1 Machine Learning of Phases and Mechanical Properties in Complex Concentrated Alloys

- 2 Jie XIONG^{b,c}, San-Qiang SHI^{b,c}*, Tong-Yi ZHANG^{a,d}*
- 3 ^a School of Materials Science and Engineering, Harbin Institute of Technology, Shenzhen, China
- 4 ^bDepartment of Mechanical Engineering, The Hong Kong Polytechnic University, Hong Kong, China
- 5 ^cShenzhen Research Institute, The Hong Kong Polytechnic University, Shenzhen, China
- 6 ^dMaterial Genome Institute, Shanghai University, Shanghai, China
- 7

8 Abstract

The mechanical properties of complex concentrated alloys (CCAs) depend on their formed 9 10 phases and corresponding microstructures. The data-driven prediction of the phase formation and associated mechanical properties is essential to discovering novel CCAs. The present work 11 collects 557 samples of various chemical compositions, comprising 61 amorphous, 167 single-12 phase crystalline, and 329 multi-phases crystalline CCAs. Three classification models are 13 developed with high accuracies to category and understand the formed phases of CCAs. Also, 14 two regression models are constructed to predict the hardness and ultimate tensile strength of 15 CCAs, and the correlation coefficient of the random forest regression model is greater than 0.9 16 17 for both of two targeted properties. Furthermore, the Shapley additive explanation (SHAP) values are calculated, and accordingly four most important features are identified. A significant 18 finding in the SHAP values is that there exists a critical value in each of the top four features, 19 which provides an easy and fast assessment in the design of improved mechanical properties of 20 CCAs. The present work demonstrates the great potential of machine learning in the design of 21 advanced CCAs. 22

23

24 Keywords

25 Materials Informatics, SHAP, Complex Concentrated Alloys, High Entropy Alloys

- 26
- 27
- 28
- 29

30 1. Introduction

The conventional alloys contain only one principal element in each alloy, which is usually 31 called the principal element based alloy. The mechanical properties of a principal element based 32 alloy are generally predominated by the principal element[1]. In contrast, complex concentrated 33 alloys (CCAs) are composed of multiple principal elements with more or less same percentage 34 35 and have excellent mechanical properties and unexpected micro-structures. The multiple principal elements render high configuration entropy if the CCAs in solid solution, thereby 36 called high-entropy alloys. Actually, CCAs encompass medium-entropy alloys (MEAs) 37 consisting of three or four principal components and high-entropy alloys (HEAs) consisting of 38 more than four principal components [2-7]. 39

According to the Gibbs phase rule F = C - P + 1 (where F is the degree of freedom, C is 40 the number of components and P is the number of phases), multiple elements might produce (C 41 + 1) equilibrium phases in CCAs, a maximum of six equilibrium phases are expected in the 42 case of a five-component CCA[8]. However, the high configuration entropy in a CCA can 43 enhance the formation of a single phase rather than multiple phases[3,9]. Thus, some CCAs 44 45 were reported to form single-phase disordered solid solutions with simple body-centered-cubic (BCC)[10], face-centered-cubic (FCC)[11], or hexagonal close-packed (HCP)[12] lattice 46 structure, and some CCAs were found to form single intermetallic phase, such as C14 Laves 47 phase[13] and B2 phase[14]. Multiple disordered solid-solutions[15] and multiple intermetallic 48 phases[16] are termed here as multi-phase CCAs. Besides, some amorphous CCAs have been 49 developed in the past decades[17]. Therefore, in the present work, a CCA belongs to one of 50 51 three crystalline sets of amorphous (AM), single-phase (SP), and multi-phases (MP).

Figure 1 shows that the mechanical properties of CCAs depend on their formed phases; the 52 MP CCAs are stronger than the SP CCAs, the CCAs that contain intermetallic phases (termed 53 as IM) are stronger than the CCAs that only consists of solid solutions (termed as SS), the SP-54 BCC and MP-BCC CCAs are harder and stronger than the SP-FCC and MP-FCC CCAs. In 55 materials computation, researchers conduct thermodynamic modeling, density functional 56 theory (DFT) calculations, and molecular dynamics (MD) simulation in order to design CCAs 57 from elements. For example, Gorsse et al.[18] applied the CALPHAD method to predict the 58 phase formation of CCAs, Huhn et al.[19] utilized the DFT approach to predict the phase 59

transformation in CCAs. The large number of elements in CCAs, however, gives rise to the 60 difficulty and cost of those computation techniques. Recently, the materials community 61 employs Machine learning (ML) to investigate and design novel materials, including steels[20-62 22], metallic glasses[23–27], shape memory alloys[28], and CCAs[29–31]. For instance, Wen 63 et al.[29] proposed a ML-based strategy to find new CCAs with high hardness in the Al-Co-Cr-64 Cu-Fe-Ni system. Islam et al.[30] employed a neural network to classify the corresponding 65 phase selection in 118 CCAs with an average test accuracy of higher than 80%. Huang et al.[31] 66 adopted and compared three different ML algorithms on 401 CCAs to predict the crystalline 67 phases. The distinction between single-phase and multi-phases and the formation of HCP solid 68 solutions, however, have not been considered yet in ML of CCAs. The hardness of only one 69 specific CCA system has been studied by ML [29], although big hardness data are available in 70 the literature for many CCA systems. The yield strength of CCAs has been investigated by the 71 ML approach [32], whereas the ultimate tensile strength of CCAs has not been investigated by 72

73 ML at all.



Figure 1. Ashby plot of ultimate tensile strength versus hardness of CCAs with different crystalline phases, where the first phase in each label of multiphases is the matrix, e.g., BCC is the matrix for BCC+FCC+B2.

In this work, we proposed a ML framework to investigate the phase formation ability and predict the mechanical properties of CCAs. A dataset of CCAs was constructed by collecting data from related studies. These data covered variables such as processing conditions, resulting phases, hardness, and ultimate tensile strength (UTS). A large number of features were initially introduced, and various feature selection algorithms were utilized to select key features during

model construction. A random forest classifier (RFC) was used to classify the formed phases in
CCAs, while a random forest regressor (RFR) was employed to predict the hardness and UTS
of CCAs.

