

1     **Continuous effectiveness of replacing catalytic converters on liquified petroleum gas-**  
2                                   **fueled vehicles in Hong Kong**

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11    **Abstract**

12    To mitigate the concentrations of air pollutants in the atmosphere, an intervention program of  
13    replacing the converters of liquefied petroleum gas (LPG) fueled vehicles was implemented by  
14    the Hong Kong government between October 2013 and April 2014. Data of ambient volatile  
15    organic compounds (VOCs) and other trace gases continuously monitored from September  
16    2012 to April 2017 at a roadside site were used to evaluate the continuous effectiveness of the  
17    replaced catalytic converters on the reduction of air pollutants. The measurement data showed  
18    that LPG-related VOCs (propane and *n/i*-butanes) and several trace gases (CO, NO and NO<sub>2</sub>)  
19    decreased significantly from before to after the program ( $p < 0.01$ ). To further assess the  
20    efficiency of the program, five periods covering before the program, during the program, 1<sup>st</sup>  
21    year after the program, 2<sup>nd</sup> year after the program and 3<sup>rd</sup> year after the program were  
22    categorized. The values of propane and *n/i*-butanes decreased from Period-1 (before the  
23    program) to Period-2 (during the program), and from Period-2 to Periods 3-5 (after the program)  
24    ( $p < 0.01$ ). In addition, the reduction rates of propane and *n/i*-butanes remained high and  
25    constant in Periods 3-5, suggesting that either had the vehicle owners themselves routinely  
26    replaced the converters at suitable interval afterwards, or were their vehicles caught by a remote  
27    sensing program checking excessive emissions. Source apportionment analysis indicated that  
28    LPG-fueled vehicular emissions were the top contributor to ambient VOCs in the roadside  
29    environment while the VOCs emitted from LPG-fueled vehicles indeed decreased at a rate of  
30     $4.21 \pm 2.38$  ppbv/year (average  $\pm$  95% confidence interval) from Period-1 to Period-5 ( $p <$   
31    0.01). Furthermore, the photochemical box model simulations revealed that the **net negative**

contribution of VOCs and NO<sub>x</sub> emitted from LPG-fueled vehicles to O<sub>3</sub> production strengthened at a rate of  $1.9 \times 10^{-2}$  pptv/day from Period-1 to Period-5 ( $p < 0.01$ ). The findings proved the continuous effectiveness of the intervention program, and are of help to future control strategies in Hong Kong.

**Keyword:** Continuous effectiveness; catalytic converter; LPG-fueled vehicle; VOCs; O<sub>3</sub>

## 1 Introduction

During the past decades, many cities in the world including Hong Kong have been suffering from severe ozone (O<sub>3</sub>) pollution (Jacob, 1999; NARSTO, 2000; So and Wang, 2003; Lam et al., 2005; Wang et al., 2017, 2018). It is essential to mitigate O<sub>3</sub> pollution due to its adverse effects on the environment and human health (Sillman, 1999; Seinfeld and Pandis, 2006). Being key O<sub>3</sub> precursors, volatile organic compounds (VOCs) and nitrogen oxides (NO<sub>x</sub>) are important chemicals contributing to high O<sub>3</sub> production in Hong Kong and the adjacent Pearl River Delta (PRD) region (Zhang et al., 2007; Shao et al., 2009; Guo et al., 2007, 2013; Wang et al., 2017). Previous studies revealed that O<sub>3</sub> formation is sensitive to VOCs in urban areas of this region (Cheng et al., 2010; Ling et al., 2011), and subsequently it is a foreseeably effective way to control O<sub>3</sub> pollution through constraining the VOCs concentrations in the atmosphere (Ling et al., 2011). Hence, understanding the characteristics of VOCs and quantifying their source apportionments are fundamental for the formulation and implementation of O<sub>3</sub> control strategies.

To improve air quality, the Hong Kong government has implemented a series of control measures in the past decades, mainly focusing on the reduction of air pollutants from emission sources such as motor vehicles and solvent usage (HKEPD, 2007, 2014). With the execution of these control measures, VOCs emissions from some sources have been under control (Ling and Guo, 2014; Huang et al., 2015; Lyu et al., 2017). For example, Huang et al. (2015) found that VOCs emitted from diesel and gasoline vehicles decreased by at least 37% in 2011 compared to those in 2003 at the same roadside site in Hong Kong. Lyu et al. (2017) reported that VOCs emitted from diesel exhaust and solvent usage decreased whereas the gasoline and liquefied petroleum gas (LPG) vehicle emissions elevated from 2005 to 2013. Indeed, due to the widespread LPG use in public transport since 2001, VOCs emissions from LPG-fueled vehicles and their contributions to O<sub>3</sub> production gradually increased in the last decade (Guo et al., 2011; Ho et al., 2013; Ou et al., 2015; Lyu et al., 2016a, 2017). Specifically, LPG usage

63 even accounted for 30% of the ambient non-methane hydrocarbons (NMHCs) in suburban  
64 Hong Kong (Ou et al., 2015). With the aid of the positive matrix factorization (PMF) model,  
65 Lyu et al. (2016) revealed that LPG fuel consumption was the largest source of VOCs at a  
66 roadside site during October 2012-September 2013, accounting for 50% of the total VOCs in  
67 Hong Kong.

68 To specifically reduce air pollutant emissions from LPG-fueled vehicles in Hong Kong, an  
69 intervention program of replacing the catalytic converters on LPG-fueled vehicles was  
70 undertaken from October 2013 to April 2014 (HKEPD, 2014). In this program, a total of about  
71 17,000 taxis and light buses took part in (Environment Bureau, 2017). To assess the immediate  
72 effectiveness of this program on pollutant emissions from LPG-fueled vehicles, Lyu et al.  
73 (2016a) compared the data at a roadside site before the program (June – September 2013) with  
74 that after the program (October 2013 – May 2014), and claimed that VOCs and nitrogen oxides  
75 ( $\text{NO}_x$ ) emitted from LPG-fueled vehicles significantly reduced, notwithstanding very minor  
76 increase of  $\text{O}_3$  production (Lyu et al., 2016a). Despite instant usefulness, it was reported that  
77 the catalyzing capacity of the converters would gradually fade away with the time of use (Heck  
78 and Farrauto, 2001; Sharma, et al., 2014). Since the mileages of many LPG-fueled vehicles are  
79 very high (e.g. several hundred kilometers per day for many taxis), the catalytic converters  
80 need to be replaced approximately every 18 months (Simpson et al., 2015). Hence, it is  
81 necessary to evaluate the continuous effectiveness of the intervention program on the reduction  
82 of air pollutant emissions from LPG-fueled vehicles.

