

# Impact of NO<sub>x</sub> reduction on long-term surface ozone pollution in roadside and suburban Hong Kong: Field measurements and model simulations

Lewei Zeng<sup>1,2</sup>, Jin Yang<sup>2</sup>, Hai Guo<sup>\*,2</sup>, Xiaopu Lyu<sup>2</sup>

<sup>1</sup> College of Chemistry and Environmental Engineering, Shenzhen University, Shenzhen, China

<sup>2</sup> Department of Civil and Environmental Engineering, The Hong Kong Polytechnic University, Hong Kong, China

\*Corresponding author: [ceguohai@polyu.edu.hk](mailto:ceguohai@polyu.edu.hk)

## Abstract

Continuous measurements of ozone (O<sub>3</sub>) and nitrogen oxides (NO<sub>x</sub>=NO+NO<sub>2</sub>) were conducted from 2007 to 2019 in Hong Kong in order to evaluate the effectiveness of control strategies for NO<sub>x</sub> emission from diesel commercial vehicles (DCV). DCV control programs were periodically applied in three phases starting from 2007, 2010 and 2014. It was found that NO and NO<sub>2</sub> levels decreased during the study period but more dramatically after the implementation of DCV Phase III than pre-DCV Phase III. Source apportionment analysis confirmed that the ambient NO and NO<sub>2</sub> in Hong Kong attributed to the regulated DCV emissions in Phase III reduced at rates of 5.1-14.4 ppbv/yr in roadside environment and 1.6-3.1 ppbv/yr in suburban area. Despite overall NO<sub>x</sub> reduction, increased NO<sub>2</sub>/NO<sub>x</sub> ratios were recorded during the study period possibly due to the application of diesel particulate filter (DPF) in DCVs. However, after introducing DCV Phase III, observed O<sub>3</sub> values experienced more

dramatic increasing trends in most areas of Hong Kong than pre-DCV Phase III. Model simulations revealed that O<sub>3</sub> production rate kept increasing and turned to be less sensitive to NO<sub>x</sub> from 2014 to 2019. On the roadside, net O<sub>3</sub> production rate was more than doubled during 2014-2019 owing to NO<sub>x</sub> reduction. Moreover, the levels of oxidants (OH, HO<sub>2</sub> and RO<sub>2</sub>) were 1.5-5 times those before 2014. In suburban environment, NO<sub>x</sub> reduction also facilitated O<sub>3</sub> production and radical cycling, but made smaller contributions than those on the roadside during 2014-2019. This study unraveled that NO<sub>x</sub> reductions benefited from DCV regulations caused increase in surface O<sub>3</sub> and fueled O<sub>3</sub> photochemistry in various environments. More stringent control measures on emissions of VOCs, especially those with high OH reactivity, might help to better mitigate O<sub>3</sub> pollution.

**Keywords:** *Nitrogen oxides, long-term variation, source apportionment, ozone photochemistry, ozone-precursors relationship, diesel commercial vehicles*

## 1 Introduction

Nitrogen oxides (NO<sub>x</sub>), the reactive nitrogen in the atmosphere, adversely affect human health (Hoek et al., 2002; Samoli et al., 2006) and cause eco-environmental problems, such as acid rain and algae blooms (Liu et al., 2020). They are also important precursors for the formation of tropospheric ozone (O<sub>3</sub>), which further leads to human health risks and reduced vegetation yields (Bell et al., 2004). As the most common air pollutant, O<sub>3</sub> is generated through photochemical reactions with the participation of volatile organic compounds (VOCs), NO<sub>x</sub> and carbon monoxide (CO)

in the presence of sunlight ([Atkinson, 2000](#)). In view of the nonlinear relationship between  $O_3$  and precursors, mitigating  $O_3$  pollution has become a complex issue.

In urban area,  $O_3$  production is generally in a VOC-limited regime due to intensive anthropogenic emissions ([Lu et al., 2019](#)). Road transport is considered as a significant contributor to ambient  $NO_x$ , accounting for more than 40 %, as well as VOCs, according to emission inventories in different countries ([EEA, 2015](#); [HKEPD, 2020](#)). In order to alleviate surface  $O_3$  and  $NO_x$  pollution, many parts of the world, including Hong Kong, which is a densely populated city in China, have implemented stringent control strategies to reduce vehicular emissions, especially diesel vehicle emissions. The design and installation of exhaust gas recirculation (EGR) and selective catalytic reduction (SCR) devices in diesel vehicles aimed to reduce  $NO_x$  emissions can be traced back to the 1970s ([Teshirogi et al., 1974](#)). Since then, stricter fuel and emission standards have been adopted, and substandard vehicles have been progressively phased out worldwide ([Guan et al., 2014](#); [Jiang et al., 2016](#); [Reşitoğlu et al., 2015](#); [Takekawa et al., 2013](#); [Wu et al., 2012](#)). For example, the Chinese government promoted the Clean Air Action Plan in 2013, in which the control of vehicular emissions is an important component ([Zhang et al., 2019](#); [Zhao et al., 2013](#)).

In Hong Kong, emission control strategies for diesel commercial vehicles (DCVs) began in April 2007 ([HKEPD, 2019a](#)). The first stage ended in March 2010, and pre-Euro and Euro I DCVs were replaced or upgraded. The measures in the second stage (July 2010 – June 2013) emphasized the upgrade of Euro II DCVs while the third stage (March 2014 – December 2019) was implemented to phase out all pre-Euro IV

DCVs. Meanwhile, SCRs and diesel particulate filter (DPF) were applied on Euro II and III franchised buses. [Table S1](#) summarizes all stages of DCVs control measures in Hong Kong. [Lyu et al. \(2017\)](#) evaluated the impact of DCVs control measures in Phases I and II on VOCs and O<sub>3</sub> formation; however, the effectiveness of control measures in DCV Phase III on NO<sub>x</sub> reduction and the subsequent impact on O<sub>3</sub> pollution has not been investigated yet.

Owing to strict regulations, NO<sub>x</sub> values in urban environments have mainly shown downward trends in the past two decades in Europe ([Beevers et al., 2012](#); [Carslaw et al., 2011](#); [Macdonald et al., 2021](#); [Vohra et al., 2021](#)), in North America ([Geddes et al., 2009](#); [Simon et al., 2015](#)), and in part of East Asia ([Itano et al., 2007](#); [Pandey et al., 2008](#); [Shon et al., 2011](#); [Zhang et al., 2014](#)). However, contrary to the decreasing trends of O<sub>3</sub> in most cases in Europe and North America, long-term observations revealed that O<sub>3</sub> in cities in East Asia, especially in China, increased during this period ([Li et al., 2020](#); [Lu et al., 2020](#)). With the aid of chemical transport model simulations, previous studies indicated that, among all possible driving factors, NO<sub>x</sub> reduction was the main cause of O<sub>3</sub> increase in China ([Liu and Wang, 2020](#); [Wang et al., 2019](#)). However, the underlying chemical mechanisms are missing.

In this study, O<sub>3</sub> and NO<sub>x</sub> were continuously monitored from 2007 to 2019 over the whole territory of Hong Kong. Temporal variations of O<sub>3</sub> and NO<sub>x</sub> were investigated and the effectiveness of the DCV Phase III on NO<sub>x</sub> variations was evaluated based on measurement data and source apportionment analysis. Moreover, by adopting a photochemical box model coupling with near-explicit master chemical mechanism

(PBM-MCM), we for the first time elucidated the impact of NO<sub>x</sub> variations on O<sub>3</sub> formation and attempted to provide a scientific basis for mitigating O<sub>3</sub> pollution in Hong Kong.

