

1 **Causes of ozone pollution in summer ~~of~~in Wuhan, Central China**

2 P. Zeng<sup>1,2</sup>, X.P. Lyu<sup>2</sup>, H. Guo<sup>2\*</sup>, H.R. Cheng<sup>1\*\*</sup>, F. Jiang<sup>3</sup>, W.Z. Pan<sup>2</sup>, Z.W. Wang<sup>1</sup>, S.W. Liang<sup>3</sup>,  
3 Y.Q. Hu<sup>3</sup>

4 1. School of Resource and Environmental Sciences, Wuhan University, Wuhan 430072, China

5 2. Department of Civil and Environmental Engineering, The Hong Kong Polytechnic University,  
6 Hong Kong, China

7 3. International Institute for Earth System Science, Nanjing University, Nanjing 210023, China

8 4. Wuhan Environment Monitoring Center, Wuhan 430022, China

9 \*Corresponding author. [ceguohai@polyu.edu.hk](mailto:ceguohai@polyu.edu.hk);

10 \*\*Second corresponding author. [chenghr@whu.edu.cn](mailto:chenghr@whu.edu.cn)

11 **Abstract**

12 In August 2016, continuous measurements of volatile organic compounds (VOCs) and trace gases  
13 were conducted at an urban site in Wuhan. Four high-ozone (O<sub>3</sub>) days and twenty-seven non-high-  
14 O<sub>3</sub> days were identified according to the China's National Standard Level II (~100 ppb). The  
15 occurrence of high-O<sub>3</sub> days was accompanied by tropical cyclones. Much higher concentrations  
16 of VOCs and carbon monoxide (CO) were observed on the high-O<sub>3</sub> days ( $p < 0.01$ ). Model  
17 simulations revealed that vehicle exhausts were the dominant sources of VOCs, ~~with the~~  
18 ~~contribution of~~  $45.4 \pm 5.2\%$  and  $37.3 \pm 2.9\%$  during high-O<sub>3</sub> and non-high-O<sub>3</sub> days, respectively.  
19 Both vehicle exhausts and stationary combustion made significantly larger contributions to O<sub>3</sub>  
20 production on high-O<sub>3</sub> days ( $p < 0.01$ ). Analysis ~~of~~using a chemical transport model found that  
21 local photochemical formation accounted for  $74.7 \pm 5.8\%$  of the daytime O<sub>3</sub>, around twice the  
22 regional transport ( $32.2 \pm 5.4\%$ ), while the nighttime O<sub>3</sub> was mainly attributable to regional  
23 transport ( $59.1 \pm 9.9\%$ ). The local O<sub>3</sub> formation was generally limited by VOCs in urban Wuhan.  
24 To effectively control O<sub>3</sub> pollution, the reduction ratio of VOCs to NO<sub>x</sub> concentrations should not  
25 be lower than 0.73, and the most efficient O<sub>3</sub> abatement could be achieved by reducing VOCs from  
26 vehicle exhausts. This study contributes ~~ed~~ to the worldwide database of ~~the~~-O<sub>3</sub>-VOC-NO<sub>x</sub>

27 sensitivity research. ~~The-Its findings of this study are~~ will be helpful ~~to-in~~ formulating and  
28 implementing ~~the~~ emission control strategies ~~against for dealing with~~ O<sub>3</sub> pollution in Wuhan.

29 **Keywords:** VOCs, Source apportionment, O<sub>3</sub> formation, Vehicle exhausts, Control measures

30 **Capsule:** Local formation dominated ~~the~~ daytime O<sub>3</sub>, while ~~the~~ nighttime O<sub>3</sub> was attributable to  
31 regional transport in Wuhan.

### 32 1. Introduction

33 Over the past decades, tropospheric ozone (O<sub>3</sub>) pollution has become a major problem around  
34 the world because of its deleterious impact on human health, vegetation, and climate (Krupa and  
35 Manning, 1988; Tilton, 1989; Stevenson et al., 2013; Monks et al., 2015). In recent years, ~~the issue~~  
36 ~~of~~ severe O<sub>3</sub> pollution ~~has attracted broad attention in many~~ megacities and fast-developing regions  
37 in China ~~has attracted much attention~~ (Wang et al., 2009; Xue et al., 2014; Verstraeten et al., 2015;  
38 Gao et al., 2017; Wang et al., 2017). ~~To effectively control the-O<sub>3</sub> pollution, it is~~ ~~urgently~~  
39 ~~necessary~~ essential to ~~investigate-understand~~ the characteristics and formation mechanisms of O<sub>3</sub>  
40 pollution in Chinese cities with different energy structures and geographic features, such as Wuhan  
41 in Central China.

42 The global background, intrusion of stratospheric O<sub>3</sub>, and photochemical formation are the main  
43 sources of tropospheric O<sub>3</sub>, among which photochemical formation generally dominates O<sub>3</sub> levels  
44 in urban areas (Monks et al., 2015; Wang et al., 2016; Gao et al., 2017; Wang et al., 2017). In ozone  
45 chemistry, O<sub>3</sub> is formed through the ~~photolysis of NO<sub>2</sub>, following by the~~ combination ~~between of~~  
46 ~~an~~ oxygen atom from NO<sub>2</sub> photolysis ~~and-with~~ oxygen (O<sub>2</sub>) (Logan, 1985; Roelofs and Lelieveld,  
47 1997). ~~Con~~Reversely, O<sub>3</sub> reacts with NO, regenerating NO<sub>2</sub>. NO, NO<sub>2</sub>, and O<sub>3</sub> can reach a photo-  
48 equilibrium, which is driven by ~~the~~ oxidative radicals (e.g. OH, HO<sub>2</sub>, and RO<sub>2</sub>) (Monks, 2005).  
49 NO<sub>x</sub> plays ~~a~~ dual roles in O<sub>3</sub> formation. In ~~the-a~~ low NO<sub>x</sub> environment, it stimulates O<sub>3</sub> formation,  
50 while in ~~a~~ NO<sub>x</sub>-rich environment, O<sub>3</sub> is depleted by NO<sub>x</sub> mainly due to the titration of NO to O<sub>3</sub>  
51 (Wang et al., 2016). The relationships ~~s~~ between O<sub>3</sub> and its precursors in ~~these~~ complicated  
52 photochemical processes have been well documented in many previous studies. For example, it  
53 has been extensively reported that VOCs are the limiting factors of O<sub>3</sub> production in ~~the~~ urban

Formatted: French (France)  
Field Code Changed  
Formatted: French (France)  
Field Code Changed  
Formatted: French (France)

Formatted: Not Superscript/ Subscript

Formatted: French (France)  
Formatted: French (France)  
Field Code Changed

54 areas of many Chinese cities, ~~spanning~~ from Beijing in the north to Hong Kong in the south (Wang  
55 et al., 2010; Lyu et al., 2016b; Xu et al., 2017). However, in ~~the~~ rural or background regions, O<sub>3</sub>  
56 formation is generally NO<sub>x</sub>-limited due to the relatively low NO<sub>x</sub> (high VOCs/NO<sub>x</sub> ratio) in the  
57 atmosphere (Pan et al., 2015; Xu et al., 2015). ~~The~~ NO<sub>x</sub>-limited regime ~~is~~ ~~has~~ also ~~been~~ observed  
58 ~~or~~ simulated in some cities of western China (Xue et al., 2014; Feng et al., 2016). Regardless of  
59 the regimes controlling O<sub>3</sub> formation, the responses of O<sub>3</sub> formation to ~~the~~ precursors are generally  
60 non-linear, ~~which is~~ ~~has been~~ revealed by ~~the~~ empirical kinetic modeling approach (EKMA)  
61 curves (Lin et al., 1988; Jia et al., 2016; Lyu et al., 2016a). In ~~the~~ a VOC-limited regime, alkenes,  
62 aromatics, and biogenic VOCs ~~are~~ ~~have been~~ repeatedly confirmed as the main contributors to O<sub>3</sub>  
63 production. An et al. (2012) indicated that aromatics and alkenes contributed 73 - 84% to ~~the~~ O<sub>3</sub>  
64 production in summer 2008 in Beijing. Han et al. (2013) found that alkenes were the largest  
65 contributors to O<sub>3</sub> production (53.3%), followed by aromatics (35.1%) and alkanes (9.2%), in  
66 autumn 2009 in Tianjin. Geng et al. (2007) reported that aromatics were the main contributor (79%)  
67 to ~~the~~ O<sub>3</sub> production in November 2005 in Shanghai. The sources of VOCs ~~in China were~~ ~~have~~  
68 also ~~been~~ widely investigated ~~in China~~ (Guo et al., 2007; Yuan et al., 2010; Lyu et al., 2016b; Gao  
69 et al., 2017). ~~The contributions of different sources to VOCs vary significantly among cities and~~  
70 regions, but ~~the~~ vehicular and industrial emissions are generally the ~~most~~ predominant sources of  
71 ambient VOCs in China (see Table S1 for details). For instance, Shao et al. (2016) indicated that  
72 ~~in summer 2013 in Nanjing~~, industry-related sources (52%) ~~was~~ ~~were~~ the largest contributor to  
73 VOCs, followed by vehicular emissions (34%) ~~in summer 2013 in Nanjing~~. Yuan et al. (2009)  
74 reported that vehicular emissions made the largest contribution to ambient VOCs ~~in~~ ~~at~~ both urban  
75 (62%) and rural sites (36%) in 2006 in Beijing. Ling and Guo (2014) found that vehicular-related  
76 sources, including gasoline exhaust (22%), diesel exhaust (20%), LPG usage (21%), and gasoline  
77 evaporation (8%), were the dominant contributors to VOCs in autumn 2010 in Hong Kong.

78 Wuhan is the largest city in Central China; ~~it is located in an area which is~~ dominated by the  
79 ~~moist~~ subtropical ~~moist~~ monsoon climate. ~~With~~ ~~Due to~~ rapid industrialization and urbanization,  
80 ~~the~~ O<sub>3</sub> pollution has become increasingly severe in Wuhan. Ozone was identified as the main air

Formatted: French (France)

Field Code Changed

Formatted: French (France)

Field Code Changed

Formatted: French (France)

Formatted: French (France)

Formatted: French (France)

Field Code Changed

Formatted: French (France)

81 pollutant on 44 days in 2015 and on 34 days in 2016 in Wuhan, when the O<sub>3</sub> un-attainment rate  
82 ~~is not being attained~~ was 15.7% and 9.3% according to China's National Ambient Air Quality  
83 Standard Level II (200 μg m<sup>-3</sup>), respectively (accessible at  
84 <http://www.whepb.gov.cn/hbHjzkgb/19431.jhtml>). Higher O<sub>3</sub> values were generally observed in  
85 summer, with ~~the a~~ peak concentration of 121 ppbv in 2015 and 116 ppbv in 2016. As such, Wuhan  
86 ~~is faces major ing great challenges in for~~ O<sub>3</sub> abatement, ~~just like the~~ other fast-developing cities  
87 and regions in China.

