POLYMER-BONDED MAGNETIC MATERIALS

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See application file for complete search history.

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ABSTRACT

A magnetic composition for power conversion includes a thermoplastic polymer and magnetic powders. The composition has a tensile strength of greater than 20 N/mm².

10 Claims, 5 Drawing Sheets
FIG. 5

MAX LOAD EXT.
(N)   (mm)
3080.0  6.868
FIG. 6

FIG. 7
POLYMER-BONDED MAGNETIC MATERIALS

BACKGROUND

The design of magnetic power converters depends on factors including the permeability, loss factor, and size and shape of the magnetic material in the converter. Specifically, the loss for magnetic material usually accounts for 30–40% of the total loss of the converter. Conventional magnetic materials, such as Ferrites and Molybdenum Permalloy Powder (MPP), are known for their low loss characteristics and high frequency operation. Therefore, these magnetic materials may be used in power converters, such as inductors and transformers. However, these magnetic materials suffer from a number of disadvantages, including limited size, brittleness, high loss and high cost.

For example, it is difficult to provide a transformer or inductor useful in high power conversion, such as a system of more than 20 kW, because of the complexity and expense in the formation of Ferrites or powder iron. In addition, conventional materials must be screened in this application. A metal and plastic material chassis is often used to screen electromagnetic emissions, which increases the cost and the weight of the electronic product.

The loss in power conversion can be divided into conductor loss and core loss. The conductor loss, or winding loss, is the resistive loss due to the current passed through the winding around the magnetic material. Because of the current distribution in the conductor at high frequency, this loss can increase dramatically as the frequency increases. The core loss is usually caused by the hysteresis, eddy loss and/or residual loss of the magnetic materials. The hysteresis loss and eddy loss can be decreased by using powder iron core for high frequency application. The introduction of polymer into the conventional core could also lower the eddy loss to some extent, which could extend their applications to a broader range at high frequency area.

The technology and engineering domains constantly set demanding requirements of magnetic materials. Recently, polymer bonded magnetic materials have attracted a great deal of attention in the fields of magneto-electrics and magneto-optics. These materials are composed of polymer matrices and magnetic powders, which may be produced using traditional polymer processing methods. Polymer bonded magnetic materials offer significant advantages over conventional materials. For example, polymer bonded magnetic materials can be molded more easily, lowering the cost of manufacturing and of quality control. Nonetheless, the polymer-bonded magnetic materials have not typically been applicable in power conversion or electromagnetic interference. The outstanding work needed in the optimization and the permeability study has prevented developing the materials into a product.

It is desirable to produce a magnetic material that could be easier to form into device cores for application in high power conversion (over 20 kW). Ideally, these magnetic materials would have sufficient flexibility. It is also desirable to manufacture the magnetic material at a low cost. It is further desirable to produce a magnetic material that is light weight. It is also desirable to produce a magnetic material useful for high frequency power conversion, such as over 100 s kHz operation. It is further desirable to that the magnetic material is applicable in power transformers and inductors.

BRIEF SUMMARY

According to one aspect, a magnetic composition for power conversion includes a thermoplastic polymer and magnetic powders. The composition has a tensile strength of greater than 20 N/mm².

According to another aspect, a method of making a magnetic composition for power conversion includes mixing Fe₃O₄, NiO and ZnO into a mixture in a high-speed blender, coating the mixture in a high-speed blender into magnetic powders, mixing dried PMMA pellets with Stearic acid (Ca₃H₇O₄) in a high-speed blender to form polymer powders with appropriate size, and mixing the magnetic powders and the polymer powders in a high-speed blender.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1A depicts the ring cores based on 10 wt % PE-90 wt % FeNiZn(50:20:30 mol) (130°C, 950 minutes).

FIG. 1B depicts the measurement results of FIG. 1A, with initial values: f = 5 kHz, L = 56.00 μH, μ₀ = 42.17 and peak values: f = 14.30 MHz, L = 125.80 μH, μ₀ = 94.80.

FIG. 2A depicts the ring cores based on 10 wt % PE-90 wt % FeNiZn(50:30:20 mol) (130°C, 950 minutes).