## 2 Data and methods

### 2.1 Data source

The Hong Kong government set up a complete monitoring network in the 1990s, measuring ambient trace gases, including O<sub>3</sub>, nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), CO and sulfur dioxide (SO<sub>2</sub>). Considering the percentage of valid data, we selected two roadside sites, eleven general ambient sites spreading in urban/suburban areas, and one rural site in this study. We acquired hourly monitoring data of the above five air pollutants during April 2007 – March 2019 from the website of Hong Kong Environmental Protection Department (HKEPD, <https://cd.epic.epd.gov.hk/EPICDI/air/station/>). Geographical locations of the selected 14 monitoring stations are shown in Figure 1, including two roadside sites, 11 general urban/suburban sites and one general rural site. Table S2 describes each monitoring station in detail, including the invalid data period.

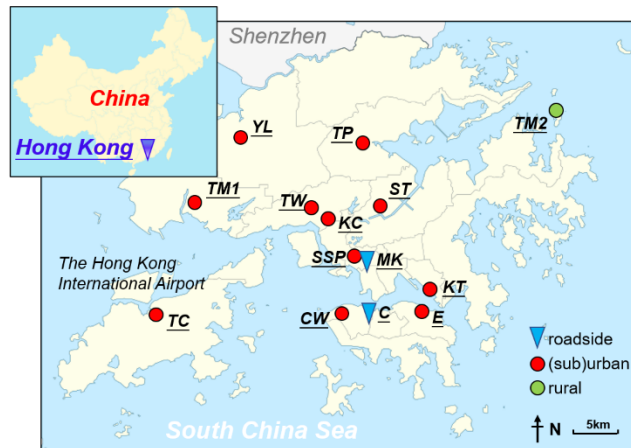


Figure 1. Geographical locations of monitoring stations in Hong Kong. Full names

and abbreviations of monitoring stations are listed in [Table S2](#).

In addition, real-time VOCs were continuously measured at one roadside site (MK) and one suburban site (TC) from 2013 to March 2019. Totally 27 C<sub>2</sub>-C<sub>9</sub> hydrocarbons, including 10 alkanes, 9 alkenes/alkynes and 8 aromatics, were identified and quantified by an online gas chromatography system (Syntech Spectra GC 955, Series 600/800) with a resolution of 30 min. Data were acquired and averaged into hourly values for analysis. Analytical instruments and quality control procedures were presented in previous studies ([HKEPD, 2019b](#); [Lyu et al., 2016](#)).

## 2.2 Simulation of in-situ photochemistry

In this study, we applied a PBM-MCM model developed by our own group to simulate the in-situ photochemistry at TC and MK sites in Hong Kong. Mixing ratios of trace gases (*i.e.*, NO, NO<sub>2</sub>, CO, SO<sub>2</sub> and O<sub>3</sub>) and VOCs from 7 am to 7 pm were input into this observation-based model to reproduce the real atmospheric conditions in the study area. Meteorological parameters, such as atmospheric pressure, temperature and relative humidity, were acquired from the Hong Kong Observatory and constrained in the model during the simulations. The developed PBM-MCM model has been widely utilized in Hong Kong and the model performance has been repeatedly testified in previous studies ([Lyu et al., 2016, 2017](#); [Wang et al., 2017](#); [Zeng et al., 2018](#)).

Index of agreement (IOA) represents the agreement between observed values ( $O_i$ ) and simulated values ( $S_i$ ), as described in [Equation 1](#).  $\bar{O}$  stands for the mean of observed data over  $n$  samples. Higher IOA value ( $0 < \text{IOA} < 1$ ) indicates better model

performance.

$$IOA = 1 - \frac{\sum_{i=1}^n (O_i - S_i)^2}{\sum_{i=1}^n (|O_i - \bar{O}| + |S_i - \bar{S}|)^2} \quad (1)$$

Since regional transport is not fully considered in the model, there might be differences between the observed and simulated values. However, because local contribution dominated the photochemical formation in urban Hong Kong, the discrepancy should not be large (Wang et al., 2017). In this study, the IOA reached 0.94 at the roadside MK site, and it decreased to 0.85 at the suburban TC site, possibly due to more regional transport of O<sub>3</sub> and VOC precursors from inland China to TC (Wang et al., 2017). High IOA values at both sites indicated high reliability of the in-situ photochemistry simulation in the study area.

### 2.3 O<sub>3</sub>-precursors relationship

In this study, the O<sub>3</sub>-precursors relationship was evaluated using the relative incremental reactivity (RIR), estimated based on the change in O<sub>3</sub> formation induced by the change in precursors levels (*i.e.*, NO<sub>x</sub>, CO and VOCs). The RIR values are calculated from Equation 2 (Wang et al., 2017; Zeng et al., 2018):

$$RIR(X) = \frac{[P_{O_3}(X) - P_{O_3}(X - \Delta X)] / P_{O_3}(X)}{\Delta S(X) / S(X)} \quad (2)$$

where X refers to one specific precursor, S(X) represents the mixing ratio of the specific precursor, P<sub>O<sub>3</sub></sub> represents the O<sub>3</sub> production rate in the unit of ppbv/h, ΔX is a hypothetical change in the mixing ratio of precursors. In this study, we assumed 10 % change in each type of precursor.

### 2.4 Scenario analysis

The impact of NO<sub>x</sub> variation on in-situ photochemistry in urban environments was

obtained from the difference between a base scenario and a hypothetical scenario. In the base scenario, hourly data of air pollutants, including trace gases and VOCs, and meteorological parameters from 2013 to 2019 (2556 days in total) were input into the model to simulate the O<sub>3</sub> photochemistry during this period. In the hypothetical scenario, the NO<sub>x</sub> values in 2013 remained unchanged in the following years (2014 - 2019), while the data of other pollutants and meteorological parameters were annually updated in the model using real measurement data from 2013 to 2019 as the settings in the base scenario. Radical concentrations (*i.e.*, OH, HO<sub>2</sub> and RO<sub>2</sub>) and pathway reaction rates in O<sub>3</sub> photochemistry from 7 am to 7 pm in each sampling day were output in both scenarios, and converted into monthly averages for further analysis. The difference between base scenario and hypothetical scenario represented the impact of NO<sub>x</sub> variation during 2014-2019 in urban Hong Kong, and offered the quantitative long-term effect of NO<sub>x</sub> variations on O<sub>3</sub> photochemistry.

### **3 Results and discussion**

#### **3.1 NO<sub>x</sub> observations in Hong Kong**

[Figure 2](#) and [Figure 3](#) display the monthly average NO and NO<sub>2</sub> mixing ratios at 14 monitoring stations in Hong Kong during the study period. 95% confidence intervals were calculated for averaged values. Overall, significantly higher NO ( $135.7 \pm 0.60$  ppbv,  $p < 0.01$ ) and NO<sub>2</sub> ( $54.0 \pm 0.2$  ppbv,  $p < 0.01$ ) averages during 2007-2019 were captured at the roadside sites (MK and C) than those at other 12 general ambient sites ( $44.0 \pm 0.09$  ppbv for NO,  $27.9 \pm 0.02$  ppbv for NO<sub>2</sub>). This was mainly attributed to the dense vehicular emissions on the roadside, especially diesel-fueled vehicles and



buses, which are well known to emit large amounts of NO<sub>x</sub> and particulate matter into the atmosphere. In general, mixing ratios of NO and NO<sub>2</sub> declined from 2007 to 2019, but the changing rates were not uniform across Hong Kong, as shown in Table S3. The reduction rates of NO ranged from  $11.9 \pm 0.6$  to  $14.9 \pm 0.8$  ppbv/yr ( $p < 0.01$ ) at the roadside sites, and from  $1.6 \pm 0.2$  to  $4.4 \pm 1.6$  ppbv/yr ( $p < 0.01$ ) at urban/suburban sites. Further, NO level at the rural site decreased insignificantly ( $p = 0.64$ ). As for NO<sub>2</sub>, it declined with decreasing rates of  $1.6 \pm 0.3$  -  $3.3 \pm 0.3$  ppbv/yr at the roadside sites, faster than those at the other sites (decreasing rates:  $0.4 \pm 0.1$ – $1.6 \pm 0.7$  ppbv/yr). The results indicated that the effectiveness of vehicle emission control measures was more obvious at the roadside sites.

