Modulation doping of transition metal dichalcogenide/oxide heterostructures†

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Control of carrier type and carrier density provides a way to tune the physical properties of two-dimensional (2D) semiconductors. Modulation doping of heterostructures can effectively inject carriers into or extract carriers from the 2D semiconductors, and eliminate the adverse effect from the ionized dopants. Here we first investigate the layer-dependent negative trion PL of 2D MoS2, and further construct heterostructures with transition metal dichalcogenides (TMDs) and transition metal oxides (TMOs). By choosing the oxide with different charge neutrality levels (CNLs), we demonstrate effective electron injection into MoS2 by TiO2 doping, and electron extraction from MoS2 by MoO3 doping. Photoluminescence (PL) spectra and electrical characterization show that thicker MoS2 flakes are more easily n-doped by TiO2, while thinner MoS2 flakes are more easily p-doped by MoO3. Our experimental results are in good agreement with theoretical calculations. The modulation doping with TMO is compatible with conventional Si processing and highly air-stable. This method can also be extended for the controllable doping of other 2D materials.

Introduction

Two-dimensional (2D) layered transition metal dichalcogenides (TMDs) are of great research interest because of their unique electronic, optical and optoelectronic properties.1–5 The modulation of carrier density in the TMDs provides a way to change their physical properties in a controllable manner. For typical TMD materials, the pristine carrier type is predominated by their intrinsic characteristics. For example, pristine MoS2 is of the n-type while WSe2 shows p-type or ambipolar characteristics (the carrier type of WSe2 is dependent on its thickness).6–8 Controllable injection and extraction of electrons/holes are important for effective n- and p-doping of TMDs, which are crucial for building complementary logic circuits in the future. On the other hand, the interaction between carriers and excitons enables an additional path for the PL modulation of 2D layered materials. Controlling the carrier type and density through modulation doping provides a way to switch between exciton- and trion-dominant PL and tune the PL intensity of 2D materials.9

Various methods have been developed to dope 2D layered TMDs,10 including substitutional doping during growth,11 ion implantation,12 plasma treating,13 etc. Although these doping methods have been demonstrated to be effective, they inevitably result in the distortion of TMD crystal structures, introduce ionized impurity scattering centers, and degrade the charge mobility. Modulation doping has been widely used in III–V compound heterojunction superlattices to eliminate the influence of ionized impurity scattering.14–16 In the heterojunction device, the ionized dopants and electrons are confined to two different adjacent layers. As a result, both the mobility and carrier concentration are well maintained. A variety of dopants have been studied for the surface charge transfer to TMDs, including gas molecules,17 organics,18 metal particles19 and oxides2,20 Among these kinds of dopants, oxide is one of the promising species due to its stability in ambience and compatibility with conventional Si processing technology.

In this work, we deposit transition metal oxide (TMO) on top of 2D MoS2 to form a heterojunction device. We choose TiO2 with a high charge neutrality level (CNL) and MoO3 with a low CNL for injecting and extracting electrons, respectively. Our optical and electrical characterization, together with density functional theory (DFT) calculations, shows that TiO2 and MoO3 deposition gives rise to effective n-doping and p-doping, respectively. In addition, we also reveal layer-dependent negative trion photoluminescence (PL) and doping effects in 2D layered MoS2. This approach opens up a way to control the characteristics of PL and the electrical transport of 2D layered materials.

