

## 1 **In-situ measurements of molecular markers facilitate understanding of dynamic sources of** 2 **atmospheric organic aerosols**

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14 **Abstract:** Reducing the amount of organic aerosol (OA) is crucial to mitigation of particulate  
15 pollution in China. We present time and air-origin dependent variations of OA markers and source  
16 contributions at a regionally urban background site in South China. The continental air contained  
17 primary OA markers indicative of source categories, such as levoglucosan, fatty acids and oleic  
18 acid. Secondary OA (SOA) markers derived from isoprene and monoterpenes also exhibited higher  
19 concentrations in continental air, due to more emissions of their precursors from terrestrial  
20 ecosystems and facilitation of anthropogenic sulfate for monoterpenes SOA. The marine air and  
21 continental–marine mixed air had more abundant hydroxyl dicarboxylic acids (OHDCA), with  
22 anthropogenic unsaturated organics as potential precursors. However, OHDCA formation in  
23 continental air was likely attributable to both biogenic and anthropogenic precursors. The  
24 production efficiency of OHDCA was highest in marine air, related to the presence of sulfur  
25 dioxide and/or organic precursors in ship emissions. Regional biomass burning (BB) was identified  
26 as the largest contributor of OA in continental air, with contributions fluctuating from 8% to 74%.  
27 In contrast, anthropogenic SOA accounted for the highest fraction of OA in marine (37±4%) and  
28 mixed air (31±3%), overriding the contributions from BB. This study demonstrates the utility of

29 molecular markers for discerning OA pollution sources in the offshore marine atmosphere, where  
30 continental and marine air pollutants interact and atmospheric oxidative capacity may be enhanced.

31 **Keywords:** Secondary organic aerosols; Biomass burning; Source apportionment; Ship emission;  
32 Hydroxyl dicarboxylic acids

33

## 34 **1. Introduction**

35 Organic aerosol (OA) constitutes a considerable fraction (20-90% globally) of airborne fine  
36 particulate matters (PM) <sup>[1, 2]</sup>, and is considered to significantly account for haze occurrence <sup>[3, 4]</sup>,  
37 climatic consequences <sup>[1, 5]</sup> and impairments to health <sup>[6]</sup>. Despite successful efforts to decrease  
38 nationwide fine PM loadings in China, reducing OA has been a major obstacle for China to win  
39 the battle against air pollution <sup>[3, 7]</sup>. This is mainly due to insufficient understanding of OA,  
40 particularly the rapidly changing and spatially diverse contributions of primary (POA) and  
41 secondary (SOA) sources <sup>[8, 9]</sup>.

42 To identify sources of OA and quantify their contributions, receptor models based on known  
43 tracers of specific sources are widely adopted <sup>[3, 10]</sup>. The filter-based technique makes use of fine  
44 PM collected and accumulated on filters, which are chemically characterized following solvent  
45 extraction (SE) <sup>[11, 12]</sup> or thermal desorption (TD) <sup>[13]</sup>. Data obtained from filters are generally of  
46 low time resolution (12 – 24 hours) <sup>[11-13]</sup>. While filter-TD can substantially enhance ease of  
47 laboratory analysis, it performs poorly in analysis of polar organic compounds, unless online  
48 derivatization is utilized <sup>[13, 14]</sup>. As an alternative, Aerosol Mass Spectrometry (AMS) is extensively  
49 used in source apportionment of OA, based on the fast detection of OA at resolution of minutes <sup>[15,</sup>  
50 <sup>16]</sup>. However, the AMS-based technique usually fails to identify specific sources of OAs <sup>[15, 16]</sup>, due  
51 to the lack of compound based separation during analysis and the chemical similarity of many  
52 source markers with each other. Overall, there is a present challenge to determine the continuously  
53 changing (rather than average) contributions of specific (rather than lumped) sources to ambient  
54 OA.

55 Hong Kong in South China is an Asian metropolis, with tertiary industries (*e.g.* financial service  
56 and tourism) as the main economic pillars. Restaurants are scattered throughout every corner of  
57 the city, supplying a mix of Chinese and western foods. The city is also known for its heavy traffic,

58 *i.e.* more than 300 licensed vehicles per km of road. With a population of 7.4 million and a land  
59 area of merely  $1.1 \times 10^3 \text{ km}^2$ , the air in Hong Kong is polluted by many local sources. Additionally  
60 weather driven forces affect Hong Kong air quality with southern winds from the South China Sea  
61 (SCS) providing cleaner air to Hong Kong in the summer, while the continent-originated northern  
62 winds deteriorate air quality in winter [17-19]. The meso-scale circulations and emissions in  
63 microenvironments also modulate air quality [18-20]. These factors may explain the discrepancies in  
64 the sources of OA or source contributions reported in previous studies [21-24]. For example, cooking  
65 emissions have been determined to explain more than 30% ( $\sim 3.6 \mu\text{g m}^{-3}$ ) of OA at a roadside site  
66 in Hong Kong using AMS and Aerosol Chemical Speciation Monitoring data [21-22], which  
67 however were not resolved or identified with much lower contributions (1.2%,  $\sim 0.2 \mu\text{g m}^{-3}$ ) in  
68 many other places by filter-based source apportionment [23-24]. Hu et al. [23] argued that biomass  
69 burning (BB) contributed  $\sim 38\%$  ( $3.6 \mu\text{g m}^{-3}$ ) to OA at 4 suburban sites in Hong Kong with the  
70 intrusion of regional air, including  $2.2 \mu\text{g m}^{-3}$  of BB-related POA and  $1.4 \mu\text{g m}^{-3}$  of BB-induced  
71 SOA. In cool seasons when Hong Kong received regional air from mainland China, the BB  
72 contributions ( $1.6 \mu\text{g m}^{-3}$ ) were even comparable to those of vehicle emissions ( $1.5 \mu\text{g m}^{-3}$ ) at a  
73 suburban site [24]. Nevertheless, this finding was not supported by AMS-based source  
74 apportionment which did not identify BB as a significant source of OA in either suburban or  
75 roadside environments [15, 21-22]. Besides, the sum of SOA components resolved by filter-based  
76 approach was not always comparable in concentration to the total oxygenated organic aerosol  
77 (OOA) determined by AMS [15, 21-24]. In addition to the changing atmospheric conditions and  
78 emissions, methodological differences might be another reason for the discrepancies. Hence, to  
79 better understand the driving forces of poor air quality in Hong Kong, it is necessary to determine  
80 specific sources of OA with higher time- and space-resolution, by combining the advantages of  
81 filter-based and AMS-based source apportionment techniques.

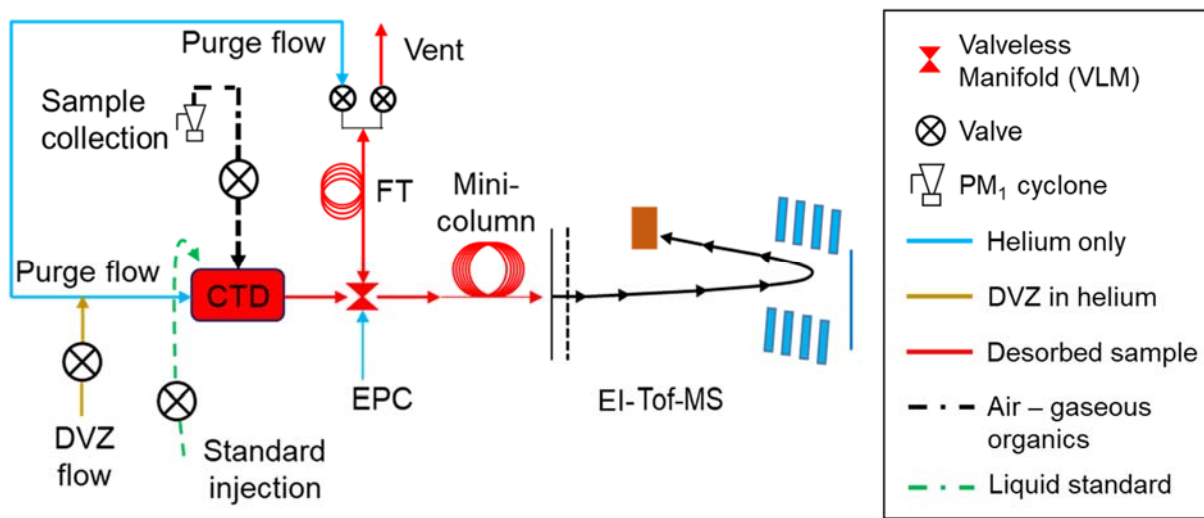
82 Thermal-desorption Aerosol Gas-chromatograph (TAG), an in-situ chemical marker oriented  
83 instrument, has been developed for time resolved analysis of organic molecular markers and is  
84 increasingly applied in OA studies [25-28]. Table S1 lists the development and representative  
85 application of different versions of TAG, including impactor TAG, 2D-TAG, SV-TAG and TAG-  
86 AMS developed by US groups [29-35], and Q-TAG, Dual-trap TAG and 2D-Q-TAG developed by  
87 a China group [36-38]. The main findings of the application are summarized in Table S1, and we  
88 also briefly commented on the performance of the instruments. Overall, TAG families have

89 contributed to comprehensive measurements of semi-volatile, intermediate-volatility and  
90 condensed organic compounds [27], study on gas/particle partitioning [25], identification of sources  
91 of airborne organics [28], and research on SOA formation mechanisms [26]. In this study, a modified  
92 impactor TAG with in-situ calibration, on-line derivatization and high-resolution time-of-flight  
93 mass spectrometry analysis, which provided in-situ measurement of both polar and non-polar OA  
94 markers every 1.5 hours, was applied to analyze the molecular composition of OA in Hong Kong.  
95 The specific sources of OA and their dynamics were determined in air masses of different origins.  
96 Intense formation of anthropogenic SOA in relatively clean air was also demonstrated. This paper  
97 presents a reliable method of tracking the sources of POA and formation processes of SOA in  
98 abrupt and transient pollution events, which should assist policy-makers in taking actions against  
99 spikes in fine PM pollution.