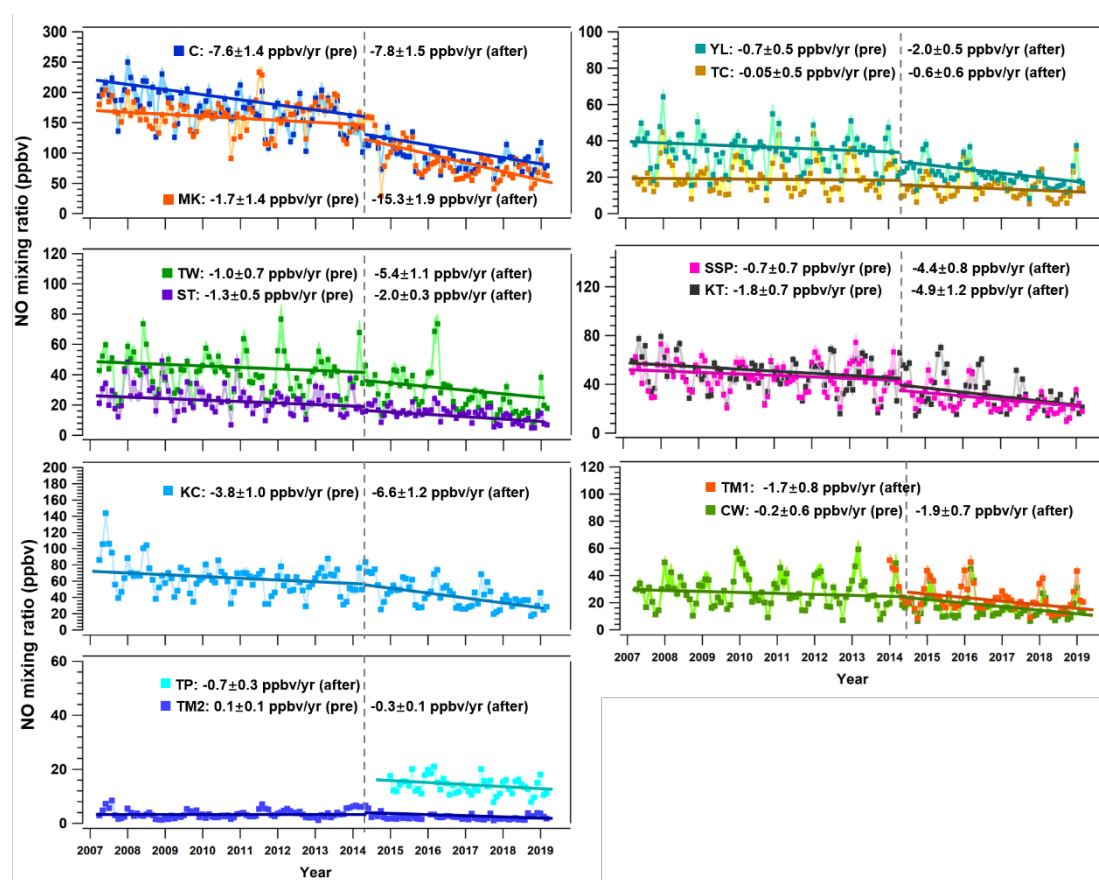


Figure 2. Monthly NO mixing ratios at 13 monitoring stations in Hong Kong during

Apr. 2007 – Mar. 2019. The grey dotted lines mark the start of the DCV Phase III program in Mar. 2014. The colored solid lines are the linear regressions of data points.

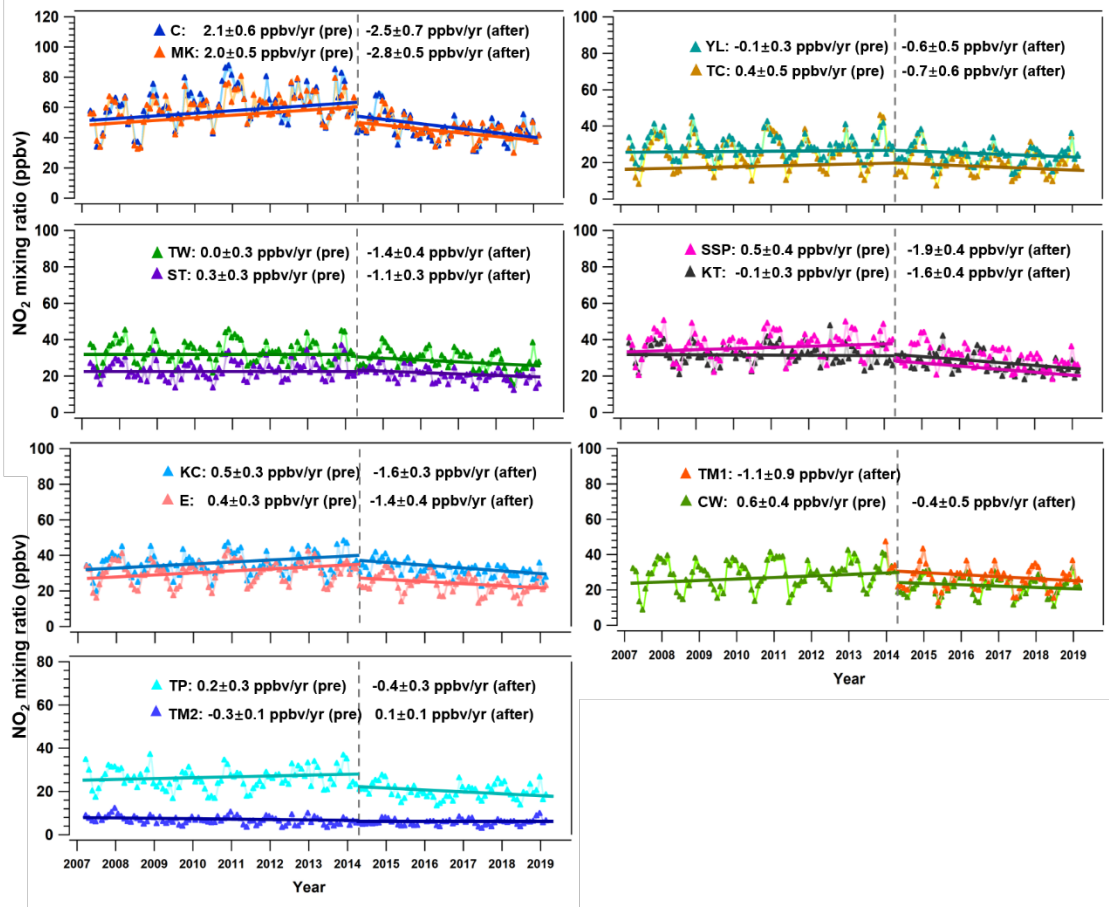


Figure 3. Monthly NO<sub>2</sub> mixing ratios at 14 monitoring stations in Hong Kong during Apr. 2007 – Mar. 2019. The grey dotted lines mark the start of the DCV Phase III in Mar. 2014. The colored solid lines are the linear regressions of data points.

