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Semiconductor-to-metal transition in platinum dichalcogenides induced by niobium dichalcogenides

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Abstract

Metallizing 2D semiconductors is a crucial research area with significant applications, such as reducing the contact resistance at metal/2D semiconductor interfaces. This is a key challenge in the realization of next-generation lowpower and high-performance devices. While various methods exist for metallizing Mo- and W-based 2D semiconductors like MoS₂ and WSe₂, effective approaches for Pt-based ones have been lacking. This study demonstrates that platinum dichalcogenides (PtX_2 , X = Se or Te) undergo a semiconductorto-metal transition when grown on niobium dichalcogenides (Nb X_2 , X = Se or Te). PtX₂/NbX₂ heterostructures were fabricated using molecular beam epitaxy (MBE) and characterized by Raman spectra, scanning transmission electron microscopy (STEM) and scanning tunneling microscopy/spectroscopy (STM/STS). Raman spectra and STEM confirm the growth of 1T-phase PtX₂ and 1H-phase NbX₂. Both 2D STS mapping and layer-dependent STS show that regardless of their layer numbers, both pristine semiconducting PtSe2 and PtTe₂ are converted to metallic forms when interfacing with NbSe₂ or NbTe₂. Density functional theory (DFT) calculations suggest that the metallization of PtSe₂ on NbX₂ and PtTe₂ on NbTe₂ results from interfacial orbital hybridization, while for PtTe₂ on NbSe₂, it is due to the strong p-doping effect caused by interfacial charge transfer. Our work provides an effective method for metallizing PtX₂ semiconductors, which may lead to significant applications such as reducing the contact resistance at metal electrode/2D semiconductor interfaces and developing devices like rectifiers, rectenna, and photodetectors based on 2D Schottky diodes.

KEYWORDS

density functional theory calculations, niobium dichalcogenides, platinum dichalcogenides, scanning tunneling microscopy/spectroscopy, semiconductor-to-metal transition, two-dimensional materials

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1 | INTRODUCTION

Metal/semiconductor interface is crucial for both conventional and 2D semiconductor electronics as it plays a vital role in device operations. Metallizing 2D semiconductors is a key research area, with one of its primary applications being the reduction of contact resistance at the metal electrode/2D semiconductor interfaces. This remains a significant challenge in 2D electronics, as achieving Ohmic contact to 2D semiconductors is essential for fully exploiting their potential in device applications. Besides, metallizing 2D semiconductors has other applications, such as 2D Schottky diode-based rectifiers and rectenna.²⁻⁴ Several approaches have been explored to metallize conventional 2D semiconducting transition metal dichalcogenides (TMDs), particularly Mo- and doping, 5-8 including TMDs. engineering, 9-16 and hybridization with semimetals like bismuth (Sb) and antimony (Bi). 17-19 However, these approaches often encounter limitations due to a lack of versatility, and the degree of success typically depends on the type of 2D materials. For example, heavy doping is a mature technology in 3D semiconductors, but it is less effective for 2D semiconductors because their atomically thin nature makes it difficult to accommodate dopants. Phase engineering applies only to materials capable of stably existing in both semiconducting and metallic phases, such as 2H/1T-MoS₂ and 2H/1T'-MoTe₂. 9-16 Up to now, state-of-the-art technology has successfully metallized MoS₂ by hybridizing it with semimetals such as Sb and Bi, pushing its contact resistance close to the limit. 17,19 Noble quantum transition metal dichalcogenides (NTMDs) are an emerging subclass of TMDs attracting attention due to their unique properties, such as facile synthesis, giant thickness-mediated band gaps, extra-high carrier mobility and conductivity, and robust stability in air. These characteristics position NTMDs as promising materials for applications in electronics and optoelectronics.²⁰⁻²² However, effective metallization methods for NTMDs remain much less studied. Hence, developing efficient methods for metallizing NTMDs holds great importance toward enhancing performance levels across devices based on this material class.

Unlike conventional TMDs that can be metallized by microscale phase engineering, PtX₂ predominantly exists stably in the semiconducting 1T phase. This indicates that traditional phase engineering methods are ineffective for PtX₂. Nevertheless, in contrast to conventional TMDs such as MoX₂ and WX₂, NTMDs possess a unique characteristic wherein their electronic properties are highly influenced by interfacial interactions. For instance, due to enhanced interlayer hybridization with

increasing layer number, both PtS2 and PdSe2 exhibit widely tunable bandgaps from \sim 1.6 and \sim 1.3 eV for the monolayer to ~ 0.25 and 0 eV in bulk, respectively, $^{23-27}$ while PtSe2 transitions from being a semiconductor with a bandgap of \sim 2.0 eV to a metal when its thickness exceeds five layers, 28 and PtTe2 even becomes semimetallic starting from its second layer.^{29–32} This motivates us to explore an innovative approach to metallize PtX₂ by leveraging the strong interfacial interaction. Specifically, interfacing PtX2 with other 2D materials that have stronger interactions with them may lead to metallization of PtX₂.³³ This method may offer dual benefits: Firstly, enabling metallization of PtX2, and secondly, potentially reducing or eliminating the Schottky barrier commonly existing at sharp 2D metal/2D semiconductor interfaces because it does not involve any phase change.

In this article, we demonstrate that platinum dichalcogenides (PtX2, X = Se or Te) undergo a semiconductor-to-metal transition induced by niobium dichalcogenides (Nb X_2 , X = Se or Te). High-quality vertical and lateral PtX2/NbX2 heterostructures were grown on highly oriented pyrolytic graphite (HOPG) using molecular beam epitaxy (MBE). The structures were studied by Raman spectra, aberration-corrected scanning transmission electron microscopy (STEM) and scanning tunneling microscopy/spectroscopy (STM/STS) as well as density functional theory (DFT) calculations. Raman spectra and STEM confirm the growth of 1T-phase PtX₂ and 1H-phase NbX₂. STM reveals that both PtSe₂ and PtTe₂ adopt a monolayer-dominated growth mode on NbX₂, whereas PtSe2 forms stacked layers and PtTe2 forms a bilayer film when grown on HOPG. STS results indicate that pristine PtSe₂ exhibits a decreasing band gap with an increasing layer number: 2.0 ± 0.1 , 1.1 ± 0.1 , 0.6 ± 0.1 , and 0.20± 0.1 eV for 1-4 layers, respectively, and 0 eV for 5 or more layers, while PtTe2 is a semiconductor with a bandgap of ~ 0.8 eV in monolayer form and becomes a semimetal from the second layer onwards. Both PtSe2 and PtTe₂ are converted to metallic when interfacing with NbSe₂ and NbTe₂. Such semiconductor-to-metal transition occurs not only in monolayer PtX2 directly in contact with NbX₂, but also in multilayer PtX₂ regardless of their layer numbers. DFT calculations suggest that the metallization of PtSe₂ on NbX₂ and PtTe₂ on NbTe₂ results from interfacial orbital hybridization, while for PtTe₂ on NbSe₂ it is due to the strong p-doping effect caused by interfacial charge transfer. This work provides an effective strategy for metallizing NTMDs and offers a deep insight into the metallization at an atomic level, which may lead to significant applications such as reducing contact resistance at metal electrode/2D semiconductor interfaces and developing devices like rectifiers, rectenna, and photodetectors based on 2D Schottky diodes.

