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MoSe₂ quantum dots modified hole extraction layer enables binary organic solar cells with improved efficiency and stability

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In this paper, we demonstrate a solution-processed MoSe₂ Quantum Dots/PEDOT:PSS bilayer hole extraction layer (HEL) for non-fullerene organic solar cells (OSCs). It is found that introduction of MoSe₂ QDs can alter the work function and phase separation of PEDOT:PSS, thus affecting the morphology of active layer and improving the performance of OSCs. The MoSe₂ QDs/PEDOT:PSS bilayer HEL can improve the fill factor (FF), short-circuit current density (*I_{sc}*) and power conversion efficiency (PCE) of OSCs based on different active layers. The best PCE of up to 17.08 % was achieved based on a recently reported active layer binary system named SZ2:N3, which is among the highest reported value to date for OSCs using 2D materials as the interface modifier. Our study indicates that this simple and solution-processed MoSe₂ QDs/PEDOT:PSS bilayer thin film could be a potential alternative HEL to the commonly used PEDOT:PSS conducting polymer.

Introduction

Bulk heterojunction (BHJ) organic solar cells (OSCs) have received notable scientific and industrial interest because of their merits of being light weight, low-cost, easy-flexibility, and the promise for mass production by roll-to-roll print process.^[1-6] To date, due to the continued progresses in the exploration of low band-gap polymer donors and non-fullerene acceptor, the maximum power conversion efficiencies (PCE) of the single-junction and tandem non-fullerene OSCs have reached over 17%, respectively, allowing them one step closer to commercial application.^[7,8] In addition to the advances in synthesizing novel materials, interface engineering is currently recognized as one of the most efficient strategies to boost device efficiency and stability. In particular, the hole extraction layer (HEL) between the anode and active layer plays a key role in anode modification, energy level alignment, charge collection and morphology controlling, *etc.*^[9-11]

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Therefore, various kinds of HEL are being used to further improve the performance of OSCs. The presently reported HEL were mainly transition-metal oxides, metal sulfides, conducting polymers, conjugated polyelectrolytes (PCPDTK_{0.50}H_{0.50}-TT, PCPDTffPhSO₃K), graphene oxides, etc.^[12,14] Among, the conducting conjugated polymer poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS) has been widely used as a HEL in organic optoelectronic devices, because of its appropriate work function, excellent conductivity, high optical transparency and compatibility with the solution processes.^[15-18] However, as is well known, the high acidity and hygroscopicity of PEDOT:PSS can corrode the indium tin oxide (ITO) electrode, which will eventually degrade the performance and long-term stability of OSCs. Furthermore, the poor electron blocking ability and anisotropy in the conductivity of PEDOT:PSS films usually lead to serious leakage current and inhomogeneous charge extraction in some locations. In terms of the above issues, the main solutions focus on substituting PEDOT:PSS with transition-metal oxides, such as molybdenum oxide (MoO₃), vanadium oxide (Va₂O₅), nickel oxide (NiO), and tungsten oxide (WO₃), etc.^[19-24] Unfortunately, these materials are usually fabricated by high vacuum techniques, which are not only high-cost and have complicated processing steps but also not compatible with the roll-to-roll manufacturing technology. Another effective approach is to insert an interfacial modifier between the HEL/ITO interface for improving the interfacial contact quality and reducing electrical losses.^[25]

Many efforts have been devoted to the HEL/ITO interface engineering of single-junction OSCs. Solution-processed transition metal dichalcogenides (TMDs) (MoS₂, WS₂, TaS₂ and NbSe₂, *etc.*) have recently attracted considerable attention due to their unique optical and electronic properties. In particular, single- and fewlayered TMDs or TMDs quantum dots (QDs) have been used to

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modify HEL, or within the active layer of OSCs to enhance device performance and stability.^[26-32] In 2013, Gu et al. proposed the integration of chemically exfoliated solution-processed MoS₂ nanosheets and NbSe₂ nanoplates as efficient anode buffer layer for performance optimization of OSCs.^[33,34] The as-reported PCE with the PTB7:PC71BM system as the active layer is superior to that obtained by using traditionally vacuum deposited e-MoO₃. In 2014, Le et al. reported the use of spin-coated ultraviolet ozone (UVO)treated, sonication exfoliated MoS₂ nanosheets as an interlayer between the anode and the PEDOT:PSS in a P3HT:PC₆₁BM-based OSC.^[35] The device exhibited enhanced stability without degradation of its photovoltaic performance compared to the device employing only PEDOT:PSS. It is noteworthy that the mechanism of performance improvement based on MoS₂ nanosheets as interfacial modifiers for HEL has never been systematically explored yet.