Results and discussion

Layer-dependent negative trion PL of 2D MoS2

Because of the extremely strong Coulomb interaction in ultrathin 2D MoS2, optically generated excitons (i.e., electron–hole pairs)
and trions (i.e., charged excitons) can exist stably at room temperature.\textsuperscript{21} Electrons are excited to the conducting band, leaving holes in the valence band. Electrons and holes are then bound together by Coulomb interactions to form neutrally charged excitons (electron–hole pairs). Due to the Coulomb attraction between the hole in an exciton and a third electron, the formation process of exciton to trion is energy favourable.\textsuperscript{22} Trions are formed in the presence of residual free charges. Fig. 1(a) shows the PL spectra of a 3L MoS\textsubscript{2} as an example. Two pronounced peaks around 1.87 eV and 2.05 eV are clearly observed, defined as the A exciton peak and the B exciton peak, respectively. As schematically depicted in the inset, the A exciton has originated from the radiative recombination of the excited electrons in the conduction band minimum (CBM) with the holes in the valence band minimum (VBM); and the B exciton has originated from the radiative recombination of the excited electrons in the CBM with the lower branch of the split VBM.\textsuperscript{23} The A exciton of ultrathin MoS\textsubscript{2} can be further devolved into a neutron exciton (X) peak around 1.9 eV and a negative trion (X\textsuperscript{-}) peak around 1.85 eV.\textsuperscript{21}

Fig. 1(b) shows the PL spectra of MoS\textsubscript{2} evolved with the layer number from 1L to 4L. The PL spectra of over 60 MoS\textsubscript{2} flakes were collected and are summarized in Fig. 1(c) and (d). Fig. 1(c) shows the layer-dependent spectral weight of the X peak in the whole A exciton peak, which rapidly decreases from 90% in the 1L MoS\textsubscript{2} to 30% in the 3L–5L MoS\textsubscript{2}. There are two possible reasons accounting for the layer-dependent characteristic of the spectral weight of the X\textsuperscript{-} peak. First, electrons in MoS\textsubscript{2} originated from donor states introduced by surface sulfur vacancies,\textsuperscript{24–27} which are expected to become less prominent with the increase of body thickness. Second, when the thickness of MoS\textsubscript{2} increases, the dielectric screening of MoS\textsubscript{2} increases,\textsuperscript{28,29} which attenuates the Coulomb interaction\textsuperscript{30} and reduces the binding energy of trions. With the decrease of binding energy, trion systems are less stable and can be easily decomposed into excitons. Fig. 1(d) shows the photon energy of X and X\textsuperscript{-} peaks as a function of MoS\textsubscript{2} thickness. The PL peaks exhibit a clear red-shift with an increase of MoS\textsubscript{2} body thickness due to the decreased bandgap. In the inset of Fig. 1(d), we present the layer-dependent energy difference (binding energy of the negative trion from the exciton) between X and X\textsuperscript{-} peaks. This binding energy decreases with the increase of MoS\textsubscript{2} thickness.

**Layer-dependent n-doping of MoS\textsubscript{2} by TiO\textsubscript{2}**

Fig. 2 shows the PL spectra of 1L–4L MoS\textsubscript{2} before and after TiO\textsubscript{2} deposition. After 6 nm thick Atomic Layer Deposition (ALD) TiO\textsubscript{2} doping, we observe the characteristics of layer-dependent PL. The thicker MoS\textsubscript{2} sample has a smaller X\textsuperscript{-} spectral weight before the TiO\textsubscript{2} deposition, which is consistent with our previous discussions; while the X\textsuperscript{-} spectral weight increases more after the TiO\textsubscript{2} deposition. These changes in PL spectra are ascribed to the injection of electrons from TiO\textsubscript{2} into MoS\textsubscript{2}, where the electron density in MoS\textsubscript{2} is increased by TiO\textsubscript{2} doping.

To rule out the possibility that this PL change has originated from in-plane lattice strain due to oxide deposition on the surface, we also present the corresponding Raman spectra before and after doping in Fig. S1 (ESI\textsuperscript{†}). As widely reported by previous research studies, the in-plane lattice strain induces the split of the E\textsubscript{1g} peak and the blue-shift of both E\textsubscript{2g} and A\textsubscript{1g} peaks.\textsuperscript{31} However, from our Raman spectra, we observe no such Raman spectral change. Thus, it is suggested that the interlayer coupling between the TiO\textsubscript{2} layer and the MoS\textsubscript{2} surface is very weak, and this weak interaction induces negligible lattice strain to MoS\textsubscript{2}. So the PL spectral change has not originated from the lattice strain. To confirm this speculation, we further deposited 30 nm thick TiO\textsubscript{2} onto 1L–3L MoS\textsubscript{2}. The Raman and PL spectra are shown in Fig. S2 (ESI\textsuperscript{†}). We still observe a more significant increase of the X\textsuperscript{-} spectral weight in thicker MoS\textsubscript{2} and a negligible shift of Raman peaks. It is noteworthy that the attenuation of Raman intensity after the doping is possibly due to the block of light of the 30 nm thick TiO\textsubscript{2} layer. On the other hand, the change of the dielectric environment introduced by TiO\textsubscript{2}

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**Fig. 1** PL analysis of ultrathin MoS\textsubscript{2} flakes. (a) PL spectrum of 3L MoS\textsubscript{2}. The inset shows the radiative recombination path of A and B excitons, respectively. (b) PL spectra of 1L–4L MoS\textsubscript{2}. (c) Layer-dependent spectral weight of the X\textsuperscript{-} peak. (d) Layer-dependent photon energy of the X peak and the X\textsuperscript{-} peak in the MoS\textsubscript{2} samples from 1L to 5L. The inset shows the layer-dependent energy difference between the X peak and the X\textsuperscript{-} peak.

**Fig. 2** PL spectra of ultrathin MoS\textsubscript{2} flakes before and after depositing 6 nm thick TiO\textsubscript{2}. The thicknesses of MoS\textsubscript{2} samples are (a) 1L, (b) 2L, (c) 3L and (d) 4L.
deposition plays an important role in the change of the PL spectra of MoS2. In both cases (6 nm thick and 30 nm thick TiO2), the A exciton peak is found to be red-shifted after the TiO2 deposition, which is more obvious in the case of 30 nm thick TiO2 deposition. This is because the bandgap of MoS2 is decreased in the environment with a higher dielectric constant (the dielectric constant of TiO2 is ~80). Furthermore, from the point of view of enhanced dielectric screening in MoS2 due to the deposition of the TiO2 layer, the Coulomb interaction in MoS2 is attenuated (similar to the effect of increasing MoS2 body thickness). The attenuated Coulomb interaction decreases the binding energy of the negative trion, thus decreasing the stability of the trion system and the PL intensity of the X peak. However, in our experiments, the PL intensity of the X peak is obviously increased after the TiO2 doping. Therefore, the increase of the X peak PL is due to the increase of electron density in MoS2, which originated from the n-doping effect by TiO2.

To further confirm and quantify the n-doping effect, we fabricated a back-gated MoS2 field effect transistor (FET) and doped it with 6 nm thick ALD TiO2. Fig. 3(a) depicts the cross-sectional schematic of a back-gated MoS2 FET with its channel covered by TMO. Fig. 3(b) shows the transfer curve of a 5L MoS2 FET before (black line) and after (red line) the 6 nm thick TiO2 doping. The pristine device shows typical n-type behaviour, consistent with previous reports. After n-doping, the ON current (I_on) is increased by a factor of 4 at V_bg = 50 V. Fig. 3(c) and (d) show the output curves of the same device before and after doping, respectively. Again, I_on at V_bg = 50 V is found to be amplified about 5 times. It is noteworthy that the device retains its semiconducting characteristics after the doping, indicating that the n-doping effect is non-degenerate. Furthermore, we can extract 2D sheet carrier density \( n_{2D} \) and field-effect mobility \( \mu_{FE} \) according to eqn (1) and (2), where \( L \) and \( W \) are the channel length and the channel width, respectively, and \( C_{ox} \) is the capacitance of the gated oxide (i.e., \( 1.16 \times 10^{-8} \text{ F cm}^{-2} \) for 300 nm thick SiO2 in this study). The extracted \( n_{2D} \) before doping is \( 1.17 \times 10^{12} \text{ cm}^{-2} \), which is consistent with previous reports, while the extracted \( n_{2D} \) after the doping is \( 1.95 \times 10^{13} \text{ cm}^{-2} \). This doping level (0.78 \( /C_0 \)) is comparable to previous reports, while the extracted \( n_{2D} \) after the doping is \( 1.95 \times 10^{13} \text{ cm}^{-2} \) for TiO2-doped graphene.

To further support our findings, DFT calculations were utilized to simulate the doping effect. The anatase phase TiO2 structure was used in this simulation since it is a stable phase at a relatively low temperature. Fig. 4(a) and (b) show the atomic and orbital projected density of states of pristine 1H-MoS2, 1H-MoS2 in the TiO2-doped MoS2 system, anatase TiO2, and anatase TiO2 in the TiO2-doped MoS2 system. The bandgap of pristine 1H-MoS2 is calculated to be \( \sim 1.86 \text{ eV} \). And both the conduction band minimum (CBM) and the valence band maximum (VBM) of MoS2 are populated by the 4d orbital of the Mo atom, which are highly hybridized with the 3p orbital of the S atom. After the TiO2 doping, the Fermi level of 1L MoS2 is up-shifted approximately 0.76 eV towards the CBM, indicating

![Fig. 3](image-url) - Electrical characterization of a MoS2 FET for n-doping by TiO2.

(a) Cross-sectional schematic of a TMO doped MoS2 FET. (b) Transfer curve of a 5L MoS2 FET before (black line) and after (red line) being doped by 6 nm thick TiO2. (c) Output curves of the FET before doping; (d) output curves of the FET after doping.

![Fig. 4](image-url) - DFT simulation results of the n-doping process and band structures of MoS2 and TiO2. (a) DOS of 1L 1H-MoS2 before (upper panel) and after doping (lower panel). (b) DOS of anatase TiO2 before (upper panel) and after doping MoS2 (lower panel). (c) Band alignment of 1L MoS2 and TiO2 before doping. (d) Band alignment of 1L MoS2 and TiO2 after doping. The dashed lines denote the CNL.
the n-doping effect. Also, the bandgap of MoS$_2$ is obviously shrunk, which is due to the enhanced dielectric screening introduced by TiO$_2$ deposition. From Fig. 4(b), we observe a significant change in the DOS of TiO$_2$ after the n-doping. Compared with the pristine state, the presence of the Ti-5d level in the conduction band is dramatically decreased. This change indicates that electrons are mainly transferred from the Ti-5d level of TiO$_2$ to MoS$_2$, leaving the Ti-5d level anti-bonding. Similar findings regarding the simulation of the TiO$_2$ doping process can be found in the work of N. Kaushik et al.\textsuperscript{40}

Fig. 4(c) shows the schematic band structures of 1L MoS$_2$ versus anatase TiO$_2$. The CBM and CNL of MoS$_2$ are about 0.5 eV lower than those of TiO$_2$. Driven by this energy difference, electrons in the CB of TiO$_2$ flow into the CB of MoS$_2$, until an interface dipole pointing from MoS$_2$ to TiO$_2$ is built up (Fig. 4(d)) and then the doping process reaches its balance state. As widely acknowledged, electrons in ultrathin MoS$_2$ flakes have originated from surface sulfur vacancies. It is reasonable to expect that this surface effect has a smaller influence on the electronic structure of thicker MoS$_2$ samples. The CNL of MoS$_2$ thin flakes lowers when the thickness of MoS$_2$ increases. Considering that the bandgap of MoS$_2$ decreases with the increase of body thickness, the CNL of thicker MoS$_2$ samples quickly lowers down. For thicker MoS$_2$ samples, the energy difference between its CNL and that of TiO$_2$ enlarges, thus the n-doping effect is more obvious.

**Layer-dependent p-doping of MoS$_2$ by MoO$_3$**

MoO$_3$ is one type of TMOs with a relatively low CNL. We choose it for p-doping MoS$_2$. Fig. 5 shows the PL spectra of 1L and 3L MoS$_2$ before and after 6 nm thick MoO$_3$ doping. In both cases, a significant decrease of the X$^+$ peak spectral weight is observed. The corresponding Raman spectra of these samples are shown in Fig. S3 (ESI†). No obvious shift in the positions of Raman peaks is observed, suggesting that the interlayer coupling between the MoO$_3$ layer and the MoS$_2$ surface is also very small. Also, it is noteworthy that the dielectric constant of MoO$_3$ is small (~5.5). Therefore, it has a negligible influence on the PL spectra of MoS$_2$.

Electrical characterization of MoS$_2$ FETs is used to further quantify the p-doping effect. Fig. 6(a) is the transfer curve of a 5L MoS$_2$ FET before and after p-doping. After the p-doping, the $I_{ON}$ at $V_{bg} = 50$ V is decreased 5 times and the threshold voltage ($V_{th}$) is increased by ~24 V. The extracted $n_{2D}$ values are $2.3 \times 10^{12}$ cm$^{-2}$ before the doping and $1.7 \times 10^{12}$ cm$^{-2}$ after the p-doping. The electron density in MoS$_2$ is depleted by about $0.6 \times 10^{13}$ cm$^{-3}$, which is comparable to the doping level of 6 nm thick TiO$_2$ ($0.78 \times 10^{12}$ cm$^{-2}$). The corresponding output curves of the device are illustrated in Fig. 6(b) (before doping) and Fig. 6(c) (after doping). As expected, we observe a decrease of $I_{ON}$ in the positive gate range (0–50 V). The gate voltage required to turn on the channel ($V_{on}$) is clearly increased from 20 V to 30 V. The $n_{2D}$ and $\mu_{FE}$ values of MoS$_2$ devices before and after the p-doping are summarized in Fig. 6(d). For thicker MoS$_2$ samples, the depleted electron density is smaller.

This layer dependence can also be ascribed to the decrease of the CNL of MoS$_2$ with the increase of body thickness, as discussed in the case of n-doping by TiO$_2$. Due to the reduced carrier density, electrons undergo a stronger scattering effect from the interface defects, which will significantly reduce the carrier mobility.

DFT calculation results of this p-doping process are shown in Fig. 7. In Fig. 7(a), the Fermi level of 1L MoS$_2$ is moved down towards the VB, demonstrating the p-doping effect. The DOS of MoO$_3$ in Fig. 7(b) shows that the population of the Mo-4d orbital is increased after the doping, indicating that electrons are transferred from the Mo-4d orbital in MoS$_2$ to the Mo-4d orbital in MoO$_3$. Fig. 7(c) illustrates the schematic band structure of 1L MoS$_2$ versus MoO$_3$. The CNL of MoS$_2$ is 2.2 eV higher than that of MoO$_3$. Due to this energy difference, electrons in the CB of MoS$_2$ are extracted to the CB of MoO$_3$ until an interface dipole is built up (Fig. 7(d)). When the CNL of MoS$_2$ is lowered down by increasing the body thickness, the energy difference between it and the CNL of MoO$_3$ is decreased, thus leading to a decreased p-doping effect.

**Experimental**

Fabrication of MoS$_2$ FETs

MoS$_2$ crystals were purchased from a commercial producer (2D semiconductors). Mechanically exfoliated MoS$_2$ thin flakes
DFT simulations of the doping process

DFT calculations were conducted using Quantum ESPRESSO to simulate the charge transfer process between oxide and TMDs. The ion cores were described using projector-augmented wave functions. The kinetic energy cut-off for charge density and potential was 200 Ry.

Conclusions

In summary, we investigate the layer dependence of n- and p-doping of MoS2 by TMO. TiO2 and MoO3 are used for the n- and p-doping of MoS2, respectively. PL and electrical characterization show that thicker MoS2 samples are more easily n-doped by TiO2 while thinner MoS2 samples are more easily p-doped by MoO3. Furthermore, the doping levels of 6 nm thick TiO2 and MoO3 are extracted to be $0.78 \times 10^{12}$ cm$^{-2}$ and $0.6 \times 10^{12}$ cm$^{-2}$, respectively. Our experiments characterized the doping effects and doping levels of the two mostly used oxide dopants (TiO2 and MoO3) and shed light on optimizing the doping strategy of TMDs by oxide layers.

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References

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