2 | RESULTS AND DISCUSSION

Based on previous research, both PtSe2 and PtTe2 crystallize in an octahedral (1T) structure with $P3\overline{m}1$ space group (Figure 1A) within a growth or annealing temperature range of 200-300°C, while above this range, they lose some chalcogen atoms and transform into other Ptchalcogenides with lower chalcogen stoichiometries than the PtX₂.^{30,34-38} For NbSe₂, growth occurs at temperatures <400°C for the trigonal prismatic (1H) phase with P63/mmc space group (Figure 1B), between 400 and 500°C for the mixed 1T and 1H phases, and >500°C for the 1T phase. ^{39–42} Therefore, to avoid complications arising from mixing different crystal phases or phase transitions due to temperature change, we selected 280°C as the growth temperature for both PtX2 and NbX2. While prior reports indicated NbTe₂ to be in the 1T phase, 43-45 it was found that it crystallizes in the 1H phase in our experiment, as will be discussed later. This finding is supported by a very recent work, which provides strong evidences for the existence of incognizant H phase in monolayer NbTe2 and suggests the preferred growth of H phase at low growth temperatures. 46 Vertical and lateral heterostructures were prepared by sequential growths of NbX₂ and PtX₂, as schematized in Figure 1C: NbX₂ was firstly grown by co-deposition of Nb and Se or Te onto HOPG at 280°C, then followed by the growth of PtX₂ by co-deposition of Pt and Se or Te atoms onto the NbX₂/HOPG at 280°C. By adjusting the amounts of each material to submonolayer levels, both (Figure 1D) and vertical (Figure 1E) heterostructures were achieved for STEM and STM/STS analyses. For clearance and simplicity, we refer to the n-layer as nL (where n = 1, 2, 3, ...), the lateral heterostructure of 1L-PtX₂/1L-NbX₂ as 1L-PtX₂-1L-NbX₂, and the vertical heterostructure of $1L-PtX_2/1L-NbX_2$ as $1L-PtX_2\bot 1L-PtX_2\bot 1L$ NbX₂ hereafter. When there is no need to distinguish between lateral and vertical heterostructures, we still use PtX₂/NbX₂ to refer to heterostructures. In certain vertical heterostructures, the second layer of PtX2 extends over the 1L-PtX2-1L-NbX2 below as schematized in Figure 1E, providing an intuitive comparison of the influence of 1L-PtX2 and 1L-NbX2 on the electronic properties of the upper PtX₂ layer.

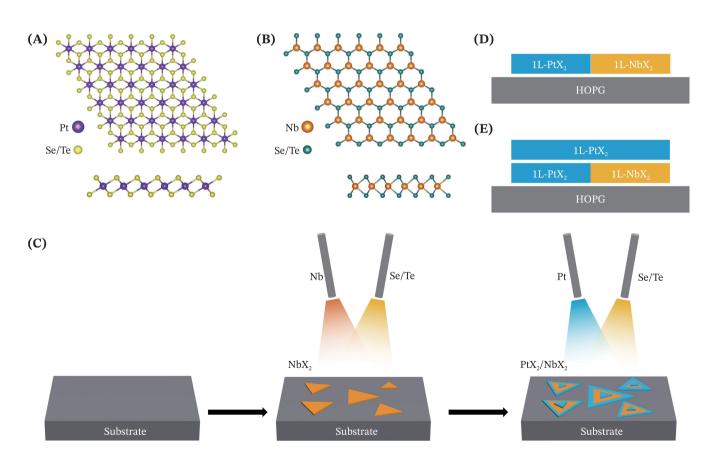


FIGURE 1 (A) and (B) Lattice structures of 1T-phase PtX_2 and 1H-phase NbX_2 , respectively. (C) Schematic of sample preparation: NbX_2 is first grown on HOPG substrate, then PtX_2 is grown, forming lateral and vertical heterostructures. (D) Schematic of a lateral heterostructure. (E) Schematic of a vertical heterostructure, in which the upper PtX_2 layer extends over the $1L-PtX_2-1L-NbX_2$ lateral heterostructure below.

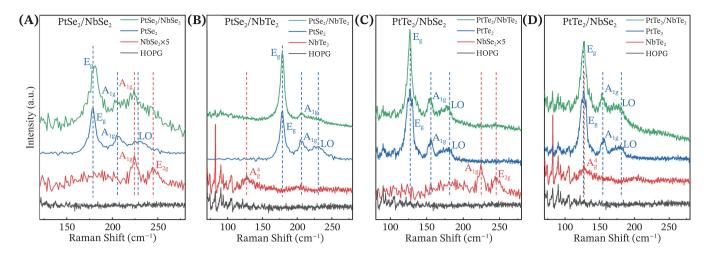


FIGURE 2 Raman spectra. (A)–(D) Raman spectra of 1L-PtSe₂ $\pm 1L$ -NbSe₂, 1L-PtSe₂ $\pm 1L$ -NbTe₂, 1L-PtTe₂ $\pm 1L$ -NbSe₂, and 1L-PtTe₂ $\pm 1L$ -NbTe₂, respectively. In each panel, Raman spectra of PtX_2 , NbX₂, and HOPG were added as a comparison.