83 The main objective of this study was to assess the continuous effectiveness of replacing  
84 catalytic converters on LPG-fueled vehicles in Hong Kong. To achieve it, the evolution of air  
85 quality at a roadside station in Hong Kong was studied between 2012 and 2017, a long period  
86 covering the three stages of before, during and after the implementation of the replacement  
87 program. The long term trends of LPG tracers were examined, and a detailed comparison was  
88 conducted among different stages of this program. Furthermore, a PMF model and a  
89 photochemical box model incorporating Master Chemical Mechanism (PBM-MCM) were  
90 applied to understand the changes of the contribution of LPG-fueled vehicles to ambient VOCs  
91 and the ground-level  $\text{O}_3$  production caused by the replacement program.

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## 2 Methodology

### 2.1 Sampling site

Continuous measurements of target air pollutants were conducted at an urban roadside location in Hong Kong, namely Mong Kok (MK) site. MK site (22.32°N, 114.17°E, 3 m a.g.l.) is one of the three urban roadside monitoring stations (MK, Causeway Bay and Central) set up by the Hong Kong Environmental Protection Department (HKEPD). The site is located at the junction of Nathan Road and Lai Chi Kok Road, Mong Kok, Kowloon (Figure 1), which is surrounded by residential and commercial building blocks with heavy daily traffic (Lee et al., 2002). As an urban roadside station, MK site is less influenced by regional air masses and the main local sources are traffic emissions (Lee et al., 2002; Cui et al., 2016).



**Figure 1.** Location of (a) the sampling site and (b) the surrounding environment

### 2.2 Measurements of target air pollutants

#### 2.2.1 Trace gases

Trace gases including SO<sub>2</sub>, CO, O<sub>3</sub>, NO and NO<sub>2</sub> were continuously monitored at MK from September 2012 to April 2017. CO was analyzed using the method of non-dispersive infrared absorption with gas filter correlation; SO<sub>2</sub> was detected with UV fluorescence analyzers; NO–NO<sub>2</sub>–NO<sub>x</sub> was measured by chemiluminescence technique; An UV photometric analyzer was used to monitor O<sub>3</sub> mixing ratio. The detection limits for CO, SO<sub>2</sub>, NO, NO<sub>2</sub> and O<sub>3</sub> were 50.0, 1.0, 0.5, 0.5 and 2.0 ppbv, respectively. All the instruments were regularly calibrated and tested with the instruments and quality control and assurance (QC/QA) procedures identical to those in the US air quality monitoring program (Ou, et al., 2015; Lyu et al., 2016a; Wang et al., 2017).

## 2.2.2 Volatile Organic Compounds (VOCs)

Non-methane hydrocarbons (NMHCs) were measured by an online analyzer (Syntech Spectras GC 955, Series 600/800, Netherlands). The instrument is an integrated separating and analytical system consisting of two sampling systems and two columns for separation of the C<sub>2</sub>-C<sub>5</sub> hydrocarbons and the C<sub>6</sub>-C<sub>10</sub> hydrocarbons, respectively (HKPU, 2012). The target species included 30 VOCs (10 alkanes, 10 alkenes, 1 alkyne and 9 aromatics). Ambient air samples were collected and analyzed every 30-minute continuously. The 30-minute concentrations were averaged into hourly values for further analysis (Ou, et al., 2015; Lyu et al., 2016a; Wang et al., 2017).

## 2.3. Quality assurance and quality control (QA/QC)

Strict QA/QC procedures were followed to assure the data quality. Built-in computerized programs of QC system, e.g., auto-linearization and auto-calibration were developed for the analyzer. Weekly calibrations were conducted by injecting certified calibration gas (National Physical Laboratory span gas). Additionally, independent comparisons with canister samples analyzed by UC-Irvine were carried out. Overall, the detection limits of the target NMHCs ranged from 2 to 787 pptv. The accuracy of the measurements was about 1-10% and the measurement precision was about 2.5-20%. More details about the QA/QC protocols can be found in the previous studies (Xue et al., 2014; Ou et al., 2015; Lyu et al., 2016a).

## 2.4 Theoretical calculation

The ratios of VOCs with different photochemical reactivity against hydroxyl radical (OH) are calculated as a measure of photochemical age (Guo et al., 2007). As *m,p*-xylene and toluene are more reactive than ethylbenzene and benzene, respectively, higher ratios of *m,p*-xylene/ethylbenzene and toluene/benzene indicate the relatively fresher air masses (Guo et al., 2004, 2007). During the study period, the average ratios of *m,p*-xylene/ethylbenzene and toluene/benzene were  $2.27 \pm 0.11$  (average  $\pm$  95% confidence interval) and  $4.17 \pm 0.32$  ppbv/ppbv, respectively. Compared to the values (1.3–1.8) in urban/rural areas in Hong Kong (Guo et al., 2007), the ratio of *m,p*-xylene/ethylbenzene in this study was higher, but comparable to the level ( $2.61 \pm 0.30$ ) measured in a tunnel in Hong Kong (Ho et al., 2009) where freshly vehicular emissions dominated. For the ratio of toluene/benzene ( $4.17 \pm 0.32$ ), it was much higher than that in aged air masses ( $\sim 0.3$ ) in Hong Kong (Lau et al., 2010), and comparable to the typical value from vehicular exhaust ( $\sim 2$ ) (Scheff and Wadden, 1993).

Hence, the air masses at MK site were generally fresh and predominantly influenced by local vehicular emissions. Thus, characterizing the VOC mixing ratios at MK site would provide reliable information of the influences of the emissions from local sources on air quality.

## 2.5 Description of models

### 2.5.1 Positive matrix factorization (PMF) model

PMF is a mathematical receptor model developed by U.S. Environmental Protection Agency (EPA) that can analyze a wide range of environmental sample data. As the large number of variables in the actual data set obtained, PMF could reduce the variables and combine them into similar source types according to the specific research problems. The source types are identified by comparing them to measured profiles of each source and source contributions are used to determine how much each source contributed to a sample (<https://www.epa.gov/air-research/positive-matrix-factorization-model-environmental-data-analyses>). These results are calculated through the function of the PMF model by analysis based on the measured data at the receptor site. The equation of the model is presented by Eq. (1), with two metrics, i.e., factor contributions and factor profiles, included and exported in the PMF results:

$$x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (1)$$

where  $x_{ij}$  is the  $j$ th species concentration measured in the  $i$ th sample,  $g_{ik}$  is the species contribution of the  $k$ th source to the  $i$ th sample,  $f_{kj}$  is the  $j$ th species fraction from the  $k$ th source,  $e_{ij}$  is the residual for each sample/species, and  $p$  is the total number of independent sources (Paatero, 1997, 2000). In this study, the hourly concentrations of VOCs and CO with the proportion of missing values less than 40% were included in the input file. Values lower than the detection limit (DL) were replaced with DL/2. The uncertainties were set as  $\sqrt{(10\% \times \text{concentration})^2 + DL^2}$  and  $5/6 \times DL$  for the samples with their concentrations higher and lower than the DL, respectively. The model ran 20 times with a random seed, and the optimum solution automatically recommended by the model was accepted. To keep consistency of the source profiles, the data from the entire study period were input into the model as a whole. Note that to reduce the interference of high rates of missing data on the model performance, not all the measured VOCs were included in source apportionment (Wang et al., 2017). After data filtration, the hourly values of 14 VOCs, CO and NO at MK were applied to PMF for source apportionment. The source contributions extracted from the model were separately calculated for each period.



