



# Tetrathienylethene-based porous framework composites for boosting photocatalytic antibacterial activity

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In order to reduce the risk of high-threat pathogens, a photocatalytic antibacterial method with a reputation for high efficiency and sustainability has attracted widespread attention. Recently, metal-organic frameworks (MOFs) have emerged as desirable platforms for photocatalytic applications by virtue of their structural diversity and functional adjustability. Herein, we report that we have synthesized a stable and photosensitive zirconium-based MOF (Zr-MOF) with a photoactive tetrathienylethene-based organic linker, Zr-TSS-1. Compared with all-carbocyclic Zr-MOF counterparts, Zr-TSS-1 shows a substantial improvement in visible-light harvesting and free-carrier generation, enabling it to be a promising candidate for photocatalytic antibacterial applications. In order to validate the advantages of this framework as an antibacterial protective material, a composite was fabricated by incorporating robust Zr-TSS-1 onto sustainably accessible bacterial cellulose (BC) using an in situ growth method. This composite exhibits near-complete lethality toward typical Gram-negative *Escherichia coli* and Gram-positive *Staphylococcus aureus* within 1 h under mild irradiation and preserves outstanding antibacterial capability after five cycles of reutilization. In addition, the high biocompatibility is confirmed by the low cytotoxicity toward human skin fibroblast, suggesting its potential for biomedical and healthcare applications. This research demonstrates the efficacious integration of a purposely designed photosensitive porous framework onto a sustainable substrate for synergistic functionality, paving a practical way for the development of the next-generation high-efficiency antimicrobial technology.

metal-organic frameworks | reticular chemistry | photocatalytic antibacterial method

Infectious pathogens like bacteria and viruses have posed a huge risk to the health of humankind for a long time. The latest case—the sudden outbreak of COVID-19—caught people off guard and posed a serious threat to public security and economic progress (1, 2). Moreover, it has been reported that many patients, infected clinically with COVID-19, suffered from secondary bacterial infections and attack from drug-resistant bacteria (3). As a result, efficient inactivation of infectious pathogens and inhibition of their secondary or even multiple transmissions are in high demand (4). The photocatalytic antimicrobial method is considered an attractive solution to lessen the menace of high-risk pathogens (1, 5). The efficient inactivation of microbes is achieved by photoinduced reactive oxygen species (ROS) with good oxidative capacity and little side effects. Also, the photocatalysts play a pivotal role in the generation of ROS, which determines the antibacterial efficiency during photocatalysis (1, 5). Therefore, it is of importance to design high-performance and biologically compatible materials for photocatalytic sterilization.

Metal-organic frameworks (MOFs), composed of organic and inorganic building blocks, have recently attracted much attention in various applications owing to their controllable structures and adjustable functionalities, including gas storage and separation, sensing and catalysis, as well as biomedical applications (6–11). Zirconium-based MOFs (Zr-MOFs) are one of the most explored MOFs owing to their high thermal, chemical, and mechanical stabilities (12). Recently, Zr-MOFs have been applied extensively in the photocatalysis field thanks to their tunable structures (13–16). Organic molecules containing multiple thiophene units are employed commonly in the design of efficient perovskite solar cells and fluorescent sensors (17–19). It is worth noting that photosensitive Zr-MOFs, constructed with ligands containing functional tetrathienylethene units, have yet to be explored (20, 21). In addition, the realization of functional composites for practical antibacterial applications requires the successful integration of photosensitive materials onto suitable substrates (22–24). In some previous studies, light-driven protective composites have been prepared by incorporating typical ZIF-8 and CAU-1 onto fabrics (25, 26). Another photocatalytic antibacterial composite has been prepared by embedding microcrystalline NU-1012 onto cotton cloths (27). It remains a challenge, however, to fabricate desirable MOF-based composites based on

## Significance

Infectious pathogens, such as viruses and bacteria, pose a huge threat to human daily life, photocatalytic antibacterial method has been considered as an effective solution. Metal-organic frameworks (MOFs), known as attractive photocatalysts, are desirable candidates for photocatalytic antibacterial applications. Herein, we have designed and prepared a photoactive and stable MOF-based composite by introducing photosensitive ligand with tetrathienylethene unit. Compared with all-carbocyclic MOF counterparts, the tetrathienylethene-based MOF exhibits stronger visible-light harvesting and quicker production of abundant free carriers. Additionally, the biosafe composite shows near-complete bacterial-killing activity against *Staphylococcus aureus* and *Escherichia coli* within 1 h. This investigation provides a good proof-of-principle for constructing a biocompatible antibacterial composite and offers a direction to shape the next era of antibacterial arsenals.

The authors declare no competing interest.

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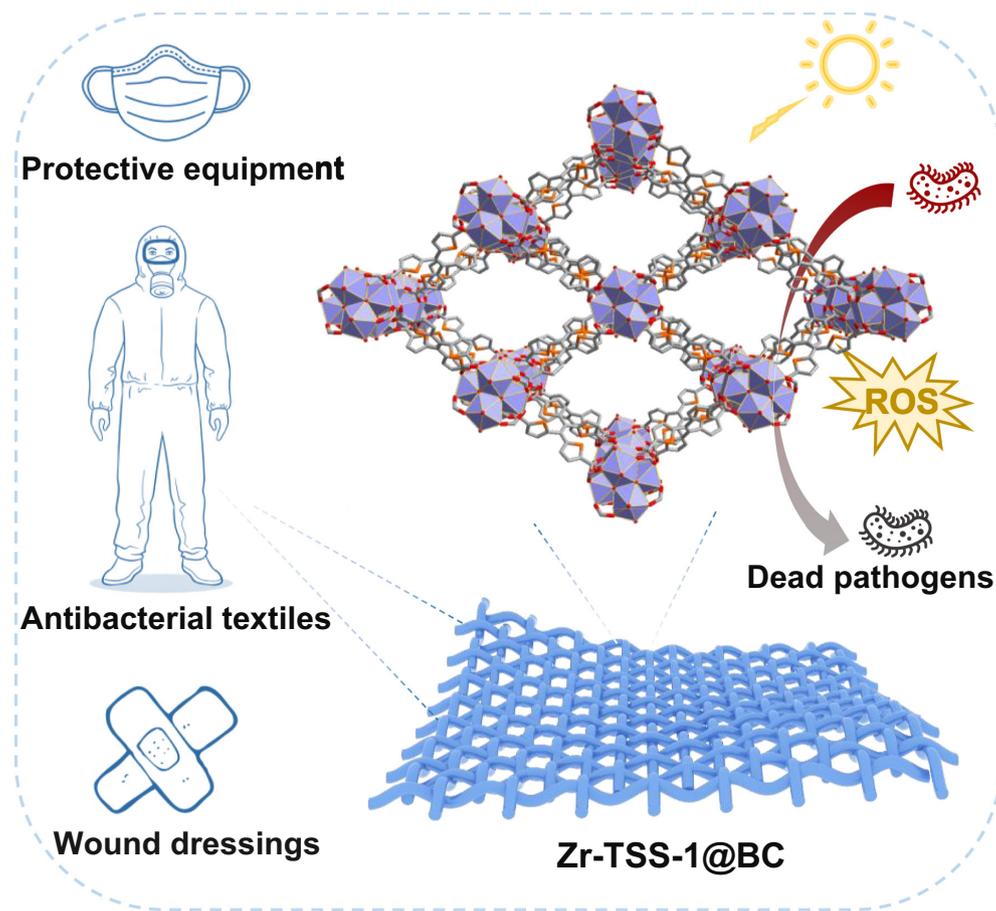
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**Scheme 1.** The schematic diagram of antibacterial fibers containing Zr-TSS-1 with promising practical applications.

molecular-level design (1, 28, 29). Moreover, the bioavailability of MOF-based composites, which has been less explored, is significant when it comes to their practical adoption in healthcare-related applications (1, 29).

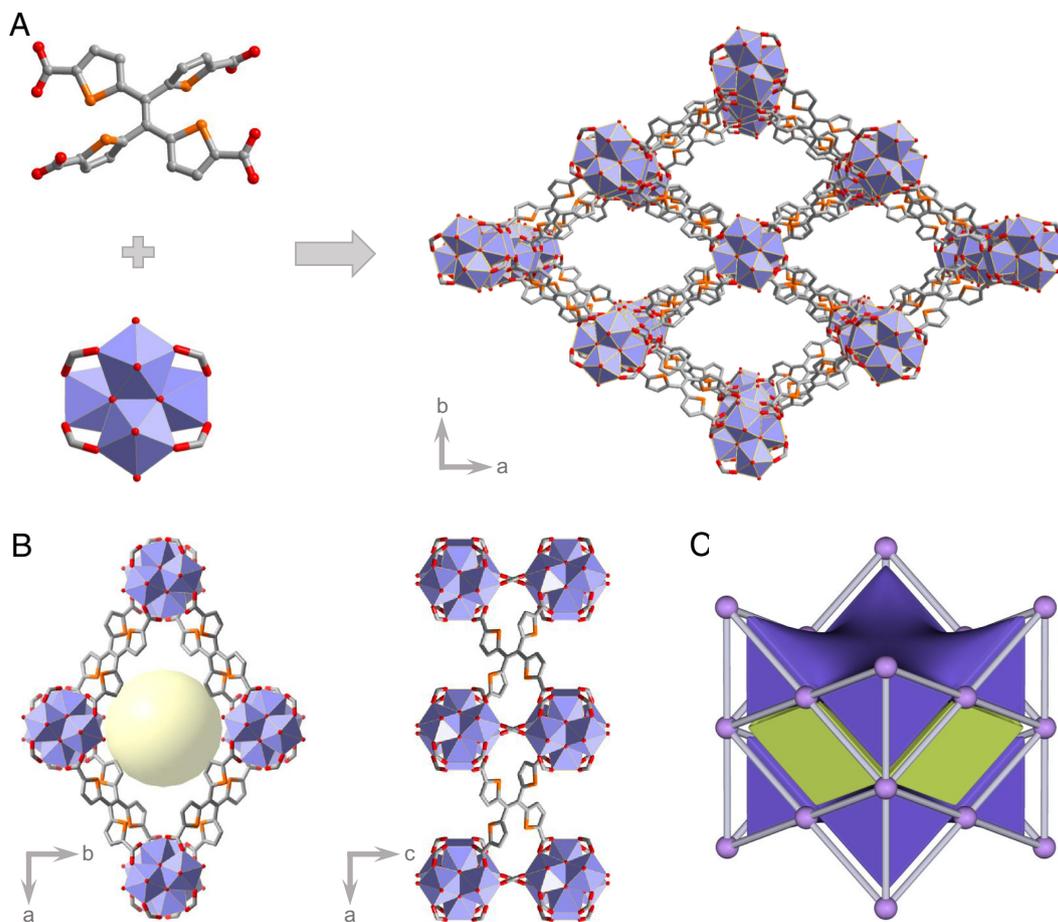
In this context, we have designed and constructed a stable and photoactive Zr-MOF—Zr-TSS-1—by introducing a photoresponsive tetrathienylethene-based linker [TSS-1 = 5,5',5'',5'''-(ethene-1,1,2,2-tetrayl)tetrakis(thiophene-2-carboxylic acid)]. In comparison to the representative all-carbocyclic MOFs with analogous topological networks, like Zr-TCPE and NU-903 [TCPE = 1,1,2,2-tetra(4-carboxyl-phenyl)ethylene, NU = Northwestern University], the as-prepared Zr-TSS-1 with multiple functional thiophene units exhibits stronger visible-light capture and quicker carrier separation and transport, implying excellent photocatalytic applicability. In order to solve the inconvenience of employing powdery MOFs in practical applications, we embedded microcrystalline Zr-TSS-1 onto porous bacterial cellulose (BC) in order to fabricate (Scheme 1) a photoactive composite—Zr-TSS-1@BC. In line with our expectations, the Zr-TSS-1@BC displays outstanding inactivation to *Escherichia coli* and *Staphylococcus aureus*. In addition, insignificant changes in antibacterial activity of the composite were found after consecutive reuse. More importantly, this composite causes negligible damage to living mammalian cells in cytotoxicity assays, confirming its suitability for the manufacture of protective equipment, antibacterial textiles, and wound dressings.

