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ARTICLE

The recent progress of metal-halide perovskite-based tandem solar cells

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Tandem solar cells (TSCs) are devices made of multiple junctions with complementary absorption ranges, aiming to overcome the Shockley–Queisser limit of single-junction solar cells. Currently, metal-halide hybrid perovskite solar cells have risen as an ideal candidate for various tandem cells due to their high open-circuit voltage, tunable optical bandgap, low energy loss, and low-temperature solution processibility. Here, the critical advances of perovskite-based tandem cells are presented, including perovskite-silicon, perovskite-perovskite, perovskite-copper indium gallium selenide, and perovskite-organic TSCs. Electrode engineering, interface engineering, bandgap engineering, and processing techniques are discussed in detail. The future directions and challenges of perovskite TSCs are also presented.

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Introduction

As energy resource depletion, the greenhouse effect, and environmental contamination intensify, developing clean, and low-carbon-emission energy has become an urgent task. Photovoltaics, which directly convert light energy into electricity, is a practical manner to address these issues. To date, solar cells based on silicon (Si), gallium arsenide (GaAs), copper indium gallium selenide (CIGS), and cadmium telluride have been commercialized by their excellent device performance and mature fabricating techniques. Nonetheless, their fabrication cost-power ratio, rigid form-factor, and deployment cost still restrict the competitiveness of solar cells with conventional fossil energy. The emerging metal-halide perovskite solar cells (PSCs) offer advantages of solution processability, compatibility with large-area deposition techniques, inexpensive and abundant raw materials, and desirable optoelectronic properties such as large extinction coefficient, long ambipolar diffusion lengths, excellent defect tolerance, and low exciton binding energy, rendering PSC a promising candidate for the next-generation photovoltaic technology. The metal-halide perovskite has the ABX₃ perovskite structure. A stands for monovalent cations, such as methylammonium (MA), formamidinium (FA), cesium (Cs), or their combination; B refers to divalent metal elements, such as lead (Pb), tin (Sn), copper (Cu), manganese (Mn), and germanium (Ge); and X denotes halogen elements, including chlorine(Cl), bromine (Br) and iodine(I). By far, the power conversion efficiency (PCE) of PSCs has achieved 25.5%, 2-5 thereby attracting substantial attention from both academia and industry.

Since the invention of the solar cell in 1954, intensive efforts have been invested in improving the PCE of photovoltaic cells. Early in 1961, Shockley and Queisser published the principle of detailed balance which unveiled the thermodynamic upper limit of p-n junction solar cells. They premised four commonly accepted assumptions. 1) One incident photon generates one electron-hole pair; 2) Photons with larger energy than the E_g of photoactive materials are absorbed, while those with lower energy than the E_g do not contribute to photogeneration of charge carriers; 3) Photovoltaic device is free of non-radiative recombination; 4) The incident light is neither concentrated nor anisotropic. As the Shockley–Queisser (S-Q) limit points out, a single-junction solar cell cannot exceed 33.16% efficiency under standard AM 1.5G illumination. The thermalization loss and transmission loss principally limit the efficiency of single-junction solar cells. Thermalization loss is mainly caused by the thermal relaxation of the excited electron to the band edge when the absorbed photon energy is more than E_g ; transmission loss is resulted from that the photons with energy smaller than E_g are not absorbed.

Tandem architecture is recognized as a promising method to overcome the S-Q limit of the single-junction cells. In a tandem device, the incident light impinging on the tandem device penetrates top subcells and bottom subcells sequentially. Typically, the upper subcell employs a wide- E_g absorber that harvests photons with higher energy, while lower-energy photons are absorbed by the bottom subcell using narrow- E_g materials. Consequently, the theoretical PCE limit for a two-subcell tandem device is up to \sim 42% and can be further improved to \sim 68% for the infinite-subcell device.

Metal-halide perovskites are an ideal component of tandem solar cells (TSCs) for the following reasons. Firstly, metal halide perovskites are sound light absorbers for the top and bottom subcells because their optical bandgaps can be adjusted in the range of 1.17-3.10 eV through composition engineering. Secondly, the high structural tolerance of perovskite reduce the stringent requirements for lattice matching between subcells. The electrical tolerance of perovskite effectively reduce the deep defect centers coming from surface and bulk defects caused by lattice mismatch between subcells. PSCs are compatible with solution and vacuum fabrication methods, and n-i-p and p-i-n device structures. The abovementioned merits of PSCs faciliate the construction of all-perovskite TSCs, and the collaboration of perovskites and established crystalline Si (c-Si), CIGS solar cells, or organic solar cells to construct hybrid TSCs with respectable PCEs.