FIG. 2B depicts the measurement results of FIG. 2A, with initial values: f = 5 kHz, L = 50.49 μH, μ₀ = 38.02 and peak values: f = 13.70 MHz, L = 108.62 μH, μ₀ = 81.79.

FIG. 3A depicts the ring cores based on 10 wt % PE-90 wt % FeNiZn(50:40:10 mol) (130°C, 950 minutes).

FIG. 3B depicts the measurement results of FIG. 3A, with initial values: f = 5 kHz, L = 38.30 μH, μ₀ = 28.94 and peak values: f = 19.30 MHz, L = 135.91 μH, μ₀ = 102.34.

FIG. 4A depicts the ring cores based on 10 wt % PE-90 wt % FeNiZn(50:20:30 mol) (110°C, 20 hours).

FIG. 4B depicts the measurement results of FIG. 4A, with initial values: f = 5 kHz, L = 43.38 μH, μ₀ = 33.12 and peak values: f = 20.2 MHz, L = 120.15 μH, μ₀ = 90.47.

FIG. 5 depicts the typical curves of load (N) vs. extension (mm) in tensile strength measurements.

FIG. 6 depicts the typical curves of load (N) vs. deflection (mm) in resistance to compression measurements.

FIG. 7 depicts the typical curves of load (N) vs. deflection (mm) in radial crushing strength measurements.

FIG. 8 depicts how external field may be applied to aid the alignment of a magnetic dipole by using a hot-press machine.

FIG. 9 depicts constructed embodiments of the polymer-bonded magnetic core.

DETAILED DESCRIPTION

Reference will now be made in detail to a particular embodiment of the invention, examples of which are also provided in the following description. Exemplary embodiments of the invention are described in detail, although it will be apparent to those skilled in the relevant art that some features that are not particularly important to an understanding of the invention may not be shown for the sake of clarity.

Furthermore, it should be understood that the invention is not limited to the precise embodiments described below, and that various changes and modifications thereof may be effected by one skilled in the art without departing from the spirit or scope of the invention. For example, elements and/or features of different illustrative embodiments may be combined with each other and/or substituted for each other within the scope of this disclosure and appended claims. In addition, improvements and modifications which may become apparent to persons of ordinary skill in the art after reading this.
Disclosure, the drawings, and the appended claims are deemed within the spirit and scope of the present invention.

A magnetic composition may include a thermoplastic polymer and magnetic powders. The composition has a tensile strength of greater than 20 N/mm². Formed magnetic cores including the magnetic composition may have better mechanical properties than conventional cores. For example, magnetic cores including the magnetic composition may have a tensile strength of greater than 20 N/mm², and may have a resistance to compression of greater than 40 N/mm². Magnetic cores including the magnetic composition may be used for power conversion applications, such as power transformers, power inductors, and ferrites screens.

Composition

A magnetic composition for power conversion may include a thermoplastic polymer and a magnetic powder. The thermoplastic polymer may be taken from the group consisting of poly(methyl methacrylate) (PMMA) and polyethylene (PE). Other polymers, such as Nylon 6, may also be used, which may vary the operational temperature of the products. The magnetic powder may be taken from the group consisting of Nickel, Cobalt, NiZn Ferrite, and MnZn Ferrite. Optionally, a coupling agent of Titanium (IV) Isopropoxide (C₂H₃O₂Ti) may be included in the composition. For example, the composition may contain from about 10 to 40 weight percent of the thermoplastic polymer, from about 60 to 90 weight percent of the magnetic powder, and the magnetic powder may contain about 15 weight percent of the coupling agent.

Methods

To produce the polymer-bonded magnetic material, appropriate amounts of Fe₂O₃, NiO and ZnO in different mole ratios (50:20:30, 50:30:20, or 50:40:10) may be vigorously mixed in a high-speed blender for about 2 minutes. The mixture may then be sintered in a high-temperature calcination furnace. The furnace may be heated at a rate of 8°C/min to 1300°C, and maintained at this temperature for 950 minutes. The melted mixture may be taken out immediately, which may then be placed at about 20°C, and allowed to cool down to room temperature rapidly. The cooled mixture may then be crushed in a high-speed blender to provide magnetic powders.