## 2.5.2 PBM-MCM model

A photochemical box model incorporating the master chemical mechanism (PBM-MCM) was utilized to simulate the local O<sub>3</sub> formation. The PBM-MCM model has been successfully applied in Hong Kong and been verified reliable in simulating O<sub>3</sub> and photochemical processes (Lam et al., 2013; Ling et al., 2014). In this study, the hourly measurements of VOCs, trace gases and meteorological parameters (temperature and relative humidity) at MK were used to construct the model. The daytime (07:00- 19:00) simulations were conducted for the whole study period. Figure S1 shows the diurnal patterns of the simulated and observed O<sub>3</sub> in Periods 1-5 at MK. In general, the simulated O<sub>3</sub> fit the observed values well in terms of magnitude and diurnal patterns. The index of agreement (IOA), which indicates the matching degree between the simulated and observed O<sub>3</sub> (Eq. (2)), was used to evaluate the model performance (Huang et al., 2005; Lyu et al., 2016b; Wang et al., 2017).

$$IOA = 1 - \frac{\sum_{i=1}^n (o_i - s_i)^2}{\sum_{i=1}^n (|o_i - \bar{o}| + |s_i - \bar{o}|)^2} \quad (2)$$

where  $o_i$  and  $s_i$  are observed and simulated hourly values, respectively,  $\bar{o}$  represents the average observed values, and  $n$  is the sample number. The range of IOA is 0-1. The higher the IOA value is, the better agreement between simulated and observed values is (Huang et al., 2005). In this study, the IOA is 0.73 for all the results, suggesting good performance of the model in O<sub>3</sub> simulation.

## 3. Results and Discussion

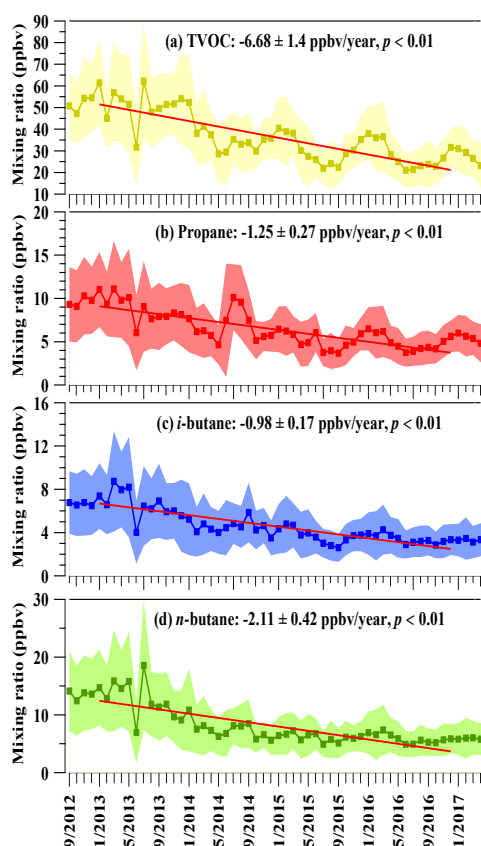
### 3.1 Temporal variations of trace gases and LPG-related VOCs

To better understand the variations of VOCs and trace gases, the study period was divided into three stages: before (September 2012-September 2013), during (October 2013-April 2014) and after the replacement program (May 2014-April 2017). Due to the widespread usage of LPG as vehicle fuel, the ambient VOC profile in Hong Kong changed substantially and the most abundant species switched from toluene to propane and *n/i*-butanes (Guo et al., 2007; Ou et al., 2015). Besides, the source profiles acquired in previous studies showed that propane and *n/i*-butanes were the most dominant components in LPG-fueled vehicular exhausts in Hong Kong (Ho et al., 2009; Guo et al., 2011, 2015; Lyu et al., 2015). Hence, propane and *n/i*-butanes were referred as LPG-related VOCs hereafter. Figure 2 shows the temporal variations of the mixing ratios of total VOCs and LPG-related VOCs from September 2012 to April 2017. The TVOC concentrations decreased with a rate of  $-6.68 \pm 1.40$  ppbv/year ( $p < 0.01$ ) at this roadside site

210 in the study period, which is contradictory to an increasing rate of  $0.34 \pm 0.02$  ppbv/year at a  
 211 suburban site from 2005 to 2013 reported by [Lyu et al. \(2015\)](#), implying some control measures  
 212 on VOCs might be effective since 2013. For the LPG-related VOCs, it was found that propane,  
 213 *i*-butane and *n*-butane decreased significantly, with a rate of  $-1.25 \pm 0.27$  ppbv/year ( $p < 0.01$ ),  
 214  $-0.98 \pm 0.17$  ppbv/year ( $p < 0.01$ ) and  $-2.11 \pm 0.42$  ppbv/year ( $p < 0.01$ ) respectively, perhaps  
 215 implying the continued effectiveness of the replacement program even after 36 months. [Table](#)  
 216 [1](#) summarizes the average mixing ratios of trace gases and LPG-related VOCs at MK before,  
 217 during and after the replacement program. The mixing ratio of propane, *i*-butane and *n*-butane  
 218 was  $5.51 \pm 0.55$ ,  $3.75 \pm 0.26$  and  $6.21 \pm 0.33$  ppbv after the program, respectively, significantly  
 219 lower than that before the program (propane:  $9.27 \pm 0.76$ ; *i*-butane:  $6.85 \pm 0.62$ ; *n*-butane:  $13.57$   
 220  $\pm 1.5$  ppbv) ( $p < 0.01$ ). Since the regional influence was minor in the roadside environment,  
 221 changes in source emissions were believed to be the main factor leading to the decreases of  
 222 LPG-related VOCs (See Section 2.4). In fact, prior to this replacement program (before  
 223 September 2013), the levels of LPG-related VOCs (propane and *n*/*i*-butanes) were even  
 224 enhanced significantly at the same MK site since 2003 ([Huang et al, 2016](#); [Lyu et al., 2016a](#)),  
 225 while their values were remarkably reduced during the replacement of the catalytic converters  
 226 on LPG-fueled vehicles ([Lyu et al., 2016a](#)), reflecting the immediate usefulness of this control  
 227 policy. The continued reduction of the LPG-related VOCs after 3-year replacement of the  
 228 converters in this study revealed the sustained effectiveness of the program. In addition to LPG-  
 229 related VOCs, the mixing ratios of primary pollutants, i.e., CO, NO<sub>2</sub>, NO and SO<sub>2</sub>, all decreased  
 230 from before to after the program ( $p < 0.01$ ). However, the levels of secondary O<sub>3</sub> increased  
 231 from  $5.5 \pm 1.5$  ppbv before the program to  $9.4 \pm 2.0$  ppbv during the program and  $8.8 \pm 1.1$   
 232 ppbv after the program. [Figure S2](#) in the Supplementary Material shows the temporal variations  
 233 of CO, SO<sub>2</sub>, NO, NO<sub>2</sub> and O<sub>3</sub> at MK from September 2012 to April 2017. Consistent with the  
 234 variations at different stages in [Table 1](#), CO, NO<sub>2</sub>, NO and SO<sub>2</sub> showed decreasing trends with  
 235 a rate of  $-47.5 \pm 26.1$ ,  $-1.9 \pm 0.5$ ,  $-28.3 \pm 3.9$  and  $-6.0 \pm 1.5$  ppbv/year, respectively.  
 236 Nevertheless, these decreases should not be fully counted by LPG-fueled vehicles as they  
 237 accounted for a small fraction of on-road vehicles ([HKEPD, 2014](#)). Indeed, apart from this  
 238 replacement program, Hong Kong government also launched other three control programs  
 239 including 1) “Reducing the emissions from diesel commercial vehicle (DCV)” started in March  
 240 2014; namely, all DCVs in Hong Kong must comply with the Euro IV emission standard by  
 241 the end of 2019 ([HKEPD, 2013](#)); 2) “Retrofitting selective catalytic reduction (SCR) devices  
 242 on Euro II and III franchised buses” started in May 2014; and “Franchised bus low emission