## Results and Discussion

**Synthesis and Structural Analysis.** In contrast to a plethora of TPE-based MOFs (TPE = tetraphenylethene) (30), tetrathienylethene (TTE) with a similar structure to classic TPE (31) has yet to be

introduced into MOF-based materials, potentially because of the challenging synthetic procedure. In addition, TTE with its four photosensitive thiophene units is ideal for the use of visible light, making it a promising building block for the construction of porous MOF-based photocatalysts (32, 33). Herein, we have designed a carboxylate-based organic ligand (TSS-1) using TTE as the core unit. It was prepared by the reaction of CO<sub>2</sub> with lithiated TTE under anhydrous and anaerobic conditions. Next, we decided to construct a stable Zr-MOF according to the principles of reticular chemistry and the theory of Hard–Soft Acid–Base (6, 13) (HSAB). A solvothermal reaction of TSS-1 and ZrCl<sub>4</sub> in *N,N*-dimethylformamide with formic acid as a modulator affords (Fig. 1*A* and *SI Appendix, Fig. S2*) yellowish single crystals of Zr-TSS-1 under optimal conditions (*SI Appendix, Table S1*). Single-crystal X-ray diffraction analyses reveal that Zr-TSS-1 crystallizes in the orthorhombic space group *Ibam* with the lattice parameters  $a = 28.2838(6)$ ,  $b = 18.9646(4)$ ,  $c = 19.9599(4)$  Å (*SI Appendix, Table S2*) at 193 K. Each TSS-1 is connected to four Zr<sub>6</sub> clusters, while each cluster is coordinated with eight carboxylates from eight different linkers. Adjacent Zr<sub>6</sub> clusters are reinforced by two in situ inserted formates (FAs), forming a robust network (Fig. 1*A* and *SI Appendix, Fig. S3–S5*) similar to that in the reported (34–36) NU-901-NDC, PCN-224-BPDC, and MTV-MOFs, synthesized by postmodification. The unsaturated sites in the Zr<sub>6</sub> clusters are occupied by four FA molecules. We identified the formula of Zr-TSS-1 to be [Zr<sub>6</sub>O<sub>8</sub>(FA)<sub>6</sub>(TSS-1)<sub>2</sub>], which is also supported by <sup>1</sup>H NMR spectroscopy (*SI Appendix, Fig. S6*) carried on digested Zr-TSS-1. In addition, the structure contains (Fig. 1*B*) two pockets, one of which is a vacant diamond-shaped channel viewed along the *c*-axis, while the other is segmented by FAs viewed in the direction of the *b*-axis.

Topologically, if the FAs bridging adjacent Zr<sub>6</sub> clusters are omitted, the TSS-1 linkers can be simplified as square blocks which link



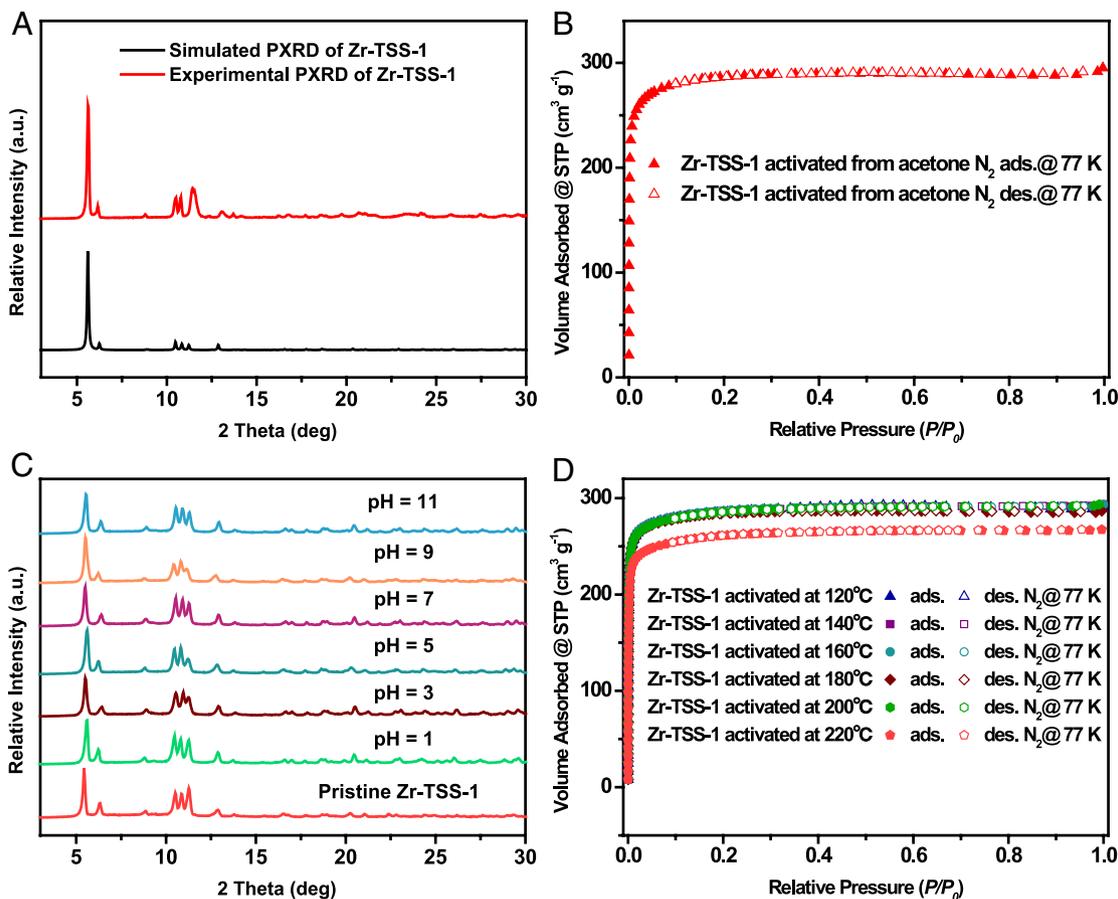
**Fig. 1.** Structural illustration of Zr-TSS-1. (A) The construction of Zr-TSS-1 from Zr<sub>6</sub> cluster and TSS-1 ligand. The single net of Zr-TSS-1 viewed along the c-axis. (B) The illustration of pockets in Zr-TSS-1 viewed along the c- and b-axis. (C) The natural tiling of 3D structure in Zr-TSS-1. Zr light purple, C gray, O red, and S orange. H atoms are omitted for the sake of clarity.

