

Structural characterization and electron-energy-loss spectroscopic study of pulsed laser deposited LiNbO₃ films on *a*-sapphire

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(Received 3 May 2004; accepted 22 August 2004)

Highly *c*- and *a*-oriented LiNbO₃ films were deposited on *a*-sapphire substrates by pulsed laser deposition. The film microstructure and crystal orientation were studied by transmission electron microscopy, and the mechanism forming the different film orientations was interpreted in terms of the adatom energy and oxygen pressures. The electron-energy-loss functions derived from the electron-energy-loss spectra exhibit characteristic energy-loss peaks at about 7.0, 11.0, and 14.0 eV. These peaks correspond to electron transitions from the maximum density of the states in the valence band of O 2*p* to the split conduction band. The dielectric functions of the LiNbO₃ films were also derived from the energy-loss function. © 2004 American Institute of Physics. [DOI: 10.1063/1.1806993]

I. INTRODUCTION

LiNbO₃ is a well-known ferroelectric oxide displaying prominent piezoelectric, pyroelectric, electro-optic, photoelastic, and nonlinear optical effects.¹ LiNbO₃ has been widely applied for the fabrication of optical waveguide, optical switch, optical modulator, and surface acoustic wave (SAW) devices, and it exhibits important technical value and profound developing prospect in the fields of acoustics, optical communications, integrated optics, and nonlinear optics.^{2,3} The deposition of LiNbO₃ films and the orientation control of the films on sapphire substrates have been studied by many researchers,^{4,5} and it has been shown that for different applications, different orientations of LiNbO₃ epitaxial layers may be needed. For example, SAW devices require [0001]-oriented (*c*-oriented) LiNbO₃ epitaxy to fully utilize the *d*₃₃ piezoelectric coefficient,^{6,7} and the electro-optical modulator devices need the film to be [11 $\bar{2}$ 0] oriented (*a*-oriented) so that the in-plane piezoelectric property can be utilized to modulate the refractive index.⁸ Lee and co-workers have systematically studied the epitaxial deposition of LiNbO₃ on *c*- and *a*-plane sapphire substrates.^{8,9} Even though the deposition of LiNbO₃ on sapphire by pulsed laser deposition (PLD) has been widely studied, surprisingly, the study of the interfacial detail structure and, thus, the understanding of the growth mechanisms are still limited. In this paper, a transmission electron microscopy (TEM) was carried out for a detailed study of the interfacial structure be-

tween LiNbO₃ films deposited at different oxygen pressures on sapphire substrates, and the orientation control mechanism will be discussed.

For optical device applications, the electronic structure and dielectric function of LiNbO₃ films are of fundamental and practical interest. Previously reported electron-energy-loss functions measured by an electron-energy-loss spectrum (EELS) and x-ray photoelectron spectroscopy analysis were mainly on single crystals,^{10–12} but the energy-loss function of LiNbO₃ thin films containing crystal defects and oxygen deficiencies has not been well studied. In this paper, we also report the energy-loss function of the LiNbO₃ films on sapphire substrates by means of an EELS-equipped in-field-emission TEM. The EELS in-field-emission TEM possesses an excellent spatial resolution, and the dielectric function derived through the Kramers-Kronig transformation reflects the dielectric property from the visible to the soft x-ray region, whereas the ellipsometry measurement only provides the dielectric function from the visible to the ultraviolet range with a finer energy resolution.

II. EXPERIMENTS

The LiNbO₃ films were deposited on 10×10×0.5 mm³ *a*-plane sapphire (Al₂O₃) substrates by PLD with a KrF excimer laser ($\lambda=248$ nm) at a 250 mJ and 5 Hz repetition rate. A LiNbO₃ single-crystal target was used in the deposition. The LiNbO₃ films were deposited with different oxygen pressures at 600 °C, i.e., 500 mtorr and 100 mtorr, respectively. After deposition, the films were *in situ* annealed at 600 °C for 30 min in an oxygen pressure of 11 torr. The same pressure was maintained until the sample was cooled down to room temperature. The growth rate of

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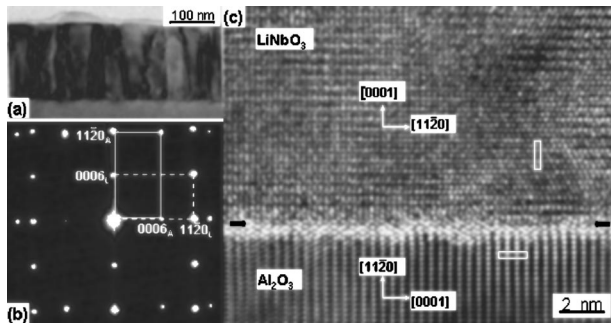