In this work, we firstly reported a solution-processed composite HEL within MoSe₂ QDs and PEDOT:PSS to improve the device performance and stability of OSCs. Meanwhile, the working principle of two-dimensional (2D) MoSe₂ QDs as HEL interface modifier in OSCs was studied thoroughly for the first time. Compared to the device with only PEDOT:PSS, the MoSe₂ QDs/PEDOT:PSS bilayer HEL efficiently reduced the charge recombination and enhanced the charge extraction and transport due to its well-matched work function (WF) and superior conductivity. We also investigated the stability of OSCs without and with MoSe₂ QDs, and found that the stability of OSC with MoSe₂ QDs is dramatically improved due to the separation of ITO and PEDOT: PSS caused by insertion of MoSe₂ QDs. The champion device based on MoSe₂ QDs/PEDOT:PSS showed a PCE of 17.08 %, which is the highest value among OSCs with 2D materials as the interface modifier so far. The excellent PCE of MoSe₂ QDs/PEDOT:PSS based device is attributed to its higher short-circuit current density (Jsc), which is mainly derived from the superior conductivity and suitable work function of MoSe₂ QDs/PEDOT:PSS.

Results and discussion



Fig. 1. (a) Schematic diagram of preparation process of monolayer MoSe₂ QDs using a lithium intercalation and ultrasonication method. The magnified area shows the layered structure of MoSe₂ QDs, with the chalcogen atoms (Se) in purple and the metal atoms (Mo) in green. (b) TEM and (c) HR-TEM images measured for the MoSe₂ QDs.

The MoSe₂ QDs were synthesized through lithium intercalation and ultrasonication method as depicted in Fig. 1a. It is worth noting that MoSe₂ QDs prepared by most methods suffer from intrinsic structural defects, which are detrimental to device performance. The crystal structure of the layered MoSe₂ QDs is presented in Fig. 1a. It is a typical sandwiched structure and the interlayer effect is relatively weak and easy to peel off, leading to specific anisotropic mechanical and electrical properties. In addition, some in situ derived structural defects in MoSe₂ QDs are usually difficult to avoid. Transmission electron microscopy (TEM) and high-resolution TEM (HR-TEM) images were used to measure the crystal structures of MoSe₂ QDs. Fig. 1b is a conventional bright field TEM image showing many crystalline QDs. The dark contrasts correspond to Bragg contrasts and the inserted selected area electron diffraction (SAED) pattern reveals the presence of MoSe₂ (hexagonal, a=b=0.329 nm and c=1.293 nm). To further analyze the atomic structure of individual MoSe₂ QDs, we measured the HR-TEM image of a single MoSe₂ QD (as presented in Fig. 1c), revealing that these as-prepared MoSe₂ QDs possess high crystallinity.^[36] The fast Fourier transform (FFT) of the QD (inset of Fig. 1c) is in perfect agreement with a [-2110] projection of the MoSe₂ structure. To investigate the composition of the resultant MoSe₂ QDs, the corresponding energy-dispersive X-ray spectroscopy (EDS) elemental mapping images were measured, as shown in Fig. S1. The results indicate that Mo (green) and Se (yellow) elements were uniformly distributed over the entire QDs.

It was pointed out in previous literatures that hightemperature annealing could tune the WF of HEL, improving the performance of OSCs.^[37,39] However, the underlying mechanisms have not yet been studied. Herein, the MoSe₂ QDs film was annealed at high-temperature and its structural composition was confirmed by the X-ray photoelectron spectra (XPS), as depicted in Fig. S2a, S2b and S2c. Notably, the characteristic peaks around 235.7 and 232.2 eV, as well as the peaks at 59.5 and 58.7 eV, corresponding to the Mo⁶⁺ 3d_{5/2} and Mo⁶⁺ 3d_{3/2}, and Se⁴⁺ 3d_{5/2} and Se⁴⁺ 3d_{3/2}, respectively, were also observed. This indicates that Mo and Se elements in MoSe₂ QDs are partially oxidized to MoO₃ and SeO₂ species. Generally, SeO₂ seldom exists in HEL, given that it has a low sublimation temperature.^[36] The Mo⁶⁺ 3d spectrum exhibits another two contributes, located at respectively 235.2 and 231.9 eV, which can be assigned to O-Mo-Se band due to the participated lattice oxygen atoms.^[40] It is deduced that the O incorporation not only could partially fill the Se vacancies and passivate the structural defects, but lead to a change of interfacial energy level alignment (change n-type MoSe₂ to p-type), which is beneficial to device efficiency and stability (see detail in Supplementary Information and Fig S2d). Then, we studied the morphology and optical properties of the thin films. The as-prepared MoSe₂ QDs solution and PEDOT:PSS were spin-coated onto ITO substrates, respectively. Fig. S3(a-c) shows the atomic force microscopic (AFM) images of ITO substrate, PEDOT:PSS and MoSe₂ QDs thin films as coated on ITO, respectively. It can be observed distinctly that the root-meansquare (R_q) surface roughness of ITO, ITO/PEDOT:PSS and ITO/MoSe₂ QDs are 3.70, 2.62 and 1.27 nm, respectively, indicating that MoSe₂ QDs can decrease the RMS of ITO and then result in a smaller leakage current. The normalized UV-vis absorption spectrum of post heat-treated MoSe₂ QDs film is shown in Fig. S3d.