243 zones” started on 31 December 2015  
 244 ([https://www.epd.gov.hk/epd/english/environmentinhk/air/prob\\_solutions/cleaning\\_air\\_atroad.html](https://www.epd.gov.hk/epd/english/environmentinhk/air/prob_solutions/cleaning_air_atroad.html)). Thus, the decreases of CO, SO<sub>2</sub>, NO and NO<sub>2</sub> particularly after March 2014 were likely  
 246 the integrated impacts of these programs. Reversely, the O<sub>3</sub> levels increased with a rate of  $1.18 \pm 0.61$  ppbv/year ( $p < 0.01$ ) in the study period. This is not surprising because NO titration  
 248 effect was weakened with the decrease of NO mixing ratio from  $170.3 \pm 11.0$  ppbv before the program to  $135.7 \pm 12.2$  ppbv during the program, and then to  $92.1 \pm 10.5$  ppbv after the  
 249 program, resulting in the enhancement of O<sub>3</sub> levels at the roadside site. The results implied the  
 250 continuous effectiveness of the intervention program on LPG-related VOCs reduction and  
 251 possible combined impact on trace gases of both LPG and the other three programs. In  
 252 particular, the converters on LPG-fueled vehicles replaced between October 2013 and April  
 253 2014 seemed still effective after more than three years.



255  
 256 **Figure 2.** Temporal variations of total VOCs and LPG-related VOCs from September 2012 to  
 257 April 2017. The shaded area indicates the standard deviation of the data.

**Table 1.** Mixing ratios of trace gases and LPG-related VOCs (ppbv) at MK (average  $\pm$  95% confidence intervals)

	Before	During	After
<b>CO</b>	912.3 $\pm$ 59.2	944.5 $\pm$ 105.1	771.1 $\pm$ 44.4
<b>SO<sub>2</sub></b>	10.5 $\pm$ 1.5	12.9 $\pm$ 2.3	6.5 $\pm$ 0.8
<b>NO<sub>2</sub></b>	62.2 $\pm$ 5.3	65.2 $\pm$ 6.5	47.1 $\pm$ 2.3
<b>NO</b>	170.3 $\pm$ 11.0	135.7 $\pm$ 12.2	92.1 $\pm$ 10.5
<b>O<sub>3</sub></b>	5.5 $\pm$ 1.5	9.4 $\pm$ 2.0	8.8 $\pm$ 1.1
<b>propane</b>	9.27 $\pm$ 0.76	7.18 $\pm$ 0.80	5.51 $\pm$ 0.55
<b><i>i</i>-butane</b>	6.85 $\pm$ 0.62	5.13 $\pm$ 0.56	3.75 $\pm$ 0.26
<b><i>n</i>-butane</b>	13.57 $\pm$ 1.5	9.2 $\pm$ 1.26	6.21 $\pm$ 0.33
<b>TVOC</b>	53.04 $\pm$ 4.1	48.97 $\pm$ 5.7	32.46 $\pm$ 2.0

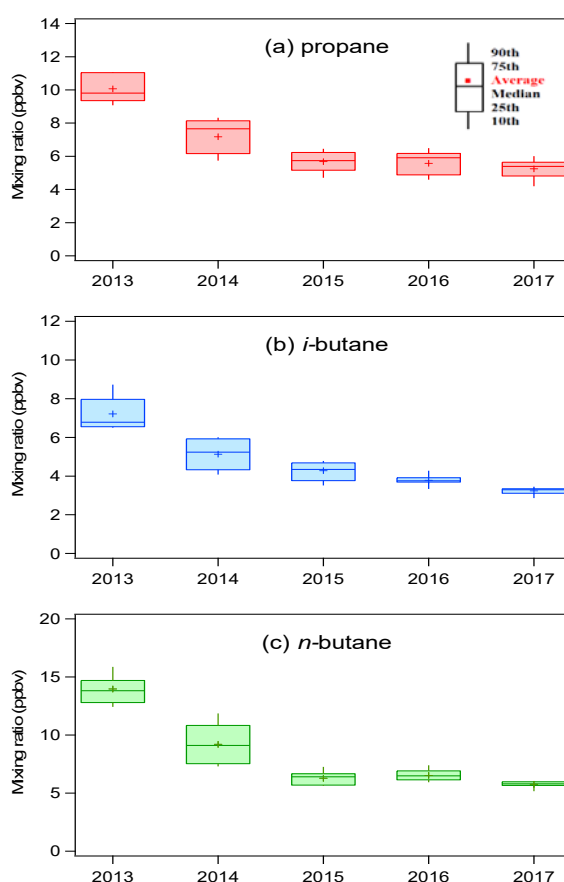
## 3.2 LPG-related VOCs in the same months of different years

### 3.2.1 LPG-related VOCs in the atmosphere at various stages

To better evaluate the replacement program, the influence of the weather conditions on the changes of trace gases and VOCs needs to be minimized. To do so, the mixing ratios of trace gases and VOCs in the same months of different years were used for comparison, because the inter-annual differences of meteorological conditions were relatively minor for the same months (Lyu et al., 2015). Since the replacement program was from October 2013 to April 2014, we categorized the study period into five time spans: Period-1: October 2012 - April 2013 (before); Period-2: October 2013 - April 2014 (during); Period-3: October 2014 - April

2015 (first-year after); Period-4: October 2015 - April 2016 (second-year after); and Period-5: October 2016 - April 2017 (third-year after the program).