Herein, we first introduce the basic concepts of perovskite solar cells, four-terminal (4-T) and two-terminal (2-T) tandem structure. Then we present the recent advancement of perovskite TSCs, including perovskite-c-Si, perovskite-perovskite, perovskite-CIGS, and perovskite-organic tandems, where the aspects of electrode engineering, interface engineering, bandgap engineering, composition engineering, as well as processing techniques, are discussed in detail. The architecture and performance of the perovskite-based TSCs are summarized in Table 1. Lastly, we discuss the future directions and challenges of perovskite TSCs.

1. Configuration of TSCs

Generally, TSCs can be mainly designed as a mechanically stacked 4-T cell or a monolithic 2-T device (Figure 1). In a 4-T configuration, top and bottom subcells are connected through an electrical circuit, enabling the independent fabrication of top and bottom subcells. Circuit management is simplified, and the total efficiency is simply the sum PCE of two subcells. In a 2-T monolithic tandem architecture, the two subcells are directly joined through an interconnecting layer (ICL) instead of an electrical circuit to form a tandem device. The ICL layer between two subcells, such as the tunneling junction and recombination layer, strongly influences the charge carrier transport and current loss. Compared with the 4-T structure, the monolithic 2-T architecture has fewer fabrication steps, transparent conducting electrodes, and encapsulation materials, resulting in much lower manufacturing cost and less parasitic absorption. However, monolithic tandems require strict process compatibility: (i) Both subcells should be optimized to produce the same/similar current density at maximum power point (MPP), as the tandem current density is limited by the lower current density of two subcells. (ii) The later formed cell should be processed at low temperatures and via orthogonal solvent for temperature and solvent-sensitive former formed cell.

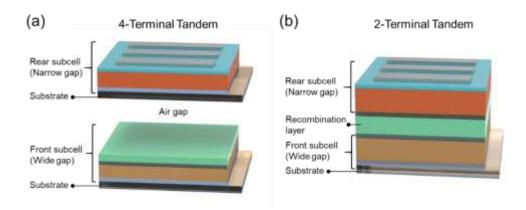


Figure 1. Structures of (a)mechanical stacked four-terminal cell and (b) monolithic two-terminal cell

2. Perovskite-Silicon TSCs

c-Si solar cells have occupied over 90% share of the global photovoltaic market with highly mature manufacturing lines and supply chains. Currently, the single-junction c-Si cell has reached a high PCE of 26.7%, 7.8 closely approaching the theoretical limit of 29.4%. Furthermore, c-Si presents a low E_g of 1.12 eV, which renders Si solar cells highly suitable as the narrow bandgap bottom cell in TSCs. The perovskite-c-Si tandems, where perovskites can effectively improve efficiency with little additional cost, are presumably the most direct pathway to commercializing perovskite solar cells. It is a shortcut to improving commercial c-Si efficiencies, associated with significant financial benefit considering the global solar PV Panels market size has been steadily growing and surpassed \$130 billion in 2020, and every 1% higher module efficiency means \sim \$5 billion profit.

2.1 Perovskite-Silicon 4-terminal TSCs

Semi-transparent perovskite solar cells (ST-PSCs) are the prerequisite and challenge of perovskite-silicon 4-T TSCs. The transparent electrode of ST-PSCs should have both high transparency to ensure the photon harvest of the bottom cell and high lateral conductivity to minimize current loss. Besides, transparent electrodes should have processing compatibility with the layers beneath. More specifically, the fabrication of top transparent electrodes should use low-temperature deposition and mild precursors to prevent or reduce the damage to the underlayer interlayer and perovskite layer. Besides, they should have appropriate energy levels. Transparent electrodes of optoelectronic devices can be made of thin metal films, conductive oxides, metal nanowires, dielectric/metal/dielectric (DMD) photonic structures, conductive polymers, carbon nanotubes, graphene, and so on.

In 2014, the Ballif group pioneered constructing the perovskite- c-Si TSC with a MoO_x /indium tin oxide (ITO) transparent electrode. The 30-nm MoO_x layer functioned as a hole-collecting buffer layer to prevent the harm of ion bombardment during ITO sputtering, and the 100-nm ITO layer provided lateral conductivity and broad-band transparency. The MAPbl₃ top cell demonstrated a transmittance >55% in the near-infrared (NIR) region, allowing the bottom cell with a short-circuit current density (J_{SC}) of 13.7 mA cm⁻². The 4-T tandem device yielded a sum efficiency of 13.4% from 6.2%-efficiency MAPbl₃ top cell and 7.2%-efficiency filtered silicon heterojunction (SHJ) bottom cell, exceeding the MAPbl₃ solar cell with MoO_x/Ag electrode (11.6%). McGehee and coworkers utilized aluminum-doped (2 mol%) zinc oxide (AZO) nanoparticles to solution-deposited an electron-selective buffer layer atop PCBM. The aluminum dopant eliminates the extraction barrier