to the clusters to form a (4, 8)-connected **scu** net (37) (*SI Appendix, Fig. S4*). Alternatively, the underlying network of Zr-TSS-1 can be analyzed as a (4, 10)-connected **sqc246** net, if the Zr<sub>6</sub> clusters are simplified as 10-connected nodes while the FAs can be considered as linkers (38) (*SI Appendix, Fig. S7*). In addition, the three-dimensional (3D) structure of Zr-TSS-1, visualized intuitively by 3dt software (39), contains two types of tiles with a transitivity of [2332] in the **sqc246** net (Fig. 1C).

**Characterization and Chemical Stability.** The phase purity of bulk samples of Zr-TSS-1 was validated (Fig. 2A and *SI Appendix, Fig. S8*) by a high congruence of simulated and experimental powder X-ray diffraction (PXRD) patterns. Scanning electron microscope (SEM) images reveal the tablet-like morphology of the microcrystalline Zr-TSS-1 (*SI Appendix, Figs. S9 and S10*). SEM equipped with energy dispersive X-ray spectroscopy (EDX) system was used to determine the distribution of C, O, Zr, and S element in Zr-TSS-1 (*SI Appendix, Fig. S11*). Additionally, the X-ray photoelectron spectroscopy (XPS) illustrates the presence and chemical state of C, O, Zr, and S in Zr-TSS-1 (*SI Appendix, Fig. S12*). The high-resolution XPS of Zr 3d can be deconvoluted into two peaks for Zr 3d<sub>5/2</sub> and Zr 3d<sub>3/2</sub> at 182.6 and 184.9 eV, respectively (*SI Appendix, Fig. S12C*). The high-resolution XPS of C 1s, S 2p, and O 1s were identified to prove the structure of Zr-TSS-1 (*SI Appendix, Fig. S12*). Moreover, the permanent porosity is evaluated by N<sub>2</sub> adsorption–desorption isotherms at 77 K, confirming the microporous nature of Zr-TSS-1 (37) (Fig. 2B). The apparent Brunauer–Emmett–Teller surface area was calculated

to be 950 m<sup>2</sup> g<sup>-1</sup>, and the experimental total pore volume was estimated to be 0.45 cm<sup>3</sup> g<sup>-1</sup> at  $P/P_0 = 0.95$ , matching the theoretical pore volume of 0.46 cm<sup>3</sup> g<sup>-1</sup>. In addition, we have assessed the chemical stability and thermal resistance, as these properties are important for the photocatalytic applications of Zr-TSS-1. The PXRD patterns of Zr-TSS-1 are almost unchanged (Fig. 2C) after exposure to aqueous solutions with pH values ranging from 1 to 11. The framework, morphology, and particle size of Zr-TSS-1 after treatment were evaluated by N<sub>2</sub> sorption, PXRD, and SEM characterizations (*SI Appendix, Figs. S13–S15*). These results indicate that the crystalline framework and micromorphology of Zr-TSS-1 can generally be retained at different pH, implying that Zr-TSS-1 has good acid and base resistance (40). Furthermore, the crystalline structure and porous framework are retained after activation from water (*SI Appendix, Figs. S16 and S17*), suggesting its high hydrolytic stability (41). Additionally, the thermal stability of Zr-TSS-1 exceeds 200 °C according to N<sub>2</sub> adsorption isotherms activated at different temperatures (Fig. 2D). The thermogravimetric analysis of Zr-TSS-1 suggests that its decomposition temperature under air flow is about 300 °C (*SI Appendix, Fig. S18*). These results show that Zr-TSS-1 is a stable material for further manufacture and practical use.

**Photoelectric Properties and Band Structures.** We investigated the photoelectric properties of Zr-TSS-1, which are critical for the determination of photocatalytic capability (42, 43). For comparison, we also prepared (*SI Appendix, Fig. S19*) two typical all-carbocyclic Zr-MOFs (Zr-TCPE and NU-903) with



**Fig. 2.** (A) Experimental and simulated PXRD patterns of Zr-TSS-1. (B) Nitrogen sorption isotherm profile of Zr-TSS-1. (C) PXRD patterns of Zr-TSS-1 after exposure to aqueous solutions with pH values ranging from 1 to 11. (D) N<sub>2</sub> sorption isotherms of Zr-TSS-1 activated at different temperatures.