Formatted: Highlight

Formatted: Highlight

89 This study aimed to characterize ~~the~~ O<sub>3</sub> pollution during high-O<sub>3</sub> and non-high-O<sub>3</sub> days so as  
90 to understand the main causes of high O<sub>3</sub> days in Wuhan. ~~Specifically, the~~ concentrations and  
91 compositions of VOCs and NO<sub>x</sub> ~~were are~~ firstly compared between high-O<sub>3</sub> and non-high-O<sub>3</sub> days.  
92 ~~Second~~ ~~Nextly~~, ~~the a~~ positive matrix factorization (PMF) receptor model coupled with a  
93 photochemical box model incorporating the Master Chemical Mechanism (PBM-MCM) ~~was is~~  
94 applied to investigate the major sources and their contributions to ambient VOCs and  
95 photochemical O<sub>3</sub> formation during high-O<sub>3</sub> days and non-high-O<sub>3</sub> days. ~~Lastly~~ ~~Finally~~, ~~the~~ O<sub>3</sub>-  
96 precursor relationships ~~were are~~ evaluated ~~with using a the approach of~~ relative incremental  
97 reactivity (RIR) ~~approach~~, and a reduction ratio of VOCs/NO<sub>x</sub> ~~was is~~ recommended for effective  
98 mitigation of O<sub>3</sub> pollution in Wuhan. The outcomes are expected to improve our understanding of  
99 photochemical O<sub>3</sub> formation and transport mechanisms for subtropical regions ~~that have with~~ moist  
100 monsoon climates ~~and a~~ complex coupling of meteorology and chemistry. Our results also have  
101 implications for other subtropical inland regions around the world.

## 102 2. Methodology

### 103 2.1 Sampling site

104 The sampling site is located in the urban area of Wuhan (30.604° N, 114.278° E), which is  
105 adjacent (~70 m) to a main traffic road and surrounded by residential and commercial blocks (Fig.  
106 1). The instruments were deployed on the roof of a nine-story building (~25 m a.g.l.), ~~the of~~ Wuhan  
107 Research Institute of Environmental Protection.

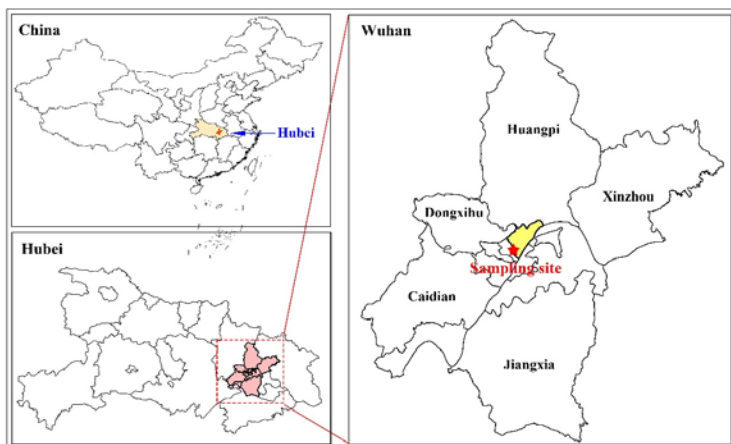


Fig. 1. Geographical location of the sampling site

## 2.2 Measurement techniques

### 2.2.1 On-line measurement of ambient VOCs

102 VOCs species, including 58 non-methane hydrocarbons (NMHCs), 31 halocarbons, and 13 oxygenated VOCs (OVOCs), were analyzed using an online gas chromatography-mass spectrometry-flame ionization detector system with one-hour time resolution (GC-MSD-FID, TH-PKU 300B; Tianhong Instrument Co., Ltd., Wuhan). This online system ~~consists of~~ has four parts: ~~the, i.e.,~~ sampling system, pre-concentrator, GC-MSD-FID system, and data processing system. Ambient air was continuously drawn through a Teflon tube with an inner diameter of 0.76 cm, and a particle filter was connected to the sampling tube inlet to remove ambient particles. The sampling tube inlet was installed at 1~2 m above the rooftop, and the outlet was connected to a PFA-made manifold with a by-pass pump drawing air at a rate of 15 L min<sup>-1</sup>. ~~When the~~ A air samples were collected and transferred to the pre-concentrator; ~~;~~ water and carbon dioxide were removed at -80 °C and VOCs were trapped at -150 °C. After preconcentration, the VOCs were desorbed by rapidly heating up the trap to 100 °C, and then introduced into the GC-FID-MSD system (Li et al., 2015; Lyu et al., 2016). The C<sub>2</sub> - C<sub>4</sub> NMHCs were analyzed by the FID, while other compounds were quantified by MSD. The quality assurance and control (QA/QC) ~~is~~ are shown in Text S1.

## 126 2.2.2 Monitoring of trace gases other than VOCs and meteorological parameters

127 Five trace gases (O<sub>3</sub>, NO<sub>2</sub>, NO, CO<sub>2</sub> and SO<sub>2</sub>) were measured continuously from ~~+~~August 1 to  
128 31, ~~August~~ 2016, with a one-hour resolution ~~of one hour~~. Trace gases were monitored using ~~the~~  
129 instruments developed by Thermo Environmental Instruments (TEI) Inc. Specifically, O<sub>3</sub> was  
130 detected with a commercial UV photometric analyzer (TEI, Model 49i), which has a detection  
131 limit of 1.0 ppbv. Nitric oxide (NO) and NO<sub>2</sub> were analyzed by a chemiluminescence trace level  
132 analyzer (TEI, Model 42i) with a detection limit of 0.5 ppbv (Geng et al., 2009). A gas filter  
133 correlation CO analyzer (TEI, Model 48i) was used to detect ~~the~~ ambient CO. The detection limit  
134 of ~~the~~ CO analyzer was 0.04 ppm. Sulfur dioxide (SO<sub>2</sub>) was analyzed using a pulse fluorescence  
135 analyzer (TEI, Model 43i) with a detection limit of 0.05 ppbv. The QA/QC of trace gases can be  
136 found in Text S1. Weather parameters, including temperature, relative humidity, pressure, wind  
137 speed, and wind direction, were also monitored at the site ~~in~~ with hourly a one-hour resolution by  
138 the integrated sensor suite (Vantage Pro TM & Vantage Pro 2 Plus TM Weather Stations, Davis  
139 Instruments).

140 In this study, high-O<sub>3</sub> days were defined as those when the highest hourly O<sub>3</sub> concentration  
141 exceeded ~~the~~ China's National Ambient Air Quality Standard Level II of 200 µg m<sup>3</sup> (~100 ppbv)  
142 (accessible at [http://english.mep.gov.cn/Resources/standards/Air\\_Environment/quality\\_](http://english.mep.gov.cn/Resources/standards/Air_Environment/quality_standard1/201605/t20160511_337502.shtml)  
143 [standard1/201605/t20160511\\_337502.shtml](http://english.mep.gov.cn/Resources/standards/Air_Environment/quality_standard1/201605/t20160511_337502.shtml)). Otherwise, the sampling days were denoted as non-  
144 high-O<sub>3</sub> days. As such, four high-O<sub>3</sub> days, (August i.e., 1, 17, 24, and 30) ~~August~~, were captured  
145 during the entire sampling period.

## 146 2.3 Model description

### 147 2.3.1. PMF model

148 The U.S. Environmental Protection Agency (EPA) Positive Matrix Factorization (PMF, version  
149 5.0) model was applied to identify the sources of ambient VOCs and to quantify their contributions  
150 to VOCs. PMF is a multivariate factor analysis tool that decomposes a matrix of speciated sample  
151 data into two matrices: factor contributions (G) and factor profiles (F). Here, ~~the~~ “factor  
152 contribution” means the contribution of a source to the total measured VOC concentration in a

153 sample, while “factor profile” indicates the percentages of individual VOC species in the total  
154 measured VOC concentration in a sample. To ascribe the concentrations of  $m$  chemical species in  
155  $n$  samples to  $p$  independent sources, the receptor model can be generally expressed as follows (Eq.  
156 1):

$$x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (\text{Eq. 1})$$

158 where  $x_{ij}$  (unit: ppbv) is the concentration of the  $j$ th species in the  $i$ th sample,  $g_{ik}$  (unit: ppbv) is the  
159 contribution of the  $k$ th source to the  $i$ th sample,  $f_{kj}$  is the fraction of  $j$ th species in  $k$ th source, and  
160  $e_{ij}$  is the residual for the  $i$ th sample of the  $j$ th species.

161 The factor profiles were derived based on the principle that the objective function  $Q$  ~~had-should~~  
162 have the lowest value, as shown in Eq. (2) (Paatero and Tapper, 1994; Paatero, 1997). In addition,  
163 the profiles should have the best interpretability.

$$Q = \sum_{i=1}^n \sum_{j=1}^m \left[ \frac{e_{ij}}{u_{ij}} \right]^2 \quad (\text{Eq. 2})$$

165 where  $u_{ij}$  is the uncertainty of the  $i$ th species in the  $j$ th sample. The function value,  $Q(\text{robust})$ , was  
166 calculated by-using the model through-by excluding the points with ~~the~~ uncertainty-scaled residuals  
167 greater than 4. The profile with the lowest value of  $Q(\text{robust})$  was selected as the optimum solution.

168 In this study, speciated VOCs including 20 NMHCs, chloromethane, and  
169 ~~trichlorethylenetrichloroethylene~~ from 565 samples ( $m$ ) were input into the PMF model. The 22  
170 species ( $n$ ) were selected because most of them are tracers of different sources. ~~Since~~Because  
171 accuracies in measurements for all species were <10% (Table S1), we used 10% as input to  
172 represent the uncertainty of each species in each sample in the model. Different numbers of factors  
173 ~~was-were~~ tested, and an optimum solution was determined based on ~~a~~ good fit to the observed data  
174 and the most meaningful results. Eventually, seven factors ( $p$ ) were extracted from the PMF model  
175 simulation.

### 176 2.3.2 PBM-MCM model

177 A zero-dimension Lagrangian photochemical box model incorporating a Master Chemical  
178 Mechanism (PBM-MCM) was applied to investigate ~~the~~ in situ photochemical  $\text{O}_3$  formation in

179 Wuhan in August 2016. This model simulates the complex photochemistry within a well-mixed  
180 boundary layer air parcel, including 5,900 species and 16,500 reactions. All ~~the~~ air pollutants were  
181 assumed to be well mixed in the atmospheric boundary layer defined by the model. It should be  
182 noted that ~~the~~ vertical and horizontal dispersions were not considered in the model. ~~The~~ This model  
183 was described in details by Lam et al. (2013) and its reliability has been repeatedly confirmed  
184 ~~reliable~~ in previous studies (Cheng et al., 2013; Ling et al., 2014; Lyu et al., 2016b). In this study,  
185 the observed hourly values of 52 VOCs and four trace gases (NO<sub>2</sub>, NO, SO<sub>2</sub>, and CO), ~~as well~~  
186 ~~as in addition to~~ temperature and relative humidity, between 00:00 local standard time (LST) and  
187 23:00 LST were used to construct the model. The photolysis rates of different chemicals in the  
188 model were parameterized using the photon flux determined from the Tropospheric Ultraviolet  
189 and Visible Radiation (TUVv5) model (Lam et al., 2013). The simulations of daytime hours (07:00  
190 - 19:00 LST) were conducted for the entire sampling period except for the rainy days, and the  
191 model output was the simulated hourly mixing ratios of O<sub>3</sub>. The in situ O<sub>3</sub> production, the O<sub>3</sub>-NO<sub>x</sub>-  
192 VOCs relationship, and the control strategies of O<sub>3</sub> were all simulated using the PBM-MCM model.  
193 ~~The Detailed descriptions~~ of the modeling results ~~were~~ are shown in section 3.2.2 and section  
194 3.4.