It can be seen clearly that MoSe₂ QDs film may absorb some amount of UV light, whereas not impacting the absorption of active layer in solar cells. This is highly desirable for the device performance and stability optimization of OSCs.^[41,42] The overall transmittance spectra of ITO, ITO/PEDOT:PSS and ITO/MoSe₂ QDs/PEDOT:PSS are presented in Fig. S3e. It can be found that the transmittance of ITO/PEDOT:PSS thin film is higher than pure ITO in the range of 300-800 nm. With the introduction of MoSe₂ QDs into PEDOT:PSS, the transmittance is nearly unchaged. This may be attributed that the energy distributions of both PEDOT:PSS and MoSe₂ QDs/PEDOT:PSS have been changed due to the different interfaces between ITO and HEL, which facilitates the enhancement of light interference and thus the increase of transmittance.^[43] This is in agreement with the analyses of Fig. S3d, and demonstrates that the MoSe₂ QDs layer can server as an effective interlayer in traditional ITO/PEDOT-based OSCs. In addition, the effect of MoSe₂ QDs on the morphology of the active layer was also measured by the AFM, as shown in Fig. S3f and S3g. Compared to R_q of PEDOT:PSS/BHJ (6.65 nm), the surface roughness of MoSe₂ QDs/PEDOT:PSS/BHJ increases to 6.94 nm after the insertion of MoSe₂ QDs. Such a change is likely caused by a positive interaction between MoSe₂ QDs and PEDOT:PSS, which is instrumental in forming solid contact with BHJ, thus preventing BHJ from spreading to HEL. The slightly increased R_q is beneficial for modulating phase separation of BHJ, increasing device efficiency and stability.^[44] The discrepancy of R_q is ascribed to the different surface energies of HEL or ITO, which can be quantitatively calculated by the water contact angle. The contact angle of different materials measured with water and ethylene glycol (EG) as testing liquids (equal volume of liquids for each test) are displayed in Fig. 2. The calculation details are described in the Supplementary Information and Table S1-S3. The MoSe₂ and MoSe₂/PEDOT:PSS films exhibit water and EG contact angles lower than the ITO and pristine PEDOT:PSS. It was pointed out in literatures that the surface energy of HEL has a marked impact on the orientation of BHJ, that is, the face-on orientation dominates at high surface energy, while the edge-on orientation is easier to form at low surface energy.^[45] The surface energy of ITO, MoSe₂, PEDOT and MoSe₂/PEDOT:PSS films are 68.8, 70.0, 85.5 and 89.0 mN m⁻¹, respectively. The corresponding interfacial energies of MoSe₂ and PEDOT:PSS, MoSe₂/PEDOT:PSS and PM6:Y6 are calculated to be 7.5 and 128.4 mN m⁻¹, respectively, which are much higher than ITO and PEDOT (4.4 mN m⁻¹), and PEDOT:PSS and PM6:Y6 (121.0 mN m⁻¹), indicating that the insertion of MoSe₂ QDs can not only alter the morphology of PEDOT:PSS and BHJ but also facilitate the phase separation in PEDOT:PSS and BHJ, which benefits the face-on orientation of BHJ and enhances the device performance.



Fig. 2. Water and EG contact angles of various materials: (a) ITO, (b) $MoSe_2 QDs$, (c) PEDOT:PSS, (d) $MoSe_2 QDs/PEDOT:PSS$, (e) PM6:Y6.

The OSCs with standard structure were fabricated based on a typical binary blend of PM6:Y6 and state-of-the-art BHJ systems of SZ2:N3 to investigate the influence of MoSe₂ QDs/PEDOT:PSS as a composite HEL on device performance. Schematic device structure and the chemical structures of the components in the active layer are presented in Fig. 3a and 3b. Since the WF of HEL plays an important role in controlling the carrier injection properties, the WFs of PEDOT:PSS and MoSe₂ QDs/PEDOT:PSS thin films coated on the ITO substrates were determined by ultraviolet photoelectron spectroscopy (UPS). The corresponding UPS spectra are shown in Fig. 3c. The secondary electron cut-off ($E_{cut-off}$) values of the PEDOT:PSS and MoSe₂ QDs/PEDOT:PSS films are measured to be 16.2 and 16.0 eV, respectively. Therefore, the WF of MoSe₂ QDs/PEDOT:PSS film was determined to be 5.2 eV (WF = 21.22 eV- $E_{cut-off}-E_{onset}$), which lies slightly higher than that of PEDOT:PSS (5.0 eV).^[46] The energy diagram of the OSCs with different HELs is illustrated in Fig. 3d. All of the energy levels of ITO, PM6, Y6, SZ2, N3, PFN-Br and Al refer to the values as reported in the literature.^[47,48] It can be observed clearly that, the insertion of MoSe₂ QDs can pull up the HOMO energy level of HEL, which is favorable for efficient hole extraction and reducing charge recombination near the anode. Moreover, the MoSe₂ QDs/PEDOT:PSS HEL can also better block the electron transport compared to the one with only PEDOT:PSS, resulting in a smaller leakage current. The increase of WF is ascribed to the formation of Mo⁶⁺ species due to the oxidation of selenium vacancies and edge termination, indicating that MoSe₂ QDs/PEDOT:PSS HEL has a ptype conductive mechanism.^[49]