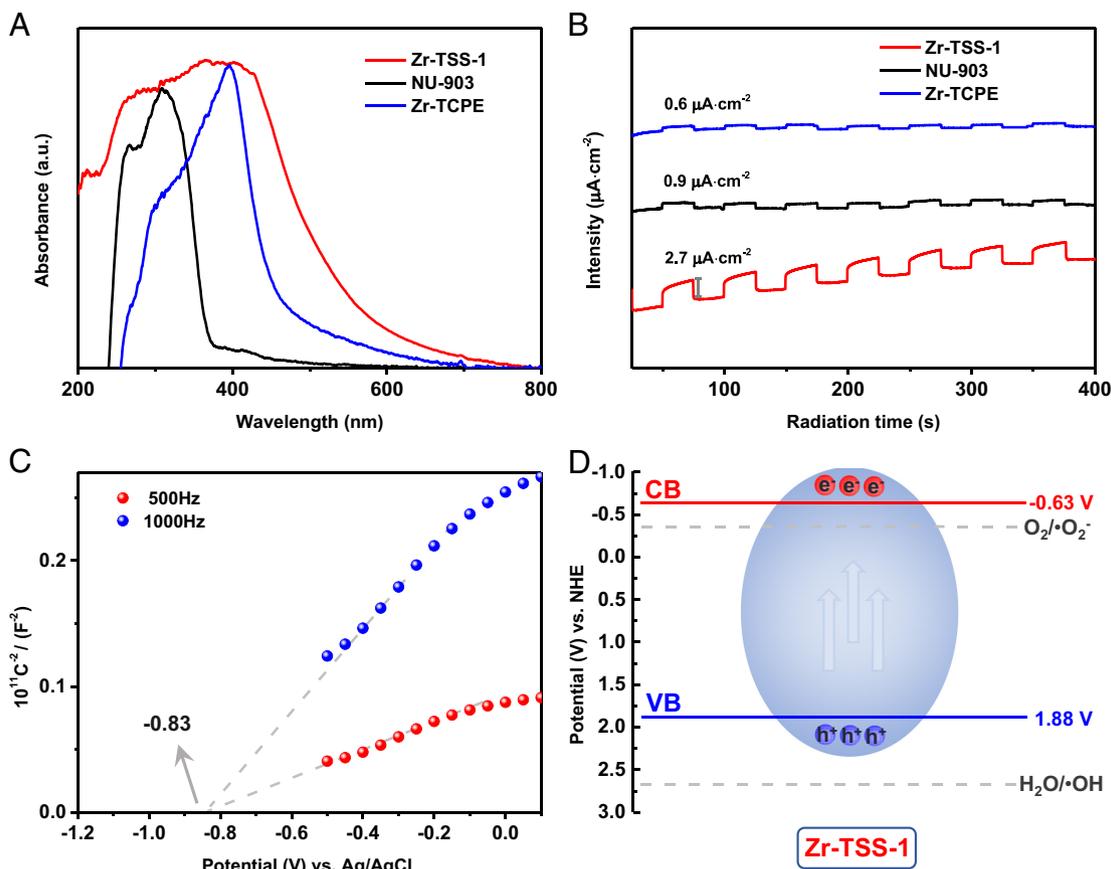
analogous topologies to Zr-TSS-1 according to the literature (44, 45). The PXRD patterns (*SI Appendix, Fig. S20*) of as-prepared microcrystalline powders are well matched with those of the simulated structures. UV/Vis diffuse reflectance spectroscopic measurements of the as-prepared samples indicate that Zr-TCPE and NU-903 exhibit (420 nm) mainly narrow optical absorbance within the blue edge of the visible light spectrum. In contrast, a broader absorbance extending to about 500 nm was acquired (Fig. 3A) by Zr-TSS-1 through the introduction of tetrathienylethene. This observation suggests that higher visible-light utilization and more photogenerated exciton formation can be achieved (46, 47). Besides, the photocurrent density of Zr-TSS-1 was measured to be about three- to four-fold higher than those of Zr-TCPE and NU-903 (Fig. 3B), implying more efficient photoinduced charge separation and transfer for photocatalysis (48, 49). Additionally, the photocurrent density of Zr-TSS-1 also surpassed that of its ligand—TSS-1 (*SI Appendix, Fig. S21*), suggesting the advantage of porous framework for high-efficiency charge transfer (50). To this end, the findings are aligned with our design concept, indicating that this Zr-TSS-1 with a strong visible-light response and good photoelectric properties is a promising photocatalyst.

Furthermore, the optical energy gap ( $E_g$ ) of Zr-TSS-1 is determined to be 2.51 eV from a Tauc plot (*SI Appendix, Fig. S22*). To confirm the band energy structure, Mott–Schottky (M–S) plots were employed to estimate the energy level of the conduction band minimum (51) ( $E_{CB}$ ). Based on the M–S plots (Fig. 3C), the  $E_{CB}$  of Zr-TSS-1 is about  $-0.83$  V vs. Ag/AgCl, corresponding to  $-0.63$  V vs. NHE (Normal Hydrogen Electrode). Thus, the energy level ( $E_{VB}$ ) of the valence band maximum of Zr-TSS-1 is calculated

(Fig. 3D) to be 1.88 V vs. NHE according to the equation (42) of  $E_g = E_{VB} - E_{CB}$ . With respect to the energy level necessary for generating common ROS-like  $\bullet O_2^-$  and  $\bullet OH$ , the conduction band (CB) of Zr-TSS-1 meets the required potential for producing  $\bullet O_2^-$ . The valence band (VB) of Zr-TSS-1, however, is unfavorable for the generation of  $\bullet OH$ . The findings imply (Fig. 3D) that Zr-TSS-1 possesses sufficient driving force for some type of photoinduced ROS generation, suggesting its potential as a photocatalytic antimicrobial agent (1). For the purpose of comparing the ability of Zr-TSS-1 to produce ROS with those of two all-carbocyclic Zr-MOFs, nitro-blue tetrazolium (NBT) and furfuryl alcohol (FFA) were selected to indicate the  $^1O_2$  and  $\bullet O_2^-$  generation kinetics (*SI Appendix, Fig. S23*), respectively. The results suggest that Zr-TSS-1 has the highest  $^1O_2$  and  $\bullet O_2^-$  generation rate, implying its high antibacterial activity as we can predict (52, 53).

**Preparation and Characterization of the Composite.** We have fabricated protective materials by integrating photosensitive Zr-MOFs onto sustainable substrates, yielding excellent light-driven self-cleaning effects. Among various substrates, BC is well known as a porous, biocompatible, and biodegradable natural polymer (54). Its 3D network possesses acceptable surface areas with hydroxyl groups, which are beneficial for water retention and filler loading. Its hydrophilic characteristics contribute to its high wettability, and thus ensuring wear comfort (54).