To remove most of the moisture in the mixture, PMMA pellets may be dried in an oven at about 60°C for about 6 hours, and the magnetic powders may be also dried in an oven at about 60°C for about 4 hours. To modify the surface properties of the magnetic powders, the dried magnetic powders may be vigorously mixed with Titanium (IV) Isopropoxide (C₂H₃O₂Ti) in a high-speed blender. The dried magnetic powders may contain about 15 weight percent of C₁₃H₂₆O₂Ti. Then, the mixture may be dried in an oven at about 60°C for about 3 hours. Such a modification may improve the compatibility between the magnetic powders and the polymer, which in turn may improve the properties of the composite.

The dried PMMA pellets may be mixed with Stearic acid (C₁₇H₃₅O₂) in a high-speed blender to form polymer powders with appropriate size. The dried PMMA may contain about 2 weight percent of C₁₃H₂₆O₂. The modified magnetic powders and polymers may then be vigorously premixed in a high-speed blender. This mixture may be blended further using a single-screw extruder operating at an appropriate rotation speed. The temperature setting may be selected as: Zone 1 at 210°C, Zone 2 at 230°C, Zone 3 at 265°C, and Zone 4 at 260°C. The extrusion may become difficult at lower temperatures, while higher temperatures may cause inconsistency in properties. The mixture may then be placed in a predefined mould and made into a magnetic core in a hot press machine operated at about 150°C at 6 to 10 ton press, which is above the melting point of the polymer to be used in the magnetic core.

To increase the permeability of the magnetic cores, an external magnetic field may be applied to aid the alignment of the magnetic dipole inside the magnetic materials. The external magnetic field may be supplied from a permanent magnetic or an electric-winding. The magnetic field applied may be in the same direction as the main field direction of the core under construction as depicted in FIG. 8.

While not being bound by theory, it is believed that an evenly distributed air-gap in the magnetic composition may reduce the fringe field and/or reduce the eddy current loss, which is desirable for high frequency power electronics.

Products

Magnetic cores based on the polymer-bonded magnetic materials may be fabricated on a hot-press machine operated at 180°C and at about 6 to 10 tons, with a mould of desired shape. Magnetic cores fabricated utilizing the magnetic composition may have various shapes, such as a ring, an EE, an EI, and a U shape, as depicted in FIG. 9. An EE-shape refers to when the shape of the magnetic core is in the geometry of two letter Es. An EI-shape refers to when the shape of the magnetic core is in the geometry of a letter E and a letter I. A U-shape refers to when the shape of the magnetic core is in the geometry of a letter U. A ring-shape refers to the shape of the magnetic core is similar to a circular ring. Other shapes, such as irregular geometry, may also be used.

A formed magnetic core includes the magnetic composition may have better mechanical properties than the conventional magnetic cores. For example, the magnetic cores may have a tensile strength of greater than 20 N/mm² and a resistance to compression of greater than 40 N/mm².

Magnetic cores including the magnetic composition may be used, for example, as power transformers, power inductors, and ferrites screens.

EXAMPLES

Example 1

The Magnetic Properties of the Composition

The magnetic properties measurements were carried out on a ring core (ø30xø15xh112 mm). Inductance (I) was measured on a HIOKI 3530 LCR Hi Tester, and then the relative permeability μr was calculated. From the equivalent circuit of the ring core shown above, the impedance Z should be Z = R + jωL. And Z = jωL (μr + jμm). Herein, μr and μm is the real part and the imaginary part of the magnetic permeability, respectively. ω is the rotational frequency. Then,

\[ R + jωL = jωL(μ' - jμ'') \]

Quality factor: \[ Q = \frac{ωL}{R} \]

then \[ μ' = \frac{L}{Te}, μ'' = \frac{L}{Q} \]
Also, the inductance value $L_o$ of the air

$$L_o = \frac{4N^2A_e}{I_e} \times 10^{-9} \cdot \frac{A_e}{A_e} = 2\pi h \ln \frac{r_2}{r_1}$$

Herein, N is the loop number, A_e is the effective area of the flux, r_e is the effective length of the magnetic circuit, r_1 is the inside diameter, r_2 is the inside diameter, and h is the height of the ring core.