To further look into the variation rates of NO<sub>x</sub> in different periods, we selected March 2014 as the threshold since it was the starting point of DCV Phase III, dividing the whole study period into pre-DCV Phase III (including Phases I and II) and the DCV Phase III. Changing rates at each sampling site during the two phases are listed in Table S4. In general, NO reduced more rapidly after the implementation of DCV

Phase III, especially at the roadside MK site, where NO level declined at a rate of  $15.3 \pm 1.9$  ppbv/yr, compared to  $1.7 \pm 1.4$  ppbv/yr in pre-DCV Phase III. To minimize the impact of meteorology in different years, the anomalies of NO and NO<sub>2</sub> were calculated from 2007 to 2019 by taking the average value for each month and subtracting the average for corresponding month over the whole study period (Text S1). As shown in Figure S1, NO level at the roadside sites experienced a dramatic drop-off due to the implementation of DCV Phase III, which kept obvious declining afterward, indicated by the positive anomalies (32.3-33.3 ppbv) in pre-DCV Phase III and negative anomalies (-51.7- -48.2 ppbv) during DCV Phase III. It is noteworthy that though NO at the rural site remained unchanged in pre-DCV Phase III, it significantly decreased ( $-0.3 \pm 0.1$  ppbv/yr) during the DCV Phase III. Different from the NO trend, discernible rises in NO<sub>2</sub> level were identified at the roadside sites before DCV Phase III with rates of  $2.0 \pm 0.5$  -  $2.1 \pm 0.6$  ppbv/yr ( $p < 0.01$ ). However, after the implementation of the DCV Phase III, NO<sub>2</sub> reduced at rates of 2.5 - 2.8 ppbv per year ( $p < 0.01$ ). Similar phenomena were found at other general ambient sites, where NO<sub>2</sub> level remained stable before the DCV Phase III but began to decline after that. Moreover, NO<sub>2</sub> anomalies in Figure S1 showed obvious differences before (5.5- 6.7 ppbv) and during (-8.7 - -8.2 ppbv) DCV Phase III.

Rising NO<sub>2</sub> level as well as NO<sub>2</sub>/NO<sub>x</sub> ratio were found in vehicle exhausts in previous measurements (Carslaw et al., 2015; Smit et al., 2019). Here, we further investigated the long-term variations of NO<sub>2</sub>/NO<sub>x</sub> ratio in the ambient air during 2007-2019, as shown in Figure S2. Ratios recorded in this study (0.24-0.75) were found to be above

the range of the emission ratios (0.03-0.22) in the exhaust (Preble et al., 2015; Smit et al., 2019; Tian et al., 2011), which was reasonable due to the shorter atmospheric lifetime of NO compared to NO<sub>2</sub>. Regarding temporal variations, NO<sub>2</sub>/NO<sub>x</sub> ratios were within 0.24-0.25 at roadside sites in 2017 and increased to 0.40-0.44 in 2019. The increased NO<sub>2</sub>/NO<sub>x</sub> ratio was possibly caused by intentional conversion of NO to NO<sub>2</sub> during the regeneration of diesel particulate filter (DPFs) (Preble et al., 2015; Smit et al., 2019). Higher NO<sub>2</sub>/NO<sub>x</sub> ratio at general sites than at roadside sites, increasing from 0.44-0.65 to 0.67-0.75, was mainly attributed to more secondary NO<sub>2</sub> formation in aged air masses.

To better elucidate the effectiveness of the DCV programs, source apportionment of 15 VOCs, CO, NO, NO<sub>2</sub> and SO<sub>2</sub> during 2014-2019 was conducted at the TC and MK sites (Figures S3 and S4). Model setting and the explanation of resolved factors were described in detail in Text S2. During the DCV Phase III, DCV emission on average accounted for  $69.8 \pm 0.4\%$  and  $63.0 \pm 0.4\%$  of ambient NO at MK and TC respectively, and  $55.1 \pm 0.7\%$  and  $32.8 \pm 0.3\%$  for ambient NO<sub>2</sub>, respectively. Both NO and NO<sub>2</sub> at MK experienced pronounced decline of  $-14.4 \pm 1.3$  ppbv/yr ( $p < 0.01$ ) and  $-5.1 \pm 0.5$  ppbv/yr ( $p < 0.01$ ), respectively, during DCV Phase III. Similarly, NO and NO<sub>2</sub> had dramatic drop-off with reduction rates of  $1.6 \pm 0.3$  ppbv/yr ( $p < 0.01$ ) and  $3.1 \pm 0.5$  ppbv/yr ( $p < 0.01$ ) at TC site, respectively. Therefore, NO<sub>x</sub> decreased in DCV exhaust were one of the main reasons of reductions in ambient NO<sub>x</sub>. This result was in accordance with NO<sub>x</sub> variation in the emission inventory, according to which NO<sub>x</sub> emission kept decreasing during 2000-2018 and had larger reduction rate after

2014 (HKEPD, 2021). Since SCRs for NO<sub>x</sub> removal were adopted in DCV Phase III (Table S1), it was reasonable that NO<sub>x</sub> level decreased dramatically at this stage. The results confirmed the effectiveness of regulations implemented in DCV Phase III, including tightening emission standard and application of SCRs and DPFs, on controlling real-world NO<sub>x</sub> level.

### 3.2 O<sub>3</sub> and VOCs observations in Hong Kong

Figure 4 shows the monthly O<sub>3</sub> variations at 14 monitoring stations in Hong Kong, respectively, during the study period. Generally, lower O<sub>3</sub> mixing ratio ( $p < 0.01$ ) was observed at the roadside sites ( $9.8 \pm 0.1$  ppbv), due to more intensive NO titration, compared to that at general ambient sites ( $21.6 \pm 0.02$  ppbv). Ambient O<sub>3</sub> increased throughout the whole period with significance ( $p < 0.05$ ) at most sites, with rates of  $0.5 \pm 0.2$ – $3.2 \pm 0.3$  ppbv/yr (Table S3). Furthermore, no significant trends were identifiable before the implementation of DCV Phase III as shown in Table S4, while obvious O<sub>3</sub> increases ( $0.3 \pm 0.5$ – $2.4 \pm 0.6$  ppbv/yr,  $p < 0.05$ ) were observed afterward at ten of the 14 sites. Besides, the number of O<sub>3</sub> episode days, defined as a day with maximum hourly O<sub>3</sub> value exceeding 80 ppbv, significantly ( $p < 0.05$ ) increased at six sampling sites during 2007-2019, and peaked in 2019 at all sampling sites (Table S5).

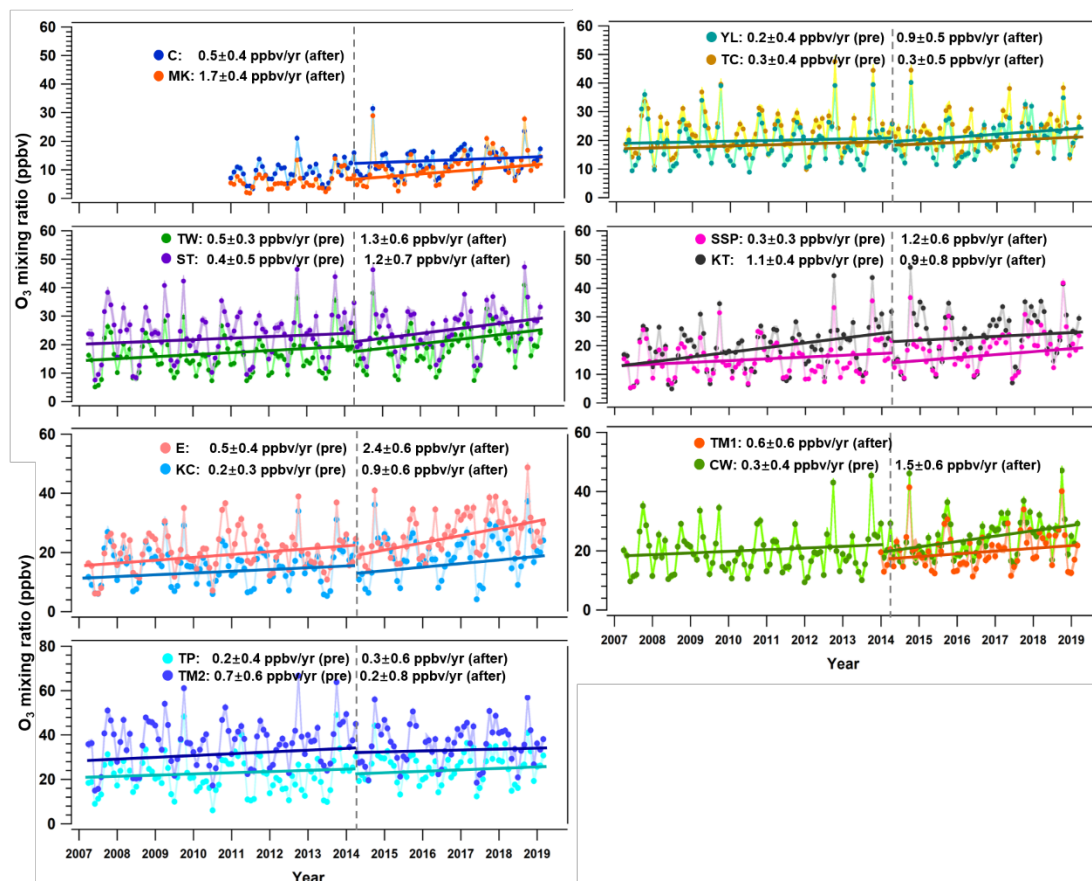


Figure 4. Monthly O<sub>3</sub> variations at 14 monitoring stations in Hong Kong during 2007-2019. Grey dotted lines indicate the start of DCV Phase III program in Mar. 2014. Colored solid lines are the linear regressions of data points.

As the O<sub>3</sub> precursors, the decline in NO<sub>x</sub> levels might be an important chemical driving factor for O<sub>3</sub> increase in a VOC control regime. In addition, the significant increase in total VOCs (TVOCs) level ( $1.3 \pm 0.5$  ppbv/yr,  $p < 0.05$ , Figure S5) at the general ambient TC site during 2014-2019 would be another possible reason. Moreover, the OH reactivity of VOCs also increased dramatically at a rate of  $0.4 \pm 0.06$  s<sup>-1</sup>/yr ( $p < 0.01$ , Figure S5). However, TVOCs level reduced at a rate of  $2.8 \pm 0.5$  ppbv/yr ( $p < 0.01$ , Figure S6) at the roadside MK site, so did the OH reactivity of VOCs ( $-0.2 \pm 0.06$  s<sup>-1</sup>/yr,  $p < 0.01$ , Figure S6). It was noteworthy that VOCs derived

from DCV emission decreased at both TC and MK sites at rates of  $-0.4 \pm 0.07$  and  $-1.7 \pm 0.2$  ppbv/yr ( $p < 0.01$ ), respectively, indicating the effectiveness of DCV Phase III in VOCs control. The difference in VOCs trends between roadside and suburban environments was probably due to different variations in local emission sources during 2014-2019, when biogenic emission increased significantly at TC ( $0.2 \pm 0.05$ ,  $p < 0.01$ ) but was invariable at MK. In addition to chemical driving factors, the impact of meteorology led to  $O_3$  increase in suburban Hong Kong before 2012, while an obvious decreasing trend was observed after 2012 (Zeren et al., 2021, under review). The impact of meteorology on the long-term  $O_3$  trend was insignificant ( $p = 0.93$ ). Bearing the effectiveness of DCV Phase III in mind, especially for  $NO_x$ , it is of high necessity to investigate the impact of  $NO_x$  chemistry on  $O_3$  trend in Hong Kong.

### 3.3 Variation trends of $O_3$ -precursors relationships in different environments

The  $O_3$ -precursors relationship is essential for the planning and formulation of emission reduction policies to control  $O_3$  pollution. Hence,  $O_3$  formation sensitivity to different groups of precursors, including anthropogenic VOCs (AVOCs), biogenic VOCs (BVOCs),  $NO_x$  and CO, in urban environments in Hong Kong from 2013 to 2019 were investigated using the PBM-MCM model. The results are shown in Figure 5. In general, RIR values of AVOCs, BVOCs and CO were positive, while negative for  $NO_x$  at both sites, which meant that  $O_3$  production was limited by VOCs and reduction in  $NO_x$  level led to  $O_3$  increase in these years. At the roadside MK site, the average absolute RIR value of  $NO_x$  ( $1.6 \pm 0.03$ ) was the highest, followed by AVOCs ( $1.2 \pm 0.04$ ) and BVOCs ( $0.08 \pm 0.02$ ), indicating the abundance of  $NO_x$  ( $86.3 \pm 2.1$  ppbv)



and its dominant impact on O<sub>3</sub> formation in such NO<sub>x</sub>-rich environment. Differently, the absolute RIRs of AVOCs ( $0.9 \pm 0.05$ ) ranked first at the suburban TC site, followed by NO<sub>x</sub> ( $0.7 \pm 0.05$ ) and BVOCs ( $0.2 \pm 0.03$ ), likely due to relatively low NO<sub>x</sub> level ( $33.1 \pm 1.0$  ppbv). The impact of CO on regulating O<sub>3</sub> formation was minor as its RIR values ( $0.02 \pm 0.01$ - $0.03 \pm 0.004$ ) at both sites were small. The findings agree with previous studies in Hong Kong (Guo et al., 2013; Wang et al., 2017). Here average RIR values before and after Mar. 2014 were calculated representing pre-DCV Phase III and DCV Phase III for intercomparison. The absolute RIR values of NO<sub>x</sub> and AVOCs in pre-DCV Phase III were  $1.7 \pm 0.06$  and  $1.3 \pm 0.09$ , larger than those ( $1.5 \pm 0.03$  and  $1.1 \pm 0.04$ ) in DCV Phase III. At TC site, the absolute RIR of NO<sub>x</sub> ( $0.9 \pm 0.09$ ) during pre-DCV Phase III was also larger than that ( $0.7 \pm 0.05$ ) in DCV Phase III. This result possibly indicated the decrease in sensitivity of NO<sub>x</sub> at both sites and in AVOCs at MK to O<sub>3</sub> production.