83

84 **2.** Methodology

85 2.1 Data Acquisition

In addition to chemical compositions, the fabrication conditions of CCAs play an essential role 86 in the phase formation of CCAs. Therefore, only the fabrication by vacuum arc melting-cast is 87 considered here, and 750 as-cast CCAs are collected from five review articles [33-37] and 88 89 several related studies [37-54]. There have been some samples with conflict reports on phases in the 750 as-cast CCAs; excluding the conflict data reduces the data number to 557. The present 90 phase dataset includes 61 AM samples, 136 SP-SS samples, 31 SP-IM samples, 94 MP-SS 91 samples, and 235 MP-IM samples. The RFC is conducted on all 557 data, called ML-A dataset, 92 to classify AM, SP, and MP. The number of crystalline-set, termed ML-B dataset, is 496, on 93 which RFC will classify SS and IM. The 230 SS CCAs form ML-C dataset, with which RFC 94 classify HCP sub-cluster, BCC sub-cluster including SP and MP BCC CCAs, FCC sub-cluster 95 96 including SP and MP FCC CCAs, and mixture (MSS) sub-cluster including the mixture of BCC, 97 FCC, and HCP. Some of the 750 CCAs have reported hardness values, and a few of them have reported UTS values. If the reported values of hardness (or UTS) of a CCA are different, the 98 average value of hardness (or UTS) is adopted for that CCA. The hardness dataset and the UTS 99 dataset in the present study consist of 290 and 71 CCAs, respectively. All the collected data are 100 101 listed in Tables S1-S3 in the Supplementary Material.

102

103 2.2 Feature Construction

Table 1 lists feature blocks, including 22 elemental parameters, 5 thermodynamic parameters, and 3 valence electron distributions. The elemental properties of a CCA can be approximately estimated by the atomic average value \bar{x} as,

$$\bar{x} = \sum_{i=1}^{n} a_i x_i \tag{1}$$

where a_i and x_i is the atomic fraction and elemental properties of the *i*th element, respectively.

108 In addition, the mismatch in elemental properties (δ_x) of constituent elements is also employed 109 to describe each CCA, which is defined by

$$\delta_x = \sqrt{\sum a_i \, (1 - \frac{x_i}{\bar{\chi}})^2}$$
(2)

110 The five thermodynamic parameters are the enthalpy of mixing (H_{mix}) [56], the entropy of 111 mixing (S_{mix}) [56], the entropy of fusion $(S_f = \overline{H_f}/\overline{T_m})$ with $\overline{H_f}$ and $\overline{T_m}$ being average heat 112 of fusion and average melting temperature, respectively [57], Gibbs free energy of mixing (G_{mix}) 113 [58], and the $1/\Omega = |H_{\text{mix}}|/\overline{T_m}S_{\text{mix}}$ parameter [12, 56] The valence electron distributions comprise 114 the fraction (f_k) of the average electrons in the *s*, *p*, *d* valence orbitals, i.e.,

$$f_k = \overline{k\text{VEC}} / \frac{1}{\text{VEC}} \quad (k = s, p, d \text{ orbitals})$$
 (3)

115 where the overbar means the average obtained by Equation 1. In this way, 30 features are 116 initially generated with the feature blocks and then normalized to [0, 1] according to:

$$x' = \frac{x - \min x}{\max x - \min x} \tag{4}$$

where x' is the normalized feature, min x and max x are the minimum and maximum values ofthe original feature x, respectively.

119

120 **2.3 Evaluation Metrics**

121 The performances of classification are evaluated by the miss rate and accuracy, which are122 defined by

$$Miss Rate = \frac{F}{T+F}$$
(5)

$$Accuracy = 1 - Miss Rate = \frac{T}{T+F}$$
(6)

where *T* and *F* stand for the number of correctly and wrongly classified samples, respectively. The performances of regression models are evaluated by the normalized root mean squared error (*NRMSE*) and correlation coefficient (r):

$$NRMSE = \frac{\sqrt{\sum_{i=1}^{n} \frac{1}{n} (\hat{y}_i - y_i)^2}}{\bar{y}}$$
(7)

$$r = \sqrt{\frac{\sum_{i=1}^{n} (\hat{y}_{i} - \bar{y})^{2}}{\sum_{i=1}^{n} (y_{i} - \bar{y})^{2}}}$$
(8)

126 y_i and \hat{y}_i is the actual and corresponding predicted value, and \bar{y} is the mean of actual values.

127 The value of r = 1 indicates perfect fitting.

	Description	Abb.	Definition
	Atomic Number	AN	
	Metallic Radius	MR	_
	Melting Point	Tm	_
Elemental parameters	Boiling Point	Tb	_
	Pauling Electronegativity	XP	$\bar{x} = \sum a_i x_i$
	Electron Affinity	Eea	
	First Ionization Potential	<i>I</i> 1	$\delta_x = \sqrt{\sum a_i \left(1 - \frac{x_i}{\bar{\chi}}\right)^2}$
	Molar Heat Capacity	Cm	
	Thermal Conductivity	Κ	_
	Valence Electron	VEC	_
	Heat of Fusion	Hf	_
Thermodynamic	Enthalpy of mixing	H _{mix}	$H_{\rm mix} = 4 \sum_{i=1}^{N} \sum_{j=1}^{N} \Delta H_{ij} a_i a_j$
	Entropy of mixing	S _{mix}	$S_{\min} = -R \sum_{i=1}^{N} a_i \ln a_i$
parameters	Entropy of fusion	S_f	$S_f = \overline{H_f} / \overline{T_m}$
	Gibbs free energy of mixing	$G_{\rm mix}$	$G_{mix} = H_{\rm mix} - S_{\rm mix} \cdot \overline{T_m}$
	The reciprocal of \varOmega	$1/\Omega$	$1/\Omega = \frac{ H_{\rm mix} }{T_m} S_{\rm mix}$
	fraction of the electrons in the <i>s</i> valence orbitals	f_s	$f_s = \overline{s \text{VEC}} / \overline{\text{VEC}}$
VEC distributions	fraction of the electrons in the p valence orbitals	f_p	$f_s = \overline{p\text{VEC}} / \frac{1}{\text{VEC}}$
	fraction of the electrons in the d valence orbitals	f_d	$f_s = \overline{d\text{VEC}} / \overline{\text{VEC}}$

128 Table 1. Feature blocks consisting of elemental parameters, thermodynamic parameters, and VEC distributions.

129

130 **2.4 Validation method**

131 The ten-fold cross-validation method is employed, where each dataset is randomly divided into132 ten folds, nine folds as the training set and one fold as the validation set. In turn, the ML model

is trained on the training set and tested on the validation set ten times. The average of ten testperformances then gives the cross-validation performance.