Finally,

$$\mu' = \frac{L_o \times 10^9}{8\pi^2 N^2 h (\ln \frac{r_2}{r_1})}$$

$$\mu = \frac{L_o \times 10^9}{8\pi^2 N^2 h (\ln \frac{r_2}{r_1}) \cdot \sqrt{1 + 1/Q^2}}$$

Typical images are shown in FIGS. 1 to 4, wherein Loop Number $N = 45$, $\mu_e = 0.753 \times A_e L_e \mu_0$. The results for the above results and those for other objects are concluded in Table 1 and Table 2.

**Table 1**

<table>
<thead>
<tr>
<th>Samples’ Composition</th>
<th>5 kHz</th>
<th>10 kHz</th>
<th>25 kHz</th>
<th>50 kHz</th>
<th>75 kHz</th>
<th>100 kHz</th>
</tr>
</thead>
<tbody>
<tr>
<td>10 wt % PE, 10% PMMA</td>
<td>43.98</td>
<td>42.55</td>
<td>42.40</td>
<td>42.03</td>
<td>41.98</td>
<td>42.29</td>
</tr>
<tr>
<td>90 wt %</td>
<td>33.12</td>
<td>32.04</td>
<td>31.93</td>
<td>31.65</td>
<td>31.61</td>
<td>31.84</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol%) (1100°C, 950 minutes)</td>
<td>20 wt %</td>
<td>42.40</td>
<td>41.45</td>
<td>41.29</td>
<td>40.91</td>
<td>40.86</td>
</tr>
<tr>
<td>10 wt %</td>
<td>31.93</td>
<td>31.21</td>
<td>31.09</td>
<td>30.80</td>
<td>30.77</td>
<td>30.90</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol%) (1100°C, 950 minutes)</td>
<td>30 wt %</td>
<td>16.60</td>
<td>16.00</td>
<td>15.91</td>
<td>15.73</td>
<td>15.70</td>
</tr>
<tr>
<td>10 wt %</td>
<td>12.35</td>
<td>12.05</td>
<td>11.98</td>
<td>11.84</td>
<td>11.82</td>
<td>11.82</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol%) (1100°C, 950 minutes)</td>
<td>40 wt %</td>
<td>13.65</td>
<td>9.95</td>
<td>9.84</td>
<td>9.69</td>
<td>9.65</td>
</tr>
<tr>
<td>10 wt %</td>
<td>10.28</td>
<td>7.49</td>
<td>7.41</td>
<td>7.30</td>
<td>7.27</td>
<td>7.26</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol%) (1100°C, 950 minutes)</td>
<td>10 wt %</td>
<td>56.00</td>
<td>55.45</td>
<td>55.26</td>
<td>54.86</td>
<td>55.01</td>
</tr>
<tr>
<td>90 wt %</td>
<td>42.17</td>
<td>41.75</td>
<td>41.61</td>
<td>41.31</td>
<td>41.42</td>
<td>41.86</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol%) (1300°C, 950 minutes)</td>
<td>10 wt %</td>
<td>50.49</td>
<td>49.38</td>
<td>49.27</td>
<td>48.98</td>
<td>49.06</td>
</tr>
<tr>
<td>90 wt %</td>
<td>38.02</td>
<td>37.18</td>
<td>37.10</td>
<td>36.88</td>
<td>31.19</td>
<td>37.48</td>
</tr>
<tr>
<td>FeNiZn(50:30:20 mol%) (1300°C, 950 minutes)</td>
<td>10 wt %</td>
<td>38.30</td>
<td>37.94</td>
<td>37.84</td>
<td>37.53</td>
<td>37.47</td>
</tr>
<tr>
<td>90 wt %</td>
<td>28.84</td>
<td>28.57</td>
<td>28.49</td>
<td>28.26</td>
<td>28.21</td>
<td>28.32</td>
</tr>
<tr>
<td>FeNiZn(50:40:10 mol%) (1300°C, 950 minutes)</td>
<td>80 wt %</td>
<td>13.50</td>
<td>12.47</td>
<td>12.30</td>
<td>11.17</td>
<td>9.36</td>
</tr>
<tr>
<td>90 wt %</td>
<td>10.17</td>
<td>9.39</td>
<td>9.26</td>
<td>8.41</td>
<td>7.05</td>
<td>5.35</td>
</tr>
<tr>
<td>PMMA</td>
<td>14.00</td>
<td>12.39</td>
<td>12.28</td>
<td>11.88</td>
<td>11.12</td>
<td>9.41</td>
</tr>
</tbody>
</table>