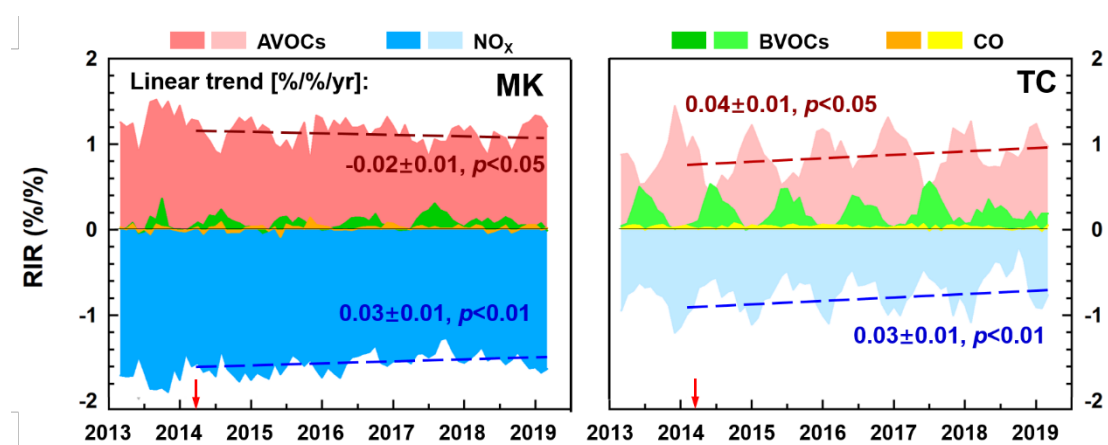


Figure 5. RIR values of O<sub>3</sub> precursors from 2013 to 2019 at MK site (left column) and TC site (right column). The linear trends during DCV Phase III are marked in colored lines and changing rates are shown inside the figures. Red arrows indicate the implementation of DCV Phase III.



The temporal variations of the RIRs were calculated after the implementation of DCV Phase III. RIRs of  $\text{NO}_x$  experienced an increase ( $p < 0.01$ ) at both sites with a growth rate of  $0.03 \pm 0.01$ . Besides, the RIR values of AVOCs decreased dramatically ( $p < 0.05$ ) at MK site at a rate of  $-0.02 \pm 0.01$ , while increased ( $p < 0.05$ ) at TC site at a rate of  $0.04 \pm 0.01$  throughout the study period. Moreover, the sensitivities of BVOCs and CO to  $\text{O}_3$  production remained unchanged ( $p > 0.05$ ) in these years. Overall, the evolution of RIRs implicated that  $\text{O}_3$  production was less sensitive to its precursors (*i.e.*,  $\text{NO}_x$  and AVOCs) at the roadside MK station, implying that reduction in same percentage of precursors might lead to less  $\text{O}_3$  decline in recent years. Consistently,  $\text{O}_3$  formation became more sensitive to AVOCs at the suburban TC site, while it turned to be less sensitive to  $\text{NO}_x$  during the study period. More rapid decrease in  $\text{NO}_x$  than in VOCs by current findings might result in enhanced  $\text{O}_3$  production. Hence, the next section will illustrate more detailed in-situ photochemistry in different environments.

### **3.4 Impact of $\text{NO}_x$ reduction on in-situ $\text{O}_3$ photochemistry**

In-situ  $\text{O}_3$  photochemistry was simulated using the PBM-MCM model and the  $\text{O}_3$  production and destruction rates were obtained during 7am-7pm for 2014-2019. Overall, the  $\text{HO}_2 + \text{NO}$  pathway contributed more, with average percentages of 55% at MK and 58% at TC, than the  $\text{RO}_2 + \text{NO}$  pathway to  $\text{O}_3$  production in Hong Kong, but their reaction rates were both on a similar level. The reaction between  $\text{RO}_2$  and NO contributed the remaining part (45% and 42%, respectively). In contrast,  $\text{OH} + \text{NO}_2$  was the dominant destruction pathway of  $\text{O}_3$  at both sites (98% at MK and 92% at TC). The results are consistent with those reported in other urban environments ([Liu](#)

et al., 2021; Lyu et al., 2019).

Figure 6 shows the monthly averaged changes in reaction rates of pathways in response to decreases of  $\text{NO}_x$  at MK and TC sites during 2014-2019 compared to 2013 in Hong Kong. Diurnal variations of changes in reaction rates were averaged during the whole period and shown in Figure 6 as well. Table S6 presents the percentage changes in reaction rates of pathways regulating  $\text{O}_3$  production and oxidative radicals induced by  $\text{NO}_x$  reduction during 2014-2019 and converted into an average value representing the situation in the study period. At MK site, two  $\text{O}_3$  formation pathways, *i.e.*,  $\text{HO}_2+\text{NO}$  and  $\text{RO}_2+\text{NO}$ , and the destruction pathway ( $\text{OH}+\text{NO}_2$ ) were all facilitated by  $\text{NO}_x$  reduction, leading to an enhancement in net  $\text{O}_3$  production rate during the whole period. Daytime net  $\text{O}_3$  production rate increased by 0.7 ppbv/h (177%) in average during 2014-2019, when reaction rates of individual pathways increased by 0.43 ppbv/h (155%) for “ $\text{HO}_2+\text{NO}$ ”, 0.37 ppbv/h (160%) for “ $\text{RO}_2+\text{NO}$ ” and 0.10 ppbv/h (99%) for “ $\text{OH}+\text{NO}_2$ ”. Variation rates of changes were calculated based on linear regression method. Changes in net  $\text{O}_3$  production rate at MK increased significantly at a rate of  $0.12\pm0.04$  ppbv/h/yr ( $p<0.01$ ), along with the continuous  $\text{NO}_x$  reduction. Besides, enhancements in “ $\text{HO}_2+\text{NO}$ ”, “ $\text{RO}_2+\text{NO}$ ” and “ $\text{OH}+\text{NO}_2$ ” increased at rates of  $0.08\pm0.02$ ,  $0.07\pm0.02$  and  $0.02\pm0.004$  ppbv/h/yr ( $p<0.01$ ), respectively, as  $\text{NO}_x$  levels kept decreasing at the site. Similar situation was found at TC, where daily net  $\text{O}_3$  production rate averagely enhanced by 4.0 ppbv/h (92%) and the enhancement increased at a rate of  $0.24\pm0.05$  ppbv/h/yr ( $p<0.01$ ) during 2014-2019. Changes in reaction rates of production pathways “ $\text{HO}_2+\text{NO}$ ” and

“RO<sub>2</sub>+NO” were 0.7 ppbv/h (18%) and 0.3 ppbv/h (15%), increasing at rates of 0.13±0.03 and 0.10±0.02 ppbv/h/yr ( $p<0.01$ ), respectively, while the change in destruction pathway “OH+NO<sub>2</sub>” was insignificant. Compared to TC site, larger enhancement in net O<sub>3</sub> production rate at MK site was caused by higher NO<sub>x</sub> reduction at MK (108.5 ppbv, 42%) than at TC (11.9 ppbv, 25%). Overall, decreases in NO<sub>x</sub> resulted in more sufficient O<sub>3</sub> formation in urban environments in Hong Kong.