The thickness of MoSe₂ QDs film is optimized by changing the spin-coating speed to obtain an optimal value (see Fig. S4a and Table S4). Fig. 4 displays the current density–voltage (*J–V*) characteristics of the fabricated devices and their detailed photovoltaic performances including J_{sc} , open-circuit voltage (V_{oc}), fill factor (FF) and PCE are summarized in Table 1. Clearly, after inserting MoSe₂ QDs as the surface modifier, the J_{sc} and FF of OSCs with MoSe₂ QDs/PEDOT:PSS composite HEL realize an obvious increase compared to the control cell. In PM6:Y6 system, the champion device based on composite HEL exhibited a maximum PCE value of 16.08 % with J_{sc} of 25.84 mA cm⁻², V_{oc} of 0.83 V, FF of 74.69% and series resistance (R_s) of 3.63 Ω cm² while the control cell with PEDOT:PSS HEL exhibited a maximum PCE value of 15.48 % with J_{sc} of 24.68 mA cm⁻², V_{oc} of 0.84 V, FF of 74.29% and R_s of 7.94

 Ω cm². It indicates that this simple and solution-processed MoSe₂ QDs/PEDOT:PSS bilayer thin film could contribute to boosting device efficiency. The impressive device performance makes MoSe₂ QDs/PEDOT:PSS bilayer thin film a promising HEL candidate for near future high-efficiency optoelectronic devices. The maximum EQE value of OSC with MoSe₂ QDs/PEDOT:PSS as HEL is approximately

80 % and the calculated J_{sc} is 25.32 mA/cm², which is consistent with the value extracted from the *J*–*V* measurements with deviation of only 5%. In SZ2:N3 system, these results are in agreement with those of PM6:Y6. Fig. 4c and 4d shows the *J*–*V* curves and EQE spectra of control cell and MoSe₂ QDs/PEDOT:PSS-based cell. Evidently, compared with the control cell, the J_{sc} and FF values of



Fig. 3. (a) Schematic device structure inserted with MoSe₂ QDs. (b) The chemical structures of BHJ. (c) UPS spectra of the cut-off region of the PEDOT:PSS and MoSe₂/PEDOT:PSS. (d) Energy level diagram of the OSCs.

the MoSe₂ QDs/PEDOT:PSS-based cell increase from 25.22 to 25.63 mA/cm² and 77.11 to 77.52%, respectively. According to the EQE spectra, the corresponding calculated J_{sc} value of the EQE spectra increases from 25.02 to 25.21 mA/cm², which agree well with the J_{sc} in J-V characteristics, showing a good reliability of our results. Especially, the OSC based on SZ2:N3 with MoSe₂ QDs/PEDOT:PSS as HEL yielded a maximum PCE of 17.08%, while the control cell only showed a PCE of 16.53%. In order to ensure the reproducibility and reliability of the PCE, the histogram of PCE based on SZ2:N3 using different HEL are plotted in Fig. S5. It is obvious that PCE values of SZ2:N3 based solar cells is in normal distribution. The average PCE of PEDOT:PSS and MoSe₂/PEDOT:PSS based solar cells are 16.28 ± 0.14% and 16.89 ± 0.12%, respectively, with a reasonably narrow standard deviation, which further indicates the good reproducibility and reliability of the PCE of corresponding OSCs. Recent reports of 2D materials as the interface modifier in OSCs are classified in Table S5. The PCE of our OSCs and those of the previously reported OSCs with 2D materials have been compared in Fig. 4e. [33-35,50-55] The PCE value in our work is the highest value among OSCs with 2D materials as the interface modifier so far.