Example 2

The Tensile Strength of the Composition

The tensile strength measurement was carried out on a Lloyd Instruments LR30KPLUS Series Universal Material Test System.
Tester. FIG. 5 shows the typical curves of load (N) vs. extension (mm). The measurement results are listed in Table 3.

<table>
<thead>
<tr>
<th>Sample’s Composition</th>
<th>Tensile Strength (N/mm²)</th>
<th>Tensile Strength (N/mm²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60 wt % FeNiZn(50:20:30 mol) - 40 wt % PMMA</td>
<td>20.15</td>
<td>29.56</td>
</tr>
<tr>
<td>70 wt % FeNiZn(50:20:30 mol) - 30 wt % PMMA</td>
<td>22.35</td>
<td>45.21</td>
</tr>
<tr>
<td>80 wt % FeNiZn(50:20:30 mol) - 20 wt % PMMA</td>
<td>60.17</td>
<td>89.28</td>
</tr>
</tbody>
</table>

Example 3

The Resistance to Compression of the Composition

The resistance to compression measurement was carried out on the rectangular samples on a Lloyd Instruments LR30 KPLUS Series Universal Material Tester. FIG. 6 shows the typical curves of load (N) vs. deflection (mm). The measurement results are concluded in Table 4.

<table>
<thead>
<tr>
<th>Sample’s Composition</th>
<th>Resistance to Compression (N/mm²)</th>
<th>Resistance to Compression (N/mm²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60 wt % FeNiZn(50:20:30 mol) - 40 wt % PMMA</td>
<td>57.91</td>
<td>39.82</td>
</tr>
<tr>
<td>70 wt % FeNiZn(50:20:30 mol) - 30 wt % PMMA</td>
<td>58.24</td>
<td>42.66</td>
</tr>
<tr>
<td>80 wt % FeNiZn(50:20:30 mol) - 20 wt % PMMA</td>
<td>45.20</td>
<td>44.01</td>
</tr>
</tbody>
</table>

Example 4

The Radial Crushing Strength of the Composition

The radial crushing strength measurement was carried out on the ring core, and was calculated based on the following formula:

\[ \sigma_r = \frac{1.906 P_r(D - t)}{2L^2} \]

Wherein, \( P_r \) is the maximal load (N), D is the outer diameter, t is the thickness, and L is the width of the sample. The measurement results are concluded in Table 5.

<table>
<thead>
<tr>
<th>Sample’s Composition</th>
<th>Radial Crushing Strength (N/mm²)</th>
<th>Radial Crushing Strength (N/mm²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60 wt % FeNiZn(50:20:30 mol)</td>
<td>207.57</td>
<td>51.11</td>
</tr>
<tr>
<td>70 wt % FeNiZn(50:20:30 mol)</td>
<td>342.36</td>
<td>75.02</td>
</tr>
<tr>
<td>80% NiZn - 20% PE</td>
<td>72.56</td>
<td></td>
</tr>
<tr>
<td>90% NiZn - 10% PE</td>
<td>29.29</td>
<td></td>
</tr>
<tr>
<td>90% NiZn - 10% UHMWPE 300 (UHMW = Ultra-High Molecular Weight)</td>
<td>83.40</td>
<td></td>
</tr>
<tr>
<td>90% NiZn - 10% UHMWPE 500</td>
<td>102.13</td>
<td></td>
</tr>
</tbody>
</table>

Example 5

The Impact Resistance of the Composition

Impact resistance of the sample was measured on a ZWICK MS25B &C D-7900 Impact Resistance Tester. The measurements were carried out on cuboid samples (50x14x9 mm) with an incision of "V" shape. The impact resistance \( A_{IKV} \) equals to the impact value \( A_r \) obtained. The measurement results are concluded in Table 6.

