (c) and (d) surface pores of SLWA.

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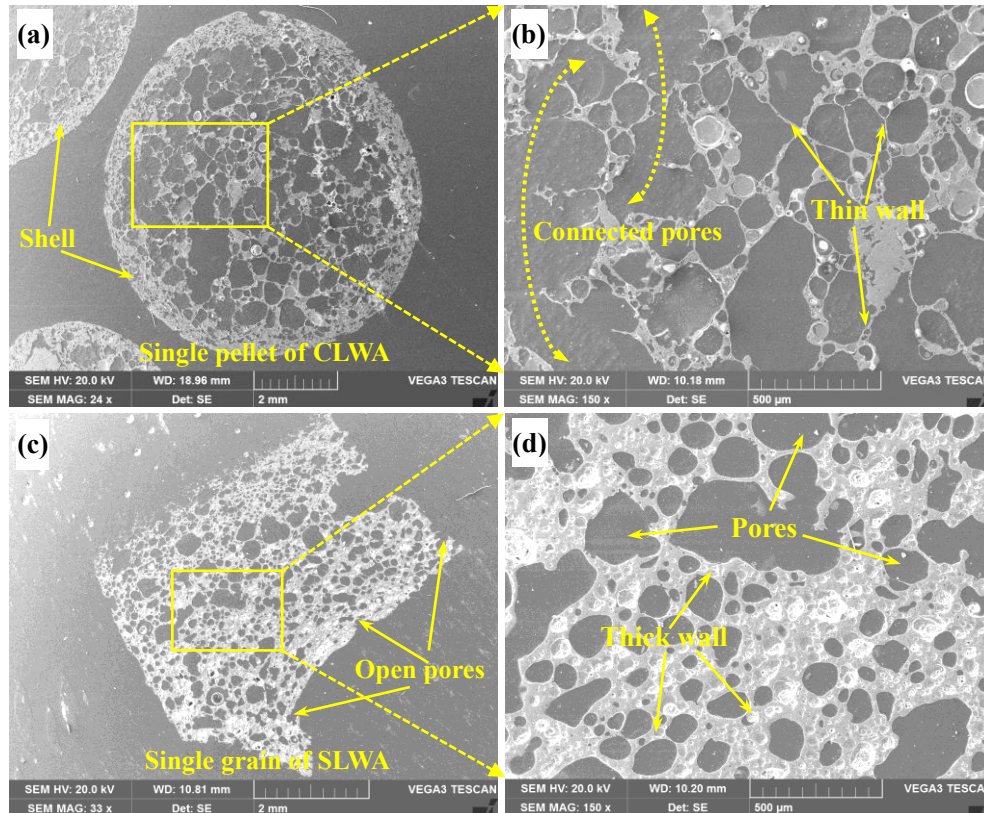


Fig. 7. BSE images of internal pores in the CLWA and the SLWA, (a) cross section of CLWA; (b) magnification of (a); (c) cross section of SLWA; (d) magnification of (c).

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3.1.3 Pozzolanic reactivity of LWAs

402 To assess the possible pozzolanic reactivity of the CLWA and SLWA in the alkaline cementitious
 403 system, Chapelle test was performed to obtain calcium hydroxide (CH) consumption by the
 404 different aggregates (which were in powdered forms). Fig. 8a shows that the RS had the least value
 405 of CH consumption and the SLWA consumed the highest amount of CH, which means that the
 406 SLWA possessed the highest pozzolanicity among the tested aggregates. Accordingly, the SLWA
 407 had a higher material activity than the CLWA. Since the SLWA was processed by rapid cooling
 408 from high temperatures and mainly composed of SiO_2 and Al_2O_3 , it could be possible that Al and
 409 Si would be released from the SLWA to react with CH [71]. It should be noted that per gram of the
 410 SLWA could react with 527 mg CH, which is similar to that of a typical pulverized fly ash
 411 (classified as a highly reactive SCM) [57]. Also, Kong et al. [72] investigated the chemical
 412 reactivity of LWAs (sintered fly ash and expanded clay/shale) and found that the finely ground
 413 LWAs had similar pozzolanic reactivity with fly ash. The powdered expanded slate and shale
 414 aggregates were also found to have pozzolanic nature, the consumption of CH was comparable to

415 or higher than fly ash [75]. Tasong et al. [76] indicated that the natural aggregates were chemically
 416 active and the ions released and absorbed by the aggregates possibly took part in the hydration of
 417 the surface layer. Therefore, the result obtained implies that the studied SLWA would exert a
 418 high/moderate pozzolanic reaction at the interfacial zone in the cement-based matrix. Certainly,
 419 the LWAs in the form of powder should be more reactive than in the form of coarse aggregates,
 420 thus if the SLWA was ground into powder form, it may be used as a potential SCM. From Fig. 8b,
 421 it can be seen that the residue of the reacted samples experienced a severe mass loss from 100 to
 422 300 °C. Thus, the bound water of the residual samples of RS, CLWA, and SLWA calculated in this
 423 temperature range was 13.0%, 19.4%, and 24.8%, respectively. These results further confirmed
 424 that the order of pozzolanic reactivity among the aggregates was $SLWA > CLWA > RS$. On the
 425 other hand, this sequence indicates that the larger amounts of Al and Si were released from the
 426 SLWA and reacted with CH to form hydration products compared to the CLWA.
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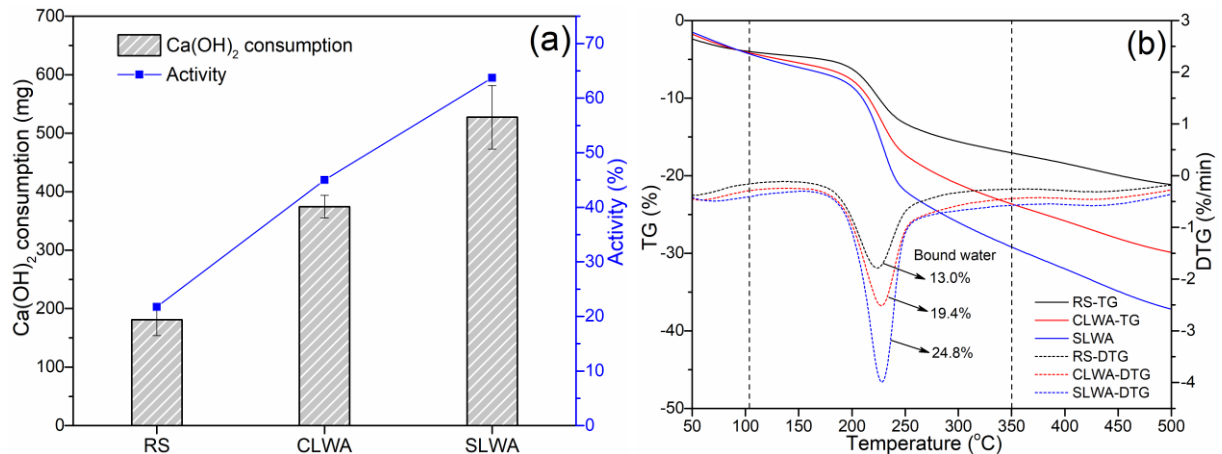


Fig. 8. Pozzolanic reactivity of aggregates determined by Chapelle tests, (a) Consumptions of Ca(OH)₂ by per g of aggregate; (b) TG/DTG analyses of residual solids after Chapelle tests.

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430 3.2 Physical and functional properties

431 3.2.1 Density and water absorption

432 The density and water absorption values of HPLACs are shown in Fig. 9. With the incorporation
 433 of the two types of LWAs, the density of samples was reduced significantly, in particular when the
 434 RS was fully replaced by the LWAs (Fig. 9a). With further increase in the content of LWAs, the
 435 reductions in the density were not remarkable, regardless of the samples prepared with the CLWA

436 and SLWA. Since the density of RS was much higher than those of LWAs (see Table 1), the density
437 of the samples dropped significantly to around 1850 kg/m^3 by replacing 100% RS. A recent study
438 [77] also reported a similar density with the use of SLWA in the concrete, regardless of quasi-
439 spherical shale and crushed shale. In general, the density of lightweight concrete is associated with
440 the density of LWAs. If the LWAs with low density were used, the lower overall density of concrete
441 would be attained. Therefore, some studies [78-80] incorporated very lightweight materials (e.g.
442 aerogel, straw and expanded polystyrene) into the UHPC system to reduce the density and ensure
443 the high strength. However, with the reduction in the density, the strength of lightweight concrete
444 would be reduced inevitably. In this study, the influence of these two types of LWAs on the
445 reduction of density seemed to be similar. This is because the pre-wetted densities of the CLWA
446 and SLWA were similar (0.96 vs 1.1 g/cm^3).

447

448 Water absorption is an essential factor which is primarily related to the durability of the concrete.
449 Fig. 9b presents the water absorption of HPLACs prepared with the two types of LWAs. In both
450 cases, the water absorption of the samples was increased with the increasing content of the LWAs.
451 Especially for the CLWA containing samples, the water absorption was increased significantly
452 with the decrease in density. Furthermore, the inclusion of the porous expanded clay aggregates
453 with interconnected pores might have a negative effect on water retention. This result is in
454 agreement with the findings of other works [28, 81], which indicated that the replacement of
455 natural aggregates by LWAs and the increased amount of LWAs in the mixture increased the water
456 absorption of the lightweight concrete. However, in this study, the water absorption values of
457 HPLACs were still much lower than those of conventional lightweight concrete [81, 82] and
458 normal weight cement mortar and concrete [83, 84] (mostly $>7\%$). As the water absorption was
459 less than 5% , the HPLACs prepared in this study were expected to have a good quality [28]. By
460 contrast, the increase of water absorption of the expanded shale-based samples was not
461 significantly changed with the increasing aggregate content. The results mean that the HPLACs
462 prepared with the SLWA had a more compact structure than those prepared with the CLWA at a
463 similar level of density.