BC, owing to its exceptional porousness, biodegradation, biocompatibility, flexibility, rigidity, and durability (55–57), is considered as a support for embedding microcrystals to produce functional materials. Next, we incorporated microcrystalline Zr-TSS-1 onto the porous BC (Zr-TSS-1@BC) by adopting an



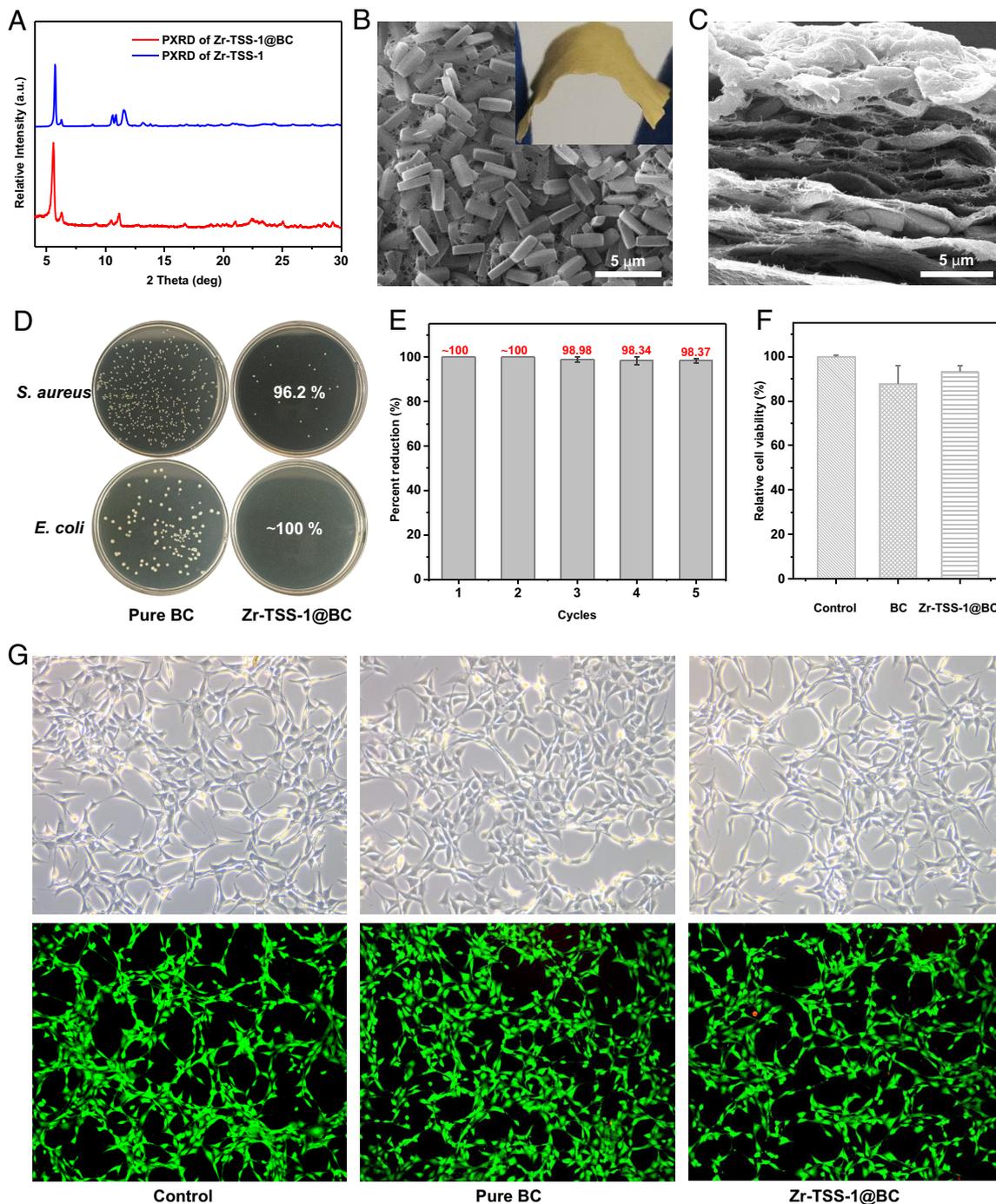
**Fig. 3.** (A) UV/Vis diffuse reflectance spectra of Zr-TSS-1, Zr-TCPE, and NU-903. (B) Photocurrent generated from Zr-TSS-1, Zr-TCPE, and NU-903 under intermittent irradiation. (C) Mott-Schottky plots for Zr-TSS-1. (D) The band structure of Zr-TSS-1 with respect to the driving force of ROS.

in situ growth method (27), suggested by the well-matched PXRD pattern (Fig. 4A) to that of Zr-TSS-1 microcrystals. SEM images taken from the surface and cross-section of Zr-TSS-1@BC show (Fig. 4B and C and *SI Appendix*, Fig. S24) that the microcrystals are distributed regularly in the 3D microstructure, which is distinct from that of pristine BC (*SI Appendix*, Fig. S25). Elemental mappings using SEM-EDX reveal the uniform distribution of Zr-TSS-1 on the BC substrate (*SI Appendix*, Fig. S26). Overall, the microfibrils and porous network of BC can lessen the agglomeration of microcrystals.