The growth of NbX2 and PtX2 as well as PtX2/NbX2 heterostructures was first monitored using Raman spectra (Figure 2). To simplify the analysis, we just focus on 1L-PtX₂⊥1L-NbX₂ to disregard the influence of 1L-PtX₂− 1L-NbX₂. This is feasible because NbX₂ predominantly forms monolayer films on HOPG, while PtX2 does so on NbX₂/HOPG, as confirmed by STM later (Figure 5). By maintaining constant growth conditions (such as temperature and evaporation rates of metals and chalcogens), the thickness in unit of layer is directly proportional to the growth time when the thickness is ≤1L. Therefore, by controlling the growth time, 1L- $PtX_2 \perp 1L-NbX_2$ with minimal $1L-PtX_2-1L-NbX_2$ can be obtained. In the case of NbSe₂ (Figure 2A,C), peaks at 224.8 and 243.5 cm^{-1} correspond to the A_{1g} and E_{2g} vibrational modes of NbSe₂. 47-49 For NbTe₂ (Figure 2B,D), only a peak at \sim 127 cm⁻¹ can be barely observed, which should represent the A_g⁴ vibrational mode of NbTe₂. ^{50–52} The Raman signals for both NbSe₂ and NbTe2 are extremely weak due to their monolayer thickness and being the bottom layer. This is consistent with previous studies. 47-49,51 In PtSe2, three peaks are clearly identified at \sim 178.0, 205.8 and 229.7 cm⁻¹, corresponding to the Eg, Alg and longitudinal optical (LO) modes of PtSe₂, respectively.^{53,54} Regarding PtTe₂, the Eg, A1g and LO Raman active modes are observed at \sim 126.6, 155 and 179 cm⁻¹, respectively, in line with previous reports. 55,56 These Raman spectra confirm the growth of PtX2 and NbX2. Due to the extremely weak Raman signals from NbX2 and partial overlap with the peaks of PtX₂ (Figure 2A,D), only the Raman peaks of PtX_2 can be observed in the $1L-PtX_2\perp 1L-NbX_2$ heterostructures. The Raman features of both PtSe2 and PtTe₂ remain unchanged in all 1L-PtX₂⊥1L-NbX₂

heterostructures, suggesting that no phase change of PtX_2 occurs when grown on NbX_2 .

During the preparation of heterostructures using the MBE method, alloying or doping may occur. Therefore, STEM was conducted to further study the microstructures of PtX₂/NbX₂ heterostructures, as shown in Figure 3. Lateral PtX₂/NbX₂ boundaries are indicated by white dashed lines. In the case of PtSe₂/NbSe₂ (Figure 3A), the boundaries appear somewhat disordered due to atom interdiffusion from PtX₂ and NbX₂, but this disorder is confined to a few nanometers near the boundaries. Away from the boundaries, clear lattice structures are observed: 1H-NbSe₂ with a lattice constant of ~0.35 nm and 1T-PtSe₂ with a lattice constant of \sim 0.37 nm, as shown in Figure 3B,C, respectively. In contrast, for PtSe₂/NbTe₂, sharper boundaries are observed. However, there are point defects present in NbTe2 regions arranged in straight lines at angles of approximately 60° between different lines (indicated by yellow lines in Figure 3D). Since during the growth of PtSe₂, previously grown NbTe2 would be inevitably exposed to Se, these defects are speculated to be Se atoms introduced during PtSe2 growth. To prove this, we used XPS to investigate how Se affects NbTe2 by subjecting NbTe2 to the same Se atmosphere as in PtSe₂ growth at 280°C. The results are presented in Figure S1 of the Supporting Information. With the increase of exposure time of NbTe₂ in the Se atmosphere, the peak intensity of Te3d gradually decreases, while that of Se3d gradually increases, showing the partial replacement of Te atoms by Se atoms. Despite these defects, the zoom-in image verifies the 1H phase of NbTe₂ with a lattice constant of \sim 0.37 nm (Figure 3E), while the PtSe2 regions show the 1T phase with a lattice constant of ~ 0.37 nm (Figure 3F). The

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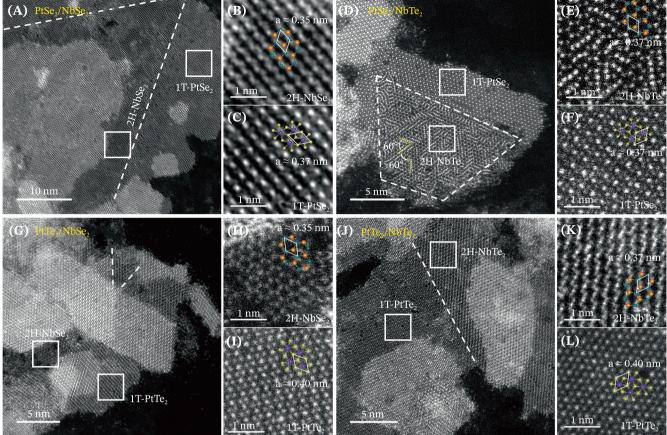


FIGURE 3 STEM images of PtX_2/NbX_2 heterostructures. Large-scale and zoom-in images of (A)–(C) $PtSe_2/NbSe_2$, (D)–(F) $PtSe_2/NbTe_2$, (G)–(I) $PtTe_2/NbSe_2$, and (J)–(L) $PtTe_2/NbTe_2$. PtX_2/NbX_2 boundaries are indicated by white dashed lines. The zoom-in images are from the areas enclosed by white squares in the corresponding large-scale images and are overlaid with lattice models and lattice constants to show the 1T phase of PtX_2 and 1H phase of PtX_2 . In panel D, the directions of typical lines of defects are indicated by yellow lines and they form \sim 60° angles. There is no apparent alloying occurring in heterostructures other than the $PtSe_2/NbTe_2$ heterostructure.

sharpest boundaries were found in $PtTe_2/NbSe_2$ and $PtTe_2/NbTe_2$ heterostructures (Figure 3G–L), where distinct phases were identified through zoom-in images with their respective lattice constants confirmed: For example, the $NbSe_2$ region (Figure 3H) shows 1H phase with a lattice constant of \sim 0.35 nm, the $NbTe_2$ region (Figure 3K) shows 1H phase with a lattice constant of \sim 0.37 nm, and the $PtTe_2$ regions (Figure 3I,L) show 1T phase with a lattice constant of \sim 0.40 nm. Overall findings from STEM analyses confirm the 1T phase of PtX_2 and 1H phase of NbX_2 and support that there is no apparent alloying occurring in heterostructures other than the $NbTe_2$ within the $PtSe_2/NbTe_2$ heterostructure.