135

136 **3. Results and Discussion**

137 **3.1 Classification of Phase-formations**

A good ML model should provide as good as possible performance with as less as possible number of features to achieve a balance between accuracy and complexity. Thus, feature selection should be conducted to remove redundant features. First, linear-correlated features are removed by using the Pearson correlation coefficient (PCC, ρ),

$$\rho_{(x_1, x_2)} = \frac{cov(x_1, x_2)}{\sigma_{x_1} \sigma_{x_2}}$$
(9)

where σ_{x_1} and σ_{x_2} denote the standard deviation of features x_1 and x_2 , and *cov* is the covariance. The absolute value of PCC bigger than 0.9 indicates a strong linear correlation between two features and then only one feature of the correlated pair is selected. After that, sequential backward selection (SBS) wrapped with RFC, sequential forward selection (SFS) wrapped with RFC, and mean decrease impurity (MDI) derived from RFC [59] are conducted to screen further the PCC initially selected features. The three feature selection methods are denoted by SBS+RFC, SFS+RFC, and MDI/RFC accordingly.

149

Table 2. The PCC selected features

Dataset	PCC selected Features
MLA	$\overline{AN}, \delta_{AN}, \overline{MR}, \delta_{MR}, \delta_{Tb}, \overline{Tm}, \delta_{Tm}, \overline{Eea}, \delta_{Eea}, \delta_{XP}, \overline{Hf}, \delta_{Hf},$
WIL-A	$\overline{I1}, \delta_{I1}, \overline{Cm}, \delta_{Cm}, \overline{K}, \delta_K, \overline{\text{VEC}}, \delta_{\text{VEC}}, f_p, H_{\text{mix}}, S_{\text{mix}}, S_f$
MUD	\overline{AN} , δ_{AN} , \overline{MR} , δ_{MR} , δ_{Tb} , \overline{Tm} , δ_{Tm} , \overline{Eea} , δ_{Eea} , \overline{XP} , δ_{XP} , δ_{Hf} ,
IVIL-D	$\overline{I1}, \delta_{I1}, \overline{Cm}, \delta_{Cm}, \overline{K}, \delta_K, \overline{\text{VEC}}, \delta_{\text{VEC}}, f_p, f_d, H_{\text{mix}}, S_{\text{mix}}, S_f, G_{\text{mix}}$
ML-C	\overline{AN} , δ_{AN} , δ_{MR} , δ_{Tb} , \overline{Tm} , δ_{Tm} , \overline{Eea} , δ_{Eea} , \overline{XP} , δ_{XP} , δ_{Hf} , δ_{I1} ,
	\overline{Cm} , δ_{Cm} , \overline{K} , δ_K , $\overline{\text{VEC}}$, δ_{VEC} , f_p , $H_{ ext{mix}}$, $S_{ ext{mix}}$, S_f , $G_{ ext{mix}}$, $1/\Omega$

Figure 2 shows the PCC map between features in the ML-A dataset with the highlighting of strong correlation features. There are six strong-correlated feature pairs, namely, $\overline{MR} - \overline{XP}$, $\overline{Tb} - \overline{Tm}$, $\overline{VEC} - f_s$, $\overline{VEC} - f_d$, $H_{\text{mix}} - G_{mix}$ and $H_{\text{mix}} - 1/\Omega$, from which the four features of \overline{MR} , \overline{Tm} , \overline{VEC} , and H_{mix} are selected [36,60,61]. Thus, the PCC selects 24 features in the

- 155 ML-A dataset, which are listed in Table 2. Similarly, the PCC approach (see supplementary
- material as Figure S1-S2 for details) screens out 26 and 24 features in the ML-B and ML-C
 datasets, respectively, which are listed in Table 2 as well.



The three feature selection methods of SBS+RFC, SFS+RFC, and MDI/RFC are thus conducted to screen further and rank the PCC selected features, and the results are shown in Figures 3(a), 3(b), and 3(c) for the ML-A, ML-B, and ML-C datasets, respectively, where the distance of each feature from the center indicates the feature rank, and the longer the feature distance, the higher the feature rank will be. The features are then ranked by integers sequentially, with one representing the most important feature.



Figure 3. The rank of features in the (a) ML-A, (b) ML-B, and (c) ML-C dataset resulting from on the SBS+RFC (in red), SFS+RFC (in blue), and MDI/RFC (in yellow).



Figure 4. The miss rate of each RFC model with 100 trees versus the feature number for the (a) ML-A, (b) ML-B, and (c) ML-C dataset.

Accordingly, Figures 4(a, b, c) show the miss rate of each RFC model versus feature rank number in the ML-A, ML-B, and ML-C datasets, respectively, where the miss rate is the averaged value of ten miss rates from the ten-fold cross-validation. As expected, when the feature number increases, the miss rate decreases, reaches a minimum and then increases. The validation minimum in the miss rate determines the feature number, which are 10, 7, and 9 corresponding to the SBS+RFC feature set for the ML-A dataset, the SFS+RFC feature set for

the ML-B dataset, and the SFS+RFC feature set for the ML-C dataset, respectively. In addition 172 to showing the determined feature numbers, Figures 4(a, b, c) all indicate the miss rate decreases 173 very slow when the feature number is larger than a critical value, which allows one to further 174 reduce the feature numbers under the cost of acceptable loss in the prediction power. Here, we 175 suggest a tolerance of 2% around the validation minimum of miss rate to balance performance 176 and complexity, which is clearly shown in each of Figures 4(a, b, c). With the 2% tolerance, we 177 select the OS-A, OS-B, and OS-C feature sets for the classifications of the ML-A, ML-B, and 178 ML-C datasets, respectively. Table 3 lists the OS-A, OS-B, and OS-C feature sets, indicating 179 each set contains 5 features. Table 3 indicates that feature H_{mix} appears in all three feature sets, 180 δ_{Cm} appears in both OS-A and OS-B, \overline{AN} appears in both OS-B and OS-C. Thus, the total 181 number of different features is 11, and the 11 features can be used to predict the mechanical 182 properties of CCAs. 183



Table 3. Finally selected features for ML-A, ML-B, and ML-C datasets.

Dataset	Feature set	Method	Features sorted by the feature rank
ML-A	OS-A	SBS+RFC	$\overline{MR}, \ \delta_{Tb}, H_{\text{mix}}, \ \overline{Eea}, \ \delta_{Cm}$
ML-B	OS-B	SFS+RFC	$\overline{\text{VEC}}, \ \overline{AN}, H_{\text{mix}}, \ \delta_K, \ \delta_{Cm}$
ML-C	OS-C	SFS+RFC	$\overline{AN}, \ \delta_{AN}, \ \delta_{XP}, \ \overline{K}, H_{\text{mix}}$

185

The OS-A feature set includes the five features of enthalpy of mixing (H_{mix}) , mismatch in 186 boiling temperature (δ_{Tb}), mismatch in molar heat capacity (δ_{cm}), average metallic radius (\overline{MR}), 187 and average electron affinity (\overline{Eea}). The previous studies indicate that a near zero of H_{mix} value 188 leads to the formation of single-phase CCAs, a medium negative H_{mix} promotes the formation 189 of amorphous; otherwise, the multi-phase CCAs exist [26,62,63]. This might be because the 190 Gibbs free energy of mixing is expressed by $G_{mix} = H_{mix} - S_{mix} \cdot \overline{T_m}$ and thus the smaller the 191 absolute value of H_{mix} is, the larger the contribution of S_{mix} will be. The boiling temperature 192 and the molar heat capacity may reflect the bond strength, thus, δ_{Tb} and δ_{cm} might be 193 194 considered as an estimator of the mismatch in metallic radius (δ_{MR}). Zhang et al. [56] found that amorphous phases are formed when δ_{MR} is large, while single-phase CCAs are more like 195 to possess a small δ_{MR} . Fukuhara et al. found that amorphous alloys tend to have more valence 196

electrons than crystalline alloys [64,65], thus the value of \overline{MR} and \overline{Eea} affects the formation of AM phase.

The OS-B feature set contains enthalpy of mixing (H_{mix}) , mismatch in thermal conductivity 199 (δ_K) , mismatch in molar heat capacity (δ_{cm}) , average valence electron (VEC), and average 200 atomic number (\overline{AN}). The presence of H_{mix} is in agreement with empirical rules [66,67]. The 201 thermal conductivity of Co, Cr, Fe, and Ni are all between 80~100 W/m/K, while Mn and Cu 202 203 have significantly different thermal conductivities of 8 and 400 W/m/K, respectively [26]. Thus, a larger δ_K might lead to the formation of solid solutions and a smaller value of δ_{cm} might 204 stabilize solid solutions. Tsai et al. [68] show that the σ phase only exists in the range of 6.88 < 205 VEC < 7.84 in as-cast Cr- and V- containing CCAs, Leong et al. [69] found that B2 phase 206 presence at $\overline{\text{VEC}} \le 6.81$ in CoCrFeNi-Al_x CCAs and C14 phase presence at $\overline{\text{VEC}} \le 7.4$ in 207 CoCrFeNi-Ti_x CCAs, these observations demonstrate that the $\overline{\text{VEC}}$ is another essential feature 208 of the phase formation in CCAs. The \overline{AN} is a coarse estimation of the average number of 209 electrons, too many electrons might destabilize the solid solutions [70]. 210

The OS-C feature set contains enthalpy of mixing (H_{mix}), mismatch in atomic number (δ_{AN}), 211 mismatch in Pauling electronegativity (δ_{XP}), average thermal conductivity (\overline{K}), and average 212 atomic number (\overline{AN}). The feature of H_{mix} appears in all the three feature sets, thereby indicating 213 that H_{mix} plays an important role in the phase formation ability. The FCC or HCP structure has 214 215 a packing fraction of 74%, while the BCC has a packing fraction of 68%. Therefore, BCC structures might have larger δ_{MR} (indicating larger δ_{AN}) than FCC or HCP structures [71]. A 216 larger δ_{XP} might promote the segregation behaviors in as-cast CCAs, which makes BCC phase 217 formation possible from the FCC matrix [72]. The phase formation depends on the cooling rate 218 during the cast and therefore \overline{K} is selected to be an important feature. The elements of Al, Cr, 219 Fe, and Mn are BCC stabilizers, while elements of Co, Ni, and Cu are FCC stabilizers in CCAs 220 [71]. Therefore, the feature \overline{AN} plays an important role in the phase classification. 221

The hyperparameters of RFC models are optimized with the selected features to the values listed in Table 4 by grid search algorithms from the presets [73]. The confusion matrices in Figure 5 (a, b, c) show the detailed cross-validation classification results on the ML-A, ML-B, and ML-C datasets, respectively. The RFC models achieve the correct predictions of 92.10%, 74.85%, and 85.25% to classify the three phases of AM, SP, and MP in the ML-A dataset, 83.04% and 92.48% to classify the two phases of SS and IM in the ML-B dataset, and 100.0%, 95.83%,

228 92.65%, and 87.50% to classify the four phases of HCP, BCC, FCC, and MSS in the ML-C

229 dataset, respectively. The results demonstrate that the RFC models are robust and reliable for

230 predicting the phase-formation in CCAs.

231

Table 4. Hyperparameters	s in the	RFC	models.
--------------------------	----------	-----	---------

Deterat	n_estimators	max_features	max_depth
Dataset	Preset: [64, 128]	Preset: [1,5]	Preset: None and [5, 10]
ML-A	88	4	None
ML-B	86	1	None
ML-C	92	3	7

232



Figure 5. The confusion matrixes of the RFC models on the (a) ML-A, (b) ML-B, and (c) ML-C datasets.

235

233

236 **3.2 Prediction of mechanical properties**



of 0.1550 with 6 features selected by BSS+RFR. Figure 7(a) plots the regression *NRMSE* in terms of UTS versus feature number obtained from the ten-fold cross-validation, showing that the *NRMSE* reaches the lowest value of 0.1886 with 5 features selected by BSS+RFR.



Figure 6. (a) The cross-validated NRMSE versus the feature number of each RFR model for hardness prediction,

(b) ML-predicted hardness versus the measured values.