195 In addition to the simulation of O<sub>3</sub> production, the PBM-MCM model was applied to investigate  
196 ~~the~~ O<sub>3</sub>-precursor relationships ~~through~~ by examining the relative incremental reactivity (RIR<sub>s</sub>) of  
197 O<sub>3</sub> precursors. The RIR is the percentage change in daytime O<sub>3</sub> production per percent change in  
198 precursor. ~~The~~ O<sub>3</sub> precursors were divided into four groups: ~~i.e.~~, anthropogenic VOCs (AVOCs),  
199 BVOCs, CO<sub>2</sub> and NO<sub>x</sub>. A positive RIR ~~of~~ for a given precursor means that reducing emissions of  
200 this precursor will significantly reduce O<sub>3</sub> production, whereas a negative RIR indicates the  
201 opposite effect, i.e., cutting emissions of the precursor will enhance O<sub>3</sub> production. In general, the  
202 RIRs of VOCs are positive, while both positive and negative RIRs are commonly seen for NO<sub>x</sub>  
203 due to its dual role in O<sub>3</sub> chemistry. The VOC-limited (VOC-sensitive) regime is indicated by the  
204 positive RIRs of VOCs and negative RIRs of NO<sub>x</sub>. Otherwise, O<sub>3</sub> formation is limited by (sensitive  
205 to) NO<sub>x</sub> or both VOCs and NO<sub>x</sub> when the RIR of NO<sub>x</sub> is positive.



206 –The RIR for precursor X at site “s” is calculated using Eq. 3 (Cardelino and Chameides, 1995):

$$207 \quad \text{RIR}^S(X) = \frac{P_{\text{O}_3\text{-NO}}^S(X) - P_{\text{O}_3\text{-NO}}^S(X - \Delta X)}{\Delta S(X)/S(X)} \quad (\text{Eq. 3})$$

208 where X represents a specific precursor (i.e. VOCs, NO<sub>x</sub>, or CO); the superscript “S” is used  
209 to denote the specific site where the measurements were made; S(X) is the measured mixing ratio of  
210 species X (ppbv); ΔS(X) is the hypothetical change in the mixing ratio of X, which is artificially  
211 set ~~as~~ at 10%; P<sub>O<sub>3</sub>-NO</sub><sup>S</sup>(X) represents the simulated O<sub>3</sub> production in the base run constructed with  
212 the observed concentrations of all ~~the~~ O<sub>3</sub> precursors, including ~~the~~ species X. P<sub>O<sub>3</sub>-NO</sub><sup>S</sup>(X – ΔX) is  
213 the simulated O<sub>3</sub> production in the constrained run with 10% reduction of species X (Ling et al.,  
214 2013; Wang et al., 2018; Xu et al., 2017), while the concentrations of other species remained  
215 unchanged. The titration of O<sub>3</sub> by NO was considered in both runs. Then, the average RIR value  
216 ( $\overline{\text{RIR}}$ ) for the species X over multiple sampling days was calculated using Eq. (4). ~~where~~ where NS  
217 means the number of days simulated.

$$218 \quad \overline{\text{RIR}} = \frac{\sum_1^{\text{NS}} [\text{RIR}^S(X) P_{\text{O}_3\text{-NO}}^S(X)]}{\sum_1^{\text{NS}} P_{\text{O}_3\text{-NO}}^S(X)} \quad (\text{Eq. 4})$$

### 219 2.3.3 WRF-CMAQ model

220 The U.S. EPA Community Multiscale Air Quality (CMAQ) model with the Carbon Bond 05  
221 (CB05) chemical mechanism was applied to simulate ~~the~~ O<sub>3</sub> concentrations during ~~the~~ high-O<sub>3</sub>  
222 days. CMAQ is a three-dimensional atmospheric chemistry and transport modeling system; ~~and~~  
223 more details ~~about the CMAQ model~~ can be found at <http://cmascenr.org/cmaq/>. In this study,  
224 CMAQ was driven by the meteorological fields from the Weather Research and Forecasting (WRF)  
225 model, which is one of the most ~~currently commonly used~~ available models ~~to apply~~ for both  
226 forecasting and atmospheric research across scales ranging from meters to thousands of kilometers.  
227 The simulation period was 00:00 LST, August 13, 2016, from 00:00 LST 13 August 2016 to 23:00  
228 LST, August 18, 2016, 23:00 LST 18 August 2016 when continuously high O<sub>3</sub> values were  
229 observed, with the highest hourly O<sub>3</sub> of 133.9 ppbv on 17 August 17 (Fig. S1). The purpose ~~to of~~  
230 running the model simulations for ~~the six~~ 6 consecutive days was to ensure that the modeling could

231 achieve ~~the a~~ balance of ~~the~~ intermediates (e.g. radicals) and reduce ~~the~~ integration error in  
232 numerical calculations (Jiang et al., 2008; Pan et al., 2015; Chen et al., 2018). Two nested domains  
233 were defined in the WRF modeling, and the corresponding grid spacings were 27 km and 9 km,  
234 ~~receptivity~~ respectively. The outer domain (DM1, 27 km) covered most ~~territory~~ of China, and the  
235 second domain (DM2, 9 km) included most cities in eastern Hubei Province, and some parts of  
236 Anhui, Henan, and Jiangxi Provinces. Vertically, there were 31 sigma levels, with the top pressure  
237 fixed at 100 hPa. The meteorological parameters were well considered by the model (Jiang et al.,  
238 2010). The anthropogenic emission inventory developed by Tsinghua University based on figures  
239 for the year of 2010 was applied for this study, which includes monthly anthropogenic emissions  
240 of SO<sub>2</sub>, NO<sub>x</sub>, CO, ammonia (NH<sub>3</sub>), PM<sub>2.5</sub>, PM<sub>coarse</sub>, black carbon (BC), organic carbon (OC)<sub>2</sub> and  
241 non-methane volatile organic compounds (NMVOCs). The data ~~has~~ have a horizontal resolution  
242 of 0.25° × 0.25°. More details are given by He (2012). The biogenic emissions were calculated  
243 offline using the Model of Emissions of Gases and Aerosols from Nature (MEGAN, version 2.04)  
244 (Guenther et al., 2006). These modeling systems have been successfully used in previous O<sub>3</sub>  
245 simulations by Jiang et al. (2010) and Wang et al. (2015).

### 246 3 Results and discussion

#### 247 3.1 Characteristics of meteorological parameters and air pollutants

248 Table S2 shows the descriptive statistics of meteorological parameters on high-O<sub>3</sub> and non-high-  
249 O<sub>3</sub> days. The occurrence of high-O<sub>3</sub> days was associated with high temperature, low wind speed,  
250 and low relative humidity. Fig. S3 presents the distributions of mean sea level (MSL) pressure and  
251 wind fields averaged over high-O<sub>3</sub> days and non-high-O<sub>3</sub> days in East Asia. It ~~is shown~~ indicates  
252 that the high-O<sub>3</sub> days were all related to tropical cyclones. ~~This is because~~ This relation exists  
253 becausee reason is that Wuhan ~~was is~~ located on the periphery of ~~the~~ tropical cyclones, where  
254 relatively high pressure and subsiding air ~~were are~~ formed, causing O<sub>3</sub> accumulation. In addition,  
255 ~~the~~ east to north winds induced by the tropical cyclones also brought ~~the~~ polluted air masses from  
256 East China and North China to Wuhan on these days. A detailed discussion is given in Text S2 in  
257 the Supplement.

Formatted: Font: Not Italic

258 The time series of observed O<sub>3</sub> concentrations in Wuhan in August 2016 is shown in Fig. S1.  
259 ~~According to~~ Based on the definition of a high-O<sub>3</sub> day, August 1, 17, 24, and 30 ~~August~~ were  
260 defined as high-O<sub>3</sub> days during the sampling period. Table S3 shows the statistical descriptions of  
261 air pollutants on high-O<sub>3</sub> and non-high-O<sub>3</sub> days. During high-O<sub>3</sub> days, the O<sub>3</sub> mixing ratio was  
262  $61.2 \pm 7.4$  ppbv, about 40% higher than ~~that ( $42.3 \pm 2.0$  ppbv)~~ on non-high-O<sub>3</sub> days ( $42.3 \pm 2.0$   
263 ppbv;  $p < 0.01$ ). The mixing ratios of O<sub>3</sub> precursors, i.e., CO and total VOCs (TVOCs) representing  
264 the sum of mixing ratios of 102 VOC species, were  $987.8 \pm 95.4$  and  $43.9 \pm 3.8$  ppbv on high-O<sub>3</sub>  
265 days, respectively, about 37% and 33% higher than ~~those~~ on non-high-O<sub>3</sub> days ( $p < 0.01$ ),  
266 respectively. However, the levels of NO and NO<sub>2</sub> were comparable between high-O<sub>3</sub> and non-high-  
267 O<sub>3</sub> days (the differences  $< 2$  ppbv,  $p > 0.05$ ). Previous studies indicated that O<sub>3</sub> formation is  
268 generally limited by VOCs and suppressed by NO<sub>x</sub> in urban areas (Wang et al., 2008; Ling et al.,  
269 2013; Lyu et al., 2016b). Hence, the more abundant TVOCs and CO during high-O<sub>3</sub> days and  
270 comparable NO<sub>x</sub> values between high-O<sub>3</sub> and non-high-O<sub>3</sub> days might indicate the significant  
271 contribution of local photochemical reactions to O<sub>3</sub> formation.

272 Fig. S2 shows the diurnal variations of O<sub>3</sub>, NO<sub>x</sub>, and *m,p*-xylene/ethylbenzene ratio on high-O<sub>3</sub>  
273 and non-high-O<sub>3</sub> days. The observed O<sub>3</sub> generally peaked in the afternoon, and had relatively low  
274 concentrations from midnight to the early morning on both high-O<sub>3</sub> and non-high-O<sub>3</sub> days. A  
275 trough of O<sub>3</sub> appeared at about 08:00 LST, which was likely caused by ~~the~~ enhanced NO titration  
276 due to ~~the~~ increased vehicular emissions of NO ~~in~~ during morning rush hours (So and Wang, 2003).  
277 In fact, the diurnal variation of NO<sub>x</sub> was opposite to that of O<sub>3</sub>: ~~Namely, the~~ high levels of NO<sub>x</sub>  
278 appeared in the morning and evening when O<sub>3</sub> was relatively low, and the increase of O<sub>3</sub> from  
279 09:00 to 15:00 corresponded to the decrease of NO<sub>x</sub>. The opposite diurnal patterns between O<sub>3</sub> and  
280 NO<sub>x</sub> indicated that O<sub>3</sub> formation in Wuhan was in a VOC-limited (NO<sub>x</sub>-suppressed) regime. The  
281 diurnal trend of *m,p*-xylene/ethylbenzene was also negatively correlated to O<sub>3</sub>. ~~Since~~ Because *m,p*-  
282 xylene is more reactive than ethylbenzene, the ratios of *m,p*-xylene/ethylbenzene ~~would~~ decreased  
283 when photochemical reaction occurred. Indeed, the lowest ratios were observed in early afternoon  
284 when O<sub>3</sub> peaked, while higher ratios were found in the morning and late afternoon, indicating fresh

Field Code Changed

Formatted: French (France)

Formatted: French (France)

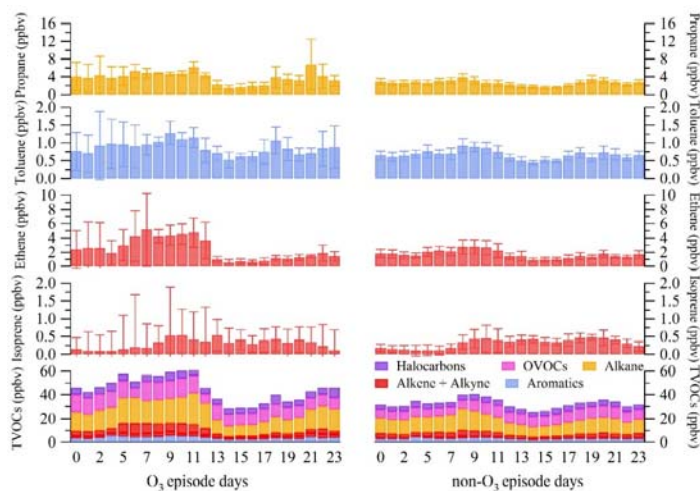
Formatted: Not Superscript/ Subscript

285 vehicular exhaust emissions and relatively weak photochemical consumption of VOCs.