The impact of NbX $_2$ on the electronic properties of PtX $_2$ was examined using STM/STS. Initially, we analyzed the morphological and electronic properties of pristine NbX $_2$ and PtX $_2$ on HOPG, as depicted in Figure 4. Both NbSe $_2$ and NbTe $_2$ form crystal islands at almost monolayer on HOPG with their first layers being \sim 0.85 and \sim 1.0 nm in apparent height, respectively

(Figure 4A,B). The atomic resolution STM (AR-STM) images show their good crystallinity with lattice constants of ~ 0.35 and ~ 0.37 nm, respectively (Figure 4E,F). The gapless STS confirms their metallic nature (Figure 4I,J), consistent with previous reports. 39,57 Unlike NbX₂, PtSe₂ forms a stack of different layers with its first layer and second layer being ~ 0.70 and ~ 0.60 nm in apparent height, respectively (Figure 4C), while PtTe₂ prefers bilayer growth mode with its first layer and second layer being ~ 0.90 and ~ 0.50 nm, respectively (Figure 4D). The AR-STM images validate their excellent crystallinity with lattice constants of ~0.37 and \sim 0.40 nm for PtSe₂ and PtTe₂, respectively (Figure 4G,H). The lattice constants of PtX₂ and NbX₂ obtained through STM match well with those by STEM in the previous section. The STS of PtSe₂ shows a layerdependent bandgap of \sim 2.0, \sim 1.1, \sim 0.6, and \sim 0.2 eV for layer numbers from 1 to 4 (Figure 4K), while for PtTe₂ it shows a bandgap of ~0.8 eV for the monolayer and 0 eV for the bilayer. The layer-dependent bandgaps of PtSe₂

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FIGURE 4 STM/STS of pristine NbX₂/HOPG and PtX₂/HOPG. (A)–(D) Large-scale STM images of NbSe₂, NbTe₂, PtSe₂, and PtTe₂, respectively. Height profiles along the white lines in each panel are included to display their apparent heights (unit: nm). (E)–(H) The corresponding atomic-resolution STM images with unit cell and lattice constants shown. (I)–(L) The corresponding STS taken at the colored dots in the large-scale images. (A)–(D) $V_{\text{sample}} = -2.0 \text{ V}$, I = 5 pA. (E)–(H) $V_{\text{sample}} = 0.01 \text{ V}$, I = 400 pA. (I) and (J) $V_{\text{sample}} = 0.5 \text{ V}$, I = 250 pA. (K) and (L) $V_{\text{sample}} = 1.0 \text{ V}$, I = 250 pA.

and PtTe₂ are consistent with our previous studies and literatures, ^{28,31,32} proving their good crystallinity and semiconducting behavior.

Figure 5A–D present the STM morphologies of PtX₂/NbX₂ heterostructures and Figure 5E–H are the corresponding zoom-in images. From an overall perspective, there are two prominent features: One is that PtX₂ mainly grows along the edge of NbX₂ islands, forming 1L-PtX₂–1L-NbX₂, and on top of NbX₂ islands, forming 1L-PtX₂±1L-NbX₂; The other one is that PtX₂ forms monolayer-dominated films on NbX₂, which are distinguished from their pristine growth behaviors on HOPG, indicating a stronger van der Waals (vdW) interaction between PtX₂ and NbX₂. As materials tend to grow where they experience stronger interactions, locations with stronger interactions will have more materials growing there, while locations with weaker interactions will

have fewer materials growing in that area. In order to qualitatively determine the ranking of vdW interactions in the 1L-PtX₂\pm1L-NbX₂, we calculated the ratio of the area where a single layer of PtX2 grows on top of another single layer of PtX2 (i.e., 2L-PtX2) and where a single layer of PtX₂ grows on top of a single layer of NbX₂ (i.e., 1L-PtX₂\lambda1L-NbX₂) (Figure S2), and the results are shown in Table S1. The results show that the ranking of the vdW interactions is $PtSe_2/NbTe_2 > PtSe_2/NbSe_2 >$ $PtSe_2/PtSe_2 \approx PtSe_2/HOPG$ for $PtSe_2$ and $PtTe_2/NbTe_2 >$ $PtTe_2/NbSe_2 \approx PtTe_2/PtTe_2 > PtTe_2/HOPG$ for $PtTe_2$. In Figure 5E-H, boundaries between PtX2 and NbX2 (indicated by white dashed lines) are visible but vary in sharpness. For PtSe₂/NbSe₂ (Figure 5E), the boundaries are less distinct due to the interdiffusion of atoms at the interface, as shown in our STEM study (Figure 3). In Figure 5F, the color, which represents the apparent



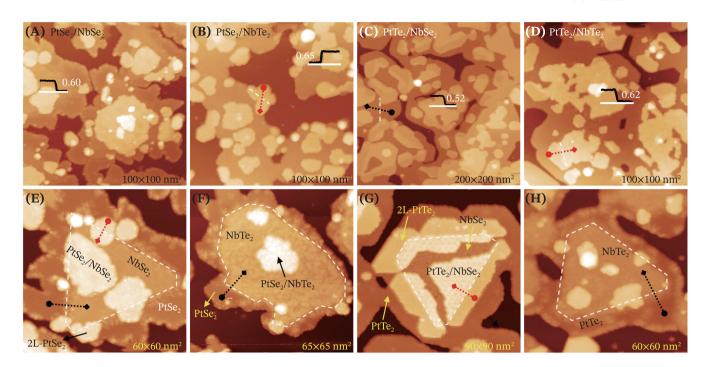


FIGURE 5 STM of PtX₂/NbX₂ heterostructures. (A)–(D) Large-scale STM images of PtSe₂/NbSe₂, PtSe₂/NbTe₂, PtTe₂/NbSe₂, PtTe₂/NbTe₂ heterostructures, respectively. Height profiles along the white lines in each panel are included to display the apparent height of 1L-PtX₂ grown on 1L-NbX₂ (unit: nm). (E)–(H) The corresponding zoom-in images, respectively. White dashed lines indicate the boundaries between 1L-PtX₂ and 1L-NbX₂. (A)–(D) $V_{\text{sample}} = -2.0 \text{ V}$, I = 5 pA. (E)–(H) $V_{\text{sample}} = -1.0 \text{ V}$, I = 20 pA.