Additionally, the mechanical properties of BC and Zr-TSS-1@BC were assessed at nanoscale by atomic force microscope (*SI Appendix*, Fig. S27). The findings demonstrate that the Young's modulus of the composite is  $\sim 13.5$  GPa while that of pure BC is about 9.5 GPa, suggesting the brittleness slightly increases after loading of Zr-TSS-1 onto BC (58, 59). The composite shows good mechanical strength (Fig. 4B, *Inset*) potentially on account of the supramolecular and interfacial interactions (4, 22) between Zr-TSS-1 and BC. Concomitantly, the color change from white to yellow was observed visually (*SI Appendix*, Fig. S28). The water contact angle of BC materials was measured at the moment of applying a water droplet onto the surfaces of specimens (*SI Appendix*, Fig. S29). BC retains good surface wettability after incorporating Zr-TSS-1, ensuring wear comfort (54). The smaller contact angle also implies better contact between bacterial suspensions and composites (60). The  $N_2$  sorption isotherms of Zr-TSS-1@BC suggest that Zr-TSS-1@BC has a similar microporosity as powdery Zr-TSS-1 (*SI Appendix*, Fig. S30). The adsorbed volume closely relates to the loading of Zr-TSS-1 on the composites, and Zr-TSS-1@BC has a mass loading of about 57.3 wt% deduced from the results measured by inductively coupled plasma-optical emission spectroscopy (27).

These results suggest that the porous BC structure allows good distribution of powdery Zr-TSS-1, and it also has no obvious negative impact on the microcrystals.

**Antibacterial Activity and Biosafety Assessment.** We have evaluated the antibacterial performance of the as-prepared composites against representative Gram-negative *E. coli* and Gram-positive *S. aureus* in the light of the modified version of AATCC 100:2019—Test Method for Antibacterial Finishes on Textile Materials (29, 61). In detail, 10  $\mu$ L of activated bacterial strains were incubated separately on pure BC and Zr-TSS-1@BC under visible-light irradiation with a light intensity of  $\sim 4.71$  W/m<sup>2</sup> for 1 h. This MOF-based composite exhibits (Fig. 4D) nearly 100% inactivation of *E. coli* and *S. aureus* while pure BC poses few threats to both microbes. Its antibacterial efficiency under mild irradiation is comparable to those of MOF-related materials under harsher conditions reported previously (25, 27, 62–65) (*SI Appendix*, Table S3). Moreover, the crystallinity of the composite remains unchanged after consecutive photocatalytic cycles as proved by the unchanged PXRD patterns (*SI Appendix*, Fig. S31). Encouraged by the good stability of Zr-TSS-1, five cycles of antibacterial experiments were conducted. The results show that Zr-TSS-1@BC demonstrates enduringly high antibacterial efficiency against *E. coli* for over five consecutive cycles (Fig. 4E and *SI Appendix*, Fig. S32), suggesting Zr-TSS-1 microcrystals are firmly anchored onto the porous BC. The photocatalytic efficiency of three different batches of composites to *E. coli* was evaluated. The findings show that different batches of composites have similar antibacterial activity, indicating good batch-to-batch reproducibility (*SI Appendix*, Figs. S33 and S34). Additionally, control experiments imply that Zr-TSS-1@BC induce limited damage to *E. coli* in the absence of light (*SI Appendix*, Fig. S35). In order to understand the



**Fig. 4.** (A) Comparison of experimental PXRD of Zr-TSS-1 and Zr-TSS-1@BC. (B) SEM images of Zr-TSS-1@BC from surface section, *Inset*: photograph of flexible composite with good mechanical strength. (C) SEM images of Zr-TSS-1@BC from cross-section. (D) Photographs of *Staphylococcus aureus* and *Escherichia coli* incubated with Zr-TSS-1@BC and pure BC under visible-light irradiation for 1 h (1:10000 dilution). (E) Antibacterial efficiency against *E. coli* for five consecutive cycles. (F) The relative cell viability of human skin fibroblasts after 24 h incubation. (G) Confocal micrographs: bright-field and merged fluorescent images of Calcein AM/PI-stained NIH-3T3 fibroblasts after 24 h incubation with blank, pure BC, and Zr-TSS-1@BC. (Scale bar, 100  $\mu\text{m}$ .)

photocatalytic mechanism, electron paramagnetic resonance (EPR) measurements were performed on Zr-TSS-1. The characteristic peaks of DMPO- $\cdot\text{O}_2^-$  and TEMP- $^1\text{O}_2$  were acquired under irradiation while no signals were observed under dark conditions (*SI Appendix, Fig. S36*). This finding suggests that  $\cdot\text{O}_2^-$  and  $^1\text{O}_2$  play a critical role in the light-driven antibacterial activity. We assume that  $\cdot\text{O}_2^-$  and  $^1\text{O}_2$  could stem mainly from the metal-to-ligand charge transfer (MLCT) process in Zr-TSS-1 according to the combined experimental and calculation results (*SI Appendix, Fig. S37 and S38*) (66, 67). The ROS most likely oxidizes the biomolecules of bacterial cell walls and cell membranes, inducing the death of the bacteria (28). Besides,

the porous framework of BC enables the Zr-TSS-1 to disperse on the substrate uniformly, exposing more surfaces for light-harvesting and bacterial contact. Thus, the excellent antibacterial efficiency of the composite is realized by the synergy of visible-light responsive Zr-TSS-1 and porous BC.

The biosafety of MOFs should be given careful consideration when it comes to biomedical applications in healthcare-related fields (1). Hence, we have evaluated the bioavailability of Zr-TSS-1@BC via 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay, an extensively adopted method for sensitive detection of cell growth and survival (68). The relative cell viability of human

skin and NIH 3T3 fibroblasts after 24 h incubation with the composite reveals that Zr-TSS-1 has weak cytotoxic activity toward living mammalian cells (69) (Fig. 4F and *SI Appendix*, Fig. S39 and S40). The bright-field confocal microscopic images of human skin and NIH 3T3 fibroblasts also confirm visually that the cell morphologies have suffered no obvious damage after exposure to the composite (Fig. 4G and *SI Appendix*, Fig. S41). Moreover, the merged fluorescent images of Calcein AM/PI-stained human skin and NIH 3T3 fibroblasts among experimental and control groups suggest that little harm is inflicted by the composites on living cells owing to its low cytotoxicity (70, 71) (Fig. 4G and *SI Appendix*, Fig. S42). Overall, the Zr-TSS-1@BC displays good biocompatibility and is a desirable photocatalytic antibacterial platform.