arising from the misaligned work functions of ZnO and ITO. A sum PCE of 18.0% is realized for 4-T tandem, based on the 12.3%-efficiency ST-PSC and 5.7%-efficiency filtered silicon solar cells, exceeding the 17.0%-efficiency mono-Si reference.¹⁰ Ballif group sputtered amorphous indium zinc oxide (IZO) layer to fabricate transparent electrode, which absorbed <3% in the 400-1200 nm wavelength range and exhibited a sheet resistance (R_{sh}) of 35 Ω /sq. Over 9% efficiency was achieved in ST-PSCs with IZO sputtered directly on Spiro-OMeTAD, which was increased to 10.3% with the insertion of MoO_x. The corresponding 4-T tandem devices with and without MoO_x buffer layer both achieved similar PCEs of 18.18% and 18.19%, revealing the trade-off between reduced sputter damage and increased parasitic absorption of MoO_x.11 The reported PCE values of tandems have so far lain below their potential owning to inefficient photon harvesting. Sargent and coworkers used MoO₃/IZO/MgF₂ transparent electrode to realize a 19.4% PCE for 1.63 eV semi-transparent perovskite cells with an average nearinfrared transmittance of 85%. With a boosted solvent extraction technique (BSE), they increased the optical path length of triple-cation perovskite (Cs_{0.05}FA_{0.81}MA_{0.14}Pbl_{2.55}Br_{0.45}) films by preserving homogeneous and smooth morphology at a large-thickness perovskite layer. ¹² In the BSE approach, the spin-coating speed is raised precisely one second before anti-solvent dripping, facilitating the lateral hydrodynamic flow and evaporation of the mixed solvents, and thus the rapid nucleation of the intermediate phase. As figure 2a shows, X-ray diffraction patterns of intermediate phases in the unannealed perovskite films prepared by the BSE method exhibited narrower peak width, which also illustrates the excess solvent in the unannealed control film, as the local structural order therein can be interrupted by DMSO/DMF over incorporation (Figure 2a). The enhanced surface morphology can improve photovoltaic performance, and the urea is further added as a lewis base to decrease trap density and increase electron diffusion length to 2.3 µm. The couple of ST-PSCs and a 9.2%-efficiency filtered silicon bottom cell yields a PCE of 28.6% (Figure 2b).

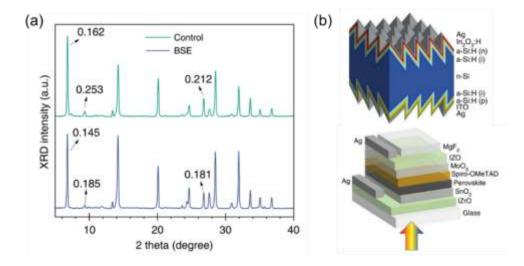


Figure 2. (a) X-ray diffraction pattern of unannealed perovskite films fabricated by non-BSE and BSE methods. The numbers refer to the full width at half maximum. (b) Device architectures of mechanically stacked TSCs, consisting of perovskite top cell and SHJ bottom cell. ¹²