height, of the NbTe₂ region is obviously non-uniform. This is because some of the Te atoms were replaced by Se atoms during the growth of PtSe2, altering surface electronic states, as discussed earlier. The sharpest boundaries appear between PtTe₂ and NbSe₂ (Figure 5G). Owing to the strong interaction between NbSe₂ and PtTe₂, a clear contrast can be seen in the second-layer PtTe₂ extending over the 1L-PtTe₂-1L-NbSe₂ below. For PtTe₂/NbTe₂ (Figure 5H), although the boundaries are sharp as revealed by STEM (Figure 3), the contrast of surface electronic states across the boundaries is much more blurred than for PtTe₂/NbSe₂, which means that the influence of the boundaries on the electronic states of these two materials extends over a longer distance. The apparent height of 1L-PtX2 grown on 1L-NbX2 was measured to be \sim 0.60 nm and \sim 0.65 nm for PtSe₂ on NbSe₂ and NbTe₂, respectively, and \sim 0.52 nm and \sim 0.62 nm for PtTe₂ on NbSe₂ and NbTe₂, respectively, as shown by the height profiles in Figure 5A-D. These apparent heights are apparently smaller than those of 1L-PtX₂ directly grown on HOPG (Figure 4A-D), further implying a stronger interaction between PtX₂ and NbX₂.

To investigate the electronic evolution in lateral PtX₂/NbX₂ heterostructures, we conducted position-dependent STS along the black dashed arrows from the PtX₂ region (round dot) to the NbX₂ region (square dot) in Figure 5 and presented the findings in Figure 6. The upper panels

of Figure 6 display 2D STS mappings from 1L-PtX2 to 1L-NbX₂, while the lower panels show typical STS spectra corresponding to the dots in the upper panels with red dots in the 1L-PtX₂ region, green dots in the boundary region, and blue dots in the 1L-NbX2 region. In the case of 1L-PtSe₂/1L-NbSe₂ (Figure 6A,E), there is a consistent bandgap value of \sim 2.0 eV observed within the area of 1L-PtSe₂, aligning with that of pristine 1T-PtSe₂ on HOPG, confirming its pure phase. Changes are confined to a narrow zone near boundaries. Repeatable STS measurements in the NbSe2 region confirm the pure phase of NbSe₂. For PtSe₂/NbTe₂ (Figure 6B,F), a constant bandgap of ~2.0 eV is also maintained within the PtSe₂ region, indicating its purity as seen with pristine PtSe₂ on HOPG. In the NbTe₂ region, metallic properties are confirmed; however, relatively random STS spectra are observed due to the partial replacement of Te by Se, as revealed by STEM (Figure 3B) and XPS (Figure S1). Regarding PtTe₂/NbSe₂ and PtTe₂/NbTe₂, STS shows a consistent bandgap of ~0.8 eV in the PtTe2 region and good repeatability of metallic character in the NbSe₂ and NbTe₂ regions, confirming their purity. The difference between these two heterostructures is that in PtTe₂/NbSe₂, the change occurs sharply within a very narrow region at the boundaries due to the sharp boundaries between PtTe₂ and NbSe₂, while in PtTe₂/ NbTe₂, the change is relatively slower at the

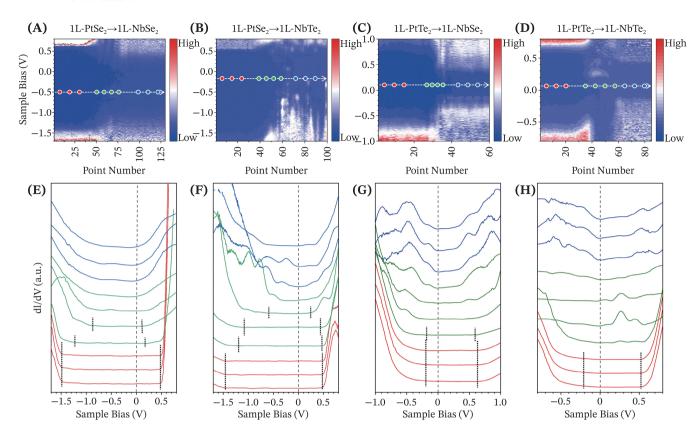


FIGURE 6 STS evolution from 1L-PtX₂ to 1L-NbX₂. (A)–(D) upper panels: 2D STS mapping from 1L-PtSe₂ to 1L-NbSe₂, 1L-PtSe₂ to 1L-NbTe₂, 1L-PtTe₂ to 1L-NbSe₂, and 1L-PtTe₂ to 1L-NbTe₂, respectively. Lower panels: Selected STS corresponding to the red, green and blue dots in upper panels. Short dashed lines indicate the band edges. (A)–(H) $V_{\text{sample}} = 1.0 \text{ V}$, I = 250 pA.

boundaries because the influence of the boundaries on the electronic states of these two materials extends over a longer distance.

In certain vertical heterostructures, the second layer of PtX₂ extends over the underlying 1L-PtX₂-1L-NbX₂, as schematized in Figure 1E. This allows for a direct comparison of the impact of the lower PtX₂ layer and NbX₂ layer on the electronic properties of the upper PtX₂ layer by comparing the STS of 1L-PtX₂\pm11L-NbX₂ with that of 1L-PtX₂\perp 1L-PtX₂ (i.e., 2L-PtX₂) through 2D STS mapping, as illustrated in Figure 7. The upper panels of Figure 7 display the 2D STS mappings from 2L-PtX₂ to 1L-PtX₂\pm1L-NbX₂, while the lower panels show typical STS spectra corresponding to the dots in the upper panels with red dots in the 2L-PtX2 region, green dots in the boundary region, and blue dots in the 1L-PtX₂\pm1L-NbX₂ region. For PtSe₂/NbSe₂ (Figure 7A,E), along the scanning path from 2L-PtSe₂ to 1L-PtSe₂\pm 11L-NbSe₂, there is initially a constant bandgap of \sim 1.1 eV in the 2L-PtSe₂ region, which then rapidly narrows upon approaching the underlying 1L-PtSe₂-1L-NbSe₂ boundary before completely disappearing within the 1L-PtSe₂\pm1L-NbSe₂ area. This not only indicates the pure phase of PtSe2 but also showcases its semiconductor-to-metal transition