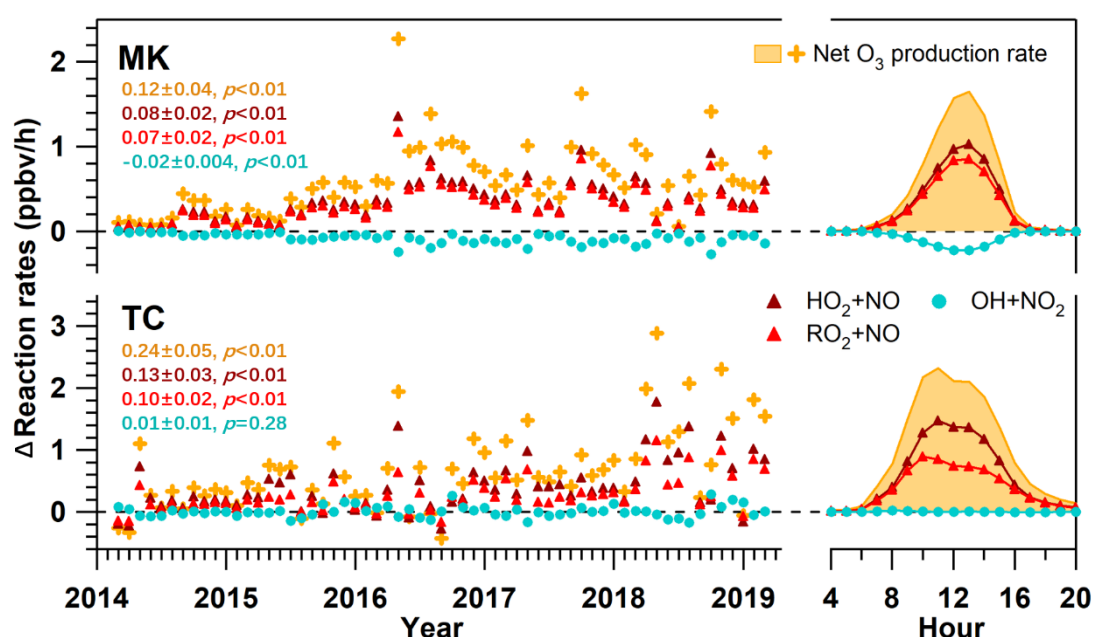


Figure 6. Monthly average changes (left) and diurnal variations of changes (right) in reaction rates of pathways in response to decreases of NO<sub>x</sub> at MK and TC sites during 2014-2019. Triangle and circular marks represent changes in production pathways and destruction pathways, respectively. Cross marks and shaded area in orange stand for the change in net O<sub>3</sub> production rate. Variation rates of changes in reaction rates were listed in the same color as the marks (unit: ppbv/h/yr). Pathways with contributions less than 5% are not shown.

To further investigate variations of atmospheric oxidative capacity caused by NO<sub>x</sub>

369 reduction, the concentrations of RO<sub>x</sub> radicals, including OH, HO<sub>2</sub> and RO<sub>2</sub> in 2014-  
 370 2019 were simulated. [Figure S7](#) shows the monthly average changes and diurnal  
 371 variations of changes in RO<sub>x</sub> radicals during 2014-2019 in response to NO<sub>x</sub> reduction  
 372 at MK and TC sites. Concentrations of OH, HO<sub>2</sub> and RO<sub>2</sub> radicals increased by  
 373  $6.1 \times 10^5$  (402%),  $9.2 \times 10^5$  (105%) and  $1.4 \times 10^6$  molecules/cm<sup>3</sup>/s (53%), respectively, in  
 374 response to NO<sub>x</sub> reduction at MK site. Besides, RO<sub>x</sub> radicals enhanced more in  
 375 concentrations but less in percentages at TC than at MK, by  $7.8 \times 10^5$  (23%),  $7.7 \times 10^6$   
 376 (35%) and  $6.3 \times 10^6$  molecules/cm<sup>3</sup>/s (32%) for OH, HO<sub>2</sub> and RO<sub>2</sub>, respectively,  
 377 induced by NO<sub>x</sub> reduction. With regard to temporal variations, the enhancement in  
 378 OH, HO<sub>2</sub> and RO<sub>2</sub> concentrations increased ( $p < 0.05$ ) at rates of  $1.0 \pm 0.2 \times 10^5$ ,  
 379  $0.7 \pm 0.3 \times 10^5$  and  $0.6 \pm 0.2 \times 10^5$  molecules/cm<sup>3</sup>/yr during 2014-2019 at MK,  
 380 respectively, and at rates of  $2.8 \pm 1.4 \times 10^5$ ,  $4.9 \pm 1.3 \times 10^6$  and  $3.7 \pm 0.8 \times 10^6$   
 381 molecules/cm<sup>3</sup>/yr ( $p < 0.05$ ) at TC, respectively. We further looked into reaction rates  
 382 in radical cycling for both OH and HO<sub>2</sub> and found that most of pathways were  
 383 enhanced during 2014-2019 in response to NO<sub>x</sub> reduction at both sites, as shown in  
 384 [Figures S8](#) and [S9](#). Moreover, those enhancements kept increasing significantly  
 385 ( $p < 0.05$ ) from 2014 to 2019 as NO<sub>x</sub> levels decreased in Hong Kong. An exception  
 386 was that the photolysis of HONO as well as its production through the reaction  
 387 between OH and NO which participated in OH radical cycling were suppressed by  
 388 NO<sub>x</sub> reduction at TC, but enhanced at MK. This could be explained by the different  
 389 variations in OH concentrations and NO levels at two sites. At MK, 42% reduction in  
 390 NO<sub>x</sub> level caused 192% increase in OH concentration, contributing to  $\sim 0.4$  ppbv/h

enhancement in pathways “HONO +  $h\nu$ ” and “OH + NO” during the study period. Differently, 25% decrease in NO<sub>x</sub> resulted in 27% increment in OH concentration at TC, which led to ~0.08ppbv/h suppression of these two pathways. Previous studies mentioned that vehicle exhaust contributed to ambient HONO, which played an important role in O<sub>3</sub> photochemistry (Kurtenbach et al., 2001). Yun et al. (2017) reported that 38% of HONO was contributed by vehicle exhaust at MK site, where DCVs accounted for 26% of the vehicle fleet. Zhang et al. (2016) found that only 2% of ambient HONO came from traffic emissions. However, very limited studies quantified the portion induced by DCV emissions. Model simulation indicated that 10% of ambient HONO reduction caused 1% and 4% decreases in O<sub>3</sub> production rate at TC and MK, respectively. In this study, we assumed an unchanged HONO emitted from DCVs in Hong Kong to solely investigate the impact of NO<sub>x</sub> reduction on in-situ photochemistry.

To further reveal the role of NO<sub>x</sub> in regulating the atmospheric oxidative capacity, a sensitivity test was conducted by cutting 10% of NO<sub>x</sub> as model input. Average changes in reaction rates modulating OH, HO<sub>2</sub> and RO<sub>2</sub> concentrations during the study period are shown in Figures S10-12. Table 1 summarizes the changes in sources, sinks and cycling of radicals. At TC site, the NO<sub>x</sub> reduction accelerated OH, HO<sub>2</sub> and RO<sub>2</sub> productions by 0.017, 0.006 and 0.159 ppbv/h, leading to an increment in radical cycling. At MK site, the decrease in NO<sub>x</sub> facilitated HO<sub>2</sub> and RO<sub>2</sub> productions by 0.017 and 0.015 ppbv/h, but restrained OH production by 0.012 ppbv/h. The increase in HO<sub>2</sub> and RO<sub>2</sub> radicals overwhelmed that in OH radical, further strengthening

radical cycling and contributing to oxidative capacity.

Table 1. Changes in reaction rates of pathways modulating OH, HO<sub>2</sub> and RO<sub>2</sub> concentrations in response to 10% reduction of NO<sub>x</sub> (Unit: ppbv/h).

<i>Site</i>	TC			MK		
<i>Radical</i>	OH	HO <sub>2</sub>	RO <sub>2</sub>	OH	HO <sub>2</sub>	RO <sub>2</sub>
Source	-0.001	0.009	0.160	-0.001	0.017	0.015
Sink	-0.018	0.002	0.001	0.011	0.0001	0.0001
Net	0.017	0.006	0.159	-0.012	0.017	0.015
Cycling	0.241	0.267	0.224	0.358	0.232	0.174

### 3.5 O<sub>3</sub> control measures

To develop science-based control measures, variations in O<sub>3</sub>-precursors relationships were clarified by plotting isopleths of net O<sub>3</sub> production rate as a function of OH reactivities of VOCs (OH reactivity<sub>VOCs</sub>) and NO<sub>x</sub> (OH reactivity<sub>NOx</sub>) based on an empirical kinetic modeling approach (EKMA) (Liu et al., 2021; Lyu et al., 2019). In this study, OH reactivity is defined as the product of O<sub>3</sub> precursors (*i.e.*, VOCs and NO<sub>x</sub>) concentrations and the reaction rate constants between O<sub>3</sub> precursors and OH. During 2013-2018, yearly average OH reactivity<sub>VOCs</sub> were within 67%-149% of the overall average. As for OH reactivity<sub>NOx</sub>, the range was 73-154%. To include OH reactivities in all sampling years at both sites, hypotheses were made to cover percentages from 0 to 200% with an interval of 10% of the OH reactivity for both VOCs and NO<sub>x</sub>. Using the PBM-MCM model, we simulated 441 scenarios and extracted maximum net O<sub>3</sub> production rates that occurred at 13:00 LT in each scenario to obtain the isopleths, as shown in Figure 7.