An ultra-thin metal film is an alternative approach for transparent electrodes by its good transmittance and conductivity. In 2014, Snaith and coworkers reported that the thermal-evaporation of 10-nm Au on spiro-OMeTAD and the creation of microporous perovskite layer by partial dewetting of solution-cast films enabled semi-transparent perovskite solar cells with ~8% efficiency and ~30% transmittance in the visible region.¹³ However, Au and Ag fail to form a uniform, continuous, and ultra-thin film due to the Volmer–Weber growth mechanism, which reduces the transmittance and increases the corresponding electrodes' sheet resistance. To improve Au's film-forming properties at a small thickness, Cheng and coworkers designed a sandwiched MoO_x-Au-MoO_x stack as a transparent electrode. ¹⁴ MoO_x has good hole injection capability and provides a good nucleation surface to form a thin yet uniform Au film. The top MoO_x layer works as an anti-reflection layer. In their work, the semi-transparent MAPbl₃ device shows a PCE of 5.3% with 31% average visible transmittance (AVT), and a PCE of 13.6% with 7% AVT. Parallelly, Yang and coworkers designed MoO_x (10 nm)-Au (1 nm)- Ag(10 nm)-MoO_x (40 nm) multi-layer transparent top electrode for PSCs.15 The gold seeded silver film exhibited superior conductivity and optical transparency to pristine Ag or Au, owning to Au's higher surface energy. The semi-transparent PSC illuminated from the DMD top electrode afforded 11.5% efficiency and exhibits almost no V_{OC} and FF loss compared with the opaque device. Likewise, Huang group designed Cu (1 nm)/Au (7 nm) metal electrode for semi-transparent MAPbl₃ PSCs.¹⁶ At a thin film, Au atoms are more strongly coupled to each other, leading to the formation of Au islands via Volmer-Weber growth during thermal evaporation. As Cu exhibits a larger surface energy of 1790 mJ m⁻² than Au (1506 mJ m⁻²), Au atoms preferentially attach to Cu, resulting in continuous ultrathin Au films via Frank-van der Merve growth (Figure 3a). Semi-transparent PSCs under front illumination exhibit a PCE of 16.5%, a high FF of 74.1%, and a J_{SC} of 20.6 mA cm⁻², illustrating that the Cu/Au transparent electrodes induce little loss during the collection and lateral transport of photogenerated charges (Figure 3b). Under rear illumination, the device exhibits a PCE of 12.1% with a smaller J_{SC} of 15.2 mA cm⁻², ascribed to the less transparency of the Cu/Au/BCP layer than ITO in the visible spectrum. As figure 3c shows, opaque PSCs and semi-transparent devices under front illumination have comparable EQE values at short wavelengths (300-550 nm), while semi-transparent devices exhibit lower EQE at longer wavelengths (600-800 nm). When illuminated through the semi-transparent PSC, a SHJ solar cell operates with a 6.5% efficiency, leading to a sum PCE of 23.0% for the 4-T tandem device.

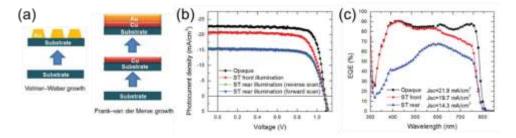


Figure 3. (a) Scheme of Volmer-Weber growth and Frank-van der Merve growth. (b) J – V curves of opaque and semitransparent (ST) PSCs. (c) EQE curves of opaque and ST-PSCs under front and rear illumination. (a)

Metal nanowire offers the advantage of easy deposition in a room-temperature solvent-free process, in addition to low sheet resistance and high optical transmission. McGehee and coworkers afforded a 12.7%-efficiency semi-transparent MAPbl₃ PSC with a AgNW electrode, which was first sprayed on a flexible polyethylene terephthalate (PET), and then mechanically transferred to the spiro-OMeTAD layer. A sum PCE of 17.0% was realized when the semi-transparent PSC was mechanically placed on multi-crystalline Si cells (unfiltered: 11.4%, filtered: 4.3%). Yang, Liu, and coworkers designed sandwiched MoO₃/gold nanowire (AuNW) mesh/MoO₃ transparent electrode. The large surface tension of MoO₃ effectively improved Au's wettability, resulting in Frank-van der Merwe growth to produce an ultrathin AuNW mesh layer. The top MoO₃ layer reduces the reflections at the gold layer to increase light transmission. A sum efficiency of 27.0% is realized when they mechanically stacked the 18.3% efficiency semi-transparent MAPbl₃ PSC on a SHJ rear cell (unfiltered: 23.3%, filtered: 8.7%).

Graphene-based electrodes have a high optical transmission of 97.4% with a sheet resistance of $100 \Omega/\text{sq.}^{19}$ Yan group pioneered laminating chemical vapor deposition (CVD) stacked graphene as a transparent electrode. They further coated a PEDOT: PSS layer atop graphene, which significantly improved the electrode's conductivity and provided adhesivity for lamination (Figure 4a and 4b). The sheet resistance (R_s) was decreased to about 260 Ω/sq by the doping effect of PEDOT:PSS (Figure 4c). As figure 4d shows, the two-layer graphene doped with PEDOT has an R_s of 140 Ω/sq and a transmittance > 90% in the visible region. When the layers of graphene further increase, the sheet resistance is further decreased and the transmittance was below 90%. The ST devices with double-layer graphene electrodes show PCEs of 12.37% and 12.03% from the FTO and graphene sides, respectively (Figure 4d).

Lang and coworkers¹⁹ showed the successful implementation of a transparent large-area, high-quality, single-layer graphene contact upon spiro-OMeTAD in PSCs. The van der Waals forces ensured the adhesion and intimate contact of graphene on spiro-OMeTAD and enabled a 6.2%-efficiency semi-transparent cells in the absence of an additional auxiliary layer. The semi-transparent PSC and the SHJ solar cell (unfiltered: 18.5%, filtