# Application of a Graphene Buffer Layer For The Growth of High Quality SnS films on $\operatorname{GaAs}(100)$ Substrate 

W. Wang ${ }^{1 *}$, K.K. Leung ${ }^{1}$, W.K. Fong ${ }^{1}$, S.F. Wang ${ }^{1}$, Y. Y. Hui ${ }^{2}$, S.P. Lau ${ }^{2}$ and C. Surya ${ }^{1}$<br>1 Department of Electronic and Information Engineering and Photonics Research Centre, The Hong Kong Polytechnic University, The Hong Kong Polytechnic University, Hong Kong, P.R. China<br>2 Department of Applied Physics, The Hong Kong Polytechnic University, Hong Kong, P.R. China


#### Abstract

Tin mono-sulfide (SnS) thin films have been grown by molecular beam epitaxy (MBE) on two different substrates, GaAs (100) and soda lime glass at $400^{\circ} \mathrm{C}$. High resolution X-ray Diffraction (HXRD) and Scanning Electron Microscopy (SEM) are used to characterize the structural properties of the as grown SnS films. By introducing a graphene buffer layer between the SnS thin film and the substrate, the XRD rocking curve's full width at half maximum (FWHM) of the SnS film grown on GaAs (100) and soda lime glass decrease from $2.92^{\circ}$ to $0.37^{\circ}$ and from $6.58^{\circ}$ to $2.04^{\circ}$ respectively, indicating a significant improvement of SnS thin films.


Index Terms - scanning electron microscopy, X-ray diffraction, molecular beam epitaxy, graphene, Gallium arsenide.

## I. InTRODUCTION

Layered materials, such as graphene and many transition metal chalcogenides, have demonstrated attractive electronic and optical properties for the application of high performance nano-electronic and optoelectronic devices. [1-3]. Tin monosulfide ( SnS ) is a layered material and is commonly referred to as a Van der Waals Epitaxy (VdWE). It has an orthorhombic structure with $\mathrm{a}=3.987 \AA, \mathrm{~b}=11.200 \AA$, and $\mathrm{c}=4.334 \AA$.[4] It is a semiconductor with a direct bandgap of $\sim 1.0 \mathrm{eV}$ and an indirect band gap of $\sim 1.3 \mathrm{eV}$.[5] The unit cell of SnS crystal consists of two unit layers, which are stacked perpendicularly along the b axis. The layers are binded together by weak van der Waals force while atoms within a unit layer are bonded by strong covalent bonds. This material shows attractive properties which may find important photovoltaic applications such as: non-toxic and cheap; appropriate band gap; high absorption coefficient $\left(\alpha>10^{4} \mathrm{~cm}^{-1}\right)$ near the band edge. [4, 5] The as-deposited SnS films exhibited p-type conductivity [6] and may potentially be used as the photovoltaic absorber layer. Although the SnS material exhibits desirable photovoltaic properties, very few studies about this material focused on the growth of high quality films.

Several methods have been used for the deposition of SnS films, including chemical bath deposition[7], electrochemical deposition[8], vacuum thermal evaporation[9], and plasma
chemical vapor deposition[10] ... etc. Low light conversion efficiency could be expected for the solar cells using such poor quality SnS films. Boonsalee.[8] reported the growth of SnS films on single crystal Au substrate by electrochemical deposition technique. The average FWHM of SnS XRD rocking curve peak is around $3.16^{\circ}$. It will be quite difficult to fabricate high performance photovoltaic cells using such poor quality films. Therefore, further improvements in the film quality are needed for the development of high efficient solar cells.

Several research groups have reported SnS-related solar cells. Although the theoretical limit in the conversion efficiency of SnS-based solar cell is $\sim 25 \%$, [11] the results of previous works were much lower than the theoretical prediction (no more than $1.3 \%$ )[12]. One of the most important reasons is the inability to grow high quality SnS films on substrates with large lattice mismatch. In this paper we demonstrate a novel growth technique that leads to drastic relaxation of requirement for lattice match between the film and the substrate by inserting a graphene buffer layer between the SnS film and the substrate. Both SnS and graphene are layered materials, therefore, high quality graphene might be a good substrate for the SnS layer.

## II. EXPERIMENTAL DETAILS

In our work, the SnS thin films were grown in an MBE system, in which $99.99 \%$ purity SnS compound was used as the evaporation source. The deposition rate of the SnS films was controlled at $\sim 1 \mu \mathrm{~m}$ per hour by adjusting the SnS k-cell temperature and the typical thickness of the as-deposited SnS films is $1 \mu \mathrm{~m}$.