In order to investigate the effect of MoSe₂ QDs on charge generation and dissociation processes, the photocurrent density (J_{ph}) and exciton dissociation probabilities P(E,T) versus effective voltage (V_{eff}) are measured and plotted, as shown in Fig. 4c and S6a.The charge extraction/collection values of these devices are extracted from the Jph-Veff curves, suggesting that MoSe2 QDs/PEDOT:PSS-based cell shows more efficient charge extraction/collection capability compared with control device. Apart from the factors of charge generation and dissociation, the charge recombination loss of OSCs is also a key parameter for analysis of the enhancement mechanism of PCE. Fig. S6b and S6c depict the variation of J_{sc} and V_{oc} versus the light intensity (1) for all cells with / ranging from 5 to 100 mW cm⁻². Herein, the α values based on MoSe₂ QDs/PEDOT:PSS HEL are significantly higher than that of control cell (0.967), demonstrating that the insertion of MoSe₂ QDs reduce the bimolecular recombination of cell (see detail in ESI). Furthermore, the transport characteristics of cells were measured to gain an insight of the mechanism responsible for the PCE enhancement. In fact, it is worth mentioning that MoSe₂ QDs/PEDOT:PSS-based cell exhibits the higher α value (0.989), which was very close to 1, indicating a negligible bimolecular recombination process. In case of V_{oc} versus *I*, V_{oc} is proportional to In *I*. From Figure S6c, it is observed that the minimum slope of MoSe₂ QDs/PEDOT:PSS-based cell is 1.04 kT/q, meanwhile, the

slope of control cell is 1.14 kT/q. The results imply that insertion of MoSe₂ QDs can effectively suppress monomolecular recombination originated from traps and defects, which devotes to improving the performance of OSCs.



Fig. 4. (a) J–V and (b) EQE curves of PM6:Y6 solar cells. (c) J–V and (d) EQE curves of SZ2:N3 solar cells. (e) A comparison of the performance of previously reported OSCs with 2D materials. (f) J_{ph}-V_{eff} curves of PM6:Y6 solar cells.

Table 1. Summary of device performances of OSCs with or without MoSe₂ QDs.

BHJ	Description	J _{sc} [mA cm ⁻²]	V _{oc} [V]	FF [%]	PCE [%]	R₅ [Ω·cm²]	J _{cal} [mA cm ⁻²]
PM6:Y6	PEDOT:PSS	24.68 (24.56±0.2)	0.84(0.84±0.002)	74.29(74.26±1.08)	15.48 (15.18±0.23) ^a	7.94	24.05
	MoSe ₂ /PEDOT:PSS	25.84 (25.68±0.3)	0.83(0.84±0.003)	74.69(74.31±1.30)	16.08 (15.75±0.24) ^a	3.63	25.32
SZ2:N3	PEDOT:PSS	25.22(25.21±0.3)	0.85(0.84±0.003)	77.11(77.09±0.09)	16.53 (16.32±0.12)ª	8.36	25.00
	MoSe ₂ /PEDOT:PSS	25.63(25.72±0.3)	0.86(0.85±0.003)	77.52(76.34±1.42)	17.08 (16.86±0.20) ^a	4.23	25.23

^aAverage PCE with standard deviations were obtained from 10 devices

The dark *J–V* characteristics of the cells are plotted in Fig. S7a, in which the device with MoSe₂ QDs/PEDOT:PSS HEL enables a lower leakage current and a higher rectification ratio, and thereby a reduced R_s in the cells, as shown in Table 1. Additionally, we fabricated two hole only devices with the device structures of ITO/MoSe₂ QDs/PEDOT:PSS/PM6:Y6/MoO₃/Al and ITO/PEDOT:PSS/PM6:Y6/MoO₃/Al, respectively, and then employed space charge limited current (SCLC) method for the evaluation of hole only transport properties. The hole mobility is evaluated by Mott–Gurney equation:^[56]

$$J_{sc} = \frac{g}{8} \varepsilon_r \varepsilon_0 \mu_h \frac{V_{eff}^3}{d^2}$$
(1)

where ε_r is the relative dielectric constant (assuming to be 3), ε_0 is the permittivity of free space (8.85× 10⁻¹² F m⁻¹), μ_h is the hole

mobility, V_{eff} is the effective voltage and d is the film thickness of active layer (150 nm). As shown in Fig. S7b, the hole mobility (μ_h) of MoSe₂ QD/PEDOT:PSS-based cell and the control cell is calculated to be 2.26×10⁻⁴ and 6.30×10⁻⁵ cm² V⁻¹s⁻¹, respectively. Evidently, OSCs with MoSe₂ QDs/PEDOT:PSS as HEL exhibit higher conductivity compared with control cell owing to its unique chemical structure and physical properties, which agrees well with the results of conductivity technique (Table S6). Since the lower frequency arc is associated with the charge recombination behavior, electric impedance spectroscopy (EIS) measurement is often used to analyze interface properties in OSCs.^[57,58] Nyquist plots of the impedance spectroscopy (IS) of the control cell and ITO/MoSe₂ QDs/PEDOT:PSS-based cells at zero bias under dark condition are