286 Fig. 2 shows the diurnal variations of TVOCs, ethene, propane, toluene, and isoprene on O<sub>3</sub> and  
287 non-high O<sub>3</sub> days. In general, the patterns between high-O<sub>3</sub> and non-high-O<sub>3</sub> days for TVOCs and  
288 individual VOC species were similar, except for the magnitude. Moreover, TVOCs showed the  
289 same trend as ethene, propane, and toluene: ~~Namely~~, a bimodal structure was observed with the  
290 first peak in the morning and the second one in late afternoon, likely due to vehicular exhausts  
291 during rush hours and strong photochemical reactions at noon and in early afternoon. Clearly, the  
292 mixing ratios of TVOCs, ethene, propane, and toluene during high-O<sub>3</sub> days were about 37%, 31%,  
293 48%, and 30% higher than ~~these~~ during non-high-O<sub>3</sub> days ( $p < 0.01$ ). It ~~was~~ is unlikely that the local  
294 emissions of these air pollutants changed substantially between high-O<sub>3</sub> and non-high-O<sub>3</sub> days.

295 ~~Since~~ Because more O<sub>3</sub> precursors were usually consumed by intensive photochemical reactions  
296 on high-O<sub>3</sub> days, the more abundant O<sub>3</sub> precursors on high-O<sub>3</sub> days might indicate poor dispersion  
297 and dilution of air pollutants. Conversely, higher levels of isoprene were observed during daytime  
298 than ~~that~~ at night, suggesting that biogenic emission (BVOCs) of isoprene was more enhanced than  
299 its photochemical oxidation and dispersion ~~at~~ during daytime hours. Isoprene is often used as a  
300 tracer of biogenic sources, as it is the most abundant compound in vegetation emissions (Guenther  
301 et al., 2006; Sharkey et al., 2008). Mixing ratios of isoprene were comparable ( $p > 0.05$ ) on both  
302 high-O<sub>3</sub> and non-high-O<sub>3</sub> days, indicating that there was no significant difference in the biogenic  
303 emissions ~~had no significant difference~~ between the two scenarios.

Formatted: Not Highlight



304 Fig. 2. Average diurnal patterns of grouped VOCs on high-O<sub>3</sub> days (78 samples) and non-high-  
 305 O<sub>3</sub> days (487 samples)  
 306

### 307 3.2 Sources apportionment of ambient VOCs

#### 308 3.2.1 Source contribution to VOCs

309 Seven factors were identified by the PMF to best describe the sources of ambient VOCs; ~~i.e.,~~  
 310 gasoline exhausts, diesel and compressed natural gas (CNG) exhausts, stationary combustion, the  
 311 petrochemical industry, solvent usage in painting, asphalt application, and biogenic VOCs  
 312 (BVOCs) (see Text S3 and Fig. S4 in the supplement for further description). Table 1 lists the  
 313 source contributions to the total mixing ratios of VOCs ~~applied-used for~~ source apportionment  
 314 on high-O<sub>3</sub> days and non-high-O<sub>3</sub> days, in the forms of mixing ratio and percentage contribution.  
 315 Noticeably, ~~the~~ vehicle emissions, including gasoline exhaust, and diesel and CNG exhausts, were  
 316 the dominant sources of VOCs<sub>s</sub> with ~~the~~ total contributions of  $45.4 \pm 5.2\%$  ( $11.0 \pm 2.1$  ppbv) and  
 317  $37.3 \pm 2.9\%$  ( $7.1 \pm 1.0$  ppbv) during high-O<sub>3</sub> and non-high-O<sub>3</sub> days, respectively. The contribution  
 318 of vehicular exhausts to VOCs in this study was comparable to ~~those-that found~~ obtained at an  
 319 urban site in Hong Kong (Ling and Guo, 2014), while lower than ~~that~~ in Beijing (Liu et al., 2005).  
 320 ~~Besides~~ In addition, stationary combustion was ~~also~~ an important contributor to ambient VOCs,  
 321 with ~~the-a~~ contribution of  $31.5 \pm 4.5\%$  ( $5.6 \pm 0.6$  ppbv) during high-O<sub>3</sub> days and  $36.5 \pm 2.7\%$  (4.9

± 0.3 ppbv) during non-high-O<sub>3</sub> days. This ~~finding was~~ is not unreasonable, in view of the considerable coal consumption in Wuhan (more than 17 million tons in 2015) and intensive ~~burning of~~ biomass (e.g. crop residues) ~~burning~~ in and around Wuhan (Lyu et al., 2016).

In comparison, VOCs emitted from diesel and CNG exhausts and solvent usage in painting on high-O<sub>3</sub> days were much higher than ~~those~~ during non-high-O<sub>3</sub> days ( $p < 0.01$ ). However, gasoline exhaust, asphalt application, stationary combustion, and BVOCs made comparable contributions to VOCs between high-O<sub>3</sub> and non-high-O<sub>3</sub> days ( $p > 0.05$ ). In addition, VOCs attributable to the petrochemical industry ~~was~~ were even lower during high-O<sub>3</sub> days ( $p < 0.01$ ). However, ~~since~~ because the reactivity of VOC species in O<sub>3</sub> formation varies ~~in~~ over a wide range (Carter, 1994), we could not conclude that the sources with higher contributions to VOCs (vehicle exhausts, stationary combustion, and solvent usage) were responsible for the occurrence of high-O<sub>3</sub> days. In other words, further investigation was needed into the contributions of VOC sources to O<sub>3</sub> production ~~needed further investigation~~.

Table 1 Source contributions to VOCs on high-O<sub>3</sub> days and non-high-O<sub>3</sub> days  
(mean ± 95% confidence interval)

	Mixing ratios (ppbv)		Contribution (%)	
	High-O <sub>3</sub> days	Non-high-O <sub>3</sub> days	High-O <sub>3</sub> days	Non-high-O <sub>3</sub> days
Gasoline exhaust	4.7 ± 0.9	3.7 ± 0.5	23.0 ± 4.0	20.9 ± 1.9
Diesel and CNG exhausts	6.3 ± 1.6	3.4 ± 0.7	22.4 ± 4.1	16.4 ± 2.2
(Vehicle exhausts)	(11.0 ± 2.1)	(7.1 ± 1.0)	(45.4 ± 5.2)	(37.3 ± 2.9)
BVOCs	0.5 ± 0.1	0.5 ± 0.1	3.0 ± 0.6	3.8 ± 0.5
Stationary combustion	5.6 ± 0.6	4.9 ± 0.3	31.5 ± 4.5	36.5 ± 2.7
Asphalt application	1.5 ± 0.6	1.6 ± 0.4	7.8 ± 2.5	9.7 ± 2.2
Solvent usage in painting	1.5 ± 0.3	0.8 ± 0.1	7.9 ± 1.2	5.3 ± 0.6
Petrochemical	0.9 ± 0.2	1.1 ± 0.1	4.4 ± 0.8	7.4 ± 0.7

337

338 **3.2.2 Source contributions to O<sub>3</sub> production**

339 To further understand the contributions of different VOCs sources to O<sub>3</sub> formation, the in situ  
340 production of O<sub>3</sub> was simulated ~~by using~~ the PBM-MCM model. ~~Since Because the~~ photolysis  
341 rates could not be well calibrated on rainy days and wet deposition was not considered in the model,  
342 there ~~generally~~ were ~~often major~~ ~~great~~ discrepancies between ~~the~~ observed and simulated O<sub>3</sub> on  
343 rainy days. Therefore, ~~the~~ days with precipitation (in total 10 days) were excluded in ~~the~~ O<sub>3</sub>  
344 simulation. In addition, the instruments were maintained on some days to guarantee ~~the~~ data quality.  
345 These days were also ~~not omitted from the~~ ~~included in~~ O<sub>3</sub> simulation. Fig. S5 in the supplement  
346 shows the model validation results. ~~It was found that~~ ~~These indicate that that~~ the PBM-MCM model  
347 well reproduced the levels and diurnal trends of O<sub>3</sub>, implying that O<sub>3</sub> on these days was mainly  
348 locally formed. However, O<sub>3</sub> was only simulated ~~in this study at for~~ daytime hours (07:00 - 19:00  
349 LST) ~~in this study~~. In fact, O<sub>3</sub> simulation by a chemical transport model in the same period  
350 indicated that while local formation was responsible for daytime O<sub>3</sub>, regional transport elevated  
351 nocturnal O<sub>3</sub> (more details ~~shown are given~~ in 3.3).

352 Two scenarios for O<sub>3</sub> simulation (~~i.e.~~ a base scenario and a constrained scenario) were run to  
353 investigate the contribution of a specific VOC source to ~~the~~ O<sub>3</sub> production. The observed VOCs  
354 were used to construct the base scenario, while in the constrained scenario VOCs emitted from an  
355 individual specific source were deducted. The difference in simulated O<sub>3</sub> between the base and  
356 constrained scenarios was the net O<sub>3</sub> production of the specific source. In this way, the contribution  
357 of each VOC source to O<sub>3</sub> production was obtained. It is noteworthy that the data on the ~~four~~  
358 high-O<sub>3</sub> days were included for source apportionment. ~~Since Because~~ species with high reactivity  
359 participate more in photochemical reactions during high-O<sub>3</sub> days, their contributions to O<sub>3</sub>  
360 production were ~~partly somewhat~~ underestimated.