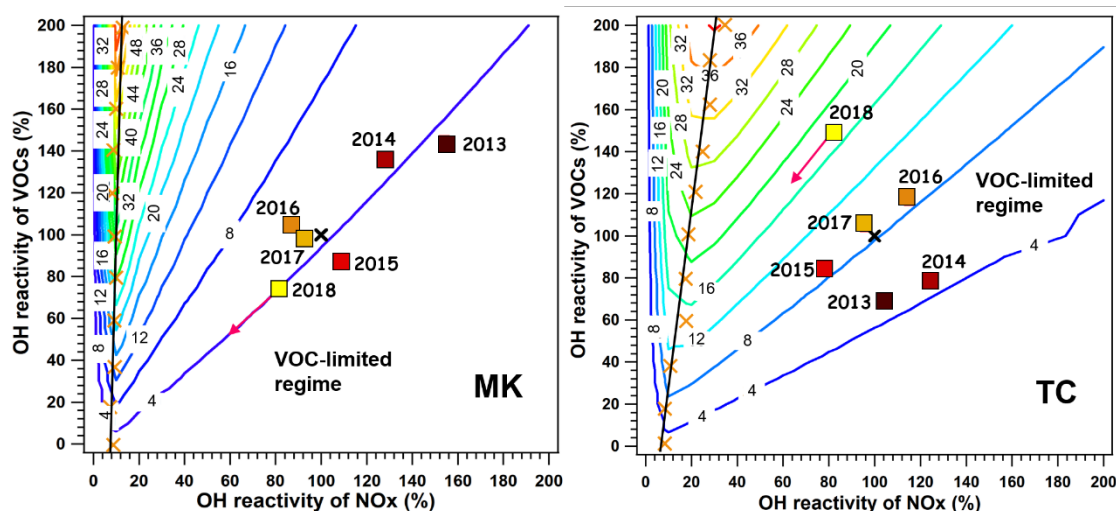


Figure 7. Average isopleth of net O<sub>3</sub> production rate (ppbv/hr) at 13:00 LT as a function of OH reactivity<sub>VOCs</sub> and OH reactivity<sub>NO<sub>x</sub></sub> at MK (left) and TC (right) sites. The colored squares represent calculated yearly average OH reactivity<sub>VOCs</sub> and OH reactivity<sub>NO<sub>x</sub></sub> at 13:00 LT from 2013 to 2018. Orange crosses denote the OH reactivity<sub>VOCs</sub> and OH reactivity<sub>NO<sub>x</sub></sub> at 13:00 LT with the highest net O<sub>3</sub> production rate at each given OH reactivity<sub>VOCs</sub>. Black cross represents the base case, referring to the overall average during the whole sampling period. Solid black line derives from the linear regression of orange crosses and divides O<sub>3</sub> formation into VOCs-limited regime and NO<sub>x</sub>-limited regime. Pink arrow represents the tangent line of the isopleth for 2018 data.

O<sub>3</sub> formation regimes were defined at both TC and MK sites during the study period. Detailed description of this approach is given in [Text S3](#). The isopleths indicated that O<sub>3</sub> formation was controlled by VOCs at both MK and TC sites from 2013 to 2018, which meant that cutting VOCs emission would be more effective to suppress O<sub>3</sub> production in Hong Kong. At MK site, the O<sub>3</sub> production rate was the lowest in 2013,

and increased during 2014-2015 due to the declined OH reactivity<sub>VOCs</sub> and OH reactivity<sub>NO<sub>x</sub></sub>. From 2016 to 2017, the combined effect of decreased OH reactivity<sub>NO<sub>x</sub></sub> and increased OH reactivity<sub>VOCs</sub> led to an obvious rise in O<sub>3</sub> production. Although OH reactivity<sub>VOCs</sub> reached the lowest value during the study period in 2018, the net O<sub>3</sub> production rate still exceeded that in 2013 due to continuous decreasing OH reactivity<sub>NO<sub>x</sub></sub>. At TC site, O<sub>3</sub> production rate increased by ~12 ppbv/h from 2013 to 2018, owing to 21% reduction in OH reactivity<sub>NO<sub>x</sub></sub> and 80% increment in OH reactivity<sub>VOCs</sub>. In view of the remarkable drop in ambient NO<sub>x</sub> in Hong Kong in the past decade, the O<sub>3</sub> control strategies in such a VOCs-limited regime have proposed stricter requirements for cutting VOCs emissions. Based on pollutant profiles in 2018, the ratio of OH reactivity reduction between VOCs and NO<sub>x</sub> should be larger than 1 at MK site and 1.3 at TC site for O<sub>3</sub> abatement (indicated by pink arrows in [Figure 7](#)).

#### **4 Implications**

In this work, we found that after the implementation of DCV Phase III, the reduction rates of NO and NO<sub>2</sub> were both greater than before. Furthermore, NO<sub>x</sub> reduction was more significant in roadside environment than in suburban/urban environments. Source apportionment results confirmed the decline in NO<sub>x</sub> from diesel vehicle emissions in these years. Besides, increasing NO<sub>2</sub>/NO<sub>x</sub> ratio were recorded, possibly due to the wide application of DPFs. Generally, O<sub>3</sub> increased faster after the implementation of DCV Phase III, largely attributed to dramatic NO<sub>x</sub> reduction. To verify the inference, the PBM-MCM model for the first time was applied to evaluate the impact of NO<sub>x</sub> reduction on O<sub>3</sub> photochemistry in two microenvironments in



Hong Kong. During 2014-2019, the net O<sub>3</sub> production rate was more than doubled because of 42% of NO<sub>x</sub> reduction in the roadside environment. Owing to NO<sub>x</sub> reduction, OH, HO<sub>2</sub> and RO<sub>2</sub> radicals were 1.5 to 5 times those in pre-DCV Phase III period. In suburban environment, declines in NO<sub>x</sub> also led to enhancement in atmospheric oxidative capacity, but the change in O<sub>3</sub> production rate was smaller than that at the roadside site. Isopleth of net O<sub>3</sub> production rate revealed that O<sub>3</sub> formation was controlled by VOCs in Hong Kong during the sampling period. To alleviate O<sub>3</sub> pollution, the result suggests the reduction ratio of OH reactivity between VOCs and NO<sub>x</sub> should be larger than 1 at the roadside site and 1.3 in the suburban area, until the control regime switches to transitional or NO<sub>x</sub>-limited regime. In other words, confining VOC emissions is a key step for Hong Kong to control O<sub>3</sub> at this stage, especially those VOCs with high OH reactivities. To sum up, this study reveals that NO<sub>x</sub> reduction due to the implementation of DCV emission control measures led to continuous increase in O<sub>3</sub> in Hong Kong. The findings contributed to a more comprehensive understanding of the O<sub>3</sub> photochemistry associated with NO<sub>x</sub> variations in various environments.

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