From the point of view of lattice match, GaAs (100) substrate might be a good candidate for the growth of high quality SnS film. The lattice mismatch between SnS and GaAs (100) is $-0.26 \%$ and $8.42 \%$ with respect to the "a" and "c" axis of SnS . Thus both GaAs (100) and soda lime glass were investigated as substrate and the growth temperature was kept
at $400^{\circ} \mathrm{C}$ to suppress the formation of other tin sulfide phases such as $\mathrm{SnS}_{2}, \mathrm{Sn}_{2} \mathrm{~S}_{3}$, and $\mathrm{Sn}_{3} \mathrm{~S}_{4} \ldots$ etc. In order to improve the crystallinity of the SnS films, a study on the growth of SnS on layered material on graphene buffer layer, was performed. This is accomplished by layer transfer of high quality, large area bi-layer graphene buffer layers onto the $\operatorname{GaAs}(100)$ and soda lime glass substrates which were deposited on Cu foil by Chemical Vapor Deposition (CVD).

## III. Results AND Analysis

According to the XRD results, when the substrate temperatures are kept at $400^{\circ} \mathrm{C}$, the crystallinity of the SnS demonstrates a strong dependency on the substrates. Fig. 1 shows two theta-omega XRD spectra of SnS films grown on four different substrates: GaAs (100), soda lime glass, graphene/GaAs (100) and graphene/glass as represented by solid lines of different colors. It is observed that SnS films deposited on GaAs exhibit stronger SnS diffraction peaks than that grown on glass and exhibit only two SnS diffraction peaks corresponding to (040) and (080) plane at $2 \theta=31.89^{\circ}$ and $66.55^{\circ}$ respectively, indicating preferential [010] orientation of the film. The results are the same with SnS films grown on graphene buffer layer. As shown in Fig.1, SnS deposited on graphene/GaAs (100) show much stronger diffraction peaks than that deposited on graphene/glass substrate and only two SnS diffraction peaks corresponding to (040) and (080) plane are observed.


Fig. 1. Two theta-omega XRD spectra of SnS grown on four kinds of substrates: GaAs (100), soda lime glass, graphene/ GaAs (100) and grapheme/glass represented in different colours.

For SnS films grown on graphene/glass, the intensity of SnS diffraction peaks are comparable with those directly grown on $\operatorname{GaAs}(100)$. The strong reflection peaks of SnS on
graphene/GaAs indicate bi-layer graphene covered $\mathrm{GaAs}(100)$ is an appropriate substrate for the growth of high quality SnS thin film.

Base on the HXRD and SEM results, a significant improvement on SnS film quality and crystallinity is observed for films deposited on graphene $/ \operatorname{GaAs}(100)$. Figure 2 shows the rocking curves of the (040) reflection of SnS deposited at $400^{\circ} \mathrm{C}$ on different substrates. The FWHMs of X-ray rocking curves of (040) reflection of SnS on GaAs, soda lime glass, graphene/GaAs, and graphene/glass are $2.92^{\circ}, \sim 6.58^{\circ}, 0.37^{\circ}$, and $2.04^{\circ}$ respectively. It can be concluded that significant improvement in the crystallinity is achieved by introducing a double layer graphene between SnS and substrate. To the best of our knowledge, $0.37^{\circ}$ is the best rocking curve FWHM value ever reported in the literature. It is worth noting that graphene covered glass substrates are not as good as graphene covered $\mathrm{GaAs}(100)$ substrate for the growth of high quality SnS films. This result could be attributed to the difference of the surface roughness between soda lime glass and $\operatorname{GaAs}(100)$.


Fig. 2. X-ray rocking curves of $\mathrm{SnS}(040)$ on GaAs, glass, graphene/GaAs, and graphene/glass with FWHMs of $2.96^{\circ}, \sim 6.58^{\circ}$, $0.37^{\circ}$, and $2.10^{\circ}$ respectively.

SEM pictures of SnS deposited on different substrates, soda lime glass, $\operatorname{GaAs}(100)$, and graphene $/ \operatorname{GaAs}(100)$, with growth temperature at $400^{\circ} \mathrm{C}$ are shown in Figure 3. It is observed that flake like structures exist on the surface of the SnS films deposited on glass, while SnS films grown on $\operatorname{GaAs}(100)$ exhibit smooth surface, the grain size of those films are very small. For SnS films deposited on graphene/ $\mathrm{GaAs}(100)$, both smooth surface and larger grains are observed.