Figure 3 shows the box plot of the mixing ratios of LPG-related VOCs in different time spans. Clearly, the values of propane and *n*/*i*-butanes decreased from Period-1 (propane:  $10.1 \pm 0.58$  ppbv; *i*-butane:  $7.21 \pm 0.64$  ppbv; *n*-butane:  $13.97 \pm 0.88$  ppbv) to Period-2 ( $7.18 \pm 0.80$  ppbv;  $5.13 \pm 0.56$  ppbv;  $9.20 \pm 1.26$  ppbv, respectively), and from Period-2 to Periods 3-5 ( $p < 0.01$ ), while propane and *n*-butane showed insignificant changes for Periods 3-5 ( $p > 0.05$ ). The unchanged levels of LPG-related VOCs from Period-3 to Period-5 suggested the immediate and sustained usefulness of this replacement program.



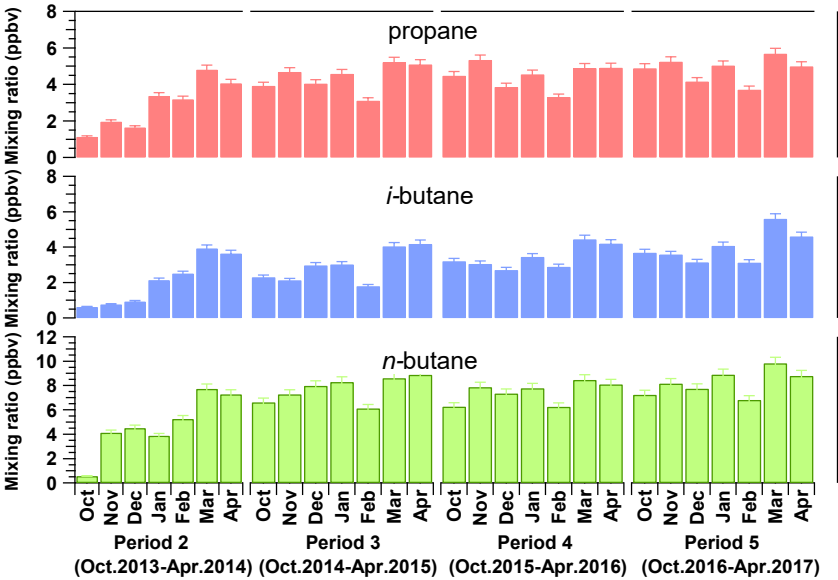
**Figure 3.** Box plot of the mixing ratios of LPG-related VOCs in different periods.

### 3.2.2 Net reduction of LPG-related VOCs

To further look into the utility of the replacement program, the net reduction of LPG-related VOCs in different periods of the replacement program was calculated and compared (Figure 4). The monthly averages of LPG-related VOCs in Period-1 were used as the baselines for the

calculation of the net reductions. Namely, for each species, its [net reduction] = [monthly average]<sub>before</sub> – [monthly average]<sub>after</sub>. Hence, positive and negative values indicated reductions and increments of the corresponding species, respectively.

Obviously, the values of the three species in Periods 2-5 were positive, indicating the net reduction of LPG-related VOCs during (Period-2) and after the program (Periods 3-5), relative to those before the program. The net reductions of propane, *i*-butane and *n*-butane in Period 2 were  $2.9 \pm 1.0$  ppbv,  $2.1 \pm 1.0$  ppbv and  $4.8 \pm 1.8$  ppbv, respectively. It was also found that the reduction amount of the three species was lower in the first several months (October – December) due to the fact that the converter replacement was done step by step and there were fewer number of LPG-fueled vehicles replaced in October-December 2013 (Lyu et al., 2016a). The results are the same as our previous study in which we found that the LPG intervention program was immediately effective on VOC reductions during the replacement of the converters (Lyu et al., 2016a). Most interestingly, the reduction rates of propane and *n*/*i*-butanes remained high and constant after the program (i.e., Periods 3-5), suggesting that the LPG replacement program was still in effect even after three years. This indicated that either had the vehicle owners themselves routinely replaced the converters at suitable interval afterwards, or might their vehicles been caught by a remote sensing program checking excessive emissions.



**Figure 4.** Net reduction of LPG-related VOCs in different periods of the intervention program.

### 3.3. Temporal variations of source contributions to LPG-related VOCs

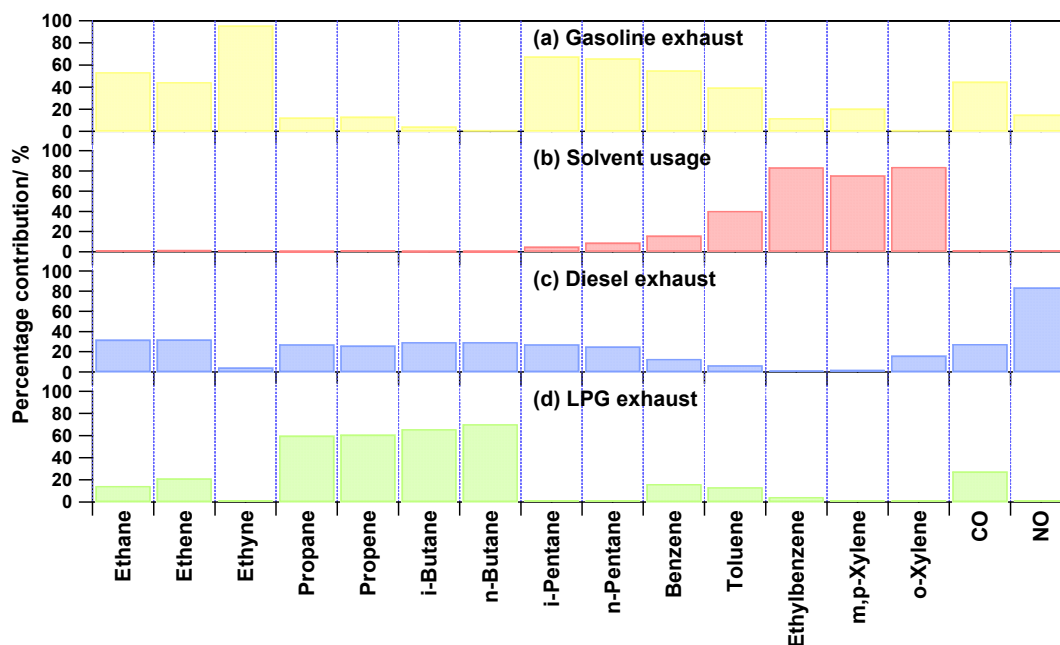
As propane and *n/i*-butanes observed in the atmosphere may not exclusively originate from LPG-fueled vehicular emissions, source apportionment is necessary to know the contributions of different sectors to them, and the variations of LPG-fueled vehicle source contributions to propane and *n/i*-butanes with time. In this way, the actual effect of the replacement program can be obtained. Hence, the contributions of the identified emission sources to VOCs and trace gases (CO and NO) in the five different periods were compared.