FIG. 1. (a) Cross-section TEM image of a *c*-LiNbO₃ on an *a*-sapphire substrate deposited at a 500 mtorr oxygen pressure; (b) the composite diffraction patterns showing the orientation relationship between the LiNbO₃ film and sapphire substrate, where “L” and “A” represent LiNbO₃ and Al₂O₃, respectively; and (c) the corresponding HREM image of the interface, where the unit cells of LiNbO₃ and Al₂O₃ are outlined by the white lines and the interface is indicated by the arrow.

the films is about 0.5 nm/s and the film thicknesses are about 200 nm. The cross-section TEM study of LiNbO₃/Al₂O₃ was carried out on a field-emission electron microscopy. The energy-loss function and dielectric function of the LiNbO₃ films were obtained from the EELS using the Kramers-Kronig transformation.¹³

III. RESULTS AND DISCUSSION

A. Interfacial structure

Figure 1(a) shows a cross-section TEM image of *c*-LiNbO₃ on an *a*-sapphire substrate deposited at a 500 mtorr oxygen pressure and the diffraction pattern, as shown in Fig. 1(b), illustrating the orientation relationship between the LiNbO₃ film and the sapphire substrate, i.e., (0001)_{LiNbO₃}∥(11 $\bar{2}$ 0)_{Al₂O₃} and [01 $\bar{1}$ 0]_{LiNbO₃}∥[01 $\bar{1}$ 0]_{Al₂O₃}. The film exhibits a columnar and epitaxial growth feature with a grain size of about 100 nm. Figure 1(c) is the corresponding high-resolution electron microscopy (HREM) image showing the interfacial structure. It can be seen that the interface is free from interfacial reaction, and the interface is sharp even with the presence of a large number of misfit dislocations at the interface due to the lattice mismatch.

Figure 2(a) is a TEM image showing an *a*-LiNbO₃ on an *a*-sapphire substrate grown at 100 mtorr. The corresponding orientation relationship between the film and substrate is illustrated by the composite-electron diffraction pattern, as shown in Fig. 2(b), i.e., (11 $\bar{2}$ 0)_{LiNbO₃}∥(11 $\bar{2}$ 0)_{Al₂O₃} and [01 $\bar{1}$ 0]_{LiNbO₃}∥[01 $\bar{1}$ 0]_{Al₂O₃}. Again, from the corresponding HREM image, as shown in Fig. 2(c), it can be seen that the interface is sharp and free from interfacial reaction. The misfit dislocations exhibit a periodic feature with a mean distance of 5.6 nm, and the induced strains at the interface can be clearly seen.

It is worthy to note that some areas in the LiNbO₃ film deposited at a 100 mtorr oxygen pressure show coexistence of the *a*- and *c*-orientations, as shown in Fig. 3. The percentage of the *c*-oriented grains is about 10%. This suggests that the interfacial mismatch energies between the *a*- and *c*-oriented LiNbO₃ with an *a*-sapphire is small and may be tolerated by other factors such as oxygen pressure, laser

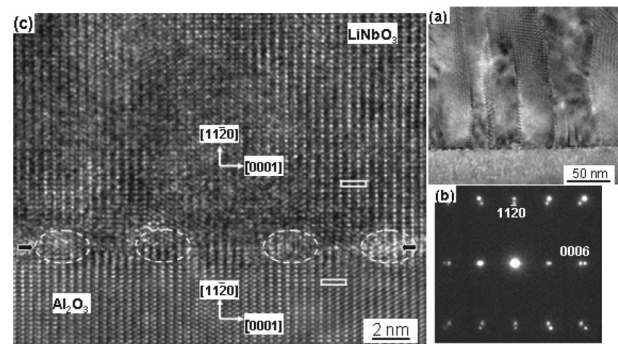


FIG. 2. (a) Cross-section TEM image of an *a*-LiNbO₃ on an *a*-sapphire substrate deposited at a 100 mtorr oxygen pressure; (b) the composite diffraction patterns showing the orientation relationship between the LiNbO₃ film and sapphire substrate, where the labeled (0006) and (11 $\bar{2}$ 0) diffractions are for LiNbO₃ and Al₂O₃; and (c) the corresponding HREM image of the interface, where the unit cells of LiNbO₃ and Al₂O₃ are outlined by the white lines and the interface is indicated by the arrow. The strained areas due to the interfacial dislocations are indicated by the white dot circles.

beam energy, etc. The film orientation does not change as the film thickness increases, but we noticed that some small *c*-oriented grains were finally “eaten-up” by the neighboring *a*-oriented domains. This suggests that the *a*-oriented grains may grow faster than the *c*-oriented grains at a low oxygen pressure. Consider also the work reported earlier by Lee and co-workers,^{8,9} where a much lower oxygen pressure resulted only in the *a*-oriented growth of the LiNbO₃ film; we believe that it should be a continuous change of the film growth orientation as the oxygen pressure changes.

The results turned out that oxygen pressure is the key parameter that affects the growth orientation of LiNbO₃ on sapphire substrates. A higher oxygen pressure favors the *c*-axis LiNbO₃ growth and a lower oxygen pressure favors the *a*-axis growth. Lee and co-workers have ascribed this to the different atomic arrangements of the oxygen layers in the *c*- and *a*-oriented LiNbO₃.⁹ However, our results showed that

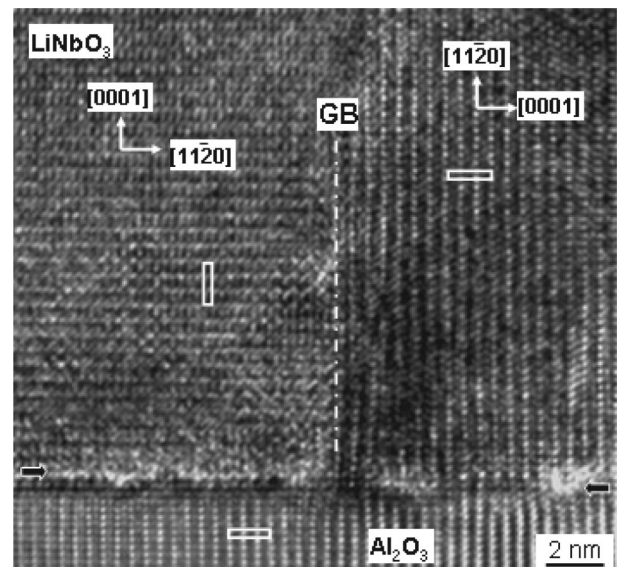


FIG. 3. HREM image of the LiNbO₃ film on an *a*-sapphire substrate deposited at a 100 mtorr oxygen pressure showing coexistence of the *a*- and *c*-oriented LiNbO₃. The white dot line indicates the grain boundary between the *a*- and *c*-oriented LiNbO₃.

this rule was broken when the laser energy changed. For example, when the laser fluence was reduced from 5 to 3 J/cm², the same low oxygen pressure (100 mtorr) still resulted in the *c*-LiNbO₃ growth. We believe that the adatom energy is an important factor that controls the crystal orientation of the LiNbO₃ films during the nucleation stage, and the oxygen pressure also affects the growth orientation through changing the adatom energy. In fact, both a higher oxygen pressure and a lower laser power result in a lower adatom energy; therefore, the adatom energy should play the most important role in controlling the film growth orientation.

The lattice similarity between LiNbO₃ and sapphire determines that the interfacial energy of an *a*-LiNbO₃/*a*-sapphire is lower than that of a *c*-LiNbO₃/*a*-sapphire. However, another competition factor is the nucleation surface free energy. The Gibbs free energy has been thought to play an important role in the nucleation stage, which directly affects the growth orientation of the films. A model based on the change of the Gibbs free energy in forming a single-crystal nucleus has been proposed to interpret the *c*-oriented LiNbO₃ growth on an amorphous SiO₂.⁶ We believe that when the oxygen pressure is higher, its rapid quenching effect results in a lower adatom energy, and the lower adatom energy makes it difficult to overcome the energy barrier to allocate to the correct lattice sites and, thus, the surface free energy becomes a dominant factor in controlling the growth orientation of the LiNbO₃ nucleus. Since LiNbO₃ has the hexagonal structure, the (0001) lattice plane has the largest atomic density and the lowest surface free energy thus grows preferentially along the *c* axis on the amorphous substrates. Therefore, if the difference of the lattice mismatches between the *c*- and *a*-oriented LiNbO₃ with an *a*-sapphire is small, in fact it is true,⁹ the *c*-oriented LiNbO₃ will be a preferred orientation when the adatom energy is relatively low. This interpretation matches the reported results by Lee and co-workers,⁹ where the higher and lower oxygen pressures resulted in the *c*- and *a*-oriented LiNbO₃ films grown on *c*-sapphire substrates, respectively. Based on the adatom energy understanding, we believe that the growth orientation of the LiNbO₃ films on a *c*-sapphire substrate may also be controlled by altering the laser fluence. This is desirable for further studies.