361 Table 2 summarizes the contributions of VOC sources to O<sub>3</sub> production on high-O<sub>3</sub> days and  
362 non-high-O<sub>3</sub> days. It was found that vehicle exhausts (exhausts from gasoline-, diesel- and CNG-

363 fueled vehicles) made the largest contribution to O<sub>3</sub> production, with 34.0 ± 1.4% (13.9 ± 2.2 ppbv)  
 364 during high-O<sub>3</sub> days and 29.5 ± 0.7% (8.3 ± 0.9 ppbv) during non-high-O<sub>3</sub> days. However,  
 365 compared to their contributions to VOCs (45.4 ± 5.2% during high-O<sub>3</sub> days, and 37.3 ± 2.9%  
 366 during non-high-O<sub>3</sub> days;  $p < 0.01$ ) the contribution of vehicle exhausts to O<sub>3</sub> production was lower  
 367 ( $p < 0.01$ ). ~~On the contrary~~In contrast, the contributions to O<sub>3</sub> production of the petrochemical  
 368 industry, solvent usage in painting, and BVOCs were obviously higher than their contributions to  
 369 VOCs ( $p < 0.01$ ). This discrepancy might be caused by the source-specific compositions of VOCs  
 370 and the different reactivity of VOCs in O<sub>3</sub> formation. More importantly, we noticed that O<sub>3</sub>  
 371 produced by vehicle exhausts (the sum of exhausts from gasoline-, diesel- and CNG-fueled  
 372 vehicles) and stationary combustion all increased significantly during high-O<sub>3</sub> days ( $p < 0.05$ ).  
 373 ~~Particularly~~In particular, O<sub>3</sub> produced by VOCs from diesel and CNG exhausts increased from  
 374 13.7 ± 0.6% (3.9 ± 0.4 ppbv) during non-high-O<sub>3</sub> days to 19.2 ± 1.3% (7.8 ± 1.3 ppbv) during  
 375 high-O<sub>3</sub> days. This finding ~~implied~~ that emission of VOCs from diesel and CNG exhausts was  
 376 the main culprit ~~of for the~~ elevated O<sub>3</sub> during-on high-O<sub>3</sub> days.

377 Table 2 Source contributions to O<sub>3</sub> production on high-O<sub>3</sub> days and non-high-O<sub>3</sub> days  
 378 (mean ± 95% confidence interval)

	Net O <sub>3</sub> increment (ppbv)		Contribution (%)	
	High-O <sub>3</sub> days	Non-high-O <sub>3</sub> days	High-O <sub>3</sub> days	Non-high-O <sub>3</sub> days
Gasoline exhaust	6.1 ± 0.9	4.5 ± 0.4	14.8 ± 0.3	15.8 ± 0.5
Diesel and CNG exhausts	7.8 ± 1.3	3.8 ± 0.4	19.2 ± 1.3	13.7 ± 0.6
(Vehicle exhausts)	(13.8 ± 2.2)	(8.3 ± 0.8)	(34.0 ± 1.4)	(29.5 ± 0.7)
BVOCs	6.9 ± 1.1	6.0 ± 0.8	16.7 ± 0.8	20.3 ± 1.1
Stationary combustion	5.3 ± 0.7	3.8 ± 0.4	13.3 ± 0.2	12.7 ± 0.3
Asphalt application	3.3 ± 0.5	2.7 ± 0.4	8.9 ± 1.1	8.5 ± 0.9
Solvent usage in painting	5.9 ± 0.9	4.8 ± 0.5	14.6 ± 0.5	16.4 ± 0.8
Petrochemical industry	4.9 ± 0.7	3.8 ± 0.4	12.5 ± 0.5	12.8 ± 0.3



379

### 380 3.3 Contributions of local formation and regional transport to O<sub>3</sub> levels

381 To understand the causes of high-O<sub>3</sub> days in Wuhan during the sampling period, the local  
382 formation and regional transport of O<sub>3</sub> during the continuous sunny days from August 13 ~~August~~  
383 to 18 ~~August~~ were simulated as a case study with using the WRF-CMAQ model ~~as a case study~~.  
384 One of the high-O<sub>3</sub> days (~~17 August~~) discussed in section 3.1.2 (August 17) was involved. This  
385 case was selected to represent the general characteristics of high-O<sub>3</sub> days during the sampling  
386 period, as all the high-O<sub>3</sub> days occurred under similar meteorological conditions (see section 3.1.2).  
387 The processes influencing O<sub>3</sub> concentration, i.e. advection and diffusion in both the horizontal and  
388 ~~vertical directions, the dry deposition,~~ and chemical reactions, were fully considered in the WRF-  
389 CMAQ model. The model configurations were consistent with those described in Jiang et al.  
390 (2010). Figure 4 shows the hourly variations of the simulated and observed O<sub>3</sub> through August 13  
391 to 18, 2016, in Wuhan. The model well reproduced the observed O<sub>3</sub> in both magnitude and diurnal  
392 pattern, with the an index of agreement of 0.79 (Willmott, 1981).

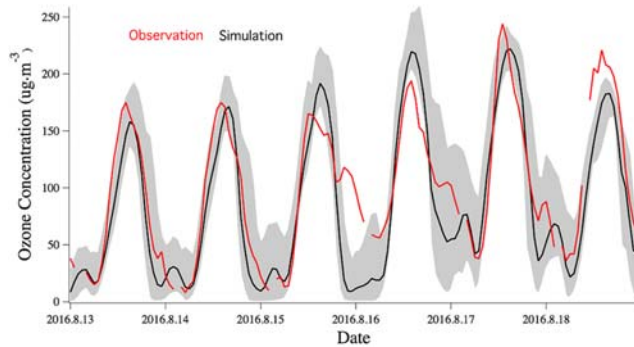
393 It is noteworthy that, the WRF-CMAQ overestimated O<sub>3</sub> on August 15-16 ~~August~~, but  
394 underestimated O<sub>3</sub> on August 13, 17, and 18 ~~August~~ (Figure 4). The PBM-MCM model  
395 overestimated O<sub>3</sub> on August 17, 19, 24, and 31 ~~August~~ (Figure S5). Despite the discrepancies, the  
396 modeling results were acceptable, in view of the uncertainties inherent in model simulations. For  
397 the WRF-CMAQ model, the uncertainty mainly derives from the uncertainty of the emission  
398 inventory and the carbon bond chemical mechanisms. Though the PBM-MCM model is near-  
399 explicit in chemical mechanisms, it does not consider the physical processes. Therefore, it is  
400 ~~expected that~~ some differences exist can be expected between the observed O<sub>3</sub> and simulated O<sub>3</sub>  
401 by for both models. In fact, this discrepancy is common in almost all the studies using that use  
402 these chemical transport models and box models (e.g., Gao et al., 2016; Wang et al., 2017; Chen  
403 et al., 2018).

Formatted: Font: Not Italic

Field Code Changed

Formatted: French (France)

Formatted: French (France)



404  
 405 Fig. 4 Time series of simulated and observed O<sub>3</sub> concentrations in Wuhan from 00:00 LST, ~~August 13-August~~  
 406 ~~August 13-August~~ to 23:00 LST, ~~August 18-August~~ 2016. The grey area defines the lowest and  
 407 highest simulated O<sub>3</sub> at ~~eight~~ air quality monitoring stations (AQMSs) surrounding the  
 408 sampling site. The AQMSs are managed by ~~the~~ China National Environmental Monitoring  
 409 Centre.

410  
 411 ~~Furthermore,~~ The contributions of different processes to O<sub>3</sub> are summarized in Table 3. Overall,  
 412 ~~the~~ horizontal advection ( $-1.3 \pm 7.0$  ppbv) and diffusion ( $-0.1 \pm 0.02$  ppbv) made negative  
 413 contributions to O<sub>3</sub> mixing ratios. In addition,  $3.4 \pm 0.6$  ppbv of O<sub>3</sub> was removed through dry  
 414 deposition. In contrast, O<sub>3</sub> was elevated by  $3.9 \pm 7.5$  and  $14.2 \pm 1.2$  ppbv due to ~~the~~ vertical  
 415 advection and diffusion, respectively. As the largest contributor to O<sub>3</sub>, chemical reactions built up  
 416 the O<sub>3</sub> mixing ratio by  $42.3 \pm 6.1$  ppbv. It is noteworthy that the chemical reactions indicated the  
 417 process of local O<sub>3</sub> formation. ~~Since~~ ~~Because~~ advection and diffusion generally represent the  
 418 transport of air pollutants, the sum of horizontal and vertical advection/diffusion is referred to  
 419 ~~hereafter~~ as regional transport ~~hereafter~~ (Jiang et al., 2010; Li et al., 2012; Chen et al., 2018). ~~As a~~  
 420 ~~result,~~ ~~R~~ regional transport, local photochemical formation, and dry deposition accounted for  $44.4$   
 421  $\pm 5.8\%$ ,  $60.2 \pm 6.1\%$  and  $-4.6 \pm 0.5\%$  of the total simulated O<sub>3</sub>, respectively. More specifically, the  
 422 contribution of regional transport ( $59.1 \pm 9.9\%$ ) to nighttime O<sub>3</sub> ~~predominated over local formation~~  
 423 ( $42.7 \pm 10.1\%$ ) (20:00 - 06:00 LST), while daytime O<sub>3</sub> was mainly derived from ~~the~~ local

Formatted: Not Superscript/ Subscript

424 photochemical formation ( $74.7 \pm 5.8\%$ ), about twice the regional transport ( $32.2 \pm 5.4\%$ ). ~~It~~  
 425 ~~T~~therefore, ~~it would seem~~ ~~was concluded~~ that the high daytime O<sub>3</sub> on high-O<sub>3</sub> days in Wuhan ~~was~~  
 426 ~~is~~ primarily attributable to local formation, while regional transport from the east made significant  
 427 contributions to ~~the~~ nocturnal O<sub>3</sub>. In fact, compared to the same time slots on other days, the  
 428 observed nighttime O<sub>3</sub> increased by 4–12 ppbv ~~on August 15--17 August~~ (Figure 4), ~~in~~ contrast  
 429 to ~~the~~ limited nocturnal O<sub>3</sub> production from photochemical reactions and the consumption of O<sub>3</sub>  
 430 by some substances (e.g. NO and alkenes). As such, ~~the increase of nocturnal O<sub>3</sub> on these days~~  
 431 ~~was is~~ most likely attributable to regional transport, consistent with the modeling results.

Formatted: Font: Not Italic

432 Table 3 Factor contributions to the simulated O<sub>3</sub> during August 13-18, 2016, in Wuhan (Unit:  
 433 ppbv).

	Average $\pm$ 95% Confidence Interval	Maximum	Minimum	Standard deviation
Horizontal advection	$-1.3 \pm 7.0$	125.0	-111.5	42.7
Vertical advection	$3.9 \pm 7.5$	127.8	-128.3	45.5
Horizontal diffusion	$-0.1 \pm 0.02$	0.1	-0.4	0.1
Vertical diffusion	$14.2 \pm 1.2$	36.6	1.5	7.5
Dry deposition	$-3.4 \pm 0.6$	-0.01	-10.9	3.4
Chemical reactions	$42.3 \pm 6.1$	130.5	-8.0	36.9

434  
 435 **3.4 Implications ~~for~~ ~~on~~ control strategies**

436 To understand ~~the~~ O<sub>3</sub> formation mechanisms, the relationships between O<sub>3</sub> and its precursors  
 437 were explored using the PBM-MCM model. ~~The~~ O<sub>3</sub> precursors were divided into four groups,  
 438 ~~namely i.e.~~ anthropogenic volatile organic compounds (AVOCs), BVOCs, CO<sub>2</sub> and NO<sub>x</sub>; ~~then~~  
 439 their RIR values were calculated using Eq. 3. According to previous studies (Zhang et al., 2006;  
 440 Lyu et al., 2016b; Wang et al., 2018), the positive RIR value of VOCs and negative RIR value of  
 441 NO<sub>x</sub> indicates VOCs-limited O<sub>3</sub> production in ~~the~~ study area, while the negative RIR value of  
 442 VOCs and positive RIR value of NO<sub>x</sub> means ~~that production is~~ NO<sub>x</sub>-limited. Fig. S6 shows the  
 443 mean RIR values of O<sub>3</sub> precursors over the high-O<sub>3</sub> and non-high-O<sub>3</sub> days. It ~~was found~~ ~~indicates~~