After the PMF running, four sources including gasoline vehicular exhaust, solvent usage, diesel vehicular exhaust and LPG-fueled vehicles were identified (see Figure 5). Source 1 was characterized by high percentages of C<sub>2</sub> species (ethane, ethene and ethyne), *n/i*-pentanes, benzene and toluene, and a considerable percentage of CO and NO, indicating vehicular emissions, likely from gasoline-fueled vehicles as *n/i*-pentane, benzene and toluene are good tracers of gasoline exhaust in Hong Kong (Tsai et al., 2006; Ho et al., 2009; Guo et al., 2011).

Source 2 was dominated by high percentages of toluene, ethylbenzene and xylenes (*m,p*-xylenes and *o*-xylene), with the proportions higher than those in other VOC source profiles. With a certain amount of these species and low contributions of C<sub>2</sub> hydrocarbons, CO and NO, the source was defined as solvent usage rather than vehicular emissions. These species could emit from paints, inks, sealant, varnish and thinner for architecture and decoration (Borbon et al., 2002; Liu et al., 2008; Ling and Guo, 2014).

Source 3 was distinguished by high emissions of C<sub>2</sub> hydrocarbons particularly ethane and ethene, and NO. Compared to source 1, this factor made much lower contributions to *n/i*-pentanes, eliminating the possibility of gasoline exhaust. In addition, a certain amount of CO confirmed this source as diesel exhaust.

Source 4 was dominated by high percentages of propane and *n/i*-butanes, which are typical tracers of LPG exhaust (Blake and Rowland, 1995; Liu et al., 2008). In addition, ethene, propene and CO showed high levels in the source profile, indicating that this source was related to combustion. Indeed, previous roadside and exhaust samples demonstrated that LPG-fueled vehicles emit significant amount of ethene and propene (Tang et al., 2007; Ling and Guo, 2014). Compared to sources 1 and 3, there were very minor emissions of benzene, toluene and ethylbenzene, which was a feature of LPG-fueled vehicular exhaust.



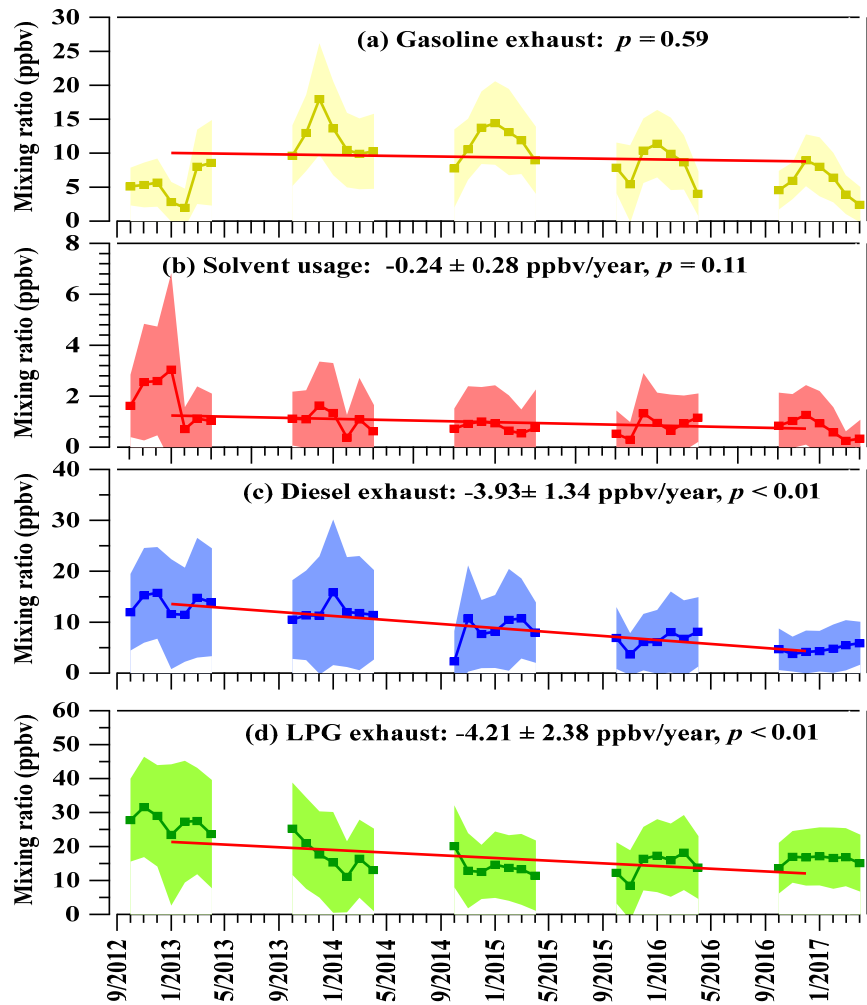
**Figure 5.** Average source profiles of VOCs and CO extracted from the PMF modeling at MK

Table S1 in the Supplementary Material shows the average mass concentration of each source and its percentage contribution to the total VOCs. The LPG-fueled vehicular emission was the most predominant contributor to VOCs in the roadside environment, with the average mass concentration of  $17.96 \pm 0.16 \mu\text{g}/\text{m}^3$ , followed by diesel exhaust ( $9.07 \pm 0.11 \mu\text{g}/\text{m}^3$ ) and gasoline exhaust ( $8.60 \pm 0.08 \mu\text{g}/\text{m}^3$ ). The contribution of solvent usage was the lowest with mass concentration of  $1.05 \pm 0.02 \mu\text{g}/\text{m}^3$ . The results of source apportionment reflected the urgent necessity of controlling LPG-fueled vehicular emissions, in line with Lyu et al (2016a) who found that LPG exhaust accounted for  $39.1 \pm 0.1\%$  -  $51.5 \pm 0.1\%$  of VOCs measured at MK.