shown in Fig. S8a. It can be seen obviously that, compared to control cell, the MoSe₂ QDs/PEDOT:PSS-based cells show smaller semicircles, indicating the lower transport resistance and interfacial barrier between the ITO and MoSe₂ QDs/PEDOT:PSS. The reduced transport resistance is favorable for charge transporting and extraction, which benefits the increase of J_{sc} and FF values. Fig. S8b shows the PL emission spectra of ITO/PM6, ITO/PEDOT:PSS/PM6 and ITO/MoSe₂ QDs/PEDOT:PSS/PM6 thin films, respectively. The MoSe₂ QDs/PEDOT:PSS/PM6 film exhibits a much lower PL quenching efficiency (9 %) than the PEDOT:PSS/PM6 film (20%), exciton indicating that the dissociation in MoSe₂ QDs/PEDOT:PSS/PM6 film are more efficient than PEDOT:PSS/PM6, which results in higher EQE and J_{sc} for the corresponding OSCs. The corresponding PL intensity of active film with MoSe₂ QDs was significantly enhanced as compared to that of without MoSe₂ QDs, primarily due to diminished exciton quenching effect after insertion of MoSe₂ QDs.^[59] The PL measurement again prove that the embedded MoSe₂ QDs can facilitate the charge transport and thus contribute to boosting the efficiency of OSCs. Fig. S8c displays the PL intensity decay profiles obtained for PEDOT:PSS/PM6:Y6 and MoSe₂ QDs/PEDOT:PSS/PM6:Y6 thin films. It is obvious that the lifetime of excitons in MoSe₂ QDs/PEDOT:PSS device (0.840 ns) is shorter than that in PEDOT:PSS device (0.962 ns), which is mainly attributed to the efficient exciton dissociation and charge collection of MoSe₂ QDs/PEDOT:PSS. So, we can infer that the introduction of MoSe₂ QDs can increase the charge extraction rate and thus reduce charge recombination, which is beneficial to the enhancement in J_{sc} of the devices.



Fig. 5. S 2p XPS spectra measured for (a) the pristine PEDOT:PSS and (b) MoSe₂ QDs/PEDOT:PSS thin films. (c) Zeta potential of MoSe₂ QDs dispersions. (d) Raman spectra of PEDOT:PSS and MoSe₂ QDs/PEDOT:PSS films.

In order to further deeply explore the intrinsic working mechanism of MoSe₂ QDs in OSCs, we measured the XPS spectra for the pristine PEDOT:PSS and MoSe₂ QDs/PEDOT:PSS thin films to examine the change of the chemical and electronic properties of PEDOT:PSS before and after the insertion of MoSe₂ QDs. Fig. 5 shows that there are two characteristic S 2p peaks with the binding energy at 167.5 eV and 163.5 eV, respectively, in the S 2p XPS spectrum of the pristine PEDOT:PSS film, which are assigned to the S species of PSS and PEDOT, respectively.^[60,61] In pristine PEDOT:PSS

films, PEDOT was attached by insulating PSS via columbic interaction which promotes the dissolution of PEDOT in solution, whereas hinders the charge transport in solid films of device. However, when MoSe₂ QDs and PEDOT:PSS form bilayer, the ratio of PSS to PEDOT gradually decreased from 6.98 of the pristine PEDOT: PSS to the 6.88 of MoSe₂ QDs/PEDOT: PSS. This reveals the interfacial interaction between MoSe₂ QDs and PEDOT:PSS, which results in a phase separation between PEDOT and PSS and then causes aggregation of conducting PEDOT.^[62] Hence, the insertion of MoSe₂ QDs helps remove partial insulated PSS with strong acidity and hygroscopicity, which enhances the conductivity of the PEDOT:PSS and device stability. To gain more insights, we investigated the zeta potential of MoSe₂ QDs (Fig. 5c), which is measured to be -38.8 mV. Consequently, 2D MoSe₂ QDs with the negatively charged surface can promote the aggregation of PEDOT to form a more regular chain-type morphology, thus weaken the shield effect of PSS on PEDOT and improve the electron conductivity of PEDOT:PSS.^[63] There is no denying that the conformational change of PEDOT is concerned with the conductivity of the PEDOT:PSS film. Therefore, the Raman spectra were conducted to examine whether the improvement of conductivity is due to the conformational change of PEDOT chains from coil (benzoid structure) to linear (quinoid structure) or expanded-coil structure or not. It is noted that no obvious change was observed between 1400 and 1500 cm⁻¹ denoting $C_{\alpha}=C_{\beta}$ double bond in the PEDOT thiophene ring (Fig. 5d), [64,65] implying that the enhanced conductivity of the PEDOT:PSS film was not owing to the conformational change of PEDOT chains. Taking the above phenomena into account, we proposed a reasonable mechanism about the conductivity enhancement of PEDOT:PSS film induced by MoSe₂ QDs. From the chemical structures,^[66] one can conclude that the PSS chain is of much heavier molecular weight than PEDOT chain, thereby more positively charged PEDOT molecules are attached to PSS chains. As a result of the positive charge on PEDOT, the repulsions between the PSS anions were weakened, allowing the PEDOT to form a coiled or core-shell structure with insulated PSS surrounded.^[62] Furthermore, the negatively charged MoSe₂ QDs can attract the PEDOT and reject PSS, accelerate the aggregation of PEDOT in the core-shell structure, promote more conductive PEDOT chains to escape from PSS encirclement and boost up the conductivity of the whole PEDOT:PSS layer.

Finally, grazing-incidence wide-angle X-ray scattering (GIWAXS) was carried out to study the molecular packing and crystallinity orientation of the PM6:Y6. Fig. S9(a–c) depict the 2D GIWAXS images of the PM6:Y6 based on different HEL and the corresponding profiles both along the in-plane (IP, dotted line) (q_{xy}) and the out-of-plane (OOP, solid line) (q_z) directions. The profiles of PM6:Y6 films with different HEL exhibit strong (100) diffraction peak in IP direction and (010) diffraction peak in OOP direction, indicating the co-existence of face-on and edge-on molecular orientation in the BHJ. It could be clearly observed clearly from **Fig. S9c** that PEDOT:PSS based PM6:Y6 film exhibits a strong (010) peak in OOP direction at 1.74 Å⁻¹. When introducing MoSe₂ QDs interlayer, the (010) peak resulting from the π - π stacking can be observed in OOP direction at 1.76 Å⁻¹. According to the formulas of d = $2\pi/q$ and L = $2\pi K/\Delta q$, where q is the peak position, K is the

Scherrer constant (~ 0.9), Δq is the half maximum of each peaks, the π - π stacking distance (d-spacing) and the coherence length L are determined to be (0.360 nm and 0.217 nm) and (0.352 and 0.403 for PEDOT:PSS/PM6:Y6 respectively, and MoSe₂ nm). QDs/PEDOT:PSS/PM6:Y6. ^[68] So, it is clearly that the π - π stacking distance of the PM6:Y6 film based on bilayer HEL is smaller than the PM6:Y6 film based on PEDOT:PSS HEL, which is favorable for intermolecular electron hopping. In addition, the coherence length of PM6:Y6 film based on bilayer HEL is larger than that based on PEDOT:PSS. Note that the increased L corresponds to higher crystallinity and better phase separation of BHJ, leading to increased charge transport and decreased bimolecular recombination.



Fig. 6. GISAXS intensity profiles of (a) ITO/PEDOT:PSS and ITO/MoSe₂ QDs/PEDOT:PSS and (b) ITO/PEDOT:PSS/PM6:Y6 and ITO/MoSe₂ QDs/PEDOT:PSS/PM6:Y6 thin films along the IP directions. The Guinier approximation of GISAXS intensity profiles of (c) ITO/PEDOT:PSS and ITO/MoSe₂ QDs/PEDOT:PSS and (d) ITO/PEDOT:PSS/PM6:Y6 and ITO/MoSe₂ QDs/PEDOT:PSS/PM6:Y6 in low-*q* region.

In order to further investigate the effect of MoSe₂ QDs on mesoscopic structure, the grazing incidence small-angle X-ray scattering (GISAXS) of PEDOT:PSS and PM6:Y6 thin films with and without MoSe₂ QDs were determined. Fig. S10a and S10b presents the 2D GISAXS images of HEL and active layer with and without MoSe₂ QDs, respectively. From these scattering images, the Yoneda peak with a high intensity peak along the IP direction can be observed.^[69] The corresponding 1D IP profiles are shown in Fig. 6a and 6b and no obvious scattering peak is detected along the IP direction, indicating the nanostructure of PEDOT:PSS was less ordered along the q_{xy} direction.^[70] Fig. S10c and S10d show the 2D GISAXS images of the active layers with and without MoSe₂ QDs. It is clear that the scattering peaks were seen along the OOP direction, implying that the PM6:Y6 thin films form a layered structure.^[71] It is believed that the domain size is another key impact factor for the performance of OSCs. The radius of gyration (R_a) of the clusters of the PEDOT:PSS and PM6:Y6 thin films with and without MoSe₂ QDs is quantitatively calculated by Guinier approximation according to the following equation:^[72,73]

$$I\left(q_{xy}\right) = I(0) \exp\left(-\frac{q_{xy}^2 R_g^2}{3}\right)$$
(2)

where $l(q_{xy})$ and l(0) are the scattering intensity and the zero-angle scattering intensity, q_{xy} is obtained by $4\pi \sin(\frac{\theta}{2})/\lambda$, θ is the scattering angle and λ is the wavelength of incidence light. So, R_g is the average of the absolute values of two slopes $(lnl(q_{xy}) vs q_{xy}^2)$ in the low-q region (ca. 0.05–0.13 nm⁻¹). From Fig. 6c and 6d, the R_g of PEDOT:PSS, MoSe₂ QDs/PEDOT:PSS, PEDOT:PSS/PM6:Y6 and MoSe₂ QDs/PEDOT:PSS/PM6:Y6 thin films are calculated to be 35.18, 22.92, 33.82 and 29.36 nm, respectively, indicating that the insertion of MoSe₂ QDs contributes to the reduction of domain sizes and phase separation of both PEDOT:PSS and PM6:Y6.

It is well established that stability is an important factor for practical application of OSCs.^[74] To evaluate the stability of our OSCs, the performances in air of the control cell and the champion cell based on MoSe₂ QDs-6000r/PEDOT:PSS after simple encapsulation are measured as shown in Fig. S11. It can be seen distinctly that the PCE based on MoSe₂ QDs/PEDOT:PSS composite HEL can maintain 85% PCE after 120 h, which is superior to the control cell (67%). The better long-term stability is ascribed to the insertion of MoSe₂ QDs which can alleviate the large area contact of ITO and active layer with PSS.

To demonstrate the universality of the strategy as discussed in this work for performance optimization of OSCs, we also applied $MoSe_2 QDs/PEDOT:PSS$ composite HEL into other nonfullerene and fullerene-based systems, namely, PBDB-T/ITIC, PTB7-Th/PC71BM, and P3HT/PC61BM. The device performances of these OSCs are plotted in Fig. S12 and S13. It can be found that, the insertion of $MoSe_2 QDs$ leads to PCE improvement from 9.51% to 10.41%, 7.88% to 8.32%, and 2.93% to 3.24%, respectively, for the PBDB-T/ITIC, PTB7-Th/PC71BM, and P3HT/PC61BM-based cells. Detailed photovoltaic parameters are listed in Table S7. The similar performance enhancement in other systems demonstrates that introducing $MoSe_2 QDs/PEDOT:PSS$ as bilayer HEL is an effective pathway for the optimization of OSCs.

Conclusions

To conclude, MoSe₂ QDs were synthesized by a lithium intercalation method and utilized as a modifier in HEL for the optimization of OSCs. We found that high temperature annealing of thin film of MoSe₂ QDs led to the generation of O-Mo-Se band due to participated lattice oxygen which partially fill the Se vacancies and passivate the structural defects, as well as the formation of MoO₃ species. Meanwhile, we also learnt that the insertion of MoSe₂ QDs can induce the phase separation of PEDOT and PSS, and thus weaken the shield effect of PSS on PEDOT and improve the electron conductivity and WF of PEDOT:PSS. The device characterization results show that, BHJ solar cell based on MoSe₂ QDs/PEDOT:PSS composite HEL in PM6:Y6 and SZ2:N3 system yielded a maximum PCE of 16.08 and 17.08%, respectively, which are significantly higher than the control cell based on PEDOT:PSS (15.48 and 16.53%). These remarkable efficiencies in our work are the highest value among single junction solar cells with 2D materials as the interface modifier so far. The enhancement of device performances is mainly attributed to the improved hole transport ability and WF of HEL, increased exciton dissociation and charge extraction probability at the electrode/function layer interface, as well as reduced charge recombination due to the better phase

separation of PEDOT and PSS and active layer as a result of introduction of MoSe₂ QDs interlayer. Moreover, the stability of OSCs with MoSe₂ QDs is improved remarkably because MoSe₂ QDs interlayer can prevent PEDOT:PSS from corroding the ITO electrode and remove the partial PSS from the interface of HEL and active layer. All these factors work synergistically and result in a significant enhancement of device performance. This strategy is also adaptable in other OSC systems, which therefore opens a new pathway for the application of TMD materials as interfacial modifiers in the 3rd generation of OSCs.

Author Contributions

Hong Lian and Mingao Pan fabricated the solar cell devices and optimized their processing parameters for photovoltaic performance. Jinba Han produced and characterized the MoSe₂ QDs. Xiaozhe Cheng performed subsequent stability tests with the help of Hong Lian. Jiaen Liang synthesized the SZ2 material. Bin Wei and Qingchen Dong supervised the materials synthesis and device fabrication. Wenqiang Hua conducted the GIWAXS measurements and analysis. Yongquan Qu performed the UPS and XPS measurements and analysis. Yucheng Wu provided discussion and suggestions. The manuscript was written by Hong Lian, Qingchen Dong and He Yan. The project was planned, directed and supervised by Qingchen Dong, He Yan and Wai-Yeung Wong. All authors discussed the results and commented on the manuscript.

Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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