B. Electron-energy-loss function and electronic structure

The EELS spectra of the *a*- and *c*-oriented LiNbO₃ films were obtained using the diffraction mode at zero-momentum transfer. The energy-loss functions of these two differently oriented LiNbO₃ films after the multiple-scattering corrections were obtained from the raw EELS spectra, and it turns out that the general shapes of the spectra are the same, and there is no significant difference to be noticed. Figure 4 represents a typical energy-loss function obtained from the LiNbO₃ films. Based on the first-principle calculation by Kohiki *et al.*¹⁰ and the experimental results reported by Kasper and Hüfner,¹¹ the electron-energy losses of about 7.0, 14.0, and the small shoulder at 11.0 eV can be attributed to the

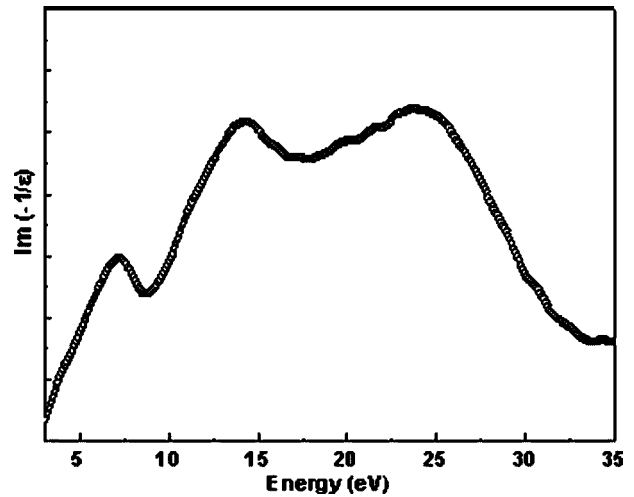


FIG. 4. (a) Represented energy-loss function of the LiNbO₃ films after the multiple scattering corrections from the raw energy-loss spectra.

electron transitions from the maximum density of the states in the valence band to the split conduction band, where the valence band is predominantly of O 2*p* character and the conduction band contains Nb *d* and *s-p* states.

Kasper and Hüfner have reported that the Li or O deficiency induced a reduction of Nb⁵⁺ to Nb⁴⁺ and resulted in an energy-loss peak at 5.2 eV. The fact that such a peak was absent in our PLD-deposited LiNbO₃ films suggests that the LiNbO₃ films are well crystallized without the significant oxygen and Li deficiencies. The peak at 24 eV in the energy-loss function is attributed to a plasmon peak.

Figures 5(a) and 5(b) show the real and imaginary parts

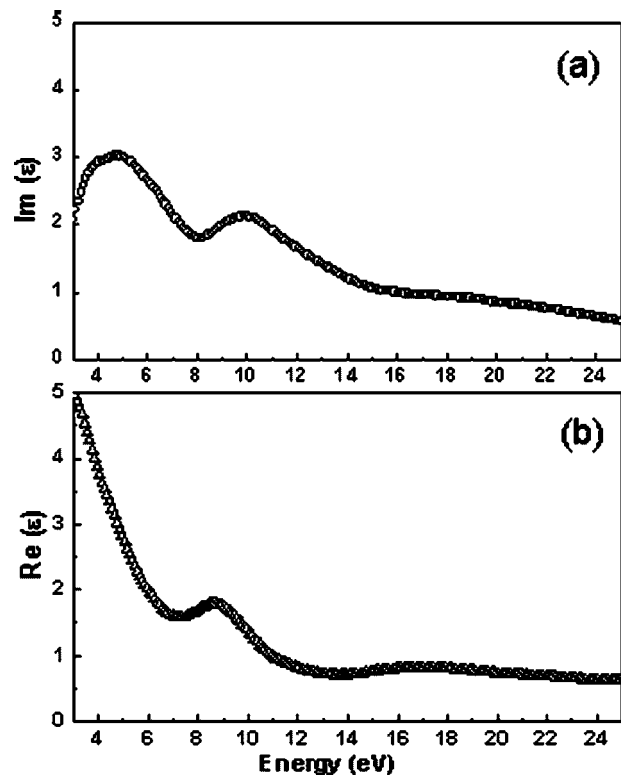


FIG. 5. Dielectric function of the LiNbO₃ films derived by using the Kramers-Kronig transformation: (a) Im(ϵ) and (b) Re(ϵ).

of the dielectric function derived from the loss function by using the Kramers-Kronig transformation. The peaks in $\text{Im}(\epsilon)$ correspond to the joint density of the states in the valence and conduction bands, and the broad peaks at 4.0–5.0 eV and 9.0–10.0 eV match fairly well with the first-principle calculation result as reported by Kohiki *et al.*¹⁰

IV. CONCLUSION

Highly *c*- and *a*-oriented LiNbO_3 films were deposited on *a*-sapphire substrates by PLD. The film microstructure and crystal orientations were studied by TEM, and the mechanism forming the different film orientations was interpreted in terms of the adatom energy and oxygen pressure. The energy-loss functions and dielectric functions derived from the EELS exhibit characteristic peaks that are similar to bulk LiNbO_3 single crystals.

ACKNOWLEDGMENTS

This work was partially supported by the Hong Kong RGC grant (Grant No. B-Q553) and also an internal grant from the Hong Kong Polytechnic University (Grant No. G-YD55).

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