when interfacing with NbSe₂. For PtSe₂/NbTe₂ (Figure 7B,F), the bandgap also remains constant in the 2L-PtSe₂ region, then rapidly narrows near the underneath 1L-PtSe₂-1L-NbTe₂ boundary and ultimately disappears in the 1L-PtSe₂⊥1L-NbTe₂ region. However, the STS appears relatively chaotic in the 1L-PtSe₂\(\pm\)1L-NbTe₂ region, likely due to the random electronic states of the underlying 1L-NbTe2 caused by the partial replacement of Te by Se, as discussed before. In PtTe2/NbSe2 (Figure 7C,G) and PtTe₂/NbTe₂ (Figure 7D,H) heterostructures, the STS reveals sharp evolutions at the boundaries from 2L-PtTe2 to 1L-PtTe2 L1L-NbX2. However, no bandgap is observed in the 1L-PtTe₂\pm11L-NbX₂ region, indicating a semiconductor-to-metal transition of 1L-PtTe₂ on NbX₂. In pristine 2L-PtTe₂ (i.e., 1L-PtTe₂\(\pm\)1L-PtTe₂), the density of states (DOS) above the Fermi level (positive sample bias) is higher in intensity than that below the Fermi level (negative sample bias). Notably, for both 1L-PtTe₂\perp11L-NbSe₂ and 1L-PtTe₂\perp11L-NbTe₂, there is a reversal where the DOS below the Fermi level becomes larger than that above the Fermi level.

To illustrate the difference in conductivities between PtX₂ on HOPG and on NbX₂, typical spectra of tunneling

FIGURE 7 STS evolution from $2L-PtX_2$ to $1L-PtX_2 \pm 1L-NbX_2$. (A)–(D) 2D STS mappings from $2L-PtSe_2$ to $1L-PtSe_2 \pm 1L-NbSe_2$, $2L-PtSe_2$ to $1L-PtSe_2 \pm 1L-NbTe_2$, respectively. (E)–(H) The corresponding STS of the red, green and blue dots in upper panel (A)–(D). Short dashed lines indicate the band edges. (A)–(H) $V_{\text{sample}} = 1.0 \text{ V}$, I = 250 pA.

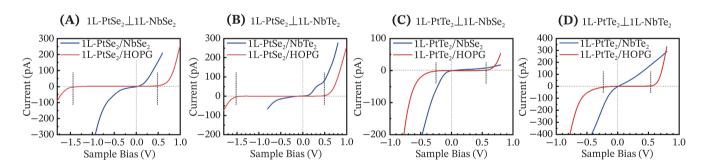


FIGURE 8 Tunneling current vs sample bias spectra of (A) 1L-PtSe₂ \pm 1L-NbSe₂ and 1L-PtSe₂/HOPG, (B) 1L-PtSe₂ \pm 1L-NbTe₂ and 1L-PtSe₂/HOPG, (C) 1L-PtTe₂ \pm 1L-NbSe₂ and 1L-PtTe₂ \pm 1L-NbTe₂ and 1L-PtTe₂/HOPG. Short dashed lines indicate the band edges. (A)–(D) $V_{\text{sample}} = 1.0 \text{ V}$, I = 250 pA.

current versus sample bias are plotted, as depicted in Figure 8. In the case of 1L-PtSe₂/HOPG, the tunneling current remains close to zero within a sample bias range from approximately -1.4 to 0.6 V, consistent with its bandgap of around 2.0 eV. Conversely, for 1L-PtSe₂ \perp 1L-NbSe₂ and 1L-PtSe₂ \perp 1L-NbTe₂, the tunneling currents deviate significantly from zero when the samples bias is non-zero, indicating the metallic nature of 1L-PtSe₂ on NbX₂ (see Figure 8A,B). Similarly, for 1L-PtTe₂/HOPG, there is nearly no tunneling current observed within a

sample bias spanning roughly -0.2 to 0.6 V corresponding to its bandgap of about 0.8 eV. However, both $1\text{L-PtTe}_2 \pm 1\text{L-NbSe}_2$ and $1\text{L-PtTe}_2 \pm 1\text{L-NbTe}_2$ demonstrate non-zero tunneling currents under non-zero sample bias, signifying a semiconductor-to-metal transition of PtTe_2 induced by NbX_2 , as shown in Figure 8C,D. To further confirm if the transition only occurs in the PtX_2 layer directly in contact with NbX_2 , we measured the layer-dependent STS of PtX_2 on NbX_2 and show the results in Figure 9. It can be seen that both PtSe_2 and

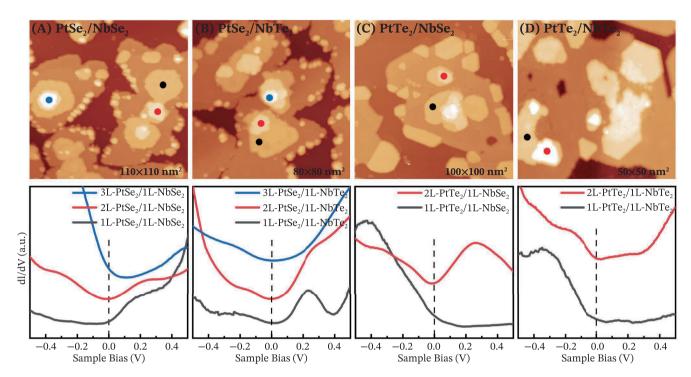


FIGURE 9 Layer-dependent STS of (A) PtSe₂ \pm 1L-NbSe₂, (B) PtSe₂ \pm 1L-NbTe₂, (C) PtTe₂ \pm 1L-NbSe₂, and (D) PtTe₂ \pm 1L-NbTe₂. The upper panels display the STM images, and the lower panels present the layer-dependent STS corresponding to the dots in the upper panels. The Black dots are on the 1L-PtX₂/1L-NbX₂, red dots on the 2L-PtX₂/1L-NbX₂, and blue dots on the 3L-PtX₂/1L-NbX₂. Both PtSe₂ and PtTe₂ exhibit metallic properties on NbX₂ regardless of their layer numbers. (A)–(D) STM images: $V_{\text{sample}} = -2.0 \text{ V}$, I = 5 pA. (A)–(D) STS: $V_{\text{sample}} = 1.0 \text{ V}$, I = 250 pA.

TABLE 1 Layer-dependent bandgaps of PtX_2 on NbX_2 and HOPG.

Substrates PtX ₂	NbSe ₂	NbTe ₂	HOPG	
PtSe ₂	Metallic	Metallic	1L	$2.0 \pm 0.1 \text{ eV}$
			2L	$1.1 \pm 0.1~\text{eV}$
			3L	$0.6 \pm 0.1 \text{ eV}$
			4L	$0.2 \pm 0.1 \text{ eV}$
			≥5L	0 eV
PtTe ₂	Metallic	Metallic	1L	$0.8 \pm 0.1 \text{ eV}$
			≥2L	0 eV

 $PtTe_2$ exhibit metallic properties on NbX_2 regardless of their layer numbers. The layer-dependent bandgaps of PtX_2 on NbX_2 and HOPG are summarized in Table 1.

DFT calculations were performed to shed light on the semiconductor-to-metal transition of 1T-PtX₂ when interfacing with 1H-NbX₂. The optimized in-plane lattice constant is 3.732 Å for 1T-PtSe₂, 4.003 Å for 1T-PtTe₂, 3.476 Å for 1H-NbSe₂, and 3.699 Å for 1H-NbTe₂, which are close to our experimental values. Following Anderson's rule, the energy levels of isolated 1T-PtX₂ were first aligned with

those of isolated 1H-NbX₂ (refer to Figure S3). The Fermi level of 1H-NbX₂ is situated within the band gap of 1T-PtSe2, indicative of weak charge transfer between them. On the contrary, the Fermi levels of 1H-NbSe₂ and 1H-NbTe₂ are respectively 0.631 eV (0.717 eV) and 0.178 eV (0.132 eV) below the valence band maximum of 1T-PtTe₂ at the optB88-vdW (HSE06) level of theory, suggesting that interfacial charge transfer from 1T-PtTe2 to 1H-NbX2 may play an important role in the semiconductor-to-metal transition of 1T-PtTe₂, especially for the 1T-PtTe₂\(\pm\)1H-NbSe₂ heterostructure. Figure 10 shows the calculated projected band structures of the 1T-PtX₂\(\pm\)1H-NbX₂ heterostructures in their respective most stable stacking configurations (see Figures S4–S7). For 1T-PtTe₂⊥1H-NbSe₂ in Figure 10C, the strong interfacial charge transfer leads to pronounced p-doping of 1T-PtTe₂. This is consistent with our experimental observations that there are more states below the Fermi level than above the Fermi level. When interfacing with 1H-NbTe₂, however, p-doping of 1T-PtTe₂ is weaker due to a smaller offset (0.178 eV) (Figure 10D). For 1T- $PtSe_2 \perp 1H-NbX_2$ heterostructures (Figure 10A,B), the interfacial orbital hybridization is likely responsible for the experimentally observed semiconductor-to-metal transition. The 1H-NbX₂ stacking-induced metallicity of 1T-PtX₂ was further revealed by the projection ratio of 1T-PtX2 to

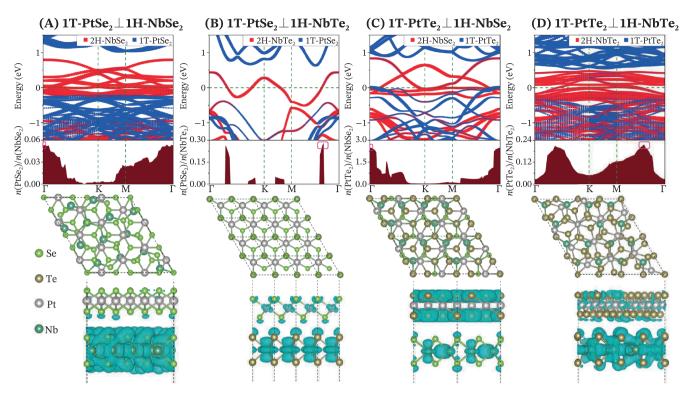


FIGURE 10 Top panels: The projected band structures of (A) 1T-PtSe₂ \pm 1H-NbSe₂, (B) 1T-PtSe₂ \pm 1H-NbTe₂, (C) 1T-PtTe₂ \pm 1H-NbSe₂ and (D) 1T-PtTe₂ \pm 1H-NbTe₂ heterostructures using the optB88-vdW functional. Middle panels: The respective total projection ratio of PtX₂ to NbX₂ for the states within 0.1 eV of the Fermi level along the high-symmetry k-path. Bottom panels: The top and side views of the respective heterostructure in the most stable stacking configuration. The partial charge associated with the Bloch state marked by a pink round rectangle in the middle panel is superimposed on the side view. The iso-value is set to (A) 0.001 e/Å³, (B) 0.004 e/Å³, (C) 0.004 e/Å³, and (D) 0.0005 e/Å³, respectively.

1H-NbX₂ near the Fermi level along the high-symmetry kpath, as shown in the middle panel of Figure 10. The projection ratio for 1T-PtTe₂\(\pm\)1H-NbX₂ is generally higher in magnitude than for 1T-PtSe₂⊥1H-NbX₂. Although the projection ratio for 1T-PtSe₂\(\pm\)1H-NbTe₂ is comparable to that for 1T-PtTe2 1H-NbTe2, the former has fewer electronic states near the Fermi level across the Brillouin zone (see the middle panel of Figure 10B,D). This relative abundance of the contact-induced 1T-PtX2 states around the Fermi level is further corroborated by the calculated projected density of states in Figure S8. Furthermore, Figure S9 shows the estimated interlayer binding energies, which suggest that 1T-PtTe₂ in general binds more strongly with 1H-NbX2 than 1T-PtSe2. These analyses may suggest that 1T-PtTe2 has better conductivity than 1T-PtSe2 when interfaced with 1H-NbX2, in line with our experiments. In the bottom panel of Figure 10, we further visualized the partial charge of the Bloch states near the Fermi level at the k-point where the projection ratio of 1T-PtX₂ to 1H-NbX₂ is the highest (marked by the pink round rectangle in the middle panel of Figure 10). In the case of the 1T-PtTe₂\(\pm\)1H-NbSe₂ heterostructure, the strong interfacial charge transfer leads to significant partial charge

within the 1T-PtTe₂ layer. On the other hand, the minor partial charge is observed within the 1T-PtSe₂ layer in 1T-PtSe₂ \perp 1H-NbX₂ and the 1T-PtTe₂ layer in 1T-PtTe₂ \perp 1H-NbTe₂, further manifesting the role of interfacial orbital hybridization in the relatively weak semiconductor-to-metal transition.