Fig. 3. SEM top view of SnS deposited at $400^{\circ} \mathrm{C}$ on a)glass, b) $\operatorname{GaAs}(100)$, and c)graphene $/ \operatorname{GaAs}(100)$

## IV. CONCLUSIONS

In conclusion, the effect of graphene buffer layer on the growth of high quality SnS films by MBE was investigated. Significant improvement on both the grain size and rocking curve FWHM of the SnS films were observed for films grown on graphene/ $\mathrm{GaAs}(100)$ at typical substrate temperature $\left(400^{\circ} \mathrm{C}\right)$. This indicates significant improvement in the crystallinity of the as-grown films. The XRD results are consistent with the experimental data on the SEM pictures of the films in which SnS film deposited on graphene/ GaAs (100) substrates demonstrate significant improvement in the morphology of the films. To further improve SnS film quality, the deposition and transfer process of graphene buffer layer should be optimized.

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## References

[9] N. K. Reddy, K. Ramesh, R. Ganesan, K. T. R. Reddy, K. R. Gunasekhar, and E. S. R. Gopal, "Synthesis and characterisation of co-evaporated tin sulphide thin films," Applied Physics a-Materials Science \& Processing, vol. 83, pp. 133-138, Apr 2006.
[10] A. Ortiz, J. C. Alonso, M. Garcia, and J. Toriz, "Tin sulphide films deposited by plasma-enhanced chemical vapour deposition," Semiconductor Science and Technology, vol. 11, pp. 243-247, Feb 1996.
[11] J. P. Singh and R. K. Bedi, "Electrical-Properties of FlashEvaporated Tin Selenide Films," Thin Solid Films, vol. 199, pp. 9-12, Apr 11991.
A. Ayari, E. Cobas, O. Ogundadegbe, and M. S. Fuhrer, "Realization and electrical characterization of ultrathin crystals of layered transition-metal dichalcogenides," Journal of Applied Physics, vol. 101, Jan 12007.
B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti, and A. Kis, "Single-layer $\operatorname{MoS}(2)$ transistors," Nature Nanotechnology, vol. 6, pp. 147-150, Mar 2011.
V. Podzorov, M. E. Gershenson, C. Kloc, R. Zeis, and E. Bucher, "High-mobility field-effect transistors based on transition metal dichalcogenides," Applied Physics Letters, vol. 84, pp. 3301-3303, Apr 262004.
A] A. Tanusevski and D. Poelman, "Optical and photoconductive properties of SnS thin films prepared by electron beam evaporation," Solar Energy Materials and Solar Cells, vol. 80, pp. 297-303, Nov 2003.
S. S. Hegde, A. G. Kunjomana, K. A. Chandrasekharan, K. Ramesh, and M. Prashantha, "Optical and electrical properties of SnS semiconductor crystals grown by physical vapor deposition technique," Physica BCondensed Matter, vol. 406, pp. 1143-1148, Mar 12011.
B. Ghosh, M. Das, P. Banerjee, and S. Das, "Fabrication of the $\mathrm{SnS} / \mathrm{ZnO}$ heterojunction for PV applications using electrodeposited ZnO films," Semiconductor Science and Technology, vol. 24, Feb 2009.
P. P. Hankare, A. V. Jadhav, P. A. Chate, K. C. Rathod, P. A. Chavan, and S. A. Ingole, "Synthesis and characterization of tin sulphide thin films grown by chemical bath deposition technique," Journal of Alloys and Compounds, vol. 463, pp. 581-584, Sep 82008.
[8] S. Boonsalee, R. V. Gudavarthy, E. W. Bohannan, and J. A. Switzer, "Epitaxial electrodeposition of tin(II) sulfide nanodisks on single-crystal $\operatorname{Au}(100)$," Chemistry of Materials, vol. 20, pp. 5737-5742, Sep 232008.
K. Hartman, J. L. Johnson, M. I. Bertoni, D. Recht, M. J. Aziz, M. A. Scarpulla, and T. Buonassisi, "SnS thin-films by RF sputtering at room temperature," Thin Solid Films, vol. 519, pp. 7421-7424, Aug 312011.