Figure 6 shows the PMF-resolved temporal trends of VOCs emitted from each individual source from Period-1 to Period-5. The VOCs emitted from LPG-fueled vehicles decreased from Period-1 (before) to Period-2 (during) then to Periods 3-5 (1 to 3 years after the LPG program) ( $p < 0.01$ ), with a rate of  $4.21 \pm 2.38 \text{ ppbv}/\text{year}$ . In fact, the reduction of VOC emissions from LPG-fueled vehicles was previously identified by Lyu et al. (2016a). Here, the steady decline of LPG emitted VOCs after the replacement program likely indicated the continuous effectiveness of the LPG program on VOC control. The VOCs from diesel vehicular exhaust decreased at a rate of  $-3.93 \pm 1.34 \text{ ppbv}/\text{year}$  ( $p < 0.01$ ), indicating that the DCV program was in force, consistent with the results of Lyu et al. (2017). Different from the findings of Lyu et al. (2017), VOCs from solvent usage did not show significant change during the study period



( $p > 0.1$ ), likely due to the fact that the MK site was a roadside site mainly affected by vehicular emissions (84 - 92%), and it was not appropriate to study the annual trend of solvent usage at this roadside site. In addition, the VOCs originated from gasoline vehicular exhaust also had no significant change at this roadside site throughout the study period, perhaps related to the stable number of gasoline-fueled vehicles running at this roadside site between 2012 and 2017 in Hong Kong (HKATC, 2016).



**Figure 6.** PMF-resolved temporal trends of VOCs emitted from each individual source from Period-1 to Period-5. The shaded area indicates the standard deviation of the data.

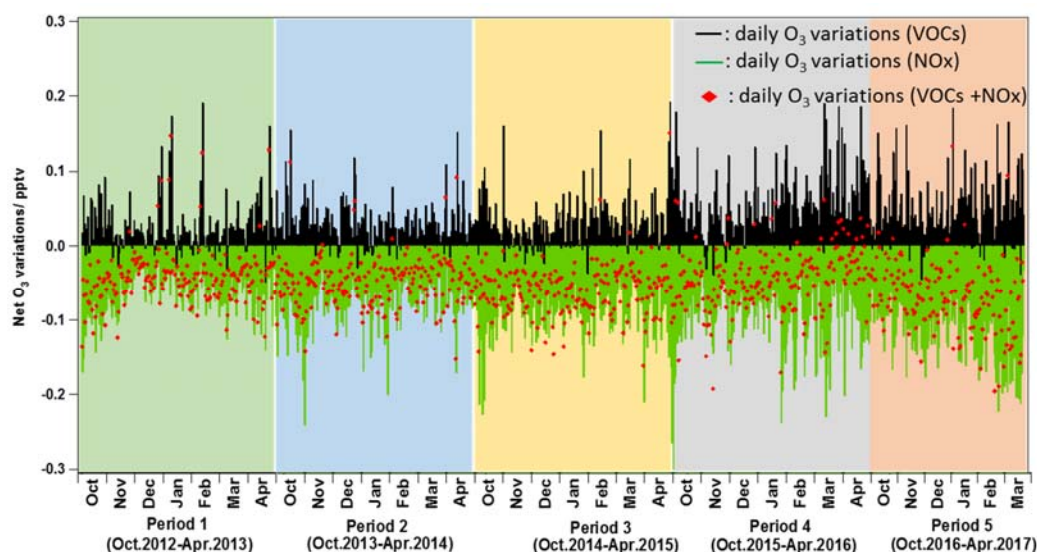
### 3.4 Influence of the replacement program on ozone production

As VOCs and  $\text{NO}_x$  are key precursors of tropospheric  $\text{O}_3$ , it is of interest to investigate how the reductions of VOCs and  $\text{NO}_x$  in LPG exhaust affected the  $\text{O}_3$  production at the roadside site (MK). Firstly,  $\text{O}_3$  was simulated by the PBM-MCM model (section 2.5.2) with the observed

concentrations of air pollutants as input, referred to as the base scenario. Then, a constrained scenario was constructed, in which VOCs and NO<sub>x</sub> emitted from LPG-fueled vehicles were excluded from the model input according to the source apportionment (section 3.3). The removal of VOCs and NO<sub>x</sub> were conducted separately or together. When only the VOCs in LPG exhaust were removed from the input of the constrained scenario, the difference of the simulated O<sub>3</sub> between the base and constrained scenario represented the contribution of LPG-related VOCs to O<sub>3</sub> production. Similarly, the contributions of LPG-related NO<sub>x</sub> and the entire LPG exhaust inclusive of VOCs and NO<sub>x</sub> to O<sub>3</sub> were also simulated, which are shown in Figure 7.

Overall, LPG exhaust made minor contributions to O<sub>3</sub> at MK, ranging from  $-4 \pm 0.7$  pptv to  $-6.7 \pm 0.7$  pptv ( $\sim 2.4\%$  -  $\sim 3.1\%$  of the observed O<sub>3</sub>). The overall negative contribution of the LPG exhaust to O<sub>3</sub> was explained by the greater O<sub>3</sub> depletion due to the titration by LPG emitted NO than the O<sub>3</sub> production by LPG emitted VOCs. From Period-1 to Period-5, the daily average O<sub>3</sub> production by LPG-related VOCs increased significantly at a rate of  $3.7 \times 10^{-2}$  pptv/day ( $p < 0.01$ ). At the same time, the O<sub>3</sub> depletion by LPG emitted NO also increased significantly at a rate  $6.1 \times 10^{-2}$  pptv/day ( $p < 0.01$ ). As a result, the net negative contribution of the LPG exhaust to O<sub>3</sub> production varied from  $-4 \pm 0.7$  pptv in Period-1 to  $-6.7 \pm 0.7$  pptv in Period-5, at a rate of  $1.9 \times 10^{-2}$  pptv/day ( $p < 0.01$ ).

It is noteworthy that VOCs and NO emitted from LPG-fueled vehicles decreased from period-1 to period-5 ( $p < 0.01$ ), inconsistent with the enhanced positive contribution of LPG emitted VOCs and negative contribution of LPG emitted NO to O<sub>3</sub> production. We found that the observed O<sub>3</sub> increased significantly at MK from 2012 to 2017, with an average rate of  $1.18 \pm 0.61$  ppbv/year. This increase was likely caused by other reasons, but not by the variation of LPG exhaust which generally made minor negative contributions to O<sub>3</sub> production. It was reasonable that the increase of O<sub>3</sub> enhanced the oxidative capacity of the atmosphere and hence the oxidation efficiency of VOCs at MK. This explained why the decreased LPG-emitted VOCs made increased contribution to O<sub>3</sub> production from period-1 to period-5. Similarly, the increase of O<sub>3</sub> also accounted for the increased O<sub>3</sub> depletion by LPG emitted NO, though in reality LPG-related NO decreased due to the replacement program. This implied that the effectiveness of some O<sub>3</sub> control measures might be masked or even reversed due to the uncontrolled increase of O<sub>3</sub> caused by other factors.