Figure 7. (a) The cross-validated *NRMSE* versus the feature number of each RFR model for UTS prediction,(b) ML-predicted UTS versus the measured values.

251

Table 5.	Hyperpar	ameters in	the	RFR	models
raore o.	11, perpar	annecers m		10110	111000010

Datasat	n_estime	ators	max_fe	eatures	max_depth		CV r
Dataset	preset	result	preset	result	preset	result	C v -/
Hardness	[64, 128]	90	[1, 6]	2	None and [5, 10]	None	0.9062
UTS	[64, 128]	74	[1, 5]	1	None and [5, 10]	None	0.9498

The grid search algorithm is then employed here to tune the hyperparameters of RFR models on selected features, and the results are listed in Table 5. Figures 6(b) and 7(b) show the predicted hardness and UTS from the trained RFR models with the selected features against the actual values, respectively, indicating r = 0.9062 for hardness and r = 0.9498 for UTS.

In addition, the Shapley additive explanation (SHAP) approach is employed here to analyze features, which can explore the contribution of each of features in the trained RFR model to the predictions of hardness and UTS. The SHAP approach is developed based on the game theory that partitions each prediction into individual feature contributions [74]. The SHAP value of the *i*-th feature is calculated by

$$\varphi_i = \sum_{S \subseteq F \setminus \{i\}} \frac{|S|! (|F| - |S| - 1)!}{|F|!} [f_{S \cup \{i\}}(x_{S \cup \{i\}}) - f_S(x_S)]$$
(10)

where *F* is the set of all features, *S* is the subset of *F*, $S \cup \{i\}$ is the union of the subset S and the *i*-th feature, $f_{S \cup \{i\}}(x_{S \cup \{i\}})$ is the prediction of the model with the *i*-th feature, and $f_S(x_S)$ is the prediction of the model trained without the *i*-th feature. Adding up contributions from all features yields the prediction, i.e.,

$$\hat{y} = \hat{y}_0 + \sum_{i=1}^n \varphi_i \tag{11}$$

where \hat{y} is the predicted value and \hat{y}_0 is the prediction without any features, Equations (10, 11) are applied to every datum, viz., the SHAP value varies with the datum. The mean absolute value of the SHAP values for each feature can be computed here to get its impact on the targeted property. Furthermore, since hardness and UTS are both such properties with positive values that when the SHAP value of a feature is positive, the feature improves the properties; conversely, a feature with a negative SHAP value weakens them.

Figure 8(a) plots and ranks the mean absolute SHAP values of 6 hardness features selected by BSS+RFR, where the mean absolute SHAP values of the three most important features ($\overline{\text{VEC}}$, H_{mix} , δ_{XP}) are much greater than other features. Figures 8 (b-g) plot the SHAP value on every datum of the hardness data, of the six selected features, $\overline{\text{VEC}}$, H_{mix} , δ_{XP} , \overline{AN} , δ_{Cm} , and \overline{K} respectively. An extremely important finding is that there exists a critical value to each of the three most important features, which almost sharply separates the SHAP values into positive and negative two regions. The SHAP values are positive when the features $\overline{\text{VEC}} < 7.67$ 278 $H_{\text{mix}} < -9.8 \text{ kJ/mol}$, and $\delta_{XP} > 0.067$, regardless there are some data located in the wrong 279 region in the SHAP value versus δ_{XP} plot. This finding is significant to the design of high 280 hardness CCAs, because it provides the right value regions of the top three features.



281

Figure 8. (a) Ranked mean absolute value of SHAP values of the 6 BSS+RFR selected features for hardness. The SHAP values (negative in blue, positive in red) of (b) $\overline{\text{VEC}}$, (c) H_{mix} , (d) δ_{XP} , (e) \overline{AN} , (f) δ_{Cm} , and (g) \overline{K} for every one of the data.

285

The same behavior regarding the separation of positive and negative SHAP values is found 286 in the plots of SHAP value versus UTS feature. Figure 9(a) shows the mean absolute SHAP 287 values of 5 UTS features selected by BSS+RFR and the two most important features of δ_{Tb} 288 and H_{mix} have the mean absolute SHAP value greater than 100 MPa. Figure 9(b-c) show that 289 the SHAP values are positive when $\delta_{Tb} > 0.15$ and $H_{mix} < -14.6$ kJ/mol, regardless there 290 is one datum located in the wrong region in the SHAP value versus H_{mix} plot. If one is going 291 to design high UTS CCAs, one can select elements with the top two features located in the right 292 293 regions.



294

Figure 9. (a) Ranked mean absolute value of SHAP values of the 5 BSS+RFR selected features for UTS. The SHAP values (negative in blue, positive in red) of (b) δ_{Tb} , (c) H_{mix} , (d) \overline{AN} , (f) \overline{K} , and (g) δ_{K} for every one of the data.

299 3.3 Design of high hardness and high UTS CCAs

The SHAP values of these top features provide guidance to the design of novel CCAs with high hardness and high UTS. As an example, we take the CoCrFeNi CCA as the parent CCA to show how to strengthen and harden it, under the guidance of the SHAP values, by adding another element, which leads to equimolar quinary CCAs with each having five elements. These transition elements of Ti, V, Mn, Cu, Zr, Nb, and Hf are selected as candidates. The feature values of $\overline{\text{VEC}}$, H_{mix} , δ_{XP} , and δ_{Tb} in the parent CoCrFeNi CCA are 8.25, -3.75 kJ/mol, 0.0531, and 0.0328, respectively, as shown in Table 6. Simple calculations give the feature values of $\overline{\text{VEC}}$, H_{mix} , δ_{XP} , and δ_{Tb} in CoCrFeNi-X (X=Ti, V, Mn, Cu, Zr, Nb, or Hf), which are shown in Table 6 also.