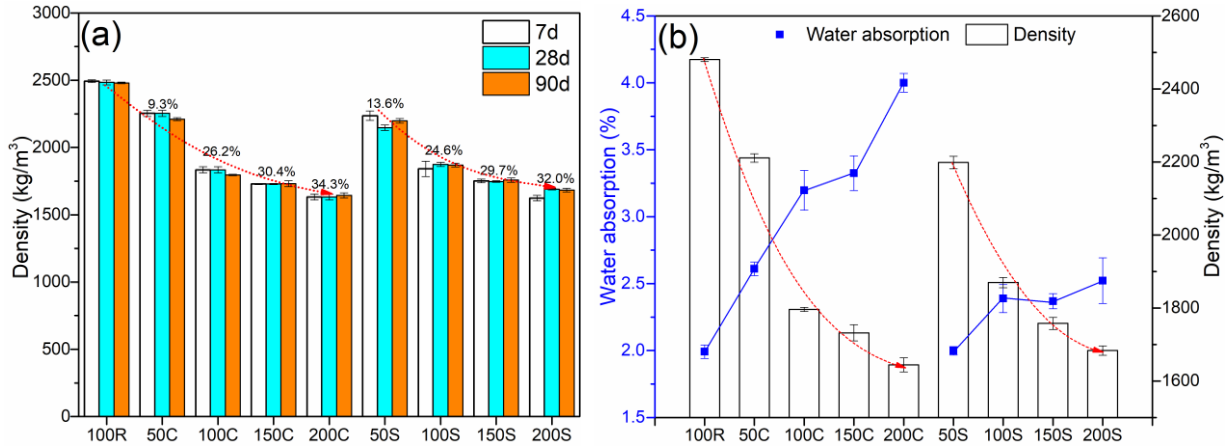


Fig. 9. SSD density (a) and water absorption (b) of HPLACs.

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3.2.2 Structural efficiency

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The compressive strength and corresponding structural efficiency of HPLACs are presented in Fig. 10. When the LWAs were used to replace RS, the compressive strength was reduced considerably due to their intrinsic weak strength. In particular, a large reduction in the strength was observed in the case of the CLWA-incorporated samples. The strength of the 100C dropped by 70% as compared with the reference sample at 90 days. Further increasing the content of the CLWA only slightly reduced the strength. This might be due to the fact that the large proportion of the CLWA had formed a network packing in the matrix to stabilize the strength reduction. For the HPLACs prepared with SLWA, the reduction of strength appeared to be less pronounced in comparison with the CLWA system, especially when the RS was fully replaced by the SLWA. The strength was substantially higher than those of the CLWA samples. A high strength exceeding 70 MPa could still be achieved for the 100S. Based on the results of ref. [77], the use of the crushed shale LWAs in concrete could attain higher strength than the use of the quasi-spherical shale LWAs because the angular shape and rough surface of the LWAs could provide a better bonding with the cement paste. Hence, the SLWA with open-pore surface in this study might contribute to the high strength as compared to the CLWA. With the increase of the SLWA content, the strength decreased as expected.

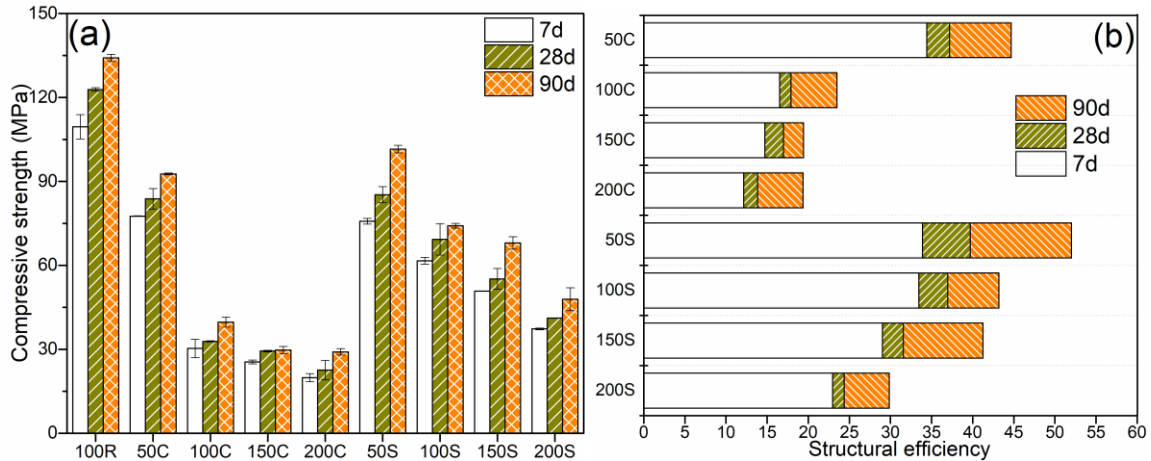


Fig. 10. Compressive strength (a) and structural efficiency (b) of HPLACs.

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Structural efficiency (strength/density, $\text{kN}\cdot\text{m}/\text{kg}$) is a parameter to represent the material efficiency for lightweight concrete. The higher value means a higher efficiency to reduce the dead load of the concrete structures and vice versa. Fig. 10b shows the structural efficiency of HPLACs prepared with two kinds of LWAs. The results indicate that the structural efficiencies of HPLACs prepared with SLWA were obviously higher than those of the CLWA incorporated samples. Therefore, the introduction of the SLWA in the HPLACs was more effective in reducing the structures' weight at the same grade of density compared to the use of the CLWA. Several physicochemical properties of the SLWA may be responsible for this behavior: (i) the finer pore structure of the SLWA (see Section 3.1.2) resulted in a higher aggregate strength (see Table 2); (ii) the open-pores surface of the SLWA (see Section 3.1.2) were conducive to improving the bonding between the aggregates and the paste (to be discussed in Section 3.3.2); (iii) the coupled effects of internal curing and high pozzolanic activity of the SLWA facilitated to form good bonds with the paste to partly counteract the weak strength of the aggregates (to be discussed in Section 3.3.3). However, it is found that the structural efficiency of the SLWA-containing HPLACs decreased significantly with the increased amount of the SLWA. This is because the strength was reduced largely while the density did not decrease too much. Another reason might be due to the incorporation of relatively larger size of the SLWA in the mixture. As reported by previous studies [34, 85], the use of LWAs with small size resulted in a higher compressive strength of lightweight concrete. This finding was confirmed by our recent work [8], which incorporated micro-sized lightweight materials into UHPC system to achieve very high structural efficiency ($>65 \text{ kN}\cdot\text{m}/\text{kg}$).

504 3.2.3 Thermal conductivity

505 The thermal conductivity test results are shown in Fig.11a. The reference sample (100R) had a
506 high thermal conductivity (2.32 W/mK at 90 days) due to the use of quartz sand with very high
507 thermal conductivity (18.37 W/mK [86]). The replacement of the RS by the LWAs reduced the
508 thermal conductivity remarkably. For the 100C and the 100S, the thermal conductivity values were
509 reduced by 68% and 62% in comparison with the reference sample, respectively. The porous
510 structure of the LWAs hindered the conduction of heat through the matrix. Also, as the UHPC
511 matrix was water-deficient, the absorbed water in the LWAs would be consumed by the hydration
512 of the surrounding UHPC matrix (internal curing effect) [87]. Therefore, the empty pores in the
513 LWAs with an extremely low thermal conductivity (0.026 W/mK for air) was beneficial to
514 lowering the thermal conductivity. This explanation is also supported by the lower thermal
515 conductivity at the longer hydration time (90 days of curing age). For the HPLACs prepared with
516 the LWAs, the thermal conductivities seemed to be similar at the same aggregate content, which is
517 due to the similar porosity of the CLWA and the SLWA (see Table 2). When the content of the
518 LWAs was increased up to 200%, the thermal conductivity only went down a little. This trend is
519 consistent with the development of density. The thermal conductivity also shows a linear
520 relationship with the density value (Fig. 11b). In other words, the thermal conductivity of concrete
521 decreased with the decreasing of mixture density. Hence, the thermal conductivity of the HPLACs
522 was much lower than that of normal weight mortar/concrete (about 2.0 W/mK [88, 89]), implying
523 a better heat insulation of mixture. If the HPLACs were used in the construction sector, energy
524 consumption in the structures would be reduced indirectly. However, when the content of the
525 LWAs exceeded 100%, the thermal conductivity tended to be stable. Thus, to ensure a high strength
526 of concrete, a proper amount of the LWAs should be considered in the design of the HPLACs.

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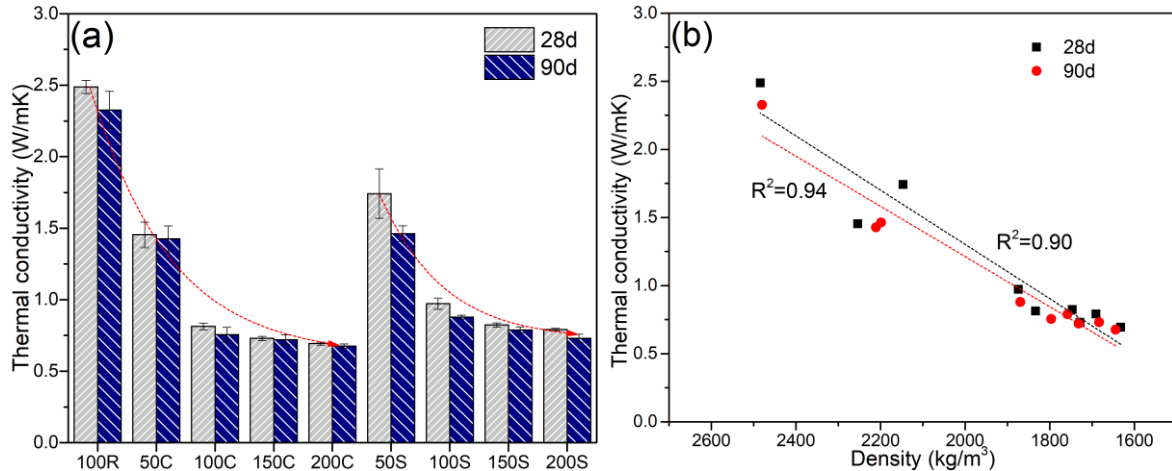


Fig. 11. Thermal conductivity of HPLACs (a) and its relationship with density (b).