<table>
<thead>
<tr>
<th>Samples’ Composition</th>
<th>Impact Resistance, ( A_{IKV} ) (J)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60 wt % FeNiZn(50:20:30 mol) - 40 wt % PMMA</td>
<td>0.25</td>
</tr>
<tr>
<td>70 wt % FeNiZn(50:20:30 mol) - 30 wt % PE</td>
<td>0.94</td>
</tr>
<tr>
<td>90 wt % FeNiZn(50:20:30 mol) - 10 wt % PE</td>
<td>0.43</td>
</tr>
</tbody>
</table>

Example 6

The Rockwell Hardness of the Composition

Rockwell hardness was measured on an ESF WAY RB Hardness Tester. The measurements were carried out on cuboid samples (50x14x9 mm). The measurement results are concluded in Table 7.
TABLE 7

<table>
<thead>
<tr>
<th>Samples' Composition</th>
<th>Rockwell</th>
<th>Rockwell</th>
<th>Rockwell</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Standard F 30-T</td>
<td>Standard B 60 kg, ½ in</td>
<td>Standard B 100 kg, ½ in</td>
</tr>
<tr>
<td>FeNiZn(50:20:30 mol)</td>
<td>94</td>
<td>119</td>
<td>121</td>
</tr>
<tr>
<td>60 wt % FeNiZn(50:20:30 mol)</td>
<td>94</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>70 wt % FeNiZn(50:20:30 mol)</td>
<td>66</td>
<td>62</td>
<td>—</td>
</tr>
<tr>
<td>30 wt % PE</td>
<td>—</td>
<td>94</td>
<td>94</td>
</tr>
</tbody>
</table>

While the examples of the magnetic composition have been described, it should be understood that the composition not so limited and modifications may be made. The scope of the composition is defined by the appended claims, and all devices that come within the meaning of the claims, either literally or by equivalence, are intended to be embraced therein.

REFERENCES


What is claimed is:

1. A method of making a magnetic composition for power conversion, comprising:
   a step of preparing magnetic powders from Fe2O3, NiO and ZnO;
   a step of mixing dried PMMA pellets with Stearic acid (C18H36O2) in a high-speed blender for forming polymer powders with appropriate size; and
   a step of mixing said magnetic powders and said polymer powders in a high-speed blender to afford a magnetic composition for power conversion.

2. The method of claim 1, wherein said step of preparing magnetic powders comprises:
   mixing Fe2O3, NiO and ZnO into a mixture in a high-speed blender;
   sintering said mixture in a high-temperature calcination furnace;
   cooling said mixture to room temperature prior to crushing said mixture; and
   crushing said mixture in a high-speed blender into magnetic powders.

3. The method of claim 2, wherein said furnace is heated at a rate of 8° C./min to 1300° C. and maintained at this temperature for 950 minutes.

4. The method of claim 1, further comprising drying said PMMA pellets in an oven at about 60° C. for about 6 hours prior to mixing with said magnetic powders.

5. The method of claim 1, further comprising drying said magnetic powders in an oven at about 60° C. for about 4 hours after crushing said mixture.

6. The method of claim 5, further comprising mixing said dried magnetic powders with Titanium (IV) Isopropoxide (C12H28O4T) in a high-speed blender.

7. The method of claim 6, wherein the mass ratio of C12H28O4T to said dried magnetic powders is about 15 weight percent.

8. The method of claim 6, wherein the mass ratio of C18H36O2 to said dried PMMA is about 2 weight percent.

9. A method for increasing magnetic permeability of a magnetic composition made according to the method of claim 1, comprising applying an external field to aid the alignment of a material dipole to increase magnetic permeability.

10. The method of claim 9, wherein said external field is applied in the same direction of a flux direction in a magnetic core.