Clearly, adding Zr, Nb, or Hf element makes the feature values of $\overline{\text{VEC}}$ and H_{mix} lower 309 than the corresponding critical values and the feature values of δ_{XP} , and δ_{Tb} higher than the 310 corresponding critical values. Then, as expected, the ML predicted hardness and UTS of 311 CoCrFeNi-X (X=Zr, Nb, or Hf) are greatly enhanced, which might be caused by Laves phases. 312 On the other hand, adding V, Mn, or Cu cannot make the feature values of $\overline{\text{VEC}}$ and H_{mix} 313 lower than the corresponding critical values and the feature values of δ_{XP} , and δ_{Tb} larger than 314 the corresponding critical values. Therefore, the ML predicted hardness and UTS of CoCrFeNi-315 X (X= V, Mn, or Cu) are not improved significantly. In addition, adding Ti makes the feature 316 values of VEC and H_{mix} lower than the corresponding critical values and the feature value of 317 δ_{XP} larger than the corresponding critical value, the ML predicted hardness of CoCrFeNiTi is 318 319 thus improved significantly, while the predicted UTS is not improved due to a small δ_{Tb} value. 320

322

321 Table 6. The feature values, ML-predicted, and mechanical properties of CoCrFeNi-X (X=Ti, V, Mn, Cu, Zr, Nb,

Alloy	VEC	<i>H</i> _{mix}	$\delta_{_{XP}}$	δ_{Tb}	Hardness (HV)	UTS (MPa)	Phases
CoCrFeNi	8.25	-3.75	0.0531	0.0328	173.6 (129.8 [75])	527.3 (480 [76])	SP-FCC (FCC [75])
CoCrFeNiTi	7.4	-16.32	0.0802	0.0623	592.6	655.0	MP-IM
CoCrFeNiV	7.6	-8.96	0.0646	0.0754	377.5	424.1 (311 [39])	MP-IM (FCC+σ [39])
CoCrFeNiMn	8	-4.16	0.0783	0.1101	175.7 (144.0 [77])	495.9 (494 [39])	SP-FCC (FCC [77])
CoCrFeNiCu	8.8	3.2	0.0502	0.0474	152.9	558.3	SP-FCC
CoCrFeNiZr	7.4	-22.72	0.1244	0.1846	616.5	1133.3	MP-IM (BCC+C15 [78])
CoCrFeNiNb	7.6	-14.88	0.0694	0.2191	615.2 (602 [79])	1293.1	MP-IM (FCC+C14 [79])
CoCrFeNiHf	7.4	-19.52	0.1313	0.2047	586.1	1209.0	MP-IM (BCC+C36 [78])

or Hf) CCAs, where mechanical properties and structures in parentheses are measured values.

323

As seen in Table 6, a lower $\overline{\text{VEC}}$ [68,69] promotes the formation of IM phases, such as σ phases or Laves phases, which are harder than solid solutions. Therefore, a lower $\overline{\text{VEC}}$ can harden the CCAs. A smaller mismatch leads to the formation of solid solutions, which are softer and weaker than IM phases, thus a larger δ_{XP} and δ_{Tb} can harden and strengthen CCAs, respectively. A near zero of H_{mix} value leads to the formation of single-phase solid solutions (SP-FCC for CoCrFeNi, CoCrFeNiMn, and CoCrFeNiCu). Thus, a much negative H_{mix} can harden and strengthen CCAs.

331

332 Concluding remarks

The present work illustrates an ML framework to predict the formed phases, the hardness, and the UTS of CCAs. A dataset of 557 samples is constructed, and 30 initial features are proposed. The adopted feature selection method gives the five features of \overline{MR} , δ_{Tb} , H_{mix} , \overline{Eea} , and δ_{Cm} in the classification of AM, SP, and MP phases, the five features of \overline{VEC} , \overline{AN} , H_{mix} , δ_K , and δ_{Cm} in the classification of SS and IM, and the five features of \overline{AN} , δ_{AN} , δ_{XP} , \overline{K} , and H_{mix} in the classification of mixed solid solutions, pure BCC, FCC, and HCP. All the ML models can classify the formed phases accurately.

The most significant finding in the present ML regressions of hardness and UTS of CCAs 340 is the critical value in each of the four most important features ranked by the absolute SHAP 341 value. The critical feature value separates the SHAP values into positive and negative regions. 342 This means that the feature values in the positive/negative SHAP value region improve/impair 343 the mechanical properties of CCAs, thereby providing a straightforward assessment in the 344 design of high hardness and high UTS CCAs. A demonstration of such a novel design 345 methodology is illustrated in the present work. Since all data collected are as-cast CCAs, the 346 ML suggested high hardness and high UTS CCAs should be fabricated by the cast. 347

It should be kept in mind that the mathematic basis of ML is probability and statistics, and thus ML predictions depend highly on data and should be understood in the sense of probability. It is a great challenge to develop knowledge from data. Understanding why there exists the critical value in each of the four most important features calls for more deep and comprehensive investigations from both materials science and ML approaches.

353

354 Acknowledgments

The work is supported by the National Key R&D Program of China (No. 2018YFB0704404),

356	the H	Iong Kong Polytechnic University (internal grant nos. 1-ZE8R and G-YBDH), and the 111
357	Proje	ect of the State Administration of Foreign Experts Affairs and the Ministry of Education,
358	Chin	a (grant no. D16002).
359		
360	Soft	ware
361	All I	ML approaches are performed on Python. The SBS and SFS algorithms are available in
362	Mlxt	end libraries, the SHAP values are calculated by the SHAP library, and other algorithms
363	are a	vailable in scikit-learn libraries.
364		
365	CRe	diT authorship contribution statement
366	Jie X	Kiong: Conceptualization, Investigation, Methodology, Data curation, Manuscript draft.
367	San-	Qiang Shi: Supervision, Manuscript review, Funding acquisition.
368	Tong	g-Yi Zhang: Supervision, Methodology, Manuscript review & editing.
369		
370	Decl	aration of competing interest
371	The	authors declare that they have no known competing financial interests or personal
372	relat	ionships that could have appeared to influence the work reported in this paper.
373		
374	Refe	rences
375	[1]	J.W. Yeh, Ann Chim Sci Des Mater 31 (2006) 633-648.
376	[2]	M.H. Tsai, J.W. Yeh, Mater Res Lett 2 (2014) 107-123.
377	[3]	C. Zhang, M.C. Gao, in: High-Entropy Alloy, Springer International Publishing, Cham,
378		2016, pp. 399–444.
379	[4]	Z. Li, D. Raabe, JOM 69 (2017) 2099–2106.
380	[5]	O.N. Senkov, D.B. Miracle, K.J. Chaput, JP. Couzinie, J Mater Res 33 (2018) 3092-
381		3128.
382	[6]	K. Pan, Y. Yang, S. Wei, H. Wu, Z. Dong, Y. Wu, S. Wang, L. Zhang, J. Lin, X. Mao, J
383		Mater Sci Technol 60 (2021) 113–127.
384	[7]	Y. Wu, F. Zhang, X. Yuan, H. Huang, X. Wen, Y. Wang, H. Wu, X. Liu, H. Wang, S. Jiang,

- 385 Z. Lu, J Mater Sci Technol 62 (2020) 214–220.
- 386 [8] B. Cantor, I.T.H.H. Chang, P. Knight, A.J.B.B. Vincent, Mater Sci Eng A 375–377 (2004)
 213–218.
- 388 [9] M.C. Gao, D.B. Miracle, D. Maurice, X. Yan, Y. Zhang, J.A. Hawk, J Mater Res 33 (2018)
 3138–3155.
- 390 [10] M. Gao, D. Alman, Entropy 15 (2013) 4504–4519.
- 391 [11] T.T. Zuo, R.B. Li, X.J. Ren, Y. Zhang, J Magn Magn Mater 371 (2014) 60–68.
- 392 [12] A. Takeuchi, K. Amiya, T. Wada, K. Yubuta, W. Zhang, JOM 66 (2014) 1984–1992.
- 393 [13] Y.F. Kao, S.K. Chen, J.H. Sheu, J.T. Lin, W.E. Lin, J.W. Yeh, S.J. Lin, T.H. Liou, C.W.
- Wang, Int J Hydrogen Energy 35 (2010) 9046–9059.
- [14] N.Y. Yurchenko, N.D. Stepanov, S. V. Zherebtsov, M.A. Tikhonovsky, G.A. Salishchev,
 Mater Sci Eng A 704 (2017) 82–90.
- 397 [15] C.M. Lin, H.L. Tsai, Intermetallics 19 (2011) 288–294.
- 398 [16] X. Yang, S.Y. Chen, J.D. Cotton, Y. Zhang, JOM 66 (2014) 2009–2020.
- 399 [17] Q.W. Xing, Y. Zhang, Chinese Phys B 26 (2017) 1–9.
- 400 [18] S. Gorsse, F. Tancret, J Mater Res (2018).
- 401 [19] W.P. Huhn, M. Widom, JOM (2013).
- 402 [20] T. ZHANG, Y. HE, J. WANG, S. SUN, Sci Sin Technol 49 (2019) 1148–1158.
- 403 [21] J. Xiong, T. Zhang, S. Shi, Sci China Technol Sci 63 (2020) 1247–1255.
- 404 [22] X. Geng, H. Wang, W. Xue, S. Xiang, H. Huang, L. Meng, G. Ma, Comput Mater Sci405 (2020).
- 406 [23] Y.T. Sun, H.Y. Bai, M.Z. Li, W.H. Wang, J Phys Chem Lett 8 (2017) 3434–3439.
- 407 [24] F. Ren, L. Ward, T. Williams, K.J. Laws, C. Wolverton, J. Hattrick-Simpers, A. Mehta,
 408 Sci Adv 4 (2018).
- 409 [25] J. Xiong, T.Y. Zhang, S.Q. Shi, Mrs Commun 9 (2019) 576–585.
- 410 [26] J. Xiong, S.Q. Shi, T.Y. Zhang, Mater Des (2020).
- 411 [27] L. Ward, S.C. O'Keeffe, J. Stevick, G.R. Jelbert, M. Aykol, C. Wolverton, Acta Mater
 412 159 (2018) 102–111.
- 413 [28] D.Z. Xue, D.Q. Xue, R.H. Yuan, Y.M. Zhou, P. V Balachandran, X.D. Ding, J. Sun, T.
 414 Lookman, Acta Mater 125 (2017) 532–541.

- 415 [29] C. Wen, Y. Zhang, C.X. Wang, D.Z. Xue, Y. Bai, S. Antonov, L.H. Dai, T. Lookman, Y.J.
 416 Su, Acta Mater 170 (2019) 109–117.
- 417 [30] N. Islam, W. Huang, H.L. Zhuang, Comput Mater Sci (2018).
- 418 [31] W. Huang, P. Martin, H.L. Zhuang, Acta Mater (2019).
- 419 [32] U. Bhandari, M.R. Rafi, C. Zhang, S. Yang, Mater Today Commun (2020) 101871.
- 420 [33] D.B. Miracle, Jom 69 (2017) 2130–2136.
- 421 [34] J.P. Couzinié, O.N. Senkov, D.B. Miracle, G. Dirras, Data Br (2018).
- 422 [35] S. Gorsse, M.H. Nguyen, O.N. Senkov, D.B. Miracle, Data Br (2018).
- 423 [36] M.H. Tsai, J.W. Yeh, Mater Res Lett (2014).
- 424 [37] W.H. Wang, Prog Mater Sci 57 (2012) 487–656.
- 425 [38] A. V. Kuznetsov, D.G. Shaysultanov, N.D. Stepanov, G.A. Salishchev, O.N. Senkov,
 426 Mater Sci Eng A 533 (2012) 107–118.
- 427 [39] J.Y. He, W.H. Liu, H. Wang, Y. Wu, X.J. Liu, T.G. Nieh, Z.P. Lu, Acta Mater 62 (2014)
 428 105–113.
- [40] Y. Ma, Q. Wang, B.B. Jiang, C.L. Li, J.M. Hao, X.N. Li, C. Dong, T.G. Nieh, Acta Mater
 (2018).
- 431 [41] H. Huang, Y. Wu, J. He, H. Wang, X. Liu, K. An, W. Wu, Z. Lu, Adv Mater 29 (2017)
 432 1–7.
- 433 [42] A. Asabre, A. Kostka, O. Stryzhyboroda, J. Pfetzing-Micklich, U. Hecht, G. Laplanche,
 434 Mater Des 184 (2019) 108201.
- [43] G. Dirras, L. Lilensten, P. Djemia, M. Laurent-Brocq, D. Tingaud, J.P. Couzinié, L.
 Perrière, T. Chauveau, I. Guillot, Mater Sci Eng A (2016).
- 437 [44] L. Liu, J.B. Zhu, L. Li, J.C. Li, Q. Jiang, Mater Des 44 (2013) 223–227.
- 438 [45] L. Liu, J.B. Zhu, C. Zhang, J.C. Li, Q. Jiang, Mater Sci Eng A 548 (2012) 64–68.
- 439 [46] C. Ng, S. Guo, J. Luan, Q. Wang, J. Lu, S. Shi, C.T. Liu, J Alloys Compd (2014).
- [47] G.A. Salishchev, M.A. Tikhonovsky, D.G. Shaysultanov, N.D. Stepanov, A. V.
 [41] Kuznetsov, I. V. Kolodiy, A.S. Tortika, O.N. Senkov, J Alloys Compd (2014).
- [48] C.J. Tong, Y.L. Chen, S.K. Chen, J.W. Yeh, T.T. Shun, C.H. Tsau, S.J. Lin, S.Y. Chang,
 Metall Mater Trans A Phys Metall Mater Sci 36 (2005) 881–893.
- 444 [49] W.H. Liu, J.Y. He, H.L. Huang, H. Wang, Z.P. Lu, C.T. Liu, Intermetallics 60 (2015) 1–

8.

- 446 [50] Y.D. Wu, Y.H. Cai, T. Wang, J.J. Si, J. Zhu, Y.D. Wang, X.D. Hui, Mater Lett 130 (2014)
 447 277–280.
- 448 [51] Y. Deng, C.C. Tasan, K.G. Pradeep, H. Springer, A. Kostka, D. Raabe, Acta Mater 94
 449 (2015) 124–133.
- 450 [52] L. Zhang, Y. Zhou, X. Jin, X. Du, B. Li, Scr Mater (2018).
- 451 [53] Z.Y. Rao, X. Wang, J. Zhu, X.H. Chen, L. Wang, J.J. Si, Y.D. Wu, X.D. Hui,
 452 Intermetallics 77 (2016) 23–33.
- 453 [54] Y. Lu, X. Gao, L. Jiang, Z. Chen, T. Wang, J. Jie, H. Kang, Y. Zhang, S. Guo, H. Ruan,
 454 Y. Zhao, Z. Cao, T. Li, Acta Mater (2017).
- 455 [55] X. Jin, Y. Zhou, L. Zhang, X. Du, B. Li, Mater Lett (2018).
- 456 [56] Y. Zhang, Y.J. Zhou, J.P. Lin, G.L. Chen, P.K. Liaw, Adv Eng Mater 10 (2008) 534–538.
- 457 [57] Y. Zhang, T.T. Zuo, Z. Tang, M.C. Gao, K.A. Dahmen, P.K. Liaw, Z.P. Lu, Prog Mater
 458 Sci (2014).
- 459 [58] J.W. Yeh, Y.L. Chen, S.J. Lin, S.K. Chen, Mater Sci Forum (2007).
- 460 [59] L. Breiman, Mach Learn (2001).
- 461 [60] Q. Hu, S. Guo, J.M. Wang, Y.H. Yan, S.S. Chen, D.P. Lu, K.M. Liu, J.Z. Zou, X.R. Zeng,
 462 Sci Rep (2017).
- 463 [61] Y.F. Ye, Q. Wang, J. Lu, C.T. Liu, Y. Yang, Mater Today (2016).
- 464 [62] Y. Zhang, Y.J. Zhou, Mater Sci Forum (2007).
- 465 [63] R. Kozak, A. Sologubenko, W. Steurer, Zeitschrift Fur Krist (2015).
- 466 [64] M. Fukuhara, M. Takahashi, Y. Kawazoe, A. Inoue, J Alloys Compd (2009).
- 467 [65] Z. Zhou, Y. Zhou, Q. He, Z. Ding, F. Li, Y. Yang, Npj Comput Mater (2019).
- 468 [66] Y. Zhang, Y.J. Zhou, J.P. Lin, G.L. Chen, P.K. Liaw, Adv Eng Mater 10 (2008) 534–538.
- 469 [67] S. Guo, C.T. Liu, Prog Nat Sci Mater Int 21 (2011) 433–446.
- 470 [68] M.H. Tsai, K.Y. Tsai, C.W. Tsai, C. Lee, C.C. Juan, J.W. Yeh, Mater Res Lett 1 (2013)
 471 207–212.
- 472 [69] Z. Leong, J.S. Wróbel, S.L. Dudarev, R. Goodall, I. Todd, D. Nguyen-Manh, Sci Rep
 473 (2017).
- 474 [70] Y. Zhang, C. Wen, C. Wang, S. Antonov, D. Xue, Y. Bai, Y. Su, Acta Mater (2020).

- [71] S.A. Kube, S. Sohn, D. Uhl, A. Datye, A. Mehta, J. Schroers, Acta Mater (2019).
- 476 [72] Z.S. Nong, J.C. Zhu, Y. Cao, X.W. Yang, Z.H. Lai, Y. Liu, Mater Sci Technol (United
 477 Kingdom) (2014).
- 478 [73] T.M. Oshiro, P.S. Perez, J.A. Baranauskas, in: Lect Notes Comput Sci (Including Subser
 479 Lect Notes Artif Intell Lect Notes Bioinformatics), 2012.
- 480 [74] S.M. Lundberg, S.I. Lee, in: Adv Neural Inf Process Syst, 2017.
- 481 [75] W.R.W.L. Wang, W.R.W.L. Wang, S.C. Wang, Y.C. Tsai, C.H. Lai, J.W. Yeh,
 482 Intermetallics 26 (2012) 44–51.
- 483 [76] W.H. Liu, Z.P. Lu, J.Y. He, J.H. Luan, Z.J. Wang, B. Liu, Y. Liu, M.W. Chen, C.T. Liu,
 484 Acta Mater (2016).
- 485 [77] N.D. Stepanov, D.G. Shaysultanov, G.A. Salishchev, M.A. Tikhonovsky, E.E. Oleynik,
 486 A.S. Tortika, O.N. Senkov, J Alloys Compd 628 (2015) 170–185.
- 487 [78] M.H. Tsai, A.C. Fan, H.A. Wang, J Alloys Compd 695 (2017) 1479–1487.
- 488 [79] H. Jiang, L. Jiang, D. Qiao, Y. Lu, T. Wang, Z. Cao, T. Li, J Mater Sci Technol 33 (2017)
 489 712–717.