**Figure 7.** Net O<sub>3</sub> variations caused by VOCs, NO<sub>x</sub> and VOCs + NO<sub>x</sub> emitted from LPG-fueled vehicles. The shaded area indicates the Periods 1-5 separately.

#### 4 Conclusions

Between October 2013 and April 2014, ~75% of the old converters on LPG-fueled vehicles were replaced to improve air quality in Hong Kong. While our previous study indicated the immediate effectiveness of this program, whether it still played a role in air quality improvement was unknown to us. In this study, online trace gases and VOCs collected at MK from September 2012 to April 2017 were examined to assess the continuous effectiveness of replacing catalytic converters on LPG-fueled vehicles. The LPG-related VOCs, particularly propane and *n/i*-butanes, decreased significantly in the study period, implying effective control of VOC emissions. The reduction rates of propane and *n/i*-butanes remained high and constant after the program (i.e., Periods 3-5), suggesting that the vehicle owners routinely replaced the converters at suitable interval afterwards, or their vehicles might be caught by a remote sensing program checking excessive emissions. Furthermore, source apportionment analysis indicated that LPG-fueled vehicular emissions were the most predominant contributor to ambient VOCs. Fortunately, VOCs emitted from LPG-fueled vehicles decreased from 2012 to 2017, which was likely benefited from the continuous effectiveness of the intervention program on VOC control. Though the O<sub>3</sub> depletion by LPG exhaust increased from Period-1 to Period-5, the observed O<sub>3</sub> increased at MK ( $p < 0.01$ ), due to the fact that LPG only made minor negative contribution to O<sub>3</sub> production which however might be enhanced by some other factors. During the study

period, measures have been taken to restrict emissions from diesel and gasoline fueled vehicles in Hong Kong, such as phase III of the DCV program, retrofitting of SCR and establishment of low emission zones. These control measures might also contribute to the decrease of LPG-related VOCs. However, it could be anticipated that the reductions of VOC emissions from LPG-fueled vehicles played more important role, as propane and *n/i*-butanes were mostly emitted from LPG-fueled vehicles according to the source apportionment. Therefore, this study implied the continuous effectiveness of the LPG replacement program and would help policy-makers formulate future control strategies on vehicle emissions in Hong Kong. The impacts of the continuous emission reductions from LPG-fueled vehicles on air quality in entire Hong Kong (spanning from the urban to rural areas) and even in the greater PRD are to be evaluated.

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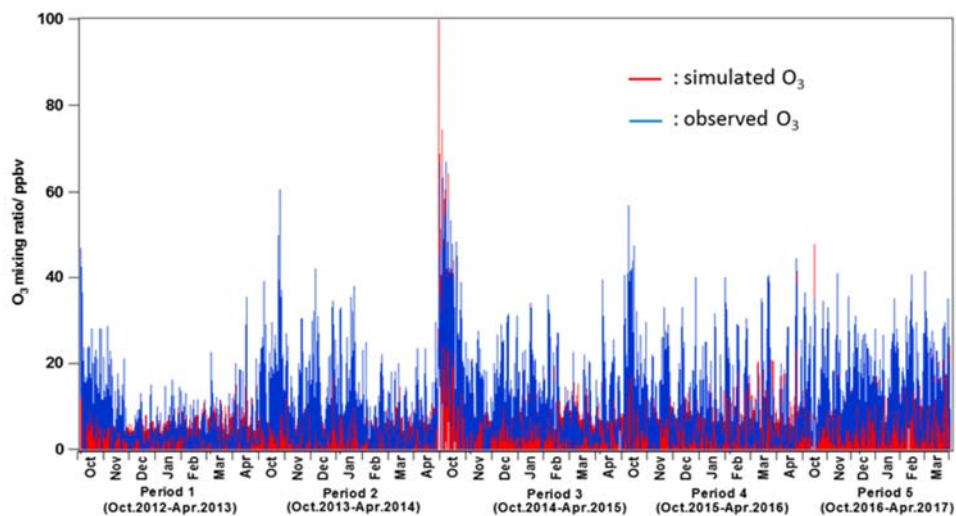
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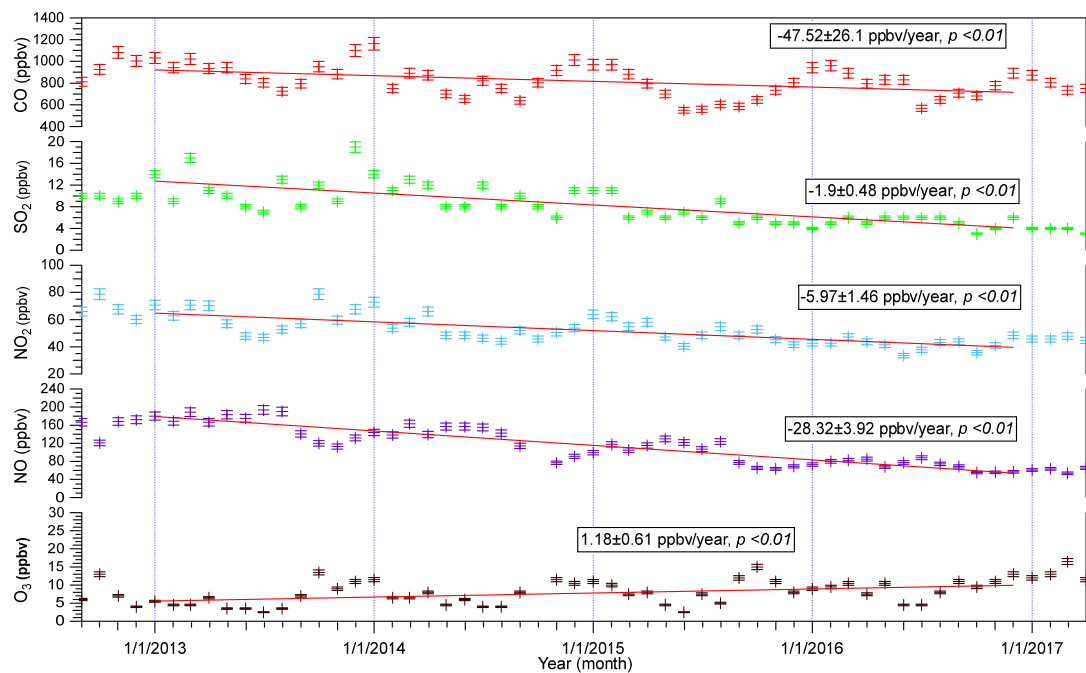
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583 **Supplementary Materials**



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585 **Figure S1:** Variations of hourly simulated and observed O<sub>3</sub> during daytime hours (07:00–19:00)  
586 at MK.



588  
589 **Figure S2:** Variations of monthly average trace gases at MK from September 2012 to April  
590 2017