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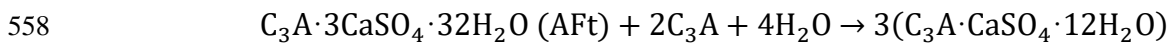
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530 3.3 Hydration kinetic and microstructure

531 3.3.1 Heat of hydration

532 Fig. 12 shows the heat flow of the HPLACs incorporated with the SLWA and CLWA as a result
 533 of hydration. As demonstrated in Fig. 12a, with the use of SLWA in the HPLACs, the exothermic
 534 rate of binder hydration was increased. The peak of silicate reaction was intensified with the
 535 increasing amount of the SLWA, which implies the accelerating hydration due to more SLWA
 536 inclusion. This result is also in agreement with a previous work [87], which showed that the heat
 537 flow of mixture was closely related to the released water from the LWAs. It should be noted that
 538 the LWAs were pre-soaked in water for 24 h prior to use. Therefore, the additional water available
 539 in the SLWA is thought to promote the hydration of the cementitious binder. Especially for the
 540 UHPC system, there were a lot of anhydrous cement clinker phases present in the matrix due to
 541 the use of an extremely low w/b ratio for mixing [90]. When the internal relative humidity of
 542 mixture started to decrease, the water desorption from the LWAs was triggered to maintain the
 543 humidity at a high level [87]. The increased humidity in the UHPC matrix accelerated the
 544 hydration of cementitious materials and delayed the occurrence of self-desiccation at the early
 545 age. Another interesting phenomenon is the occurrence of an intense shoulder peak by the use of
 546 the SLWA in the heat flow curve, which is usually associated with the formation of the AFm
 547 phase ($C_3A \cdot CaSO_4 \cdot 12H_2O$) in the hydration of OPC [91, 92]. This monosulfoaluminate phase
 548 was formed from the reaction of ettringite (AFt) with the remained C_3A (tricalcium aluminate)

549 and water [93], according to the following formula [94]. Therefore, the water released from the
550 LWAs was able to facilitate the conversion from AFt to AFm based on Le Chatelier's principle.
551 Our previous study [95] also validated a stronger third peak with the increasing water content in
552 the cement paste. Another possibility for the modified peak is that ions leaching (e.g. Al) from
553 the SLWA might enhance the intensity of the aluminate peak, similar to the effect of Al-rich
554 metakaolin [96, 97]. As indicated, a high concentration of Al was dissolved from the SLWA in
555 saturated limewater [71]. Hence, it is plausible that the SLWA with a higher reactivity might
556 provide more Al for the aluminate reaction. The Al dissolution from the SLWA will be discussed
557 in the next section.



559 Fig. 12b shows the comparison of heat evolution between the CLWA samples and the SLWA
560 samples. The replacement of the RS by 100% CLWA also expedited the hydration process.
561 However, the heat flow curve of the 100C did not exhibit the aluminate peak which inferred that
562 the presence of the CLWA suppressed the aluminate conversion. One reason may be caused by
563 the less water was available for the conversion reaction as the water in the CLWA was released
564 earlier due to the larger pores in the CLWA. The other possible explanation is that a less amount
565 of Al dissolution from the CLWA was involved in the aluminate conversion. This speculation was
566 supported by a previous study [98], which reported that the Al in the pore solution of the mixture
567 with SLWA was higher than that with CLWA. However, further studies are needed to elucidate
568 this phenomenon. Fig. 12c presents the cumulative heat output of the samples up to 7 days.
569 Obviously, due to the higher exothermic heat flow, the total heat liberation was also increased
570 with the increased amount of LWAs. It can be found that the 100S released a higher amount of
571 heat than the 100C, which means a more profound internal curing effect of the pre-wetted SLWA
572 than that of the CLWA. The increased contact areas of the SLWA and the paste due to the rough
573 and open-pore surface of the SLWA might intensify the internal curing effect.

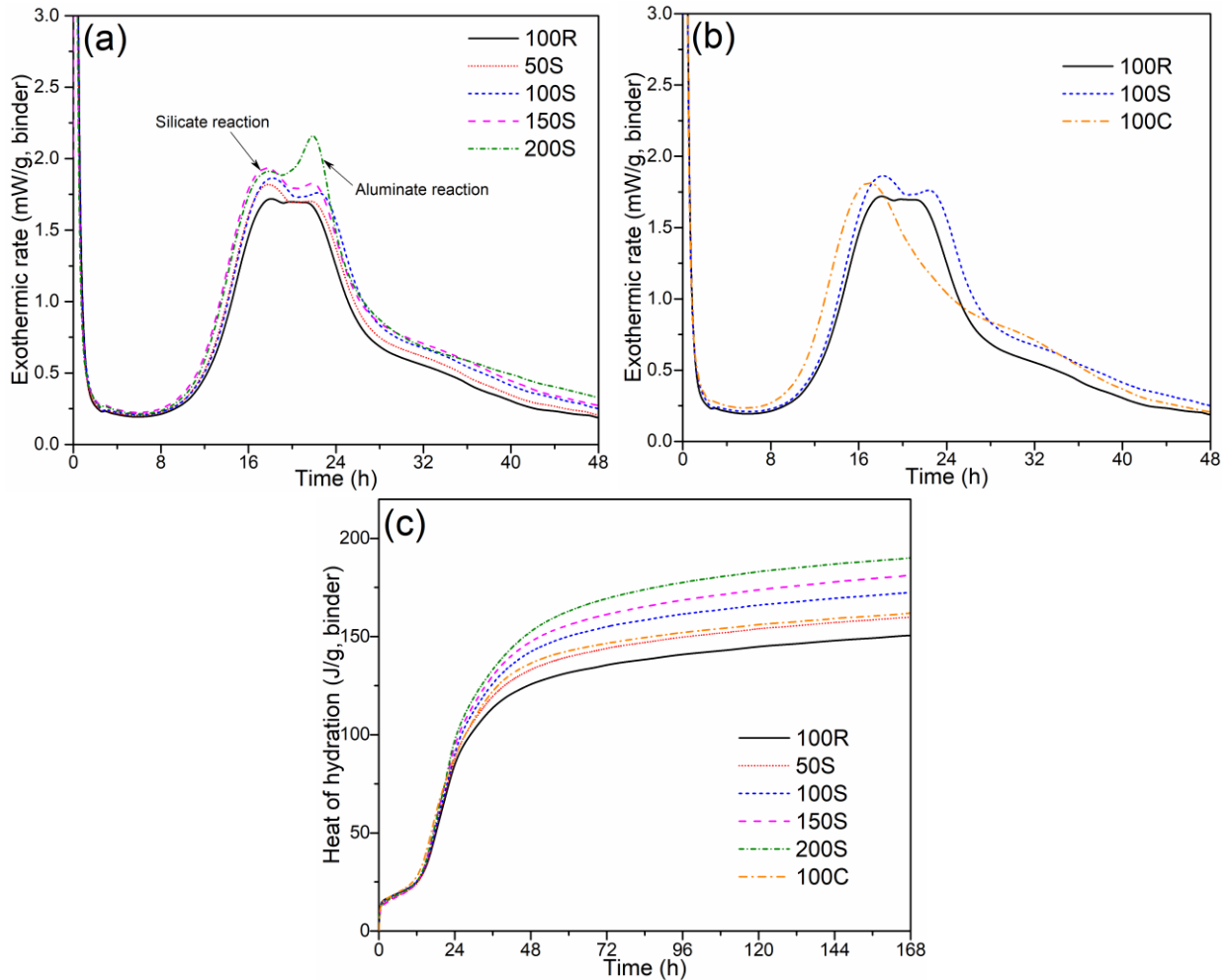


Fig. 12. Heat evolutions of HPLACs.

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576 3.3.2 Morphology and element distribution of ITZ

577 The morphologies of the interfacial transition zone (ITZ) around the different aggregates are
 578 presented in Fig. 13. Distinct boundaries were found at the interface between the RS and the bulk
 579 paste (Figs. 13a and b). There were a lot of unreacted cement grains existed indicating a low
 580 hydration degree of the UHPC matrix. Figs. 13c and d show the interface of HPLACs incorporated
 581 with CLWA. The porous CLWA was surrounded by dense paste and many unhydrated cement
 582 particles were also observed in the paste. However, a close view of the ITZ reveals that fewer
 583 unhydrated cement particles were located in the immediate vicinity of the CLWA. As the CLWA
 584 was pre-soaked before use, the desorbed water could promote the hydration of the cement around
 585 the aggregates. In the HPLACs prepared with SLWA (Figs. 13e and f), the internal curing effect
 586 induced by the pre-soaked SLWA was more pronounced because the width of the ITZ (with less

587 unhydrated cement particles) was larger. The cement grains were turned into finer particles within
588 the internal curing zone. This observation is consistent with the results of the hydration kinetic.

589
590 Infiltration of the paste into the cavities or open pores of the SLWA was observed (with uneven
591 boundaries in Figs. 13e and f), which contributed to the bonding between the two phases. Similar
592 observations were also found in other works [73, 74], for a LWA with dense outer layer, its ITZ
593 was similar to the case of normal weight aggregate, whereas more dense ITZ was formed at the
594 interphase of porous surface LWA. Therefore, due to the different microstructural characteristics
595 of the two kinds of LWAs, an improved mechanical interlocking between the SLWA and the cement
596 paste was observed. As reported by Wasserman and Bentur [99], this impregnation behavior could
597 lead to 20% increase in strength at 90 days. In conventional LWA concrete, portlandite and
598 ettringite were usually found to precipitate in the ITZ of LWAs [74, 100] since pre-wetting process
599 resulted in a layer of water amassing on the surface of LWAs, which were expected to induce
600 cracks. In this study, dense structure get rid of crystalline products deposition was observed in the
601 ITZ of LWAs. This is because the hydrophenic UHPC paste could consume most of water released
602 from the LWAs. Zhang and Gjørsv [101] also pointed out that, by reducing the w/b ratio and addition
603 of silica fume, the ITZ of the LWAs was improved. Table 5 presents the elemental compositions
604 of areas in the ITZ and paste matrix (Fig. 13e). The black point in the cement paste showed a high
605 Ca/Si ratio, while the blue and red points in the internal curing zone of SLWA had much lower
606 Ca/Si ratios. Another interesting behavior was the high concentration of Al in the ITZ of the SLWA
607 sample, which implies the chemical interaction between the SLWA and the paste in the form of
608 pozzolanic reaction.

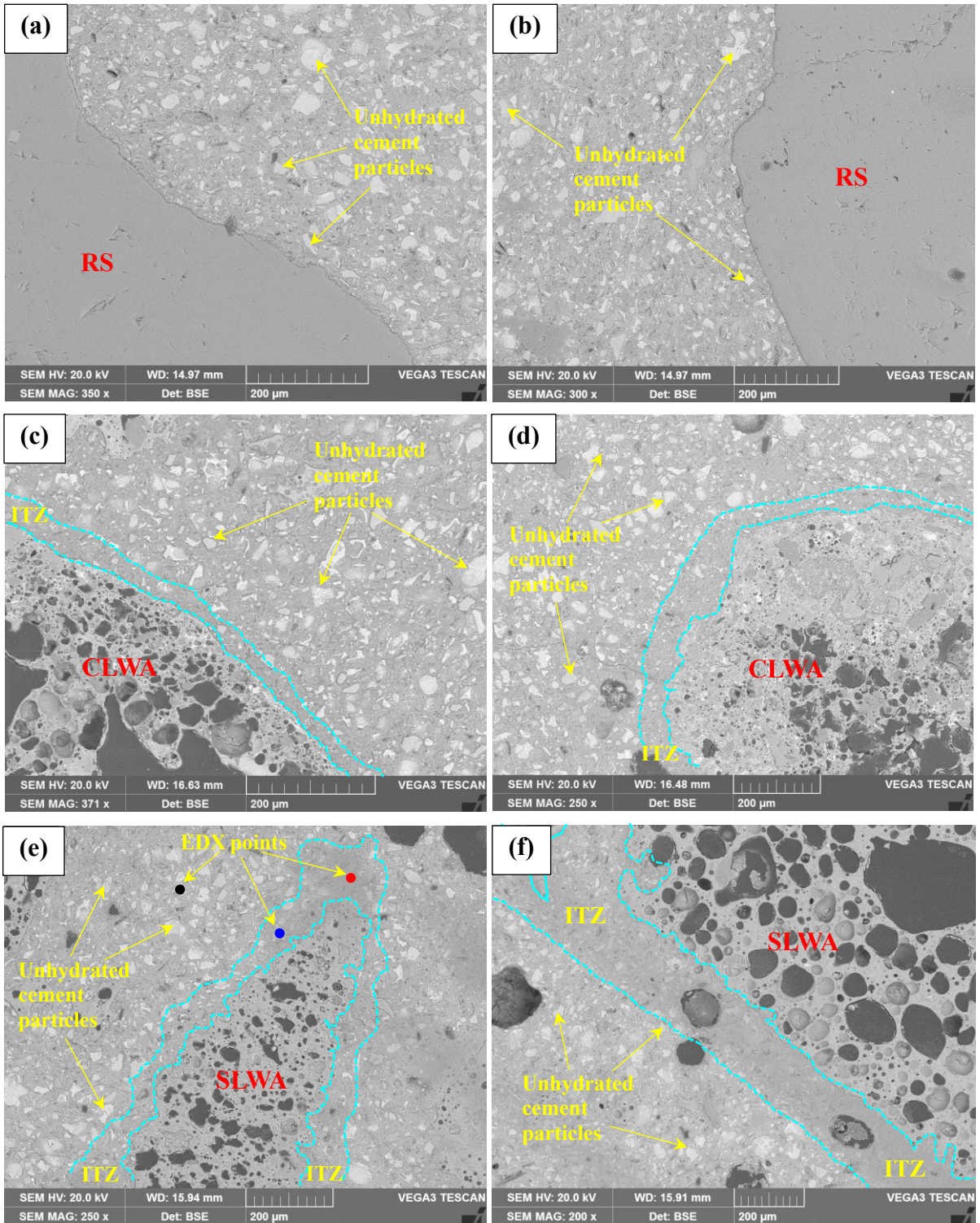


Fig. 13. SEM-BSE images of HPLACs, (a) and (b) 100R; (c) and (d) 100C; (e) and (f) 100S.

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612 Table 5 Elemental compositions of EDX points.

Element	Black point	Blue point	Red point
O	71.9	70.7	70.6
Al	4.6	11.6	20.6
Si	5.6	5.5	3.5
Ca	13.1	9.6	5.1
Ca/Si	2.34	1.76	1.46
Ca/(Si+Al)	1.28	0.56	0.21

613

614 To further understand the elemental distributions at the interface of HPLACs containing CLWA
615 and SLWA, EDX mapping was carried out and the results are shown in Fig. 14. As seen in the
616 frame mapping of 100C (Fig. 14a), the CLWA was rich in Si and Al, but deficient in Ca, which is
617 consistent with its chemical compositions (Table 1). At the interface, the Si and Al concentrations
618 were observed similar to those in the bulk paste but were lower than in the CLWA. The amount of
619 Ca in the ITZ was found lesser than that in the paste since lots of Ca-rich cement particles were
620 located at some distance from the aggregates. The elemental distribution of ITZ by line scan is
621 also shown in Fig. 14a. It can be noticed that the intensities of Si, Al, and Ca in ITZ were observed
622 close to those in the hydration products. These results imply that the internal curing caused by the
623 pre-wetted CLWA had little influence on the composition of the surrounding hydrates. Also, from
624 the linear mapping curves, an ITZ with approximately 60 μm width can be identified (but it should
625 be noted that the width of the ITZ would vary even with one LWA particle due to the distribution
626 of pore structures in the LWA). Fig. 14b presents the elemental distribution of ITZ in 100S. In the
627 frame mapping images, an interesting phenomenon can be observed: a higher concentration of Al
628 but lower concentrations of Si and Ca in the ITZ stripe in comparison with the bulk paste. The line
629 scanning curves clearly show the changes of concentration of these three elements from the paste
630 to the aggregate. A width of 150 μm ITZ could be noticed based on the elemental distributions.
631 Strong intensity of Al was observed within the ITZ with lower intensities of Si and Ca. These
632 results are in agreement with the EDX results in Table 5. The release of Al from the SLWA was
633 supported by other studies [71, 98], which demonstrated that the Al released from the shale-based
634 LWA could take part in the formation of calcium-aluminosilicate-hydrate (C-A-S-H). The study of
635 Nie et al. [71] also indicated that, as compared to the Si, the Al released from the LWA could
636 migrate more easily and farther away from the LWA. Therefore, it can be inferred that the presence
637 of pre-soaked SLWA changed the composition of its surrounding cement hydration products. As

638 mentioned in Section 3.1.3, the SLWA had a higher pozzolanic reactivity compared to the CLWA,
639 it is therefore suggested that the SLWA could react more intensely with the alkaline pore solution
640 to form C-A-S-H. Li et al. [98] found that both the expanded shale and clay could increase Al
641 content in the pore solution of concrete, but the expanded shale seemed to release more Al than
642 the expanded clay. This may be the reason of higher concentration of Al in the ITZ of the SLWA
643 mixture. The internal curing effect of the SLWA facilitated the cement hydration and pozzolanic
644 reaction in the ITZ. As a result, hydrates with a lower Ca/(Si+Al) ratio (Table 5) were formed in
645 the vicinity of the aggregates. Many studies [71, 102, 103] indicated that the incorporation of Al
646 in the C-S-H gel could lengthen the mean chain length of C-S-H structure, which might lead to a
647 denser microstructure and higher mechanical performance [104, 105]. Hence, the formation of the
648 additional C-A-S-H hydrates was expected to strengthen the ITZ (to be discussed in following
649 section). The reinforcement of the outer layer of the SLWA due to the continuity of the C-A-S-H
650 outside and inside the aggregates could also compensate for their lower initial strength. This would
651 be beneficial to ensuring the overall strength of concrete.

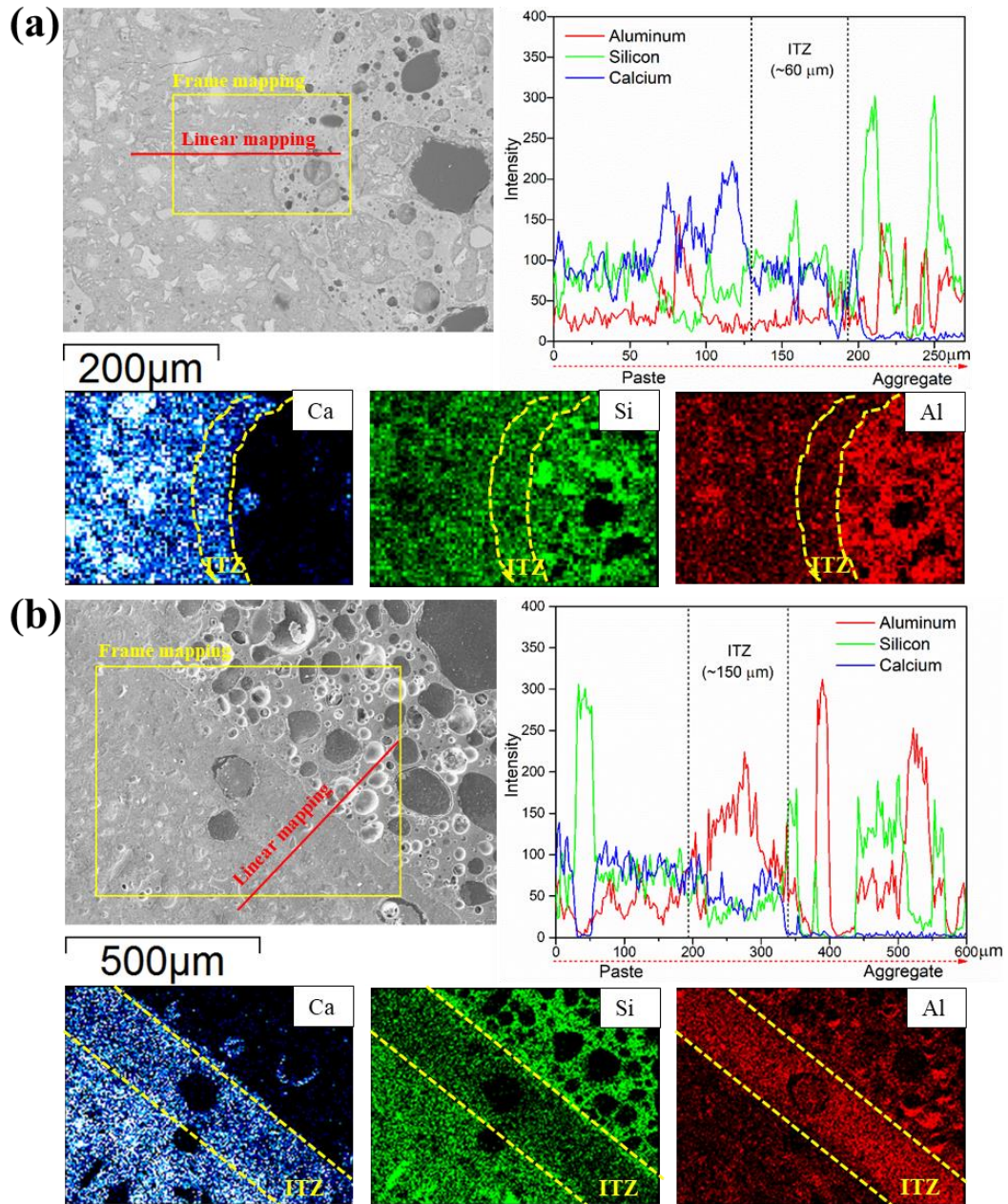


Fig. 14. Elemental mapping of ITZ, (a)100C and (b)100S.

652
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654

655 3.3.3 Micromechanical properties of ITZ

656 The nanoindentation technique was used to explore the micromechanical properties in the ITZ.
657 The ITZ boundary was demarcated by optical microscopy images and indentation matrices (Fig.
658 3). Fig. 15 shows the contour maps of micro-hardness and modulus of ITZ in the 100R, 100C, and
659 100S samples. The RS had a much higher hardness and modulus as compared to the bulk-paste,

660 thus the ITZ could be easily identified. Also, some unreacted clinker particles with high moduli
661 were found in the paste. It should be noted that a distinctive ITZ with a lower hardness and modulus
662 was present between the RS and the paste because the wall effect of the RS might increase the
663 local w/b ratio of the paste near the RS. The LWAs and the paste can be identified according to the
664 indentation images and the contour maps. The LWAs had a very low hardness and modulus since
665 they contained many pores. However, the ITZ around the LWAs seemed to have a higher hardness
666 and modulus.

667

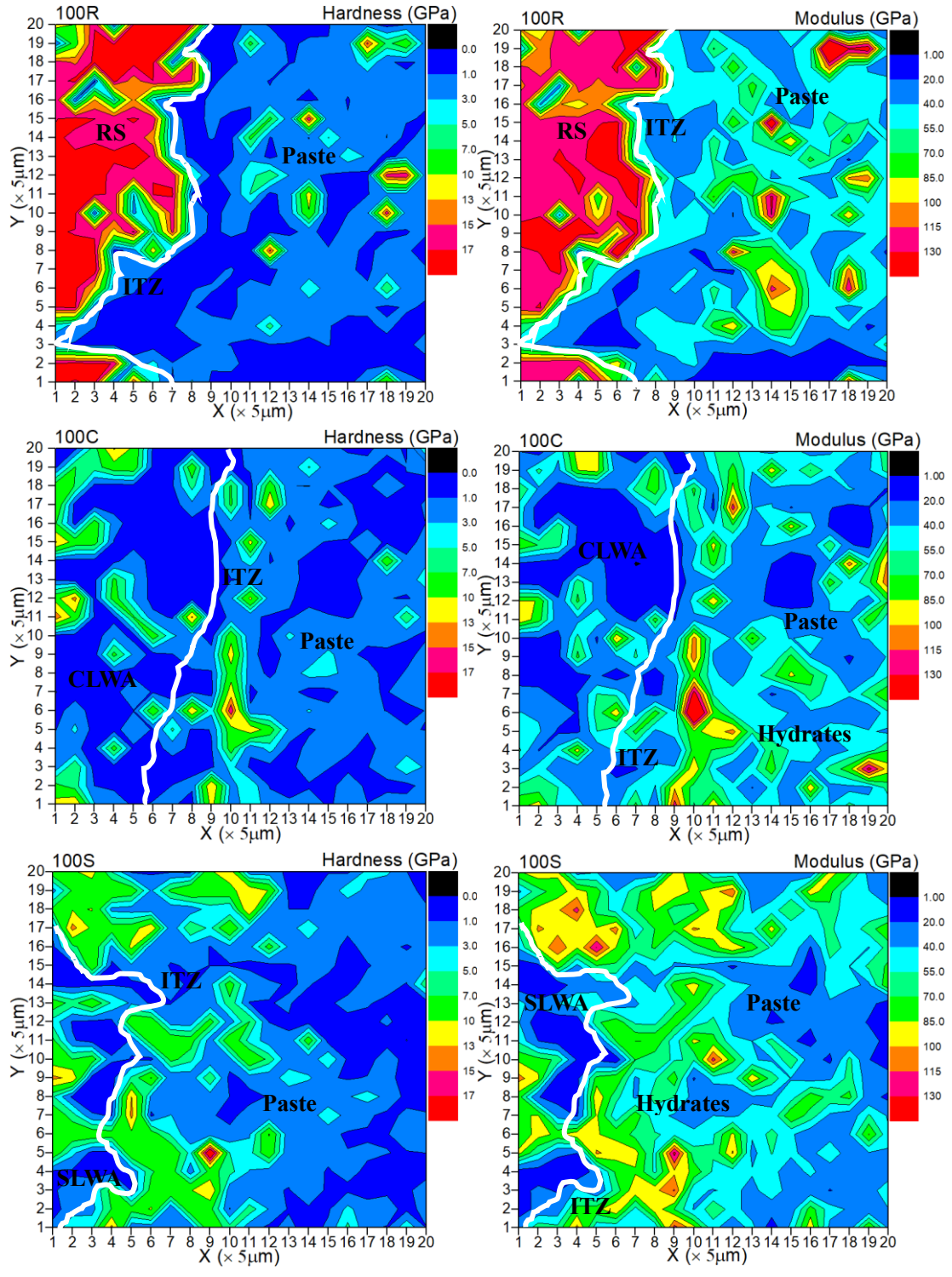


Fig. 15. Contour maps of micromechanical properties of ITZ in 100R, 100C and 100S.

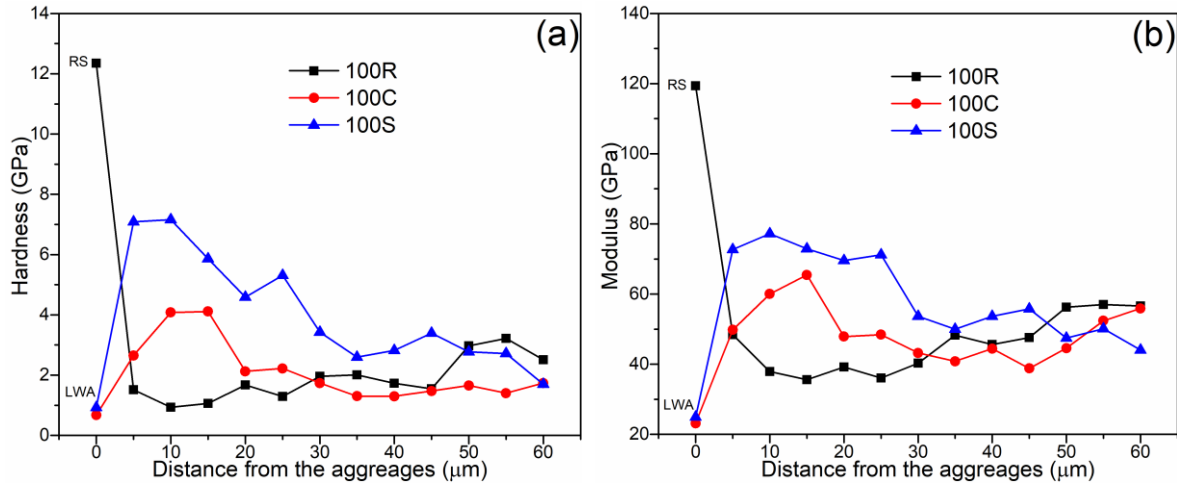


Fig. 16. Micromechanical properties of ITZ, (a) Hardness and (b) Modulus.

669
670

671 To quantitatively understand the evolution of the micromechanical properties in the ITZ, the
 672 average hardness and modulus values were plotted against the distance from the aggregates (to 60
 673 μm) in Fig 16. For the 100R sample, both the hardness and modulus dropped sharply in the ITZ
 674 and then increased gradually with the increasing distance away from the RS. On the contrary, for
 675 the samples prepared with LWAs, the hardness and modulus values at the regions adjacent to the
 676 aggregates were higher than that of the aggregate, followed by a decreasing trend reaching a stable
 677 value. Thereinto, the ITZ between the SLWA and the paste exhibited better micromechanical
 678 performance than that of the CLWA and the paste. These results suggest that the incorporations of
 679 the LWAs were beneficial to improving the micromechanical properties of the ITZ in the UHPC
 680 matrix, and the improvement derived from the SLWA was more apparent than the CLWA. The
 681 enhancement of micromechanical properties of the ITZ was also reported in conventional concrete
 682 prepared with LWAs [106, 107]. Zhang et al. [107] attributed the higher micromechanical
 683 properties of ITZ to the higher amount of CH at the surface of LWAs. However, in this study, due
 684 to the use of UHPC paste with a very low w/b ratio, dense ITZ was formed and no crystalline
 685 products were deposited in the vicinity of LWAs. Hence, the improved ITZ of both cases was
 686 probably resulted from the internal curing of the pre-soaked LWAs, especially for the UHPC
 687 system, the water desorption from the LWAs reacted with the unhydrated cement to strengthen the
 688 ITZ around the LWAs. This enhancement would, to some extent, counteract the negative influence
 689 of the porous LWAs. For the sample prepared with SLWA, the higher pozzolanic activity of the
 690 SLWA further densified the ITZ, resulting in a higher micromechanical performance of ITZ. This

691 explanation is supported by the formation of additional C-A-S-H in the ITZ. Although the ITZ was
692 strengthened by the internal curing and pozzolanic reaction of the LWAs, the compressive strength
693 of overall concrete was still controlled by the weak strength of the LWAs. However, the better
694 micromechanical performance of ITZ in the SLWA mixture as compared to that in the CLWA
695 mixture might partly contribute to the higher compressive strength of the former. Moreover, the
696 ref. [106] reported that the improved ITZ around SLWA could enhance the resistance of concrete
697 to the ingress of harmful ions penetration. In the LWA concrete, the porous aggregate would be the
698 weakest part when the concrete is subject to loading stress or harsh environment. Therefore, the
699 strengthened ITZ was expected to prevent the loading cracks or defense the intrusion of harmful
700 ions.

701
702 It is noted that the average modulus of C-S-H hydrate in this study was higher than 35 GPa, which
703 was much higher than that in other works [108, 109]. This is because of the presence of a new
704 hydration product (ultra high density C-S-H) in the matrix, formed with a low w/b ratio [110], in
705 addition to the low density and high density C-S-H phases normally found in hydrated cement
706 pastes. Furthermore, in the UHPC matrix, the high density and ultra high density C-S-H phases
707 dominated the microstructure [110, 111]. Hence, the presence of the ultra high density C-S-H is
708 believed to increase the modulus of the paste matrix. With the incorporation of the LWAs, the mean
709 modulus of the ITZ was further increased due to the increased formation of C-S-H gels in the
710 limited space, which would increase the packing and stiffness of hydrate phases. Similarly, in
711 conventional lightweight aggregate concrete, other studies [112, 113] also reported that the
712 microhardness of the ITZ around the LWAs was higher than the surrounding matrix as the newly
713 formed product densified the interface structure. The improved micromechanical properties of the
714 ITZ might contribute to the enhanced mechanical properties and durability of HPLACs.

715

716 3.3.4 Pore structure of paste

717 The pore size distributions of the paste matrices in the reference sample and HPLACs are plotted
718 in Fig. 17. It should be noted that the pores of the LWAs in the samples have been excluded via
719 testing the LWAs alone. Fig. 17a shows that most of the pores in the reference paste ranged from
720 5 to 100 nm, and by contrast, the pores in the 100C and 100S pastes were mainly concentrated in
721 a smaller-size range (<10 nm), which can be referred as gel pores [114]. As the gel pores are the

722 intrinsic porosity of the calcium silicate hydrate phase [66], the increased volume of gel pores is
 723 likely to indicate the increased hydration degree of the matrix. In this study, the internal curing of
 724 LWAs is considered to play a positive role in promoting the hydration of the UHPC matrix, thus
 725 increasing the amount of finer pores. These results are consistent with the enhanced hydration
 726 kinetic in the presence of the LWAs. A similar phenomenon was observed in the study of Liu and
 727 Wei [115], that the use of pre-wetted calcined bauxite LWAs resulted in increased amounts of gel
 728 pores and large capillary pores. Also, the amount of 10-50 nm capillary pores was found to be
 729 increased with the incorporation of the LWAs. This might be due to the water released from the
 730 LWAs at a very early age (or during the mixing) increased the water content in the paste. However,
 731 this range of mesopores has been regarded as less harmful pores, which means that they have less
 732 influence in affecting the strength and permeability of concrete [116]. The cumulative pore volume
 733 plotted in Fig. 17b shows the larger total pore volume in the pastes of the samples prepared with
 734 LWAs, which is primarily due to the larger amount of gel pores.

735

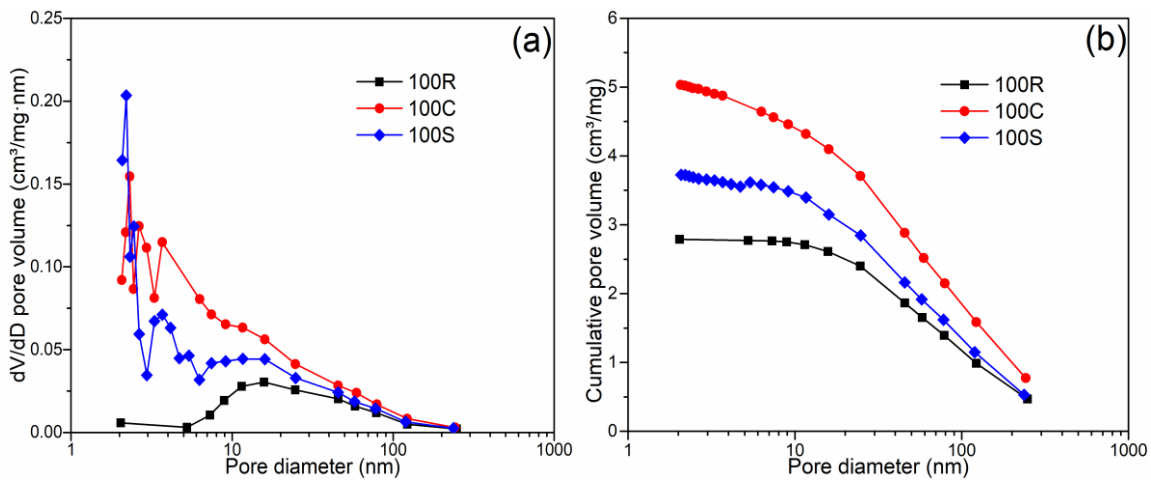


Fig. 17. Pore structure of paste in 100R, 100C and 100S determined by BET method.

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737

738 4 Discussion

739 4.1 Judicious selection of LWAs based on their physicochemical properties

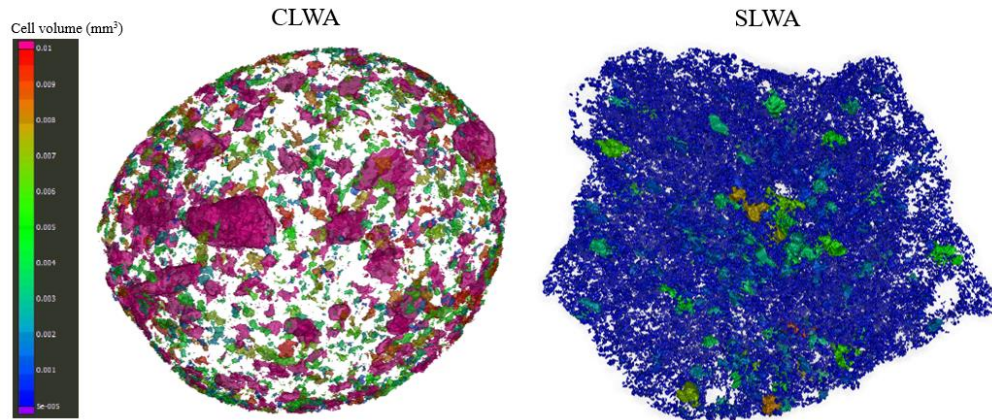
740 Besides the cement paste, the physicochemical properties of LWAs also played essential roles in
 741 affecting the performance of lightweight aggregate concrete [99]. Based on the characteristics of
 742 the LWAs and the corresponding performance of HPLACs in this study, several principles are
 743 proposed as selection criteria of LWAs for the preparation of HPLACs:

- 744 i) To make sure the HPLAC has high strength, a high-strength LWA is necessary to be adopted.
745 Although the two types of LWAs had similar density and porosity, the SLWA had much higher
746 strength than the CLWA (4.58 MPa vs. 0.76 MPa in Table 2), which partly contributed to the
747 higher strength and structural efficiency of HPLACs. The strength of LWAs was associated
748 with their pore structure [117]. As indicated by X-ray μ CT observation (Fig. 18), the presence
749 of relatively small, uniform, and isolated pores in the SLWA might account for its higher
750 strength in comparison with the CLWA with a heterogeneous pore structure. This result is
751 consistent with the finding of MIP (see Fig. 5), the different pore size distributions in two
752 types of LWAs were expected to affect the absorption and desorption of water and further
753 influence the internal curing effect of pre-wetted LWAs.
- 754 ii) The geometry of the LWAs affects the bonding of aggregates and paste (Fig. 2) and the open-
755 pore surface of the SLWA (Figs. 6 and 7) was beneficial to improving the mechanical bonding
756 between the aggregates and the paste [74]. Hence, the interlocking interface of the SLWA and
757 matrix was able to strengthen the ITZ since the porous surface increased the bonding area as
758 shown in SEM-BSE images (Section 3.3.2). Although the ITZ became better due to the
759 internal curing and interlocking bonding of the LWAs, the compressive strength of HPLAC
760 was still governed by the weak strength of the LWAs. Nonetheless, in comparison with the
761 CLWA, the improved ITZ in the SLWA mixture was expected to result in a higher compressive
762 strength (Section 3.3.3).
- 763 iii) An LWA which possesses a higher pozzolanic reactivity is preferred. As indicated by Q-XRD
764 results (see Table 4), the SLWA had a higher content of amorphous phases than the CLWA. In
765 combination with the composition of the LWAs, the amorphous silica and aluminate phases
766 may be the major components in the LWAs, thus the larger amorphous amount tended to
767 induce a higher pozzolanic reactivity. This speculation was confirmed by the Chapelle test in
768 Fig. 8, which showed that the SLWA had a higher pozzolanic reactivity than the CLWA).
769 Therefore, the chemical interaction between the SLWA and the paste was more pronounced.
770 The reinforced reaction around the surface of the SLWA enhanced the micromechanical
771 properties of ITZ, which would be able to counteract the weak strength of SLWA.
- 772 iv) An LWA with high open porosity is conducive to providing an internal curing effect by
773 desorbing sufficient water. Especially for the UHPC system, the lack of water for cement
774 hydration would make the cement superfluous. Thus, the incorporation of water-absorbing
775 LWAs could promote the hydration of cement and the pozzolanic reaction of SCMs in the

776 HPLACs (Section 3.3.1). As shown in Table 2, the LWAs used in this study had high open
777 porosity and large absorption capacity, hence the internal curing effect further benefited the
778 pozzolanic reactivity of LWAs and the physicochemical reaction in the ITZ. However, it
779 should be noted that LWAs with too large or too small pores are not endeared as the water
780 inside would migrate out too early to increase the w/b ratio of the UHPC matrix, or cannot
781 migrate out to play the internal curing role [118].

782 v) In order to produce high performance lightweight concrete, an LWA with low density or high
783 total porosity is favorable to be used. In this study, two types of LWAs with low bulk density
784 (about 500 kg/m^3 in Table 2) were selected to reduce the weight of concrete as much as
785 possible (Fig. 9). Correspondingly, the thermal insulation of the LWAs-containing HPLACs
786 was significantly reduced (see Fig. 11) due to the high porosity of the LWAs. In terms of
787 physical and functional properties, if the content of LWAs exceeded 100%, the reductions in
788 the density and the thermal conductivity tended to be less significant. Hence, to ensure high
789 structural efficiency of concrete, 100% replacement of RS by the LWAs in the UHPC system
790 seemed to be appropriate in the production of HPLACs.

791



792 Fig. 18. 3D pore structures of CLWA and SLWA obtained by X-ray μ CT.

793

794

795 Based on the above principles and performance of HPLACs, compared to CLWA, SLWA was
796 more favorable to be used in the production of HPLACs with achieving higher structural
797 efficiency. Thus, the interaction between the UHPC and the SLWA will be discussed in the
798 following part.

799

799 **4.2 Synergetic effects of UHPC and pre-soaked SLWA**

800 **4.2.1 Homogeneity of SLWA and fibers**

801 One major concern of using LWAs in concrete is that the lightweight aggregate concrete is
802 susceptible to segregation because of the lower density of LWAs [119]. However, the advantage
803 of using the UHPC system is to mitigate the potential segregation of LWAs and fibers in the
804 HPLAC. X-ray μ CT was employed to observe the distributions of aggregates in the 100R and
805 100S. As seen in Fig. 19, the distributions of RS and SLWA were homogeneous and the heavy steel
806 fibers did not sink to the bottom or agglomerate at one site in the HPLAC. The good homogeneity
807 is because the fresh UHPC paste had a high viscosity [120], which increased the adhesion between
808 LWAs/fibers and the matrix. Hence, the use of UHPC was beneficial to the homogeneity of LWAs
809 and fibers in the HPLAC.

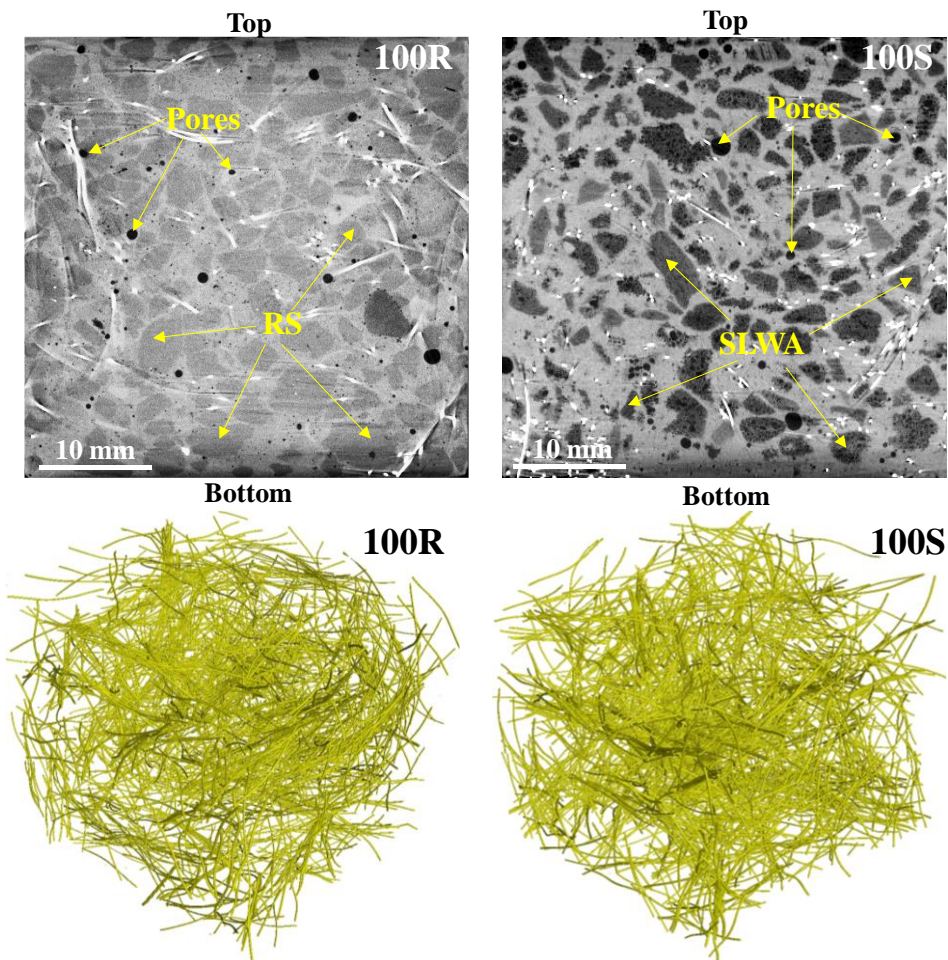


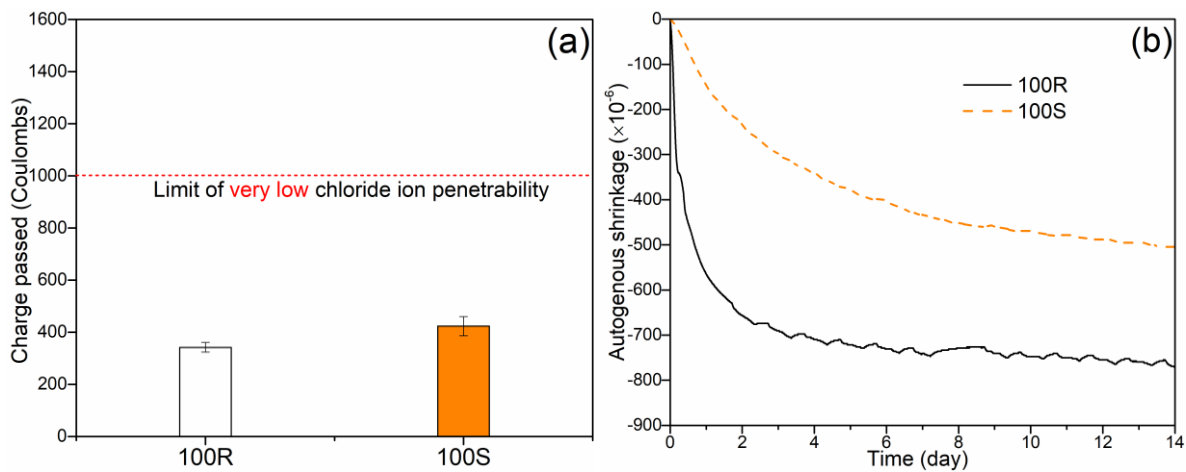
Fig. 19. 2D vertical-section images (right view) and 3D distributions of steel fibers (top-right-back view) in 100R and 100S obtained by X-ray μ CT.

810
811

812 4.2.2 Durability of HPLAC

813 To prepare a HPLAC, a good quality cement paste is needed. Hence, an UHPC was applied as the
814 binder in this study due to its excellent mechanical and durability properties. Zhang and Gjrrv [44]
815 pointed out that the permeability of lightweight aggregate concrete was mainly controlled by the
816 porosity of cement paste/mortar rather than the porous aggregates. The low porosity and fine pores
817 characteristics of the UHPC have been proven due to its extremely low w/b ratio and the presence
818 of a large amount of highly reactive powder [121, 122]. Therefore, the utilization of UHPC to bind
819 the LWAs can provide a strong skeleton for the concrete and further protect the LWAs to resist the
820 infiltration of aggressive substances. Fig. 20a shows the chloride ion penetration values for the
821 reference UHPC and HPLAC prepared with SLWA. The results indicate that the 100S had a
822 slightly higher charge passing value than the 100R, but both of them were much lower than the
823 stipulated limit of “very low ion penetrability” [63]. Also, the ability of 100S to resist chloride
824 ingress was better than conventional high strength HPLAC [123, 124] and comparable or superior
825 to high performance concrete [125, 126]. The very low water absorption of the SLWA samples
826 (Section 3.2.1) further confirmed the high resistance to external substance. Therefore, it can be
827 concluded that the combined use of the UHPC and the SLWA can produce an HPLAC with low
828 permeability. Besides the contribution from the UHPC paste, the released water from the SLWA
829 enhanced the hydration degree of the UHPC, which contributed to the improvement of ITZ
830 surrounding the SLWA by mechanical interlocking and chemical interaction (pozzolanic reaction).
831 This might be another reason for the high resistance to chloride penetration and low water
832 absorption.

833



834 Fig. 20. Permeability (a) and autogenous shrinkage (b) of 100R and 100S.

834

835 It is known that one major concern of UHPC-based matrix is its high shrinkage (mainly autogenous
836 shrinkage) due to the use of low w/b ratios, a large amount of SCMs, and the absence of coarse
837 aggregates [127, 128]. The internal curing provided by the pre-wetted LWAs had been proven to
838 successfully mitigate the autogenous shrinkage of conventional cement concrete [47, 129]. In
839 recent years, some investigations [130-132] also found that the use of pre-wetted LWAs could
840 reduce the autogenous shrinkage of UHPC. Therefore, a similar effect resulting from pre-soaked
841 SLWA is believed to reduce the autogenous shrinkage of the HPLACs since more LWAs were
842 incorporated. As shown in Fig. 20b, the HPLAC prepared with 100% SLWA had a much lower
843 autogenous shrinkage value than the reference UHPC within 14 days. It can be noted that the most
844 of autogenous shrinkage occurred at the very early age (within 24 h) for the reference UHPC. Shen
845 et al. [130] found that the internal humidity of UHPC was nearly saturated within 48 h by
846 incorporating a certain amount of LWAs. Therefore, the water desorbed from the SLWA mitigated
847 the self-desiccation of the UHPC matrix (withdrawal of water from capillary pores) by increasing
848 the internal moisture [87], and thus reducing the autogenous shrinkage of the HPLACs.

849

850 As the LWAs had a higher reactivity than the quartz sand (Section 3.1.3), alkali-silica-reaction
851 (ASR) may be one concern in the conventional LWA concrete. However, in this study, the ASR
852 would not be triggered after taking several factors into account: 1) Due to the large amounts of
853 pozzolanic materials in the mixture (i.e. SF, MK and GGBS), the ASR would be suppressed by
854 these SCMs [133]. 2) As the LWAs were porous, the internal pores were expected to accommodate
855 the expansion stress if ASR would occur. 3) Lack of water for the ASR reaction in the system
856 because of the low w/b ratio. A recent study [98] also indicated that the expanded clay and shale
857 showed superiority in limiting the ASR expansion by the reduction of alkalinity and release of Al
858 in pore solution.

859

860 4.2.3 Interaction of UHPC and SLWA

861 The physical and chemical characteristics of LWAs influenced the overall performance of
862 lightweight aggregate concrete by processes that took place in the ITZ [99]. A schematic
863 illustration of the mechanisms of interactions between the UHPC and the pre-soaked SLWA is
864 illustrated in Fig. 21. In the fresh stage, the cement paste penetrated into the cavities/pores on the
865 SLWA surface, which created a better interface bonding. This interlocking bonding of the SLWA
866 is similar to other LWAs, such as porous calcined bauxite [134] and porous pumice [131]. During

867 the hardening stage, the internal curing of the SLWA provided additional water to further hydrate
 868 the remaining cement clinkers and the pozzolanic reaction of SCMs. Moreover, a high
 869 concentration of Al was leached out from the SLWA and involved in the reaction to form C-A-S-
 870 H hydrates. The formation of C-A-S-H in the ITZ was also an evidence of the occurrence of
 871 pozzolanic activity of the SLWA. As a result, with the progress of hydration, the interaction of the
 872 SLWA and the alkaline products formed a pozzolanic product layer in the vicinity of the SLWA
 873 [71], which further strengthened the ITZ. The synergetic effects of the above reactions led to the
 874 formation of a dense rim with higher micromechanical properties in the ITZ, which protected the
 875 porous aggregates against the penetration of the detrimental ions [44]. Due to the presence of the
 876 UHPC matrix and coupled effects of internal curing and the pozzolanic reaction of the SLWA, an
 877 HPLAC with high structural efficiency and good durability was produced with the use of UHPC
 878 and pre-wetted SLWA together.

879

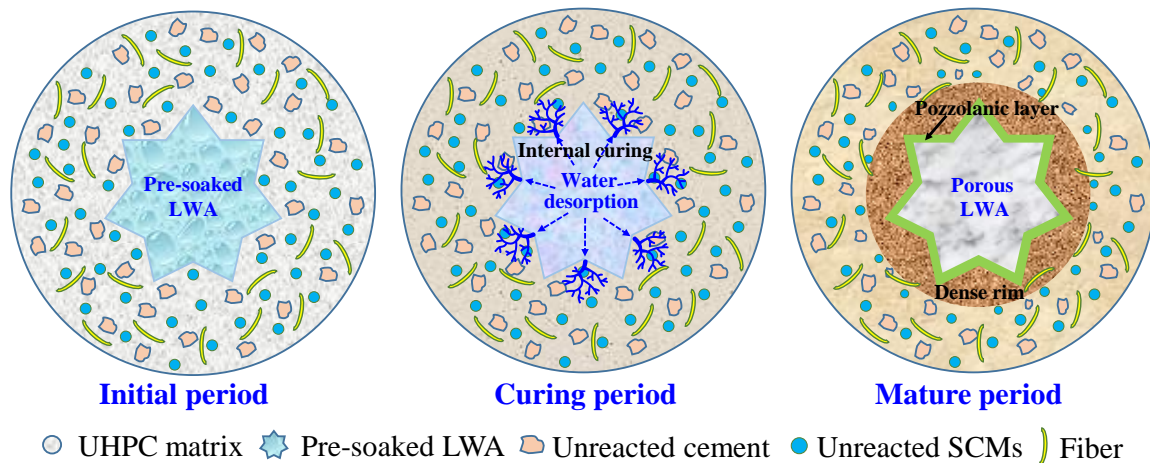


Fig. 21. Schematic illustration of interaction of SLWA and UHPC in the HPLACs.

880

881 **5 Conclusion**

882 In this study, a high performance lightweight aggregate concrete (HPLAC) was developed by
 883 combining the use of UHPC and LWAs. The role of physicochemical properties of two types of
 884 LWAs in the HPLACs was investigated with a view to selecting the optimal LWAs. The
 885 effectiveness of the combined use of the UHPC and the LWAs in affecting the physical, functional,
 886 and permeability properties was evaluated and analyzed. The physical and chemical interfacial
 887 processes between the UHPC and the LWAs were discussed in light of hydration kinetic,
 888 nanoindentation, elemental mapping, and X-ray μ CT measurements. The following conclusions

889 can be derived from this study:

890

891 (I) The shale based LWA (SLWA) exhibited a higher pozzolanic reactivity than the clay based
892 LWA (CLWA) attributed to the higher content of amorphous phase and lower quartz content
893 in the SLWA. The SLWA had an open pores surface and a more individual and fine pore
894 structure, while the CLWA had a relatively denser outer shell and well-interconnected coarser
895 pores.

896 (II) The incorporation of the two types of LWAs reduced the density of HPLACs significantly.
897 However, the reductions in the density using the CLWA and the SLWA were similar due to
898 their similar wet density values. The water absorption values of HPLACs were much lower
899 than those of conventional LWA concrete and normal weight concrete. The SLWA-
900 incorporated HPLACs had lower water absorption than the CLWA-containing samples.

901 (III) The HPLACs prepared with SLWA exhibited a higher structural efficiency than the CLWA-
902 incorporated HPLACs due to the higher strength of the SLWA, improved interlocking
903 bonding of aggregates and paste, higher pozzolanic reactivity of the SLWA. The replacement
904 of RS by the two types of LWAs in the HPLACs effectively reduced the thermal conductivity.

905 (IV) The internal curing of the pre-wetted LWAs promoted the hydration of the cementitious
906 materials in the UHPC system due to the water desorption characteristics of the LWAs. The
907 internal curing effect of the pre-wetted SLWA was more pronounced than that of the CLWA.
908 Higher intensity of Al and lower intensities of Si and Ca in the ITZ were found in the SLWA-
909 incorporated sample. Due to the pozzolanic reaction and internal curing of the SLWA, C-A-
910 S-H hydrates with a lower Ca/(Si+Al) ratio were formed in the vicinity of the SLWA. The
911 use of the LWAs in the UHPC matrix improved the micromechanical properties of the ITZ,
912 and the improvement resulted from the SLWA was more significant than the CLWA because
913 of the coupled effects of the higher pozzolanic reaction and internal curing of the SLWA.

914 (V) The preferred characteristics of the LWAs to be used for producing HPLAC are: high
915 strength with a low density, high porosity with a fine pore structure, a porous surface, and
916 high pozzolanic reactivity. The use of UHPC was beneficial to the homogeneity of the LWAs
917 and fibers in the HPLACs. The conjunct use of the UHPC and the SLWA was able to produce

918 an HPLAC with high structural efficiency, low autogenous shrinkage, good thermal
919 insulation, and low permeability.

920

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927

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