444 that the average RIR values of VOCs (AVOCs and BVOCs) were positive, while RIR for NO<sub>x</sub> was  
445 negative, indicating that O<sub>3</sub> formation was limited by VOCs at this site on both high-O<sub>3</sub> days and  
446 non-high-O<sub>3</sub> days. ~~This means~~ ~~The implication is~~ that cutting VOCs could reduce O<sub>3</sub> production,  
447 while reducing NO<sub>x</sub> would lead to O<sub>3</sub> increase. ~~The~~ A VOC-limited regime was also identified in  
448 rural areas of Delhi, India (Kumar et al., 2017), and in urban and suburban Hong Kong (Lyu et al.,  
449 2016b). However, Lin et al. (2017) indicated that ~~the~~ summertime O<sub>3</sub> pollution in the southeastern  
450 U.S. was alleviated by controlling NO<sub>x</sub> emissions, indicating ~~the~~ a NO<sub>x</sub>-limited regime in O<sub>3</sub>  
451 formation. Jin et al. (2017) ~~It~~ ~~also found~~ ~~was also revealed~~ that O<sub>3</sub> production became  
452 increasingly sensitive to NO<sub>x</sub> in mid-latitude megacities in the northern hemisphere (e.g., New  
453 York, London, and Seoul) between 2005 and 2015 (~~Jin et al., 2017~~). It should be noted that a recent  
454 study (McDonald et al., 2018) pointed out the underestimation of VOCs emissions from ~~the~~  
455 volatile chemical products in industrialized cities in the U.S., meaning that the current  
456 understandings of O<sub>3</sub> formation mechanisms based on ~~the~~ pre-existing emission inventories might  
457 need further verifications. To sum up, ~~the~~ O<sub>3</sub>-VOC-NO<sub>x</sub> sensitivity should be investigated case by  
458 case. This study expanded ~~ed~~ the worldwide database of ~~the~~ O<sub>3</sub>-VOC-NO<sub>x</sub> sensitivity to include the  
459 situation in developing regions, where NO<sub>x</sub> ~~was is on at~~ high levels and might continue to increase  
460 if no stringent control strategies ~~were are taken put in place~~ (Jin et al., 2017).

461 Further, the RIR of AVOCs was much higher than that of BVOCs. ~~Namely~~ Thus, O<sub>3</sub> formation  
462 was more sensitive to AVOCs, which should ~~thus~~ be more focused focused on in O<sub>3</sub> abatement. In  
463 addition, relatively high average RIR value was found for CO, indicating that CO had a positive  
464 impact on O<sub>3</sub> formation in Wuhan, in contrast to the findings of Wang et al. (2017), who reported  
465 that the RIR value of BVOCs was much higher than CO in summer in Hong Kong.

466 To further investigate the role of NO<sub>x</sub> in O<sub>3</sub> chemistry, ~~the~~ O<sub>3</sub> productions with different levels  
467 of NO<sub>x</sub> were simulated using the PBM-MCM model. Table S5 shows the mean net O<sub>3</sub> production  
468 in the base case with the observed NO<sub>x</sub> and in the constrained cases with NO<sub>x</sub> being reduced by  
469 10%, 20%, 40%, 60%, 80%, 90%, 95%, and 98%. The observed VOCs were applied to all the  
470 cases without any reduction. It was found that ~~with the increase of as~~ NO<sub>x</sub> reduction percentages

Formatted: Not Highlight

Formatted: Not Highlight

471 increased from 10% to 95%, the simulated O<sub>3</sub> also increased, by a ~~with the increasing~~ percentage  
472 ~~raising rising from within the range of~~ 2% to 78%. ~~This implies~~ The implication is that O<sub>3</sub> formation  
473 is limited by VOCs in these cases. However, simulated O<sub>3</sub> decreased significantly when the  
474 reduction percentage of NO<sub>x</sub> increased from 95% to 98%, indicating that O<sub>3</sub> formation switched  
475 to the NO<sub>x</sub>-limited regime. The dual roles of NO<sub>x</sub> in O<sub>3</sub> formation ~~are is~~ clearly presented in these  
476 experiments.

477 ~~Since~~ Because VOCs (particularly AVOCs) and NO<sub>x</sub> are generally co-emitted, it is nearly  
478 impossible to only reduce VOC emissions ~~with while~~ NO<sub>x</sub> remains ~~ing~~ unchanged. Therefore, it is  
479 vital to find an appropriate ~~cutting reduction~~ ratio of VOCs/NO<sub>x</sub> ~~for, to~~ effectively controlling O<sub>3</sub>  
480 pollution. With the aid of the PBM-MCM model, the net O<sub>3</sub> variations were simulated in the  
481 scenarios with different ~~cutting~~ percentage ~~cuts of in~~ VOCs and NO<sub>x</sub>, based on the measured VOCs  
482 and NO<sub>x</sub>. The positive and negative variations indicated increases and decreases of O<sub>3</sub>, respectively.  
483 ~~Both~~ The ~~percentage cuts in~~ ~~cutting percentages of~~ VOCs and NO<sub>x</sub> were 0 - 50%, ~~with the step~~  
484 ~~of at~~ 10% ~~intervals~~. As shown in Fig. S7, the net O<sub>3</sub> variation increased non-linearly with the  
485 increase of ~~percentage cuts in~~ NO<sub>x</sub> ~~cutting percentages~~ and decreased with the increase of  
486 ~~percentage cuts in~~ VOCs ~~cutting percentages~~. This ~~finding was~~ ~~is~~ consistent with the finding that  
487 O<sub>3</sub> formation ~~was occurred~~ in a VOCs-limited regime during the study period in Wuhan. ~~To~~ ~~For~~  
488 ~~reduc~~ing O<sub>3</sub> production, the ~~percentage cuts~~ ~~sting percentages of in~~ VOCs and NO<sub>x</sub> ~~could only be~~  
489 ~~accepted~~ ~~were only considered acceptable~~ when the O<sub>3</sub> variations were nil or negative. According  
490 to Fig. S7, with the increase of ~~cutting~~ percentage ~~cuts ins of~~ VOCs from 10.0% to 50.0%, the  
491 maximum ~~cutting~~ percentage ~~cut ins of~~ NO<sub>x</sub> for nil O<sub>3</sub> increment increased from 13.7% to 51.0%,  
492 and the ratio of the VOCs ~~percentage cut~~ ~~cutting percentage~~ to NO<sub>x</sub> ~~percentage cut~~ ~~cutting~~  
493 ~~percentage~~ (termed ~~the as~~ "VOCs/NO<sub>x</sub> cutting ratio") increased from 0.73 to 0.98. Therefore, to  
494 maintain a nil/negative O<sub>3</sub> increment, the VOCs/NO<sub>x</sub> cutting ratio should be higher than 0.73,  
495 given that VOCs were cut by 10% - 50%. This finding could guide the formulation and  
496 implementation of effective O<sub>3</sub> control strategies in Wuhan.

497

Formatted: Highlight

#### 498 4. Conclusions

499 In this study, ~~the~~ continuously measured data of speciated VOCs, trace gases, and  
500 meteorological parameters in August 2016 at an urban site in Wuhan were analyzed to explore the  
501 different characteristics of VOC and O<sub>3</sub> pollution between high-O<sub>3</sub> and non-high-O<sub>3</sub> days. Results  
502 indicated that ~~the~~ high temperature, low wind speed, low relative humidity, tropical cyclones, and  
503 transport of polluted air masses from northern China to Wuhan were favorable for the occurrence  
504 of high-O<sub>3</sub> days during the study period. During high-O<sub>3</sub> days, the mixing ratios of O<sub>3</sub>, CO<sub>2</sub>, and  
505 TVOCs were  $61.2 \pm 7.4$ ,  $987.8 \pm 95.4$  and  $43.9 \pm 3.8$  ppbv, about 40%, 37%, and 33% higher than  
506 ~~these~~ during non-high-O<sub>3</sub> days ( $p < 0.01$ ), respectively. However, the levels of NO ( $\Delta\text{NO}$ :  $\sim 1$  ppbv)  
507 and NO<sub>2</sub> ( $\Delta\text{NO}_2$ :  $\sim 2$  ppbv) were comparable between high-O<sub>3</sub> and non-high-O<sub>3</sub> days ( $p > 0.05$ ).

508 Seven VOCs sources were identified, ~~including~~ gasoline exhaust, diesel and CNG exhausts,  
509 stationary combustion, petrochemical industry, solvent usage in painting, asphalt application, and  
510 BVOCs. Vehicle exhausts (i.e., gasoline exhaust, diesel and CNG exhausts) made the largest  
511 contribution to VOCs and O<sub>3</sub> production during both high-O<sub>3</sub> days ( $45.4 \pm 5.2\%$ ) and non-high-  
512 O<sub>3</sub> days ( $37.3 \pm 2.9\%$ ). The contributions of vehicle exhausts and stationary combustion to VOCs  
513 and O<sub>3</sub> production increased remarkably from non-high-O<sub>3</sub> days to high-O<sub>3</sub> days, suggesting that  
514 these sources were the culprit ~~of for~~ elevated O<sub>3</sub> during high-O<sub>3</sub> days. Ozone simulation ~~by using~~  
515 ~~the~~ WRF-CMAQ model indicated that while local formation ( $74.7 \pm 5.8\%$ ) accounted for most of  
516 the daytime O<sub>3</sub>, the nighttime O<sub>3</sub> was significantly elevated by regional transport ( $59.1 \pm 9.9\%$ ).  
517 As indicated by the relationship between O<sub>3</sub> and its precursors, O<sub>3</sub> formation in urban Wuhan was  
518 VOC-limited, and the sensitivity of O<sub>3</sub> formation to AVOCs ~~pre~~dominated over BVOCs and CO.  
519 To effectively control O<sub>3</sub> pollution, the reduction ratio between VOCs and NO<sub>x</sub> should be higher  
520 than 0.73. In addition, ~~since because~~ VOCs emitted from vehicle exhausts were the main  
521 contributors to O<sub>3</sub> production, it is urgent and ~~efficient effective~~ to control vehicle emissions ~~in~~  
522 ~~order to to~~ pursue O<sub>3</sub> ~~attainment reduction~~ in Wuhan. This study ~~fill~~ed the knowledge gap ~~of~~  
523 ~~concerning~~ the causes of high O<sub>3</sub> events in Central China, particularly the characterization of local  
524 and regional contributions to ~~the~~ high O<sub>3</sub>, which ~~would will~~ be of help for O<sub>3</sub> study and control in

525 other subtropical regions with moist monsoon climates.

## 526 Acknowledgements

527 This study was supported by the National Key R&D Program of China via grant No.  
528 2017YFC0212000, the Wuhan Environmental Monitoring Center via grant No. 250000975, and  
529 The Hong Kong Polytechnic University joint supervision scheme with the Chinese mainland,  
530 Taiwan and Macao Universities (project No: G-SB63). We ~~were~~ are grateful to the Research Grants  
531 Council of Hong Kong for its financial support via grants CRF/C5004-15E, PolyU152052/14E,  
532 PolyU152052/16E, and CRF/C5022-14G.