## Conclusions

An advanced photosensitive composite—Zr-TSS-1@BC—has been designed and prepared by incorporating a stable Zr-MOF to sustainable and highly durable BC as a result of in situ growth. The as-prepared composite achieves an exceptional antibacterial efficiency against *E. coli* and *S. aureus* on account of photogenerated  $\bullet\text{O}_2^-$  and  $^1\text{O}_2$ , which can be attributed to the strong visible-light response and excellent photoelectric properties of Zr-TSS-1 as well as to its regular distribution on porous BC. Thanks to the good chemical stability and high thermal tolerance of Zr-TSS-1, this composite displays a long-lasting and effective bacterial killing serviceability in multiple reuse scenario. Despite this remarkable efficiency, cytotoxicity assessment reveals that the composite demonstrates hypotoxicity with acceptable biocompatibility. These findings prove that Zr-TSS-1@BC has a bright future for practical applications including protective equipment, antibacterial textiles, and wound dressings. Our investigation provides an example of designing a bioavailable composite at the molecular level, and offers a viewpoint to shape the next era of antibacterial arsenals.

## Materials and Methods

**Chemicals and Materials.** All reagents were obtained from commercial sources and used without further purification or treatment, unless otherwise noted. BC hydrogel was prepared from commercially available Nata de Coco (Kara Santan Pertama, Bogor 16964, Indonesia). 1,1,2,2-tetra(4-carboxylphenyl) ethylene (TCPE) and 1,2,4,5-tetrakis(4-carboxylphenyl)benzene were obtained from Shanghai Bide Pharmaceutical Technology Co., Ltd. Mouse fibroblasts (iCell-m041) were derived from mouse embryos and human fibroblasts (iCell-0051a) were derived from human dermal tissue, both of which were purchased from iCell Bioscience Inc, Shanghai.

**Characterization.** General characterization procedures are provided in *SI Appendix*.

**The Synthesis of Single Crystals of Zr-TSS-1.** TSS-1 (2 mg) and  $\text{ZrCl}_4$  (4 mg) were added into a 15 mL vial, followed with 2 mL *N,N*-dimethylformamide (DMF) and formic acid (0.6 mL). Then, the mixture was sonicated for 2 mins, heated up to 130 °C, and maintained for 20 h. After returning to room temperature, yellow single crystals were washed with fresh DMF twice for single crystal X-ray diffraction measurements.

**The Synthesis of Microcrystalline Zr-TSS-1.** TSS-1 (30 mg),  $\text{ZrOCl}_2 \cdot 8\text{H}_2\text{O}$  (60 mg), 2 mL DMF, and formic acid (0.6 mL) were added sequentially into 15 mL vial. Then, the mixture was sonicated for 2 mins, heated up to 130 °C, and maintained for 20 h. After returning to room temperature, the polycrystalline powders were collected by centrifugation and washed with fresh DMF twice. Then, the obtained samples were soaked in DMF and replaced with fresh DMF every 12 h in 2 d, further soaked in acetone and replaced with fresh acetone every 12 h in 2 d. Ultimately, the target powders were dried in a vacuum oven at 60 °C overnight and ready for subsequent characterization or performance tests.

**Synthesis of Zr-TSS-1/Bacterial Cellulose Composite (Zr-TSS-1@BC).** Under optimal conditions, BC (4 g),  $\text{ZrOCl}_2$  (60 mg, 186.19  $\mu\text{mol}$ ), and TSS-1 (30 mg, 56.33  $\mu\text{mol}$ ) were added to DMF (1 mL) in glass vial. To ensure uniform distribution of metal clusters and linkers within the BC, the mixture was first subjected to ultrasonic treatment for homogenization. Formic acid (3 mL, 65.17 mmol) was added to the mixture in a ratio of 1:0.6 (DMF: formic acid) to modulate the growth of MOF crystals. BC with precursors then reacted at 130 °C for 15 h. The product was then collected and washed with DMF and acetone several times.

**Antibacterial Efficiency Tests.** With the presence of Zr-TSS-1-loaded on BCs, we then evaluated their antibacterial performance against Gram-negative bacteria *S. aureus* (ATCC 6538) and Gram-negative *E. coli* (CMCC 44102) according to the modified version of AATCC 100:2019 – Test Method for Antibacterial Finishes on Textile Materials. The antibacterial efficiency test is conducted with three parallel groups. The activated test bacterial strains were diluted to  $\sim 1.2$  to  $4.1 \times 10^7$  colony-forming units (CFU)/mL. 10  $\mu\text{L}$  of each inoculum bacteria were added onto individual pristine BC and Zr-TSS-1@BC (1 piece, 1 cm  $\times$  1 cm), respectively. They were then irradiated under fluorescent lamp (PHILIPS TLD 18 W/865 YZ18RR25 COOLDAYLIGHT 6500 K) with a light intensity of  $\sim 4.71$  W/m<sup>2</sup> for 1 h. Immediately after light irradiation, 990  $\mu\text{L}$  of phosphate-buffered saline (PBS) was added to each specimen, and the solution was shaken vigorously for 1 min to rinse off the adhered bacteria. The solution with recovered bacteria was then serially diluted with PBS for agar plating, and the agar plates were then cultured at 37 °C for 18 h. The bacterial counts of specimens were then obtained, and the percent reduction was calculated according to the equation:  $R = 100(B - A)/B$ , where R represents the percent reduction, A is the number of viable bacteria recovered from the treated specimen over the contact period, and B is the number of viable bacteria recovered from the untreated specimen over the contact period.

**Data, Materials, and Software Availability.** X-ray crystallographic coordinates for structures reported in this study data have been deposited in Cambridge Crystallographic Data Centre (CCDC) [2328617 (Zr-TSS-1), [http://www.ccdc.ac.uk/data\\_request/cif](http://www.ccdc.ac.uk/data_request/cif)] (72). All study data are included in the article and/or *SI Appendix*.

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