3 | CONCLUSION

In conclusion, we have systematically studied the PtX_2/NbX_2 lateral and vertical heterostructures by Raman spectra, XPS, STEM, STM/STS, and DFT calculations. Both pristine semiconducting $PtSe_2$ and $PtTe_2$ are metallized when vertically interfacing with $NbSe_2$ and $NbTe_2$. The metallization of $PtSe_2$ in $1L-PtSe_2\bot1L-NbX_2$ and $PtTe_2$ in $1L-PtTe_2\bot1L-NbTe_2$ results from the interfacial orbital hybridization, while for $PtTe_2$ in $1L-PtTe_2\bot1L-NbSe_2$ it is due to the p-doping effect caused by interfacial charge transfer. This work presents an effective strategy for metallizing NTMDs and provides deep insight into the metallization at an atomic level. Our discoveries could

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have significant practical applications such as reducing contact resistance at metal electrode/2D semiconductor interfaces and developing devices like 2D Schottky diodes-based rectifiers, rectennas, photodetectors, and so forth, which will be the focus of our future research efforts.

EXPERIMENTAL SECTION/ METHODS

4.1 Sample preparation

NbX₂ and PtX₂ were grown on HOPG substrates by MBE in an ultrahigh vacuum (UHV) chamber with a base pressure of $\sim 2 \times 10^{-9}$ mbar. Before growth, the freshly cleaved substrate was annealed at 500°C for 2 h. Se and Te (Sigma, 99,99%) were evaporated from Knudsen cells at 150 and 340°C, respectively, while Nb and Pt (ESPI Metals, 99.99%) were evaporated from electron-beam evaporators with their evaporation rates controlled by the flux currents. During growth, the substrate was kept at 280°C, a temperature higher than the sublimation temperatures of Se and Te, and the whole growth process was maintained under Se- or Te-rich conditions to ensure sufficient Se or Te reacting with Nb or Pt. After growth, the sample was cooled to room temperature and capped with Se or Te for ex situ transfer to another UHV system for subsequent STM/STS characterizations. Prior to these measurements, the Se- or Te-capping layer was desorbed at 250°C for 2 h in the preparation chamber of the respective system.

4.2 STM/STS characterizations

STM/STS measurements were carried out in a multichamber UHV system housing an Omicron LT-STM system interfaced to a Nanonis controller at 77 K. The base pressure was better than 10^{-10} mbar. A chemically etched tungsten tip was used, and the bias was applied to the samples. STM images were recorded in constant-current mode.

4.3 **STEM characterizations**

The MBE-grown samples were directly lifted from the substrate using a propylene carbonate (PC) film and subsequently transferred onto a TEM grid after being dissolved in dichloromethane. Atomic-resolution STEM-HAADF imaging was conducted at 80 kV using the aberrationcorrected JEOL ARM200F equipped with a cold fieldemission gun and ASCOR corrector, with a probe convergence semiangle of approximately 30 mrad.

4.4 Raman characterization

Raman spectra were captured using commercial WITec Alpha 300 R Raman and NT-MDT NTEGR systems. A continuous wave (CW) laser wavelength of 633 nm was used as the excitation source with a spot diameter of $\sim 1 \mu m$ using a 100× objective lens (Olympus). For the Raman measurement, a laser power of 0.65 mW with an integration time of 1 s and accumulation times of 2 was used.

4.5 **DFT** calculations

All DFT calculations were performed using the Vienna ab initio simulation package (VASP).^{58,59} The projector augmented-wave (PAW) method and the optB88-vdW functional were adopted.60-63 A planewave basis with a kinetic energy cutoff of 450 eV expanded the electronic wavefunction. A Γ -centered k-mesh sampled the first Brillouin zone of 1T-PtX₂, 1H-NbX₂, and their heterostructures in such a way that the product of the number of subdivisions with the respective lattice constant is close to 60, while it was set to 1 in the out-of-plane direction. To alleviate the spurious interaction between periodic images in the out-of-plane direction, a vacuum layer of $\sim 20 \,\text{Å}$ was inserted along that direction. The energy and force convergence criteria were set to 1.0×10^{-5} eV and 0.01 eV/Å, respectively. Anderson's rule, where the vacuum level is the common energy reference, was used to align the band edges of 1T-PtX2 with the Fermi level of metallic 1H-NbX2 (Figure S3). Meanwhile, explicit 1T-PtX₂±1H-NbX₂ heterostructures were also simulated with the respective supercell models summarized in Table S2. The average in-plane lattice constant was adopted to minimize the strain experienced by both 1T-PtX2 and 1H-NbX2 due to the lattice mismatch. Potential stacking configurations of the 1T-PtX₂\(\pm\)1H-NbX₂ heterostructure were considered (Figures S4-S6). Here, the AA, AB and AC stacking of 1T-PtX₂\(\pm\)1H-NbX₂ indicates that the bottom-layer X, the middle-layer Pt, and the top-layer X of the top 1T-PtX2 is aligned with the top-layer X of the bottom 1H-NbX2, respectively. The most stable configuration for each 1T-PtX₂\pm1H-NbX₂ heterostructure was adopted to further perform the electronic structure calculations (Figure 10). The interlayer binding energy of the vertical heterostructure 1T-PtX₂±1Has $E_b(1T - PtX_2 \perp 1H - NbY_2) =$ NbX₂ is defined $\scriptstyle E(1\mathrm{T}-\underline{\mathrm{PtX}_2}\perp 1\mathrm{H}-\mathrm{NbY}_2)-E(1\mathrm{T}-\mathrm{PtX}_2)-E(1\mathrm{H}-\mathrm{NbY}_2)$ $E(1T-PtX_2 \perp 1H-NbY_2)$, $E(1T-PtX_2)$, and $E(1H-NbY_2)$ are the total energy of the vertical heterostructure 1T-PtX₂⊥1H-NbX₂, and monolayer 1T-PtX₂ and 1H-NbX₂,

respectively. A stands for the interfacial area.



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CONFLICT OF INTEREST STATEMENT

The authors declare no conflict of interest.

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SUPPORTING INFORMATION

Additional supporting information can be found online in the Supporting Information section at the end of this article.

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