## 533 References

- 534 An, J.L., Wang, Y.S., Wu, F.K., Zhu, B., 2012. Characterizations of volatile organic compounds  
535 during high ozone episodes in Beijing, China. *Environ. Monit. Assess.* 184, 1879–1889.  
536 doi:10.1007/s10661-011-2086-7
- 537 Cardelino, C.A., Chameides, W.L., 1995. An observation-based model for analyzing ozone  
538 precursor relationships in the urban atmosphere. *J. Air Waste Manage. Assoc.* 45, 161–180.  
539 doi:10.1080/10473289.1995.10467356
- 540 Carter, W.P.L., 1994. Development of ozone reactivity scales for volatile organic compounds.  
541 *Air Waste* 44, 881–899. doi:10.1080/1073161X.1994.10467290
- 542 Chen, X., Liu, Y., Lai, A., Han, S., Fan, Q., Wang, X., Ling, Z., Huang, F., Fan, S., 2018.  
543 Factors dominating 3-dimensional ozone distribution during high tropospheric ozone  
544 period. *Environ. Pollut.* 232, 55–64. doi:10.1016/j.envpol.2017.09.017
- 545 Cheng, H.R., Saunders, S.M., Guo, H., Louie, P.K.K., Jiang, F., 2013. Photochemical trajectory  
546 modeling of ozone concentrations in Hong Kong. *Environ. Pollut.* 180, 101–110.  
547 doi:10.1016/j.envpol.2013.04.039
- 548 Feng, T., Bei, N., Huang, R.J., Cao, J., Zhang, Q., Zhou, W., Tie, X., Liu, S., Zhang, T., Su, X.,  
549 Lei, W., Molina, L.T., Li, G., 2016. Summertime ozone formation in Xi'an and surrounding  
550 areas, China. *Atmos. Chem. Phys.* 16, 4323–4342. doi:10.5194/acp-16-4323-2016
- 551 Gao, J., Zhu, B., Xiao, H., Kang, H., Hou, X., Shao, P., 2016. A case study of surface ozone  
552 source apportionment during a high concentration episode, under frequent shifting wind  
553 conditions over the Yangtze River Delta, China. *Sci. Total Environ.* 544, 853–863.  
554 doi:10.1016/j.scitotenv.2015.12.039
- 555 Gao, J., Zhu, B., Xiao, H., Kang, H., Hou, X., Yin, Y., Zhang, L., Miao, Q., 2017. Diurnal  
556 variations and source apportionment of ozone at the summit of Mount Huang, a rural site in  
557 Eastern China. *Environ. Pollut.* 222, 513–522. doi:10.1016/j.envpol.2016.11.031
- 558 Gao, W., Tie, X., Xu, J., Huang, R., Mao, X., Zhou, G., Chang, L., 2017. Long-term trend of O<sub>3</sub>  
559 in a mega City (Shanghai), China: Characteristics, causes, and interactions with precursors.

Formatted: French (France)

Formatted: French (France)

- 560 [Sci. Total Environ.](#) 603–604, 425–433. doi:10.1016/j.scitotenv.2017.06.099
- 561 Geng, F., Zhang, Q., Tie, X., Huang, M., Ma, X., Deng, Z., Yu, Q., Quan, J., Zhao, C., 2009.
- 562 Aircraft measurements of O<sub>3</sub>, NO<sub>x</sub>, CO, VOCs, and SO<sub>2</sub> in the Yangtze River Delta region.
- 563 [Atmos. Environ.](#) 43, 584–593. doi:10.1016/j.atmosenv.2008.10.021
- 564 Geng, F., Zhao, C., Tang, X., Lu, G., Tie, X., 2007. Analysis of ozone and VOCs measured in
- 565 Shanghai: A case study. [Atmos. Environ.](#) 41, 989–1001.
- 566 doi:10.1016/j.atmosenv.2006.09.023
- 567 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P.I., Geron, C., 2006. Estimates of
- 568 global terrestrial isoprene emissions using MEGAN ( Model of Emissions of Gases and
- 569 Aerosols from Nature ). [Atmos. Chem. Phys.](#) 6, 3181–3210.
- 570 Guo, H., Ling, Z.H., Cheung, K., Jiang, F., Wang, D.W., Simpson, I.J., Barletta, B., Meinardi, S.,
- 571 Wang, T.J., Wang, X.M., Saunders, S.M., Blake, D.R., 2013. Characterization of
- 572 photochemical pollution at different elevations in mountainous areas in Hong Kong. [Atmos.](#)
- 573 [Chem. Phys.](#) 13, 3881–3898. doi:10.5194/acp-13-3881-2013
- 574 Guo, H., So, K.L., Simpson, I.J., Barletta, B., Meinardi, S., Blake, D.R., 2007. C<sub>1</sub>–C<sub>8</sub> volatile
- 575 organic compounds in the atmosphere of Hong Kong: Overview of atmospheric processing
- 576 and source apportionment. [Atmos. Environ.](#) 41, 1456–1472.
- 577 doi:10.1016/j.atmosenv.2006.10.011
- 578 Han, S.Q., Zhang, M., Zhao, C.S., Lu, X.Q., Ran, L., Han, M., Li, P.Y., Li, X.J., 2013.
- 579 Differences in ozone photochemical characteristics between the megacity Tianjin and its
- 580 rural surroundings. [Atmos. Environ.](#) 79, 209–216. doi:10.1016/j.atmosenv.2013.06.045
- 581 He, K., 2012. Multi-resolution Emission Inventory for China (MEIC): ~~M~~model framework and
- 582 1990–2010 anthropogenic emissions, in: AGU Fall Meeting 2012. [Abstract ID: A32B-](#)
- 583 [05](#). ~~Please give full citation information: 2012AGUFM.A32B.05H.~~
- 584 Jia, C., Mao, X., Huang, T., Liang, X., Wang, Y., Shen, Y., Jiang, W., Wang, H., Bai, Z., Ma,
- 585 M., Yu, Z., Ma, J., Gao, H., 2016. Non-methane hydrocarbons (NMHCs) and their
- 586 contribution to ozone formation potential in a petrochemical industrialized city. [Northwest](#)
- 587 [China. Atmos. Res.](#) 169, 225–236. doi:10.1016/j.atmosres.2015.10.006
- 588 Jiang, F., Guo, H., Wang, T.J., Cheng, H.R., Wang, X.M., Simpson, I.J., Ding, A.J., Saunders,
- 589 S.M., Lam, S.H.M., Blake, D.R., 2010. An ozone episode in the Pearl River Delta: Field
- 590 observation and model simulation. [J. Geophys. Res. Atmos.](#) 115, 1–18.
- 591 doi:10.1029/2009JD013583
- 592 Jiang, F., Wang, T., Wang, T., Xie, M., Zhao, H., 2008. Numerical modeling of a continuous
- 593 photochemical pollution episode in Hong Kong using WRF-chem. [Atmos. Environ.](#) 42,
- 594 8717–8727. doi:10.1016/j.atmosenv.2008.08.034
- 595 Jin, X., Fiore, A.M., Murray, L.T., Valin, L.C., Lamsal, L.N., Duncan, B., Folkert Boersma, K.,
- 596 De Smedt, I., Abad, G.G., Chance, K., Tonnesen, G.S., 2017. Evaluating a space-based
- 597 indicator of surface ozone-NO<sub>x</sub>-VOC sensitivity over midlatitude source regions and
- 598 application to decadal trends. [J. Geophys. Res. Atmos.](#) 122, 10439–10461.
- 599 doi:10.1002/2017JD026720
- 600 Krupa, S. V., Manning, W.J., 1988. Atmospheric ozone: Formation and effects on vegetation.

Formatted: French (France)

Formatted: French (France)

Formatted: Highlight



- 601 [Environ. Pollut.](#) 50, 101–137. doi:10.1016/0269-7491(88)90187-X
- 602 Kumar, A., Singh, D., Anandam, K., Kumar, K., Jain, V.K., 2017. Dynamic interaction of trace  
603 gases (VOCs, ozone, and NOx) in the rural atmosphere of sub-tropical India. *Air Qual.*  
604 *Atmos. Heal.* 10, 885–896. doi:10.1007/s11869-017-0478-8
- 605 Lam, S.H.M., Saunders, S.M., Guo, H., Ling, Z.H., Jiang, F., Wang, X.M., Wang, T.J., 2013.  
606 Modelling VOC source impacts on high ozone episode days observed at a mountain summit  
607 in Hong Kong under the influence of mountain-valley breezes. *Atmos. Environ.* 81, 166–  
608 176. doi:10.1016/j.atmosenv.2013.08.060
- 609 Li, J., Xie, S.D., Zeng, L.M., Li, L.Y., Li, Y.Q., Wu, R.R., 2015. Characterization of ambient  
610 volatile organic compounds and their sources in Beijing, before, during, and after Asia-  
611 Pacific Economic Cooperation China 2014. *Atmos. Chem. Phys.* 15, 7945–7959.  
612 doi:10.5194/acp-15-7945-2015
- 613 Li, L., Chen, C.H., Huang, C., Huang, H.Y., Zhang, G.F., Wang, Y.J., Wang, H.L., Lou, S.R.,  
614 Qiao, L.P., Zhou, M., Chen, M.H., Chen, Y.R., Streets, D.G., Fu, J.S., Jang, C.J., 2012.  
615 Process analysis of regional ozone formation over the Yangtze River Delta, China using the  
616 Community Multi-scale Air Quality modeling system. *Atmos. Chem. Phys.* 12, 10971–  
617 10987. doi:10.5194/acp-12-10971-2012
- 618 Lin, M., Horowitz, L.W., Payton, R., Fiore, A.M., Tonnesen, G., 2017. US surface ozone trends  
619 and extremes from 1980 to 2014: Quantifying the roles of rising Asian emissions, domestic  
620 controls, wildfires, and climate. *Atmos. Chem. Phys.* 17, 2943–2970. doi:10.5194/acp-17-  
621 2943-2017
- 622 Lin, X., Trainer, M., Liu, S.C., 1988. On the nonlinearity of the tropospheric ozone production. *J.*  
623 *Geophys. Res.* 93, 15879. doi:10.1029/JD093iD12p15879
- 624 Ling, Z.H., Guo, H., 2014. Contribution of VOC sources to photochemical ozone formation and  
625 its control policy implication in Hong Kong. *Environ. Sci. Policy* 38, 180–191.  
626 doi:10.1016/j.envsci.2013.12.004
- 627 Ling, Z.H., Guo, H., Lam, S.H.M., Saunders, S.M., Wang, T., 2014. Atmospheric photochemical  
628 reactivity and ozone production at two sites in Hong Kong: Application of a Master  
629 Chemical Mechanism-photochemical box model. *J. Geophys. Res. Atmos.* 119, 10567–  
630 10582. doi:10.1002/2014JD021794
- 631 Ling, Z.H., Guo, H., Zheng, J.Y., Louie, P.K.K., Cheng, H.R., Jiang, F., Cheung, K., Wong,  
632 L.C., Feng, X.Q., 2013. Establishing a conceptual model for photochemical ozone pollution  
633 in subtropical Hong Kong. *Atmos. Environ.* 76, 208–220.  
634 doi:10.1016/j.atmosenv.2012.09.051
- 635 Liu, Y., Shao, M., Zhang, J., Fu, L.L., Lu, S.H., 2005. Distributions and source apportionment of  
636 ambient volatile organic compounds in Beijing city, China. *J. Environ. Sci. Heal. Part A* ~~a~~  
637 *Toxic/Hazardous Subst. Environ. Eng.* 40, 1843–1860. doi:10.1080/10934520500182842
- 638 Logan, J.A., 1985. Tropospheric ozone: Seasonal behavior, trends, and anthropogenic influence.  
639 *J. Geophys. Res. Atmos.* 90, 10463–10482. doi:10.1029/JD090iD06p10463
- 640 Lyu, X.P., Chen, N., Guo, H., Zhang, W.H., Wang, N., Wang, Y., Liu, M., 2016. Ambient  
641 volatile organic compounds and their effect on ozone production in Wuhan, central China.

Formatted: French (France)

Formatted: French (France)

Formatted: French (France)

642 Sci. Total Environ. 541, 200–209. doi:10.1016/j.scitotenv.2015.09.093  
643 Lyu, X.P., Guo, H., Simpson, I.J., Meinardi, S., Louie, P.K.K., Ling, Z., Wang, Y., Liu, M., Luk,  
644 C.W.Y., Wang, N., Blake, D.R., 2016a. Effectiveness of replacing catalytic converters in  
645 LPG-fueled vehicles in Hong Kong. *Atmos. Chem. Phys.* 16, 6609–6626. doi:10.5194/acp-  
646 16-6609-2016  
647 Lyu, X.P., Liu, M., Guo, H., Ling, Z.H., Wang, Y., Louie, P.K.K., Luk, C.W.Y., 2016b.  
648 Spatiotemporal variation of ozone precursors and ozone formation in Hong Kong: Grid field  
649 measurement and modelling study. *Sci. Total Environ.* 569–570, 1341–1349.  
650 doi:http://dx.doi.org/10.1016/j.scitotenv.2016.06.214  
651 McDonald, B.C., De Gouw, J.A., Gilman, J.B., Jathar, S.H., Akherati, A., Cappa, C.D., Jimenez,  
652 J.L., Lee-Taylor, J., Hayes, P.L., McKeen, S.A., Cui, Y.Y., Kim, S.W., Gentner, D.R.,  
653 Isaacman-VanWertz, G., Goldstein, A.H., Harley, R.A., Frost, G.J., Roberts, J.M., Ryerson,  
654 T.B., Trainer, M., 2018. Volatile chemical products emerging as largest petrochemical  
655 source of urban organic emissions. *Science*. 359, 760–764. doi:10.1126/science.aag0524  
656 Monks, P.S., 2005. Gas-phase radical chemistry in the troposphere. *Chem. Soc. Rev.* 34, 376.  
657 doi:10.1039/b307982c  
658 Monks, P.S., Archibald, A.T., Colette, A., Cooper, O., Coyle, M., Derwent, R., Fowler, D.,  
659 Granier, C., Law, K.S., Mills, G.E., Stevenson, D.S., Tarasova, O., Thouret, V., Von  
660 Schneidmesser, E., Sommariva, R., Wild, O., Williams, M.L., 2015. Tropospheric ozone  
661 and its precursors from the urban to the global scale from air quality to short-lived climate  
662 forcer. *Atmos. Chem. Phys.* 15, 8889–8973. doi:10.5194/acp-15-8889-2015  
663 Paatero, P., 1997. Least squares formulation of robust non-negative factor analysis. *Chemom.*  
664 *Intell. Lab. Syst.* 37, 23–35. doi:10.1016/S0169-7439(96)00044-5  
665 Paatero, P., Tapper, U., 1994. Positive matrix factorization: A non-negative factor model with  
666 optimal utilization of error estimates of data values. *Environmetrics* 5, 111–126.  
667 doi:10.1002/env.3170050203  
668 Pan, X., Kanaya, Y., Tanimoto, H., Inomata, S., Wang, Z., Kudo, S., Uno, I., 2015. Examining  
669 the major contributors of ozone pollution in a rural area of the Yangtze River Delta region  
670 during harvest season. *Atmos. Chem. Phys.* 15, 6101–6111. doi:10.5194/acp-15-6101-2015  
671 Roelofs, G.J., Lelieveld, J., 1997. Model study of the influence of cross-tropopause O<sub>3</sub> transports  
672 on tropospheric O<sub>3</sub> levels. *Tellus B Chem. Phys. Meteorol.* 49, 38–55.  
673 doi:10.3402/tellusb.v49i1.15949  
674 Shao, P., An, J., Xin, J., Wu, F., Wang, J., Ji, D., Wang, Y., 2016. Source apportionment of  
675 VOCs and the contribution to photochemical ozone formation during summer in the typical  
676 industrial area in the Yangtze River Delta, China. *Atmos. Res.*  
677 doi:10.1016/j.atmosres.2016.02.015  
678 Sharkey, T.D., Wiberley, A.E., Donohue, A.R., 2008. Isoprene emission from plants: Why and  
679 how. *Ann. Bot.* 101, 5–18. doi:10.1093/aob/mcm240  
680 So, K.L., Wang, T., 2003. On the local and regional influence on ground-level ozone  
681 concentrations in Hong Kong. *Environ. Pollut.* 123, 307–317. doi:10.1016/s0269-  
682 7491(02)00370-6

Formatted: French (France)

- 683 Stevenson, D.S., Young, P.J., Naik, V., Lamarque, J.F., Shindell, D.T., Voulgarakis, A., Skeie,  
684 R.B., Dalsoren, S.B., Myhre, G., Bernsten, T.K., Folberth, G.A., Rumbold, S.T., Collins,  
685 W.J., MacKenzie, I.A., Doherty, R.M., Zeng, G., Van Noije, T.P.C., Strunk, A., Bergmann,  
686 D., Cameron-Smith, P., Plummer, D.A., Strode, S.A., Horowitz, L., Lee, Y.H., Szopa, S.,  
687 Sudo, K., Nagashima, T., Josse, B., Cionni, I., Righi, M., Eyring, V., Conley, A., Bowman,  
688 K.W., Wild, O., Archibald, A., 2013. Tropospheric ozone changes, radiative forcing and  
689 attribution to emissions in the Atmospheric Chemistry and Climate Model Intercomparison  
690 Project (ACCMIP). *Atmos. Chem. Phys.* 13, 3063–3085. doi:10.5194/acp-13-3063-2013  
691 Tilton, B.E., 1989. Health effects of tropospheric ozone. *Environ. Sci. Technol.* 23, 257–263.  
692 doi:10.1021/es00180a002
- 693 Verstraeten, W.W., Neu, J.L., Williams, J.E., Bowman, K.W., Worden, J.R., Boersma, K.F.,  
694 2015. Rapid increases in tropospheric ozone production and export from China. *Nat.*  
695 *Geosci.* 8, 690–695. doi:10.1038/ngeo2493
- 696 Wang, N., Guo, H., Jiang, F., Ling, Z.H., Wang, T., 2015. Simulation of ozone formation at  
697 different elevations in mountainous area of Hong Kong using WRF-CMAQ model. *Sci.*  
698 *Total Environ.* 505, 939–951. doi:10.1016/j.scitotenv.2014.10.070
- 699 Wang, Q., Han, Z., Wang, T., Zhang, R., 2008. Impacts of biogenic emissions of VOC and NO<sub>x</sub>  
700 on tropospheric ozone during summertime in eastern China. *Sci. Total Environ.* 395, 41–49.  
701 doi:10.1016/j.scitotenv.2008.01.059
- 702 Wang, T., Nie, W., Gao, J., Xue, L.K., Gao, X.M., Wang, X.F., Qiu, J., Poon, C.N., Meinardi, S.,  
703 Blake, D., Wang, S.L., Ding, A.J., Chai, F.H., Zhang, Q.Z., Wang, W.X., 2010. Air quality  
704 during the 2008 Beijing Olympics: Secondary pollutants and regional impact. *Atmos.*  
705 *Chem. Phys.* 10, 7603–7615. doi:10.5194/acp-10-7603-2010
- 706 Wang, T., Wei, X.L., Ding, A.J., Poon, C.N., Lam, K.S., Li, Y.S., Chan, L.Y., Anson, M., 2009.  
707 Increasing surface ozone concentrations in the background atmosphere of Southern China,  
708 1994-2007. *Atmos. Chem. Phys.* 9, 6217–6227. doi:10.5194/acp-9-6217-2009
- 709 Wang, T., Xue, L., Brimblecombe, P., Lam, Y.F., Li, L., Zhang, L., 2016. Ozone pollution in  
710 China: A review of concentrations, meteorological influences, chemical precursors, and  
711 effects. *Sci. Total Environ.* 575, 1582–1596. doi:10.1016/j.scitotenv.2016.10.081
- 712 Wang, Y., Guo, H., Zou, S.C., Lyu, X.P., Ling, Z.H., Cheng, H.R., Zeren, Y., 2018. Surface O<sub>3</sub>  
713 photochemistry over the South China Sea: Application of a near-explicit chemical  
714 mechanism box model. *Environ. Pollut.* 234, 155–166. doi:10.1016/j.envpol.2017.11.001
- 715 Wang, Y., Wang, H., Guo, H., Lyu, X.P., Cheng, H.R., Ling, Z.H., Louie, P.K.K., Simpson, I.J.,  
716 Meinardi, S., Blake, D.R., 2017. Long term O<sub>3</sub>-precursor relationships in Hong Kong: Field  
717 observation and model simulation. *Atmos. Chem. Phys. Discuss.* 1–29. doi:10.5194/acp-  
718 2017-235
- 719 Xu, Z., Huang, X., Nie, W., Chi, X., Xu, Z., Zheng, L., Sun, P., Ding, A., 2017. Influence of  
720 synoptic condition and holiday effects on VOCs and ozone production in the Yangtze River  
721 Delta region, China. *Atmos. Environ.* 168, 112–124. doi:10.1016/j.atmosenv.2017.08.035
- 722 Xu, Z., Xue, L., Wang, T., Xia, T., Gao, Y., Louie, P.K.K., Luk, C.W.Y., 2015. Measurements of  
723 peroxyacetyl nitrate at a background site in the Pearl River Delta region: production

Formatted: French (France)

Formatted: French (France)

724 efficiency and regional transport. *Aerosol Air Qual. Res.* 15, 833–841.  
725 doi:10.4209/aaqr.2014.11.0275

726 Xue, L.K., Wang, T., Gao, J., Ding, A.J., Zhou, X.H., Blake, D.R., Wang, X.F., Saunders, S.M.,  
727 Fan, S.J., Zuo, H.C., Zhang, Q.Z., Wang, W.X., 2014. Ground-level ozone in four Chinese  
728 cities: Precursors, regional transport and heterogeneous processes. *Atmos. Chem. Phys.* 14,  
729 13175–13188. doi:10.5194/acp-14-13175-2014

730 Yuan, B., Shao, M., Lu, S., Wang, B., 2010. Source profiles of volatile organic compounds  
731 associated with solvent use in Beijing, China. *Atmos. Environ.* 44, 1919–1926.  
732 doi:10.1016/j.atmosenv.2010.02.014

733 Yuan, Z., Lau, A.K.H., Shao, M., Louie, P.K.K., Liu, S.C., Zhu, T., 2009. Source analysis of  
734 volatile organic compounds by positive matrix factorization in urban and rural  
735 environments in Beijing. *J. Geophys. Res.* 114. doi:10.1029/2008jd011190

736 Zhang, J., Wang, T., Chameides, W.L., Cardelino, C., Kwok, J., Blake, D.R., Ding, A., So, K.L.,  
737 2006. Ozone production and hydrocarbon reactivity in Hong Kong, Southern China. *Atmos.*  
738 *Chem. Phys. Discuss.* 6, 8961–9002. doi:10.5194/acpd-6-8961-2006

739