## 100 **2. Overview of sampling campaign**

101 In November–December 2018, a sampling campaign was carried out at a relatively remote coastal  
102 site at the southeast tip of Hong Kong (Figure S1) to study the photochemical air pollution in this  
103 highly urbanized subtropical region. Though it is located in Hong Kong, the site has long been  
104 regarded as a regionally urban background site in South China, given that air pollutants in the  
105 adjoining Pearl River Delta reached the site within a few hours [39, 40]. A Thermal desorption  
106 Aerosol Gas chromatograph – Time of Flight – Mass Spectrometer (TAG-Tof-MS, hereinafter  
107 referred to as TAG) was deployed to measure the OA markers in airborne PM<sub>1</sub> (PM with  
108 aerodynamic diameters less than 1 μm) during 13 November–19 December. PM<sub>1</sub> was focused to  
109 match the data analyzed by Aerosol Mass Spectrometer, which is introduced below. Figure 1  
110 shows a schematic of TAG. Fine PM was sampled through a PM<sub>1</sub> cyclone to eliminate coarse  
111 particles above 1 μm before collection in an inertial impactor based collection and thermal  
112 desorption (CTD) cell. Collected samples were thermally desorbed from 30 to 315 °C into helium.  
113 To improve analysis of polar compounds, silylation derivatization was performed in-situ  
114 simultaneously with desorption using helium purge flow ~80% saturated with a silyating reagent  
115 (N-Methyl-N-(trimethylsilyl)trifluoroacetamide, MSTFA) (DVZ). All the desorbed species were  
116 directed through a secondary focusing trap (FT), where the target compounds were trapped and  
117 the high volatility species including excess MSTFA were vented. Next, the FT was heated and  
118 back-flushed with helium to transfer trapped species to the miniature gas chromatograph for

119 chemical analysis. A low-polarity metal column with length of 20 m (Restek, Cat. # 71822) was  
 120 used for analysis with the column flow controlled by an electronic pressure controller (EPC). In  
 121 this campaign, the TAG was operated at the resolution of 1.5 hours per sample. The continuous  
 122 measurement was interrupted by instrument maintenance in some time slots. Polar compounds  
 123 were not detected between 22 November 12:00 and 26 November 19:00, due to running out of  
 124 MSTFA. The operation procedures of TAG are elaborated in Text S1 and Table S2.



125  
 126 Figure 1. Schematic of TAG. The term “Air – gaseous organics” means air samples with the  
 127 gaseous organics being removed by a carbon denuder (not shown) and coarse particles removed  
 128 by PM<sub>1</sub> cyclone. DVZ: derivatization; CTD: collection and thermal desorption cell; FT: focusing  
 129 trap; EPC: electronic pressure controller; EI-Tof-MS: electron impact–time of flight–mass  
 130 spectrometer.

131 Throughout the field campaign, 66 compounds with the volatilities spanning C<sub>13</sub>–C<sub>35</sub> *n*-alkanes in  
 132 478 valid samples were quantified by TAG, with the use of internal and authentic/surrogate  
 133 standards. A mix of 41 deuterated compounds was adopted as internal standards in analysis of each  
 134 sample and in calibration, to track and correct the changes in instrumental sensitivity. In this study,  
 135 we focused on 40 out of 66 quantified organic compounds, which were routinely at detectable  
 136 concentrations and indicative of OA sources. The detection rates were lower than 75% for the  
 137 other 26 compounds. Table S3 lists the average concentrations of the 40 organic compounds, with  
 138 the quantification ions, internal standards, surrogate standards, and quality control metrics. To  
 139 facilitate discussion, the 40 OA compounds were grouped according to their chemical structures,

140 potential sources and correlations. Table S4 shows the 15 species groups, including isoprene SOA  
141 tracers (Isop-SOA-T), monoterpene SOA tracers (MT-SOA-T), saccharides, hydroxyl  
142 dicarboxylic acids (OHDCA), hydroxyl benzoic acids (OHBA), even *n*-alkanes type 1 & 2 (even  
143 *n*-alk1&2), odd *n*-alkanes type 1 & 2 (odd *n*-alk1&2), even fatty acids (even FA), 2,3-dihydroxy-  
144 4-oxopentanoic acid (DHOPA), vanillic acid (VA), mannosan, levoglucosan and oleic acid. Both  
145 the even and odd *n*-alkanes were divided into two types, due to the potentially different sources.  
146 The species in each group were at least moderately correlated, with  $R^2$  higher than 0.7. A majority  
147 of the 15 groups were used as tracers in OA source apportionment, and the rationalities and  
148 uncertainties are elaborated in section 3.3.

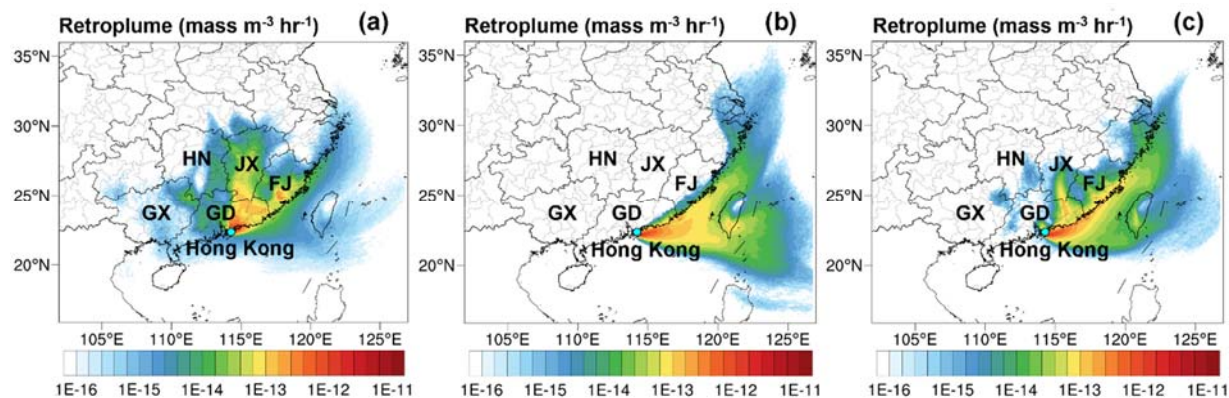
149 Additionally, a **High Resolution – Time of Flight – Aerosol Mass Spectrometer** (Aerodyne  
150 Research Inc. HR-ToF-AMS, hereinafter referred to as AMS) was applied to analyze the non-  
151 refractory composition of  $PM_{10}$ , including sulfate, nitrate, ammonium, chloride and total organic  
152 matter (OM). Text S2 illustrates the calibration and measurement configurations of the AMS. The  
153 collection efficiency of 0.73 (Text S2 and Figure S2) was applied to the concentrations of all  $PM_{10}$   
154 components. In this study, five-minute averaged W-mode data was used. It is worth noting that  
155 AMS data were not available after 7 December, due to malfunction of the chopper. Moreover,  
156 additional air pollutants and meteorological conditions were monitored by the Hong Kong  
157 Environmental Protection Department (HKEPD) and other research groups. Table S5 summarizes  
158 the instruments for these auxiliary measurements.

### 159 **3. Results and Discussion**

#### 160 **3.1 Time and air-origin dependent concentrations of OA markers**

161 Throughout the field campaign, the origins of air masses arriving at the sampling site (Hok Tsui,  
162 HT) were examined. Analysis of the daily clustered 72 hour backward trajectories (not shown)  
163 indicated that the air parcels originated from either the continent or SCS, labeled as continental  
164 and marine air, respectively. For the former, some trajectories passed over SCS before reaching  
165 HT, representing a mix of continental and marine air. Figure 2 shows the distribution of air mass  
166 concentrations for the three types of air. Continental air accounted for the largest fraction (~60%)  
167 of the sampled air, with the rest being equally constituted by marine (~20%) and mixed air (~20%).  
168 The east part of Guangdong (GD), central and southern Jiangxi (JX) and coastal area of Fujian (FJ)  
169 were the main points of origin of the continental air. While the mixed air was still partially

170 stemming from GD and JX, the back trajectories showed mainly offshore origins for marine air.  
171 The uncertainties in further analyses related to the classification of air mass origins are discussed  
172 in Text S3.



173  
174 Figure 2. Distribution of air mass concentrations (in unit of  $\text{mass m}^{-3} \text{hr}^{-1}$ ) in (a) continental, (b)  
175 marine and (c) mixed air masses within 500 m during 13 November–19 December 2018, according  
176 to the HYSPLIT Lagrangian backward particle release simulation at HT (22.209 N, 114.253 E).  
177 GD: Guangdong; GX: Guangxi; FJ: Fujian; HN: Hunan; JX: Jiangxi. Hong Kong is symbolized  
178 by the blue circle.

179 Among the quantified OA markers, OHDCA group exhibited the highest concentrations ( $266 \pm 17$   
180  $\text{ng m}^{-3}$ ), dozens to hundreds of times higher than those of the other OA markers. Malic acid and  
181 two isomers of tartaric acid were the main components of OHDCA. The concentration of malic  
182 acid (most abundant individual specie) ranged from  $9 \text{ ng m}^{-3}$  to  $517 \text{ ng m}^{-3}$ , with a mean of  
183  $119 \pm 8 \text{ ng m}^{-3}$ . These were at the same magnitude as those observed<sup>[41]</sup> at an urban site in Hong  
184 Kong in summer (mean:  $192 \text{ ng m}^{-3}$ ; maximum:  $662 \text{ ng m}^{-3}$ ) and those reported<sup>[42]</sup> at another urban  
185 site in winter (mean:  $155 \text{ ng m}^{-3}$ ; maximum:  $179 \text{ ng m}^{-3}$ ). The concentrations of two isomerized  
186 tartaric acids were  $76 \pm 5 \text{ ng m}^{-3}$  and  $67 \pm 4 \text{ ng m}^{-3}$ , respectively. Levoglucosan, as a BB tracer<sup>[43]</sup>,  
187 ranked highest in concentration ( $12 \pm 1 \text{ ng m}^{-3}$ ) among the OA markers except OHDCA. However,  
188 the average concentration of levoglucosan was lower than  $30 \text{ ng m}^{-3}$  measured at the same site in  
189 autumn of 2001<sup>[44]</sup> and  $67 \text{ ng m}^{-3}$  at a mountainous site in autumn of 2010<sup>[17]</sup>. The increasingly  
190 stringent prohibition of field fires might account for the lower concentrations of levoglucosan  
191 observed in this campaign. As listed in Table S4, the SOA tracers were of relatively low  
192 concentrations, regardless of DHOPA derived from anthropogenic precursors<sup>[45]</sup> ( $0.4 \pm 0.02 \text{ ng m}^{-3}$ )

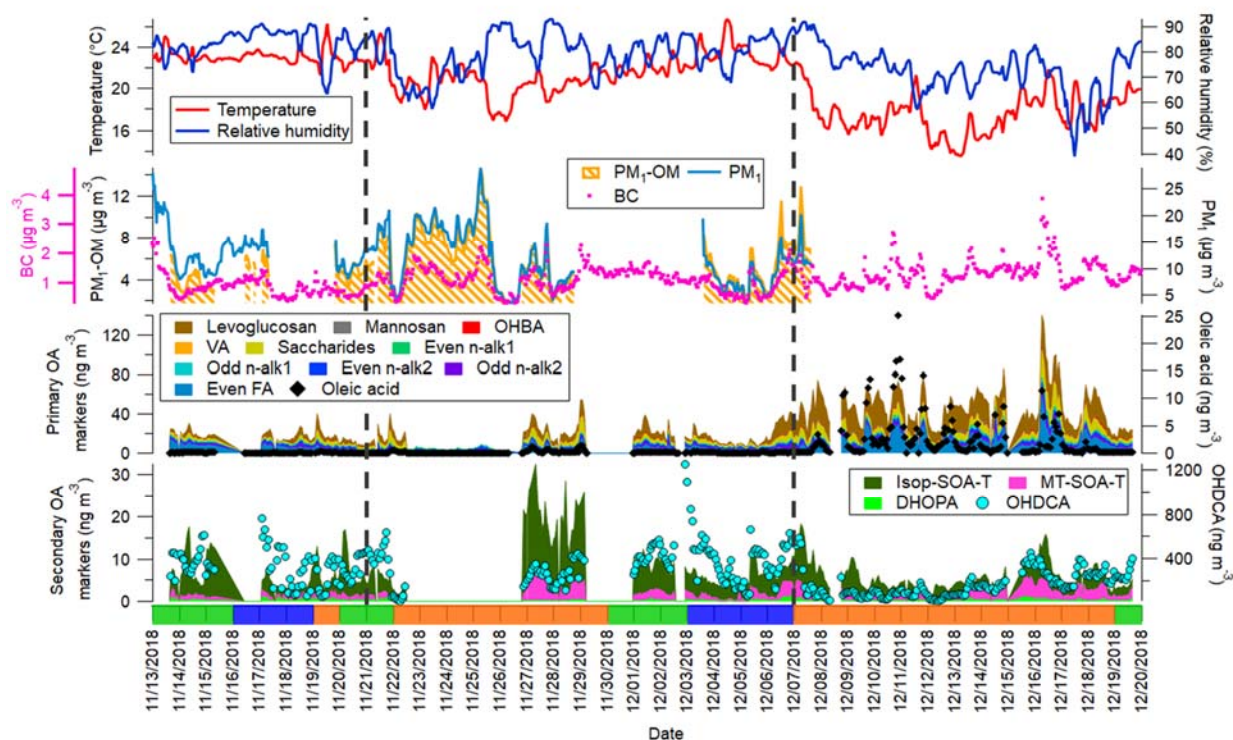
193 <sup>3</sup>) or biogenic SOA tracers (Isop-SOA-T:  $5.1 \pm 0.4 \text{ ng m}^{-3}$ ; MT-SOA-T:  $1.7 \pm 0.1 \text{ ng m}^{-3}$ ). However,  
194 this was not necessary to indicate low SOA fraction in PM<sub>1</sub>-bounded OM (PM<sub>1</sub>-OM, measured by  
195 AMS in this study), due to the incomplete analysis of SOA compounds and the unknown origin of  
196 OHDCa, as discussed later.

197 Figure 3 presents the time series of OA markers and PM<sub>1</sub>-OM, along with black carbon (BC), PM<sub>1</sub>  
198 and meteorological parameters. It is notable that OHDCa concentrations were significantly higher  
199 in the earlier phase of the sampling campaign (before 7 December), when the temperature was  
200 relatively high (mean of  $22 \pm 0.1 \text{ }^\circ\text{C}$  compared to  $17 \pm 0.2 \text{ }^\circ\text{C}$  in the later phase). The Isop-SOA-T  
201 and MT-SOA-T were also enhanced in the earlier phase. Conversely, most primary OA markers,  
202 such as levoglucosan, *n*-alkanes, fatty acids and oleic acid (cooking emission tracer <sup>[46]</sup>), were  
203 substantially built up in the later phase coincident with the dominance of continental air, indicating  
204 a rise of POA fraction in PM<sub>1</sub>-OM. In view of the concurrent spikes of BC, we infer that the POA  
205 enhancements were likely attributable to transport, associated with the passing of a cold front on  
206 7 December (Figure S3) which brought in continental air in the following days. The same  
207 phenomena were observed after another cold front passed Hong Kong on 21 November (Figure  
208 S3), with obvious enhancements of PM<sub>1</sub> and PM<sub>1</sub>-OM. Carbon preference index (CPI) and the  
209 mixing ratios of acetonitrile (gas phase BB indicator <sup>[47]</sup>) increased as well (Figure S4). Besides,  
210 no correlation with either total or individual even FA implied that the *m/z* 60 and *m/z* 73 ions  
211 detected by AMS were more related to BB in this campaign, in line with previous studies <sup>[48]</sup>. The  
212 concentrations of both ions increased notably during 22–25 November (Figure S4). All these  
213 signatures indicated the arrival of BB plumes with continental air induced by the shifting cold  
214 front. This process might also apply to the other POA components. The cross-validation between  
215 CPI (TAG data) and *m/z* 60, 73 (AMS data) during the plateau period, in combination with the  
216 quality control metrics in Table S3, confirmed the robustness of OA markers measured by TAG.  
217 Furthermore, the simultaneous increases of CPI and levoglucosan during 6–7 December (Figure  
218 S4) implied the enhanced contributions of BB to PM<sub>1</sub>-OM caused by the second cold front, though  
219 the rise in CPI was less significant due to the concurrent enhancement of vehicle emissions (section  
220 3.3).

221 Note that the correlation between levoglucosan and acetonitrile (particle and gas phase BB tracers,  
222 respectively) was undermined by high levoglucosan concentrations observed with low



223 temperatures ( $<20.4\text{ }^{\circ}\text{C}$ , median temperature), and levoglucosan inversely correlated ( $R = -0.79$ )  
 224 with temperature (Figure S5). It was plausible that the shift of gas-to-particle partitioning also  
 225 accounted for the elevated concentrations of some POA components (*e.g.* levoglucosan) in the  
 226 continental air induced by cold fronts.



227  
 228 Figure 3. Time series of meteorological conditions, black carbon (BC),  $\text{PM}_{10}$  and  $\text{PM}_{10}$  compositions  
 229 (including OA markers). Orange, green and blue bars on bottom axis denote continental, mixed  
 230 and marine air, respectively. Black dashed lines indicate dates when cold fronts passed Hong Kong.  
 231 Missing data are caused by maintenance of the instruments, except for the unavailable polar  
 232 compounds during 22 November 12:00–26 November 19:00 due to running out of MSTFA.

233 Generally, the concentrations of OA markers were closely associated with air mass type, as listed  
 234 in Table 1. An exception was isoprene, which exhibited comparable levels in all types of air masses.  
 235 This was likely attributable to the dominance of local emissions in isoprene sources regardless of  
 236 air mass origins, and/or inhibited isoprene emissions from terrestrial system in cool season. The  
 237 local emissions might also partially account for the observed mixing ratios of monoterpenes and  
 238 toluene in marine and mixed air, which were lower than those in continental air but were still of

239 significant levels. Under the assumption that local emissions were relatively consistent, the  
 240 comparisons in Table 1 still reveal different air pollution features for different air mass types.

241 The continental air was laden with mostly primary air pollutants and POA tracers, such as carbon  
 242 monoxide (CO), levoglucosan, fatty acids and oleic acid. Many other species not listed here, *e.g.*  
 243 nitrogen oxides (NO<sub>x</sub>), BC and *n*-alkanes, also exhibited higher concentrations in continental air.  
 244 BB, vehicle exhaust and cooking emissions were the most likely sources. Despite the expected  
 245 suppression of atmospheric oxidative capacity by NO<sub>x</sub> [49, 50], the continental air still contained the  
 246 largest amount of MT-SOA-T, attributable to higher background concentrations (section 3.2),  
 247 more abundant monoterpenes as precursors (159±4 pptv) and enhanced formation of MT-SOA-T  
 248 facilitated by sulfate (section 3.2). The marine air was characterized by the highest level of SO<sub>2</sub>,  
 249 indicating the impact of ship emissions. Isop-SOA-T was most deficient in marine air, despite little  
 250 dependence of isoprene mixing ratios on air mass types, perhaps suggesting minor influence of  
 251 local photo-oxidation of isoprene on Isop-SOA-T levels and/or higher background levels in  
 252 continental and mixed air (section 3.2). Ozone (O<sub>3</sub>) and OHDCA were somehow more abundant  
 253 in marine air than in continental air, though they were even more prevalent in mixed air. This  
 254 might be explained by the enhancement of oxidative capacity in offshore marine atmosphere  
 255 (OMA), with the ageing of continental air in a NO<sub>x</sub>-deficient atmosphere [51, 52]. Therefore, photo-  
 256 degradation likely dominates production of OHDCA. The slightly lower levels of O<sub>3</sub> and OHDCA  
 257 in marine air than in mixed air might be due to the lower abundance of reactants involved in photo-  
 258 oxidation.

259 Table 1. Mean levels of OA markers (*italic*, ng m<sup>-3</sup>) and trace gases including volatile organic  
 260 compounds (VOCs) (ppbv unless otherwise specified) by air mass origins.

Species	Continental air	Marine air	Mixed air
O <sub>3</sub>	39.2 ± 2.3	52.6 ± 2.0	65.3 ± 2.3
CO	372 ± 9	263 ± 11	260 ± 15
SO <sub>2</sub>	1.1 ± 0.1	1.5 ± 0.1	1.3 ± 0.1
Isoprene	1.6 ± 0.04	1.5 ± 0.05	1.7 ± 0.1
Monoterpenes (pptv)	159 ± 4	130 ± 4	142 ± 4
*			
Toluene	3.3 ± 0.1	2.0 ± 0.1	1.9 ± 0.1
<i>OHDCA</i>	<i>183 ± 16</i>	<i>355 ± 38</i>	<i>384 ± 33</i>
<i>Isop-SOA-T</i>	<i>5.9 ± 0.7</i>	<i>3.8 ± 0.3</i>	<i>5.8 ± 0.7</i>
<i>MT-SOA-T</i>	<i>2.1 ± 0.2</i>	<i>1.1 ± 0.2</i>	<i>1.3 ± 0.1</i>

<i>Levoglucosan</i>	$16.9 \pm 1.1$	$5.0 \pm 1.0$	$5.4 \pm 0.7$
<i>Even fatty acids</i>	$10.6 \pm 1.0$	$4.6 \pm 0.4$	$5.1 \pm 0.6$
<i>Oleic acid</i>	$2.0 \pm 0.5$	$0.2 \pm 0.02$	$0.2 \pm 0.02$

261 \* Sum of monoterpenes detected by a Proton Transfer Reaction Quadrupole ion Time-of-Flight  
 262 Mass Spectrometry. The molecular formula of the protonated compounds is  $(C_{10}H_{16})H^+$ , according  
 263 to the parent ion of  $m/z$  137.13248.

## 264 3.2 Diurnal patterns and correlation analysis

### 265 3.2.1 Diurnal patterns of OA markers

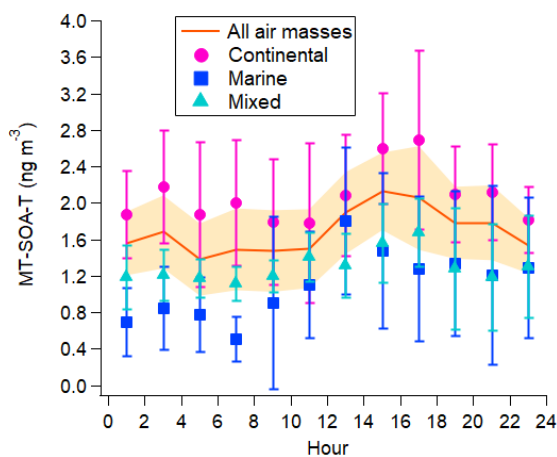
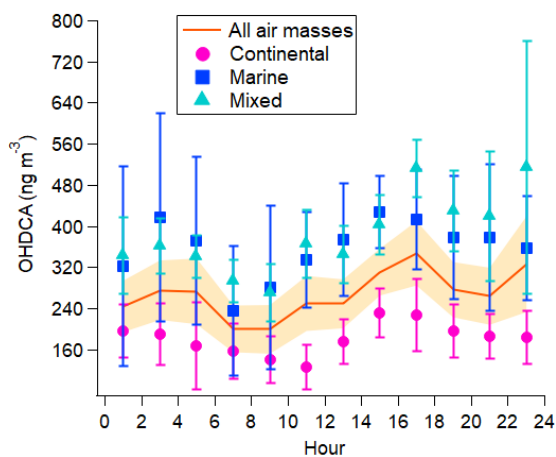
266 Diurnal pattern is a reflection of the evolution, sources and vertical mixing of air pollutants in a  
 267 day. Figure 4 shows the diurnal variations of several typical OA markers. Both OHDCAs and MT-  
 268 SOA-T followed the diurnal pattern of  $O_x$  ( $O_3 + NO_2$ ) (Figure S6), showing peaks in the afternoon,  
 269 indicating their secondary origin. Although many OHDCAs are extensively present in fruits and  
 270 are used as food additives <sup>[53]</sup>, they are also considered to be the late oxidation products of  
 271 unsaturated hydrocarbons and fatty acids, and aldehydes <sup>[54, 55]</sup>. The  $O_x$ -like diurnal pattern  
 272 suggested that OHDCAs observed in this campaign were more likely to be derived from photo-  
 273 oxidation.

274 Interestingly, Isop-SOA-T did not show analogous patterns to those of MT-SOA-T or  $O_x$ . The  
 275 Isop-SOA-T concentrations were higher at night and lowest at noon, consistent with the diurnal  
 276 development of Planetary Boundary Layer (PBL) (Figure S6). Isoprene, as the precursor of Isop-  
 277 SOA-T, however had typical peaks at around 13:00 – 15:00 (not shown). Therefore, it was  
 278 plausible that the detected Isop-SOA-T was a regional background signal more influenced by PBL  
 279 height than by daytime photo-oxidation. This was verified by the smallest differences in Isop-  
 280 SOA-T concentrations between marine air and continental/mixed air at noon, when the PBL was  
 281 fully developed. In fact, there was a discernable increase for Isop-SOA-T in marine air during  
 282 01:00 – 07:00. This was also interpretable, given that the air above marine boundary layer might  
 283 be of continental origin and the air rich in Isop-SOA-T intruded into the marine boundary layer  
 284 where the background Isop-SOA-T was less abundant. However, the possibility that some dark  
 285 reactions <sup>[56, 57]</sup> contributed to the higher nocturnal Isop-SOA-T cannot be ruled out.

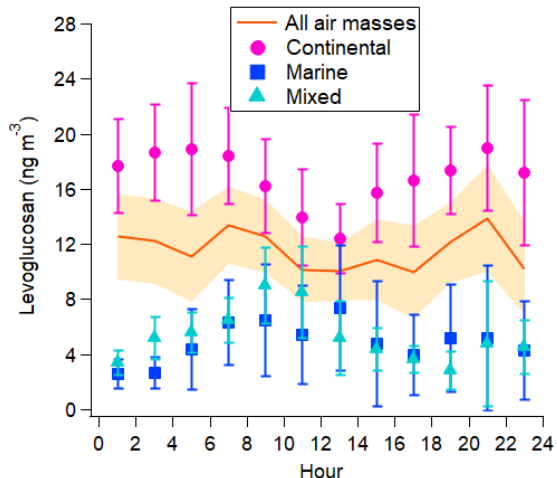
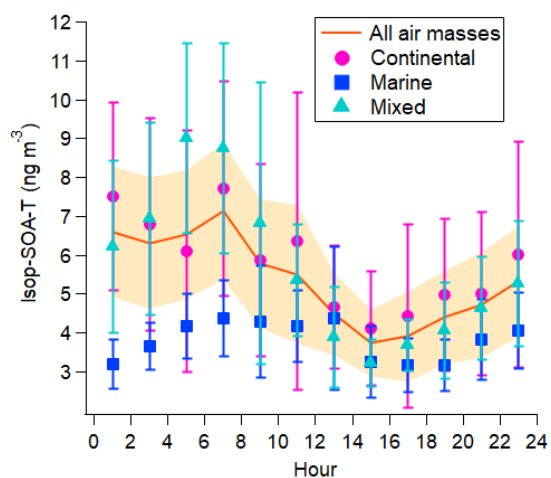
286 The diurnal patterns of levoglucosan in continental air were opposite to the variations of  
 287 temperature, likely resulting from the gas-to-particle partitioning. However, unusual increases

288 were observed for levoglucosan from 1:00 to 09:00 – 11:00 in marine and mixed air, which were  
 289 analogous to the rise of Isop-SOA-T in marine air in the early morning. Therefore, the development  
 290 of PBL and consequent air mixing caused the enhancement of levoglucosan in marine and mixed  
 291 air before 09:00 – 11:00. Again, the smallest differences of levoglucosan concentrations among  
 292 different air masses were also observed at noon, confirming the modulation of levoglucosan  
 293 concentration by PBL development. A previous study [38] implied the downward transport of  
 294 levoglucosan at the same site with the Philippines as potential source region in spring. While the  
 295 vertical mixing still existed, the source regions might be different in our study, due to the very few  
 296 air masses originating from or passing the Philippines (Figure 2).

297 Lastly, the diurnal patterns of oleic acid in continental air were also noteworthy, with an obvious  
 298 peak during 19:00 – 21:00 (Figure S6). This coincides with the dining habits of Hong Kong people  
 299 – big dinner and simpler breakfast and lunch.



300



301

302 Figure 4. Average diurnal cycles of OHDCA, MT-SOA-T, Isop-SOA-T and levoglucosan in  
303 different air masses. Error bars and shaded area represent 95% confidence intervals (CIs).

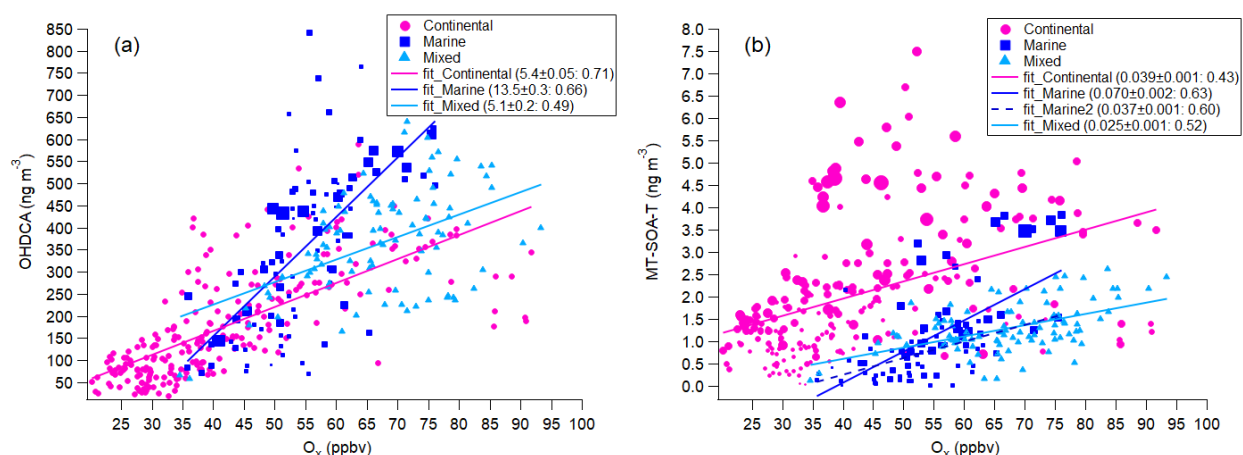
### 304 3.2.2 Correlations of SOA markers with O<sub>x</sub>

305 To further understand the formation processes of OHDCA and MT-SOA-T, correlation analyses  
306 were performed. Figure 5 shows the correlations of OHDCA *vs.* O<sub>x</sub> and MT-SOA-T *vs.* O<sub>x</sub> in  
307 different air masses. The moderate correlations between OHDCA and O<sub>x</sub> reaffirmed the secondary  
308 source of OHDCA. Further correlation analysis found that OHDCA moderately ( $R^2=0.44$ )  
309 correlated with DHOPA (aromatics SOA) and fairly well ( $R^2=0.73$ ) with MT-SOA-T in  
310 continental air. However, the correlations became much weaker in marine and mixed air. Nearly  
311 no correlation was identified between OHDCA and Isop-SOA-T in any type of air. Therefore,  
312 OHDCA in continental air might be derived from both anthropogenic and biogenic precursors,  
313 while it was likely attributable to anthropogenic precursors other than aromatics in marine and  
314 mixed air. The inferences were reasonable for individual compounds included in the group of  
315 OHDCA (Table S4), except for citramalic acid (CMA) and 2-hydroxyglutaric acid (2-HGA) which  
316 might be partially derived from biogenic VOCs in marine air due to their good correlations with  
317 MT-SOA-T ( $R^2=0.65$  and  $0.72$ , respectively). However, CMA and 2-HGA only constituted 0.5%  
318 and 1.4% of OHDCA in mass concentration, respectively. This supports the argument of Hu and  
319 Yu [42], who indicated that only in summer were biogenic VOCs the main precursors of OHDCA  
320 in Hong Kong.

321 Furthermore, the slope of OHDCA *vs.* O<sub>x</sub> in marine air was highest among the three regressions,  
322 and we found that the high levels of OHDCA in marine air were generally accompanied by  
323 elevated SO<sub>2</sub> concentrations. The SO<sub>2</sub> spikes in marine air most likely related to ship emissions,  
324 which has become the largest source of SO<sub>2</sub> in Hong Kong [58]. It was therefore plausible that the  
325 production efficiency of OHDCA was enhanced in marine air with ship emissions. On one hand,  
326 ship emissions might consist of high levels of some OHDCA precursors not detected by us. On  
327 the other hand, it remains to be verified whether SO<sub>2</sub> in ship emissions facilitates the formation of  
328 anthropogenic SOA.

329 For MT-SOA-T, the correlation with O<sub>x</sub> was weakest in continental air. It is interesting to note that  
330 the samples away from the regression line generally had high concentrations of sulfate in PM<sub>2.5</sub>,  
331 implying the well-documented catalysis of acidic aerosols to formation of biogenic SOA [59].

332 Besides, the larger intercept in “continental” regression suggested higher background MT-SOA-T  
 333 in continental air. The correlation of MT-SOA-T vs.  $O_x$  in marine air was somewhat undermined  
 334 by several samples with high BC levels, possibly due to the intrusion of continental air.  
 335 Regressions excluding these samples achieved better correlations with a slope closer to those  
 336 acquired in “continental” and “mixed” regressions. As such, the formation of MT-SOA-T was not  
 337 enhanced in marine air. This reduced the possibility that  $SO_2$  in ship emissions accelerated the  
 338 formation of SOA (*e.g.* OHDCA), but further studies are needed for confirmation. Nevertheless,  
 339 the lower concentrations of MT-SOA-T in marine and mixed air were mostly attributable to the  
 340 gaps in background levels, rather than different oxidation efficiencies.



341  
 342 Figure 5. Correlations of (a) OHDCA vs.  $O_x$ , and (b) MT-SOA-T vs.  $O_x$  by air mass origins. Size  
 343 of square markers in (a) is proportional to  $SO_2$  concentration. Size of circles and squares in (b) is  
 344 proportional to sulfate level in  $PM_{2.5}$  and BC, respectively. Numbers in parentheses are (slope:  
 345 correlation coefficient). Blue dashed line in (b) indicates the regression excluding data points with  
 346 high BC concentrations.

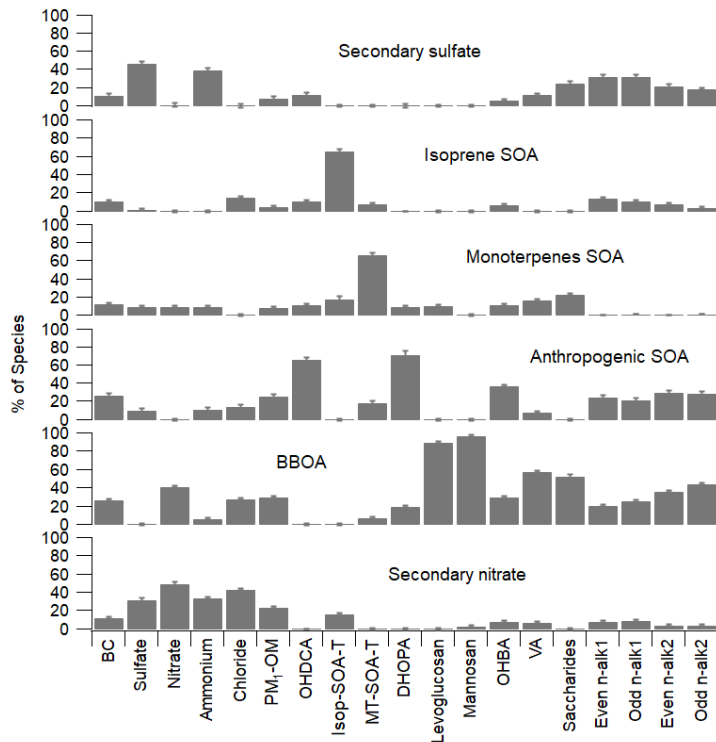
347

### 348 3.3 Sources of $PM_1$ -OM

349 With provision of time resolved concentrations of organic molecular markers, TAG data assisted  
 350 to identify dynamic and specific sources of OA [28, 60-63] and other PM bound components (*e.g.* BC)  
 351 [64]. Figure 6 shows the source profiles of  $PM_1$ -OM resolved by US Environmental Protection  
 352 Agency’s Positive Matrix Factorization model (v 5.0.12). A PMF solution of 6 factors was adopted,  
 353 according to the metrics of  $Q/Q_{exp}$  (Figure S7) and the interpretability of source profiles (Figures

354 6 & S8), as illustrated in Text S4 where the model configurations, residual analysis (Figure S9)  
355 and error estimates are also presented.

356 The first factor was defined as secondary sulfate, due to the dominance of sulfate and ammonium.  
357 8% of PM<sub>1</sub>-OM and moderate fractions of saccharides and *n*-alkanes were also allocated in this  
358 factor, partially constituting the unresolved OMs due to the lack of OA markers for some sources  
359 (such as hopanes indicative of vehicle emissions<sup>[65]</sup>). The second and third factors had exclusively  
360 high loading of Isop-SOA-T and MT-SOA-T, representing isoprene SOA and monoterpenes SOA,  
361 respectively. The fourth factor was characterized by OHDCa and DHOPA, both of which are SOA  
362 components. As discussed above, anthropogenic VOCs were most likely precursors of OHDCa  
363 in marine and mixed air. Though OHDCa in continental air might also be derived from biogenic  
364 VOCs, the concentrations were only half of those in marine and mixed air (Table 1). CMA and 2-  
365 HGA partially of biogenic origins in marine air had very low mass fractions (<2%) in OHDCa.  
366 Besides, DHOPA is generally considered to be an oxidation product of aromatics<sup>[66]</sup>. Hence, this  
367 factor was tentatively identified as anthropogenic SOA. The fifth factor was dominated by BB  
368 tracers, such as levoglucosan, mannosan and vanillic acid<sup>[67]</sup>, thereby representing biomass  
369 burning OA (BBOA). Moderate loadings of nitrate, ammonium and chloride presented in the last  
370 factor, which was termed as secondary nitrate. 24% of PM<sub>1</sub>-OM was attributed to this factor, being  
371 another part of unresolved OMs. The concentrations of oleic acid were below detection limit in  
372 51% samples, and were not included in source apportionment. Hence, cooking OA was not  
373 identifiable.



374

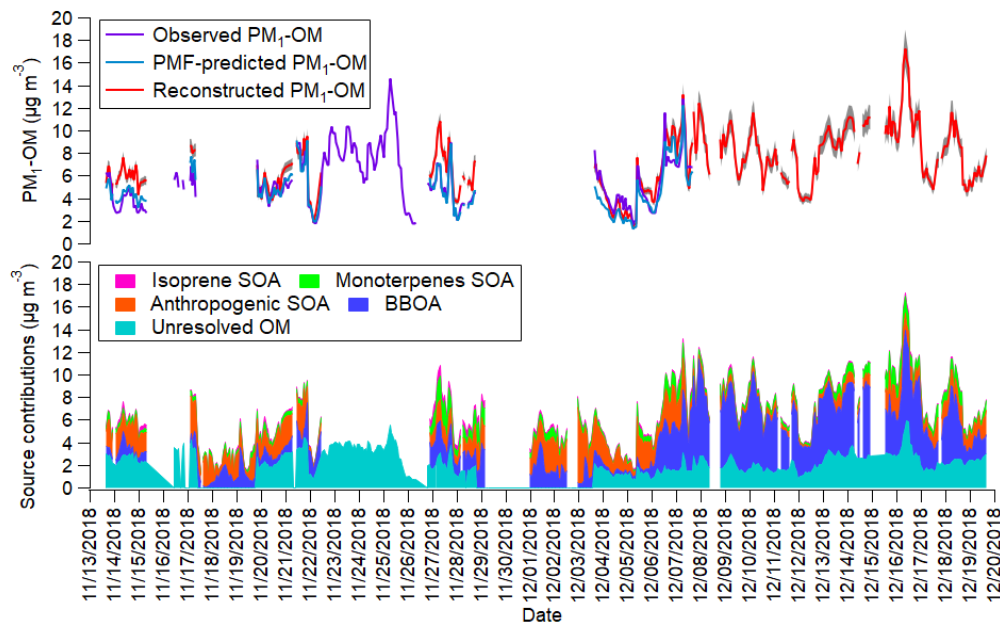
375 Figure 6. Source profiles of PM<sub>1</sub>-OM. The errors are 95% CIs estimated by Bootstrap method.

376 Since PM<sub>1</sub>-OM data were only available before 7 December, the source apportionment results  
 377 were extended to the period of 7–19 December, by reconstructing PM<sub>1</sub>-OM with individual source  
 378 contributions. This approach is illustrated in Text S5, Figures S10-S11 and Tables S5-S6. In brief,  
 379 it was based on the assumptions that the main tracers were still primarily attributed to the  
 380 corresponding sources during 7–9 December, and their mass fractions in the total emissions of the  
 381 corresponding sources did not significantly differ from those before 7 December. As a validation,  
 382 the observed, PMF-predicted and reconstructed PM<sub>1</sub>-OM are compared in Figure 7. Generally,  
 383 both PMF and the reconstruction method reasonably reproduced the observed PM<sub>1</sub>-OM. The linear  
 384 regression of PMF-predicted PM<sub>1</sub>-OM (reconstructed PM<sub>1</sub>-OM) against observed PM<sub>1</sub>-OM  
 385 indicated a slope of 0.92 (0.90) and R<sup>2</sup> of 0.80 (0.62). As such, we adopted the reconstructed PM<sub>1</sub>-  
 386 OM and the corresponding contributions from individual sources. Due to the prevalence of  
 387 continental air, the concentrations of primary air pollutants were significantly enhanced after 7  
 388 December, such as the ~4 and ~10 fold growth of NO<sub>x</sub> and oleic acid, respectively. The unresolved  
 389 POA components, including vehicle, ship and cooking emissions might make increased  
 390 contributions to PM<sub>1</sub>-OM in this period. Therefore, the PM<sub>1</sub>-OM was most likely underestimated



391 after 7 December. It is hard to determine to what extent the unresolved OM (sum of PM<sub>1</sub>-OM in  
392 factors of secondary sulfate and secondary nitrate) accounted for these POA components and their  
393 enhancements after 7 December. The uncertainties in source apportionment results related to the  
394 unresolved OM and potentially different sources during 7–19 December (*e.g.* cooking emissions)  
395 are discussed in Text S6. However, the conclusions on relative importance of the present sources  
396 would not change.

397 Bearing the uncertainties in mind, we conclude that BBOA contributed 8% – 74% (mean: 50±3%)  
398 to PM<sub>1</sub>-OM in continental air, followed by anthropogenic SOA (12±1%), monoterpene SOA  
399 (9±1%) and isoprene SOA (2±0.3%). The contributions of biogenic SOA might be underestimated,  
400 because some fractions of OHCA potentially derived from biogenic VOCs in continental air  
401 were assigned to anthropogenic SOA. While the BBOA contributions were not quantifiable during  
402 22 – 26 November due to the lack of levoglucosan data (days after passing of a cold front when  
403 continental air dominated), substantial increase in BBOA contributions occurred after 7 December.  
404 This was grounded upon the notable enhancements of levoglucosan in this period, as discussed in  
405 section 3.1. The significant contributions of BBOA in continental air induced by cold fronts were  
406 roughly comparable to those determined at a suburban site (~38%)<sup>[23]</sup> and at a mountainous site  
407 (~48%)<sup>[17]</sup> when Hong Kong was shrouded in regional air pollution. In contrast, the BBOA  
408 contributions (23±6%) were overwhelmed by those of anthropogenic SOA (37±4%) in marine air,  
409 so did those in mixed air (BBOA: 17±3% *vs.* anthropogenic SOA: 31±3%). It is interesting to note  
410 that both the time series and diurnal patterns of anthropogenic SOA were consistent with those of  
411 the more-oxidized oxygenated organic aerosols (MO-OOA) resolved by AMS, despite the overall  
412 lower concentrations of anthropogenic SOA resolved by TAG (Figure S13). This indicates the  
413 high oxidation state of anthropogenic SOA which might be inclined to stay in particles. The OMA  
414 with generally enhanced oxidative capacity might be sufficient in the passing air parcels to produce  
415 higher anthropogenic SOA in marine and mixed air. Like those in continental air, the biogenic  
416 SOA made minor contributions to PM<sub>1</sub>-OM in marine air (11±1%) and mixed air (9±1%), which  
417 might be explained by the cool weather in this field campaign.



418

419 Figure 7. Time series of observed, PMF-predicted, reconstructed PM<sub>1</sub>-OM, and contributions of  
 420 individual sources to PM<sub>1</sub>-OM. Grey area on red line of upper panel represents 95% CIs of the  
 421 reconstructed PM<sub>1</sub>-OM.

422

#### 423 4. Implications

424 As a neighbor of mainland China and a coastal city facing SCS, Hong Kong air quality is very  
 425 susceptible to regional transport of air pollution. Transient but severe air pollution episodes often  
 426 occur in Hong Kong, causing rapid deterioration of air quality and acute exposures to toxic  
 427 substances in the air. With the application of an in-situ analysis technique, this paper presents  
 428 temporally resolved in-situ measurements of OA markers and the determination of changing  
 429 contributions of specific sources to PM<sub>1</sub>-OM. BB remains a significant contributor of OA in  
 430 autumn in South China, though burning agricultural residuals had been strictly prohibited and  
 431 levoglucosan concentrations indeed decreased compared to those in early 2000s. The temperature-  
 432 dependent diurnal variations of levoglucosan in continental air suggest a shift of gas-to-particle  
 433 partitioning might be an essential factor enhancing BBOA contributions in cool seasons. Even if  
 434 marine or coastal air prevailed in Hong Kong, the residual BBOA above the PBL might still be  
 435 transported down to the ground, increasing OA burdens. Since the two cold fronts caused strong  
 436 continental outflows in this field campaign, we believe that the BBOA in continental air was

437 mainly attributable to regional rather than local sources. In addition, the “consensus” that marine  
438 air is relatively clean was challenged by our finding that the marine air was laden with  
439 anthropogenic SOA (mainly OHDCAs), whose formation was likely facilitated by SO<sub>2</sub> and/or  
440 undetected pollutants in ship emissions. The abundant OHDCAs in mixed air might also be  
441 attributable to the enhanced atmospheric oxidative capacity in OMA with the aging of continental  
442 outflows. Therefore, the OMA, where continental and marine air pollutants interacted, may be an  
443 ideal place for SOA formation. As a result, the air quality of coastal cities will be affected under  
444 sea breezes and have climatic consequences. The source apportionment failed to identify the  
445 expected vehicular, ship-emitted and cooking OA; however the findings of relative contributions  
446 of the present sources were still instructive. POA from regional BB and anthropogenic SOA  
447 possibly resulting from or at least enhanced by ship emissions should be controlled in cool seasons  
448 in Hong Kong.

449

#### 450 **Supporting Information**

451 TAG operation procedures (Text S1); AMS calibrations and configurations (Text S2);  
452 Uncertainties associated with classification of air mass origins (Text S3); PMF configurations and  
453 modeling evaluation (Text S4); Reconstruction of PM<sub>1</sub>-OM and method validation (Text S5);  
454 Uncertainties in source apportionment results (Text S6). Sampling site and instruments  
455 deployment (Figure S1); AMS collection efficiency calibration (Figure S2); Weather charts  
456 indicating cold fronts (Figure S3); Time series of BB tracers (Figure S4); Correlations between  
457 levoglucosan and acetonitrile/temperature (Figure S5); Diurnal patterns of O<sub>x</sub>, oleic acid, PBL  
458 height and temperature (Figure S6);  $Q/Q_{exp}$  versus factor number (Figure S7); Source profiles in  
459 5–7 factor solutions (Figure S8); Time series of scaled residuals (Figure S9); Relationships  
460 between source resolved PM<sub>1</sub>-OM and tracers (Figure S10); Validation of reconstructed PM<sub>1</sub>-OM  
461 (Figure S11); Diurnal patterns of unresolved OM, NO<sub>x</sub> and SO<sub>2</sub> (Figure S12); Comparisons  
462 between anthropogenic SOA and MO-OOA (Figure S13). TAG development and application  
463 (Table S1); TAG operation parameters (Table S2); Quality control metrics of TAG data (Table  
464 S3); OA marker groups (Table S4); Instruments for auxiliary measurements (Table S5);  
465 Constraints applied to base run (Table S6); Bootstrap mapping results (Table S7).

466

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478

#### 479 **References:**

480 [1] Kanakidou, M.; Seinfeld, J.H.; Pandis, S.N.; Barnes, I.; Dentener, F.J.; Facchini, M.C.;  
481 Dingenen, R.V.; Ervens, B.; Nenes, A.N.; Nielsen, C.J.; Swietlicki, E. Organic aerosol and global  
482 climate modelling: a review. *Atmos. Chem. Phys.* **2005**, 5(4), 1053-1123.

483 [2] Jimenez, J.L.; Canagaratna, M.R.; Donahue, N.M.; Prevot, A.S.; Zhang, Q.; Kroll, J.H.;  
484 DeCarlo, P.F.; Allan, J.D.; Coe, H.; Ng, N.L.; Aiken, A.C. Evolution of organic aerosols in the  
485 atmosphere. *Science*. **2009**, 326(5959), 1525-1529.

486 [3] Huang, R.J.; Zhang, Y.; Bozzetti, C.; Ho, K.F.; Cao, J.J.; Han, Y.; Daellenbach, K.R.; Slowik,  
487 J.G.; Platt, S.M.; Canonaco, F.; Zotter, P. High secondary aerosol contribution to particulate  
488 pollution during haze events in China. *Nature*. **2014**, 514(7521), 218-222.

489 [4] Goldstein, A.H.; Koven, C.D.; Heald, C.L.; Fung, I.Y. Biogenic carbon and anthropogenic  
490 pollutants combine to form a cooling haze over the southeastern United States. *Proc. Natl. Acad.*  
491 *Sci.* **2009**, 106(22), 8835-8840.

492 [5] Maria, S.F.; Russell, L.M.; Gilles, M.K.; Myneni, S.C. Organic aerosol growth mechanisms  
493 and their climate-forcing implications. *Science*. **2004**, 306(5703), 1921-1924.

494 [6] Mauderly, J.L.; Chow, J.C. Health effects of organic aerosols. *Inhal. Toxicol.* **2008**, 20(3), 257-  
495 288.

496 [7] Zhang, Q.; Zheng, Y.; Tong, D.; Shao, M.; Wang, S.; Zhang, Y.; Xu, X.; Wang, J.; He, H.; Liu,  
497 W.; Ding, Y. Drivers of improved PM<sub>2.5</sub> air quality in China from 2013 to 2017. *Proc. Natl. Acad.*  
498 *Sci.* **2019**, 116(49), 24463-24469.

- 499 [8] Chen, Q.; Fu, T.M.; Hu, J.; Ying, Q.; Zhang, L. Modelling secondary organic aerosols in China.  
500 *Natl. Sci. Rev.* **2017**, 4(6), 806-809.
- 501 [9] Cao, F.; Zhang, Y.; Ren, L.; Liu, J.; Li, J.; Zhang, G.; Liu, D.; Sun, Y.; Wang, Z.; Shi, Z.; Fu,  
502 P. New insights into the sources and formation of carbonaceous aerosols in China: potential  
503 applications of dual-carbon isotopes. *Natl. Sci. Rev.* **2017**, 4(6), 804-806.
- 504 [10] Jaeckels, J.M.; Bae, M.S.; Schauer, J.J. Positive matrix factorization (PMF) analysis of  
505 molecular marker measurements to quantify the sources of organic aerosols. *Environ. Sci. Technol.*  
506 **2007**, 41(16), 5763-5769.
- 507 [11] Fu, P.Q.; Kawamura, K.; Chen, J.; Charriere, B.; Sempere, R. Organic molecular composition  
508 of marine aerosols over the Arctic Ocean in summer: contributions of primary emission and  
509 secondary aerosol formation. *Biogeosciences*. **2013**, 10, 653-667.
- 510 [12] Ding, X.; Wang, X.; Xie, Z.; Zhang, Z.; Sun, L. Impacts of Siberian biomass burning on  
511 organic aerosols over the North Pacific Ocean and the Arctic: Primary and secondary organic  
512 tracers. *Environ. Sci. Technol.* **2013**, 47(7), 3149-3157.
- 513 [13] Wang, Q.; He, X.; Huang, X.H.; Griffith, S.M.; Feng, Y.; Zhang, T.; Zhang, Q.; Wu, D.; Yu,  
514 J.Z. Impact of secondary organic aerosol tracers on tracer-based source apportionment of organic  
515 carbon and PM<sub>2.5</sub>: A case study in the Pearl River Delta, China. *ACS Earth Space Chem.* **2017**,  
516 1(9), 562-571.
- 517 [14] Isaacman, G.; Kreisberg, N.M.; Yee, L.D.; Worton, D.R.; Chan, A.W.; Moss, J.A.; Hering,  
518 S.V.; Goldstein, A.H. Online derivatization for hourly measurements of gas-and particle-phase  
519 semi-volatile oxygenated organic compounds by thermal desorption aerosol gas chromatography  
520 (SV-TAG). *Atmos. Meas. Tech.* **2014**, 7(12), 4417-4429.
- 521 [15] Lee, B.P.; Li, Y.J.; Yu, J.Z.; Louie, P.K.; Chan, C.K. Physical and chemical characterization  
522 of ambient aerosol by HR-ToF-AMS at a suburban site in Hong Kong during springtime 2011. *J.*  
523 *Geophys. Res. Atmos.* **2013**, 118(15), 8625-8639.
- 524 [16] Sun, Y.L.; Zhang, Q.; Schwab, J.J.; Demerjian, K.L.; Chen, W.N.; Bae, M.S.; Hung, H.M.;  
525 Hogrefe, O.; Frank, B.; Rattigan, O.V.; Lin, Y.C. Characterization of the sources and processes of

526 organic and inorganic aerosols in New York city with a high-resolution time-of-flight aerosol mass  
527 spectrometer. *Atmos. Chem. Phys.* **2011**, 11(4), 1581-1602.

528 [17] Lyu, X.P.; Guo, H.; Cheng, H.R.; Wang, X.M.; Ding, X.; Lu, H.X.; Yao, D.W.; Xu, C.  
529 Observation of SOA tracers at a mountainous site in Hong Kong: Chemical characteristics, origins  
530 and implication on particle growth. *Sci. Total Environ.* **2017**, 605, 180-189.

531 [18] Guo, H.; Ling, Z.H.; Cheung, K.; Jiang, F.; Wang, D.W.; Simpson, I.J.; Barletta, B.; Meinardi,  
532 S.; Wang, T.J.; Wang, X.M.; Saunders, S.M. Characterization of photochemical pollution at  
533 different elevations in mountainous areas in Hong Kong. *Atmos. Chem. Phys.* **2013**, 13(8), 3881-  
534 3898.

535 [19] Zeren, Y.; Guo, H.; Lyu, X.; Jiang, F.; Wang, Y.; Liu, X.; Zeng, L.; Li, M.; Li, L. An ozone  
536 “pool” in South China: Investigations on atmospheric dynamics and photochemical processes over  
537 the Pearl River Estuary. *J. Geophys. Res. Atmos.* **2019**, 124(22), 12340-12355.

538 [20] Lyu, X.; Guo, H.; Simpson, I.J.; Meinardi, S.; Louie, P.K.; Ling, Z.; Wang, Y.; Liu, M.; Luk,  
539 C.W.; Wang, N.; Blake, D.R. Effectiveness of replacing catalytic converters in LPG-fueled  
540 vehicles in Hong Kong. *Atmos. Chem. Phys.* **2016**, 16(10), 6609-6626.

541 [21] Lee, B.P.; Li, Y.J.; Yu, J.Z.; Louie, P.K.; Chan, C.K. Characteristics of submicron particulate  
542 matter at the urban roadside in downtown Hong Kong – Overview of 4 months of continuous  
543 high - resolution aerosol mass spectrometer measurements. *J. Geophys. Res. Atmos.* **2015**, 120(14),  
544 7040-7058.

545 [22] Sun, C.; Lee, B.P.; Huang, D.; Li, Y.; Schurman, M.I.; Louie, P.K.; Luk, C.; Chan, C.K.  
546 Continuous measurements at the urban roadside in an Asian megacity by Aerosol Chemical  
547 Speciation Monitor (ACSM): particulate matter characteristics during fall and winter seasons in  
548 Hong Kong. *Atmos. Chem. Phys.*, **2016**, 16(3), 1713-1728.

549 [23] Hu, D.; Bian, Q.; Lau, A.K.; Yu, J.Z. Source apportioning of primary and secondary organic  
550 carbon in summer PM<sub>2.5</sub> in Hong Kong using positive matrix factorization of secondary and  
551 primary organic tracer data. *J. Geophys. Res. Atmos.* **2010**, 115(D16),  
552 doi.org/10.1029/2009JD012498.

- 553 [24] Wong, Y.K.; Huang, X.H.; Yu, J.Z. Incorporating hopane degradation into chemical mass  
554 balance model: Improving accuracy of vehicular source contribution estimation. *Atmos. Environ.*  
555 **2019**, 210, 211-219.
- 556 [25] Williams, B.J.; Goldstein, A.H.; Kreisberg, N.M.; Hering, S.V. In situ measurements of  
557 gas/particle-phase transitions for atmospheric semivolatile organic compounds. *Proc. Natl. Acad.*  
558 *Sci.* **2010**, 107(15), 6676-6681.
- 559 [26] Zhao, Y.; Kreisberg, N.M.; Worton, D.R.; Isaacman, G.; Weber, R.J.; Liu, S.; Day, D.A.;  
560 Russell, L.M.; Markovic, M.Z.; VandenBoer, T.C.; Murphy, J.G. Insights into secondary organic  
561 aerosol formation mechanisms from measured gas/particle partitioning of specific organic tracer  
562 compounds. *Environ. Sci. Technol.* **2013**, 47(8), 3781-3787.
- 563 [27] Hunter, J.F.; Day, D.A.; Palm, B.B.; Yatavelli, R.L.; Chan, A.W.; Kaser, L.; Cappellin, L.;  
564 Hayes, P.L.; Cross, E.S.; Carrasquillo, A.J.; Campuzano-Jost, P. Comprehensive characterization  
565 of atmospheric organic carbon at a forested site. *Nat. Geosci.* **2017**, 10(10), 748-753.
- 566 [28] Zhao, Y.; Kreisberg, N.M.; Worton, D.R.; Isaacman, G.; Gentner, D.R.; Chan, A.W.; Weber,  
567 R.J.; Liu, S.; Day, D.A.; Russell, L.M.; Hering, S.V. Sources of organic aerosol investigated using  
568 organic compounds as tracers measured during CalNex in Bakersfield. *J. Geophys. Res. Atmos.*  
569 **2013**, 118(19), 11388-11398.
- 570 [29] Williams, B.J.; Goldstein, A.H.; Kreisberg, N.M.; Hering, S.V. An In-Situ Instrument for  
571 Speciated Organic Composition of Atmospheric Aerosols: Thermal Desorption Aerosol GC/MS-  
572 FID (TAG). *Aerosol Sci Technol.* **2006**, 40(8), 627-638.
- 573 [30] Goldstein, A.H.; Worton, D.R.; Williams, B.J.; Hering, S.V.; Kreisberg, N.M.; Panić, O.;  
574 Górecki, T. Thermal desorption comprehensive two-dimensional gas chromatography for in-situ  
575 measurements of organic aerosols. *J. Chromatogr. A.* **2008**, 1186(1-2), 340-347.
- 576 [31] Kreisberg, N.M.; Hering, S.V.; Williams, B.J.; Worton, D.R.; Goldstein, A.H. Quantification  
577 of hourly speciated organic compounds in atmospheric aerosols, measured by an in-situ Thermal  
578 desorption Aerosol Gas chromatograph (TAG). *Aerosol Sci Technol.* **2009**, 43(1), 38-52.
- 579 [32] Worton, D.R.; Kreisberg, N.M.; Isaacman, G.; Teng, A.P.; McNeish, C.; Górecki, T.; Hering,  
580 S.V.; Goldstein, A.H. Thermal desorption comprehensive two-dimensional gas chromatography:

581 An improved instrument for in-situ speciated measurements of organic aerosols. *Aerosol Sci*  
582 *Technol.* **2012**, 46(4), 380-393.

583 [33] Zhao, Y.; Kreisberg, N.M.; Worton, D.R.; Teng, A.P.; Hering, S.V.; Goldstein, A.H.  
584 Development of an in situ thermal desorption gas chromatography instrument for quantifying  
585 atmospheric semi-volatile organic compounds. *Aerosol Sci Technol.* **2013**, 47(3), 258-266.

586 [34] Isaacman, G.; Kreisberg, N.M.; Yee, L.D.; Worton, D.R.; Chan, A.W.; Moss, J.A.; Hering,  
587 S.V.; Goldstein, A.H. Online derivatization for hourly measurements of gas-and particle-phase  
588 semi-volatile oxygenated organic compounds by thermal desorption aerosol gas chromatography  
589 (SV-TAG). *Atmos. Meas. Tech.* **2014**, 7(12), 4417-4429.

590 [35] Williams, B.J.; Jayne, J.T.; Lambe, A.T.; Hohaus, T.; Kimmel, J.R.; Sueper, D.; Brooks, W.;  
591 Williams, L.R.; Trimborn, A.M.; Martinez, R.E.; Hayes, P.L. The first combined thermal  
592 desorption aerosol gas chromatograph–Aerosol mass spectrometer (TAG-AMS). *Aerosol Sci*  
593 *Technol.* **2014**, 48(4), 358-370.

594 [36] Ren, H.; Xue, M.; An, Z.; Zhou, W.; Jiang, J. Quartz filter-based thermal desorption gas  
595 chromatography mass spectrometry for in-situ molecular level measurement of ambient organic  
596 aerosols. *J. Chromatogr. A.* **2019**, 1589, 141-148.

597 [37] Ren, H.; Xue, M.; An, Z.; Jiang, J. Improving thermal desorption aerosol gas chromatography  
598 using a dual-trap design. *J. Chromatogr. A.* **2019**, 1599, 247-252.

599 [38] An, Z.; Ren, H.; Xue, M.; Guan, X.; Jiang, J. Comprehensive Two-Dimensional Gas  
600 Chromatography Mass Spectrometry with a Solid-state Thermal Modulator for In-situ Speciated  
601 Measurement of Organic Aerosols. *J. Chromatogr. A.* **2020**,  
602 doi.org/10.1016/j.chroma.2020.461336.

603 [39] Zhang, Y.N.; Zhang, Z.S.; Chan, C.Y.; Engling, G.; Sang, X.F.; Shi, S.; Wang, X.M.  
604 Levoglucosan and carbonaceous species in the background aerosol of coastal southeast China:  
605 case study on transport of biomass burning smoke from the Philippines. *Environ. Sci. Pollut. Res.*  
606 **2012**, 19(1), 244-255.



607 [40] Wang, T.; Dai, J.; Lam, K.S.; Poon, C.; Brasseur, G.P. Twenty-Five Years of Lower  
608 Tropospheric Ozone Observations in Tropical East Asia: The Influence of Emissions and Weather  
609 Patterns. *Geophys. Res. Lett.* **2019**, 46(20), 11463-11470.

610 [41] Hu, D.; Bian, Q.; Li, T.W.; Lau, A.K.; Yu, J.Z. Contributions of isoprene, monoterpenes,  $\beta$ -  
611 caryophyllene, and toluene to secondary organic aerosols in Hong Kong during the summer of  
612 2006. *J. Geophys. Res. Atmos.* **2008**, 113(D22), doi.org/10.1029/2008JD010437.

613 [42] Hu, D.; Yu, J.Z. Secondary organic aerosol tracers and malic acid in Hong Kong: seasonal  
614 trends and origins. *Environ. Chem.* **2013**, 10(5), 381-394.

615 [43] Simoneit, B.R.; Schauer, J.J.; Nolte, C.G.; Oros, D.R.; Elias, V.O.; Fraser, M.P.; Rogge, W.F.;  
616 Cass, G.R. Levoglucosan, a tracer for cellulose in biomass burning and atmospheric particles.  
617 *Atmos. Environ.* **1999**, 33(2), 173-182.

618 [44] Zheng, M.; Hagler, G.S.; Ke, L.; Bergin, M.H.; Wang, F.; Louie, P.K.; Salmon, L.; Sin, D.W.;  
619 Yu, J.Z.; Schauer, J.J. Composition and sources of carbonaceous aerosols at three contrasting sites  
620 in Hong Kong. *J. Geophys. Res. Atmos.* **2006**, 111(D20), doi.org/10.1029/2006JD007074.

621 [45] Ding, X.; Wang, X.M.; Gao, B.; Fu, X.X.; He, Q.F.; Zhao, X.Y.; Yu, J.Z.; Zheng, M. Tracer-  
622 based estimation of secondary organic carbon in the Pearl River Delta, south China. *J. Geophys.*  
623 *Res. Atmos.* **2012**, doi.org/10.1029/2011JD016596.

624 [46] Schauer, J.J.; Kleeman, M.J.; Cass, G.R.; Simoneit, B.R. Measurement of emissions from air  
625 pollution sources. 4. C1- C27 organic compounds from cooking with seed oils. *Environ. Sci.*  
626 *Technol.* **2002**, 36(4), 567-575.

627 [47] Holzinger, R.; Warneke, C.; Hansel, A.; Jordan, A.; Lindinger, W.; Scharffe, D.H.; Schade,  
628 G.; Crutzen, P.J. Biomass burning as a source of formaldehyde, acetaldehyde, methanol, acetone,  
629 acetonitrile, and hydrogen cyanide. *Geophys. Res. Lett.* **1999**, 26(8), 1161-1164.

630 [48] Schneider, J.; Weimer, S.; Drewnick, F.; Borrmann, S.; Helas, G.; Gwaze, P.; Schmid, O.;  
631 Andreae, M.O.; Kirchner, U. Mass spectrometric analysis and aerodynamic properties of various  
632 types of combustion-related aerosol particles. *Int. J. Mass Spectrom.* **2006**, 258(1-3), 37-49.

633 [49] Logan, J.A.; Prather, M.J.; Wofsy, S.C.; McElroy, M.B. Tropospheric chemistry: A global  
634 perspective. *J. Geophys. Res. Oceans.* **1981**, 86(C8), 7210-7254.

635 [50] Kleinman, L.I. Low and high NO<sub>x</sub> tropospheric photochemistry. *J. Geophys. Res. Atmos.*  
636 **1994**, 99(D8), 16831-16838.

637 [51] Wang, H.; Lyu, X.P.; Guo, H.; Wang, Y.; Zou, S.C.; Ling, Z.H.; Wang, X.M.; Jiang, F.; Zeren,  
638 Y.Z.; Pan, W.Z.; Huang, X.B. Ozone pollution around a coastal region of South China Sea:  
639 interaction between marine and continental air. *Atmos. Chem. Phys.* **2018**, 18, 4277-4295.

640 [52] Wang, Y.; Guo, H.; Zou, S.; Lyu, X.; Ling, Z.; Cheng, H.; Zeren, Y. Surface O<sub>3</sub>  
641 photochemistry over the South China Sea: Application of a near-explicit chemical mechanism box  
642 model. *Environ. Pollut.* **2018**, 234, 155-166.

643 [53] Lobit, P.; Genard, M.; Soing, P.; Habib, R. Modelling malic acid accumulation in fruits:  
644 relationships with organic acids, potassium, and temperature. *J. Exp. Bot.* **2006**, 57(6), 1471-1483.

645 [54] Kawamura, K.; Sakaguchi, F. Molecular distributions of water soluble dicarboxylic acids in  
646 marine aerosols over the Pacific Ocean including tropics. *J. Geophys. Res. Atmos.* **1999**, 104(D3),  
647 3501-3509.

648 [55] Ervens, B.; Feingold, G.; Frost, G.J.; Kreidenweis, S.M. A modeling study of aqueous  
649 production of dicarboxylic acids: 1. Chemical pathways and speciated organic mass production. *J.*  
650 *Geophys. Res. Atmos.* **2004**, 109(D15), doi.org/10.1029/2003JD004387.

651 [56] Claeys, M.; Wang, W.; Ion, A.C.; Kourtchev, I.; Gelencsér, A.; Maenhaut, W. Formation of  
652 secondary organic aerosols from isoprene and its gas-phase oxidation products through reaction  
653 with hydrogen peroxide. *Atmos. Environ.* **2004**, 8(25), 4093-4098.

654 [57] Riva, M.; Budisulistiorini, S.H.; Zhang, Z.; Gold, A.; Surratt, J.D. Chemical characterization  
655 of secondary organic aerosol constituents from isoprene ozonolysis in the presence of acidic  
656 aerosol. *Atmos. Environ.* **2016**, 130, 5-13.

657

658 [58] *Hong Kong Air Pollutant Emission Inventory – Sulphur Dioxide Website*;  
659 [https://www.epd.gov.hk/epd/english/environmentinhk/air/data/emission\\_inve\\_so2\\_C.html](https://www.epd.gov.hk/epd/english/environmentinhk/air/data/emission_inve_so2_C.html), lastly  
660 accessed on 27 June 2020.

661 [59] Xu, L.; Guo, H.; Boyd, C.M.; Klein, M.; Bougiatioti, A.; Cerully, K.M.; Hite, J.R.; Isaacman-  
662 VanWertz, G.; Kreisberg, N.M.; Knote, C.; Olson, K. Effects of anthropogenic emissions on  
663 aerosol formation from isoprene and monoterpenes in the southeastern United States. *Proc. Natl.*  
664 *Acad. Sci.* **2015**, 112(1), 37-42.

665 [60] Williams, B.J.; Goldstein, A.H.; Millet, D.B.; Holzinger, R.; Kreisberg, N.M.; Hering, S.V.;  
666 White, A.B.; Worsnop, D.R.; Allan, J.D.; Jimenez, J.L. Chemical speciation of organic aerosol  
667 during the International Consortium for Atmospheric Research on Transport and Transformation  
668 2004: Results from in situ measurements. *J. Geophys. Res. Atmos.* **2007**,  
669 doi.org/10.1029/2006JD007601.

670 [61] Williams, B.J.; Goldstein, A.H.; Kreisberg, N.M.; Hering, S.V.; Worsnop, D.R.; Ulbrich, I.M.;  
671 Docherty, K.S.; Jimenez, J.L. Major components of atmospheric organic aerosol in southern  
672 California as determined by hourly measurements of source marker compounds. *Atmos. Chem.*  
673 *Phys.* **2010**, 10(23), 11577-11603.

674 [62] Worton, D.R.; Goldstein, A.H.; Farmer, D.K.; Docherty, K.S.; Jimenez, J.L.; Gilman, J.B.;  
675 Kuster, W.C.; De Gouw, J.; Williams, B.J.; Kreisberg, N.M.; Hering, S.V. Origins and  
676 composition of fine atmospheric carbonaceous aerosol in the Sierra Nevada Mountains, California.  
677 *Atmos. Chem. Phys.* **2011**, 11(19), 10219-10241.

678 [63] Chan, A.W.; Kreisberg, N.M.; Hohaus, T.; Campuzano-Jost, P.; Zhao, Y.; Day, D.A.; Kaser,  
679 L.; Karl, T.; Hansel, A.; Teng, A.P.; Ruehl, C.R. Speciated measurements of semivolatile and  
680 intermediate volatility organic compounds (S/IVOCs) in a pine forest during BEACHON-  
681 RoMBAS 2011. *Atmos. Chem. Phys.* **2016**, 16(2), 1187-1205.

682 [64] Lambe, A.T.; Logue, J.M.; Kreisberg, N.M.; Hering, S.V.; Worton, D.R.; Goldstein, A.H.;  
683 Donahue, N.M.; Robinson, A.L. Apportioning black carbon to sources using highly time-resolved  
684 ambient measurements of organic molecular markers in Pittsburgh. *Atmos. Environ.* **2009**, 43(25),  
685 3941-3950.

686 [65] Zielinska, B.; Sagebiel, J.; McDonald, J.D.; Whitney, K.; Lawson, D.R. Emission rates and  
687 comparative chemical composition from selected in-use diesel and gasoline-fueled vehicles. *J. Air*  
688 *Waste Manage. Assoc.* **2004**, 54(9), 1138-1150.

- 689 [66] Ding, X.; Zhang, Y.Q.; He, Q.F.; Yu, Q.Q.; Wang, J.Q.; Shen, R.Q.; Song, W.; Wang, Y.S.;  
690 Wang, X.M. Significant increase of aromatics-derived secondary organic aerosol during fall to  
691 winter in China. *Environmental science & technology*. 51(13):7432-7441, 2017.
- 692 [67] Fabbri, D.; Torri, C.; Simoneit, B.R.; Marynowski, L.; Rushdi, A.I.; Fabiańska, M.J.  
693 Levoglucosan and other cellulose and lignin markers in emissions from burning of Miocene  
694 lignites. *Atmos. Environ.* **2009**, 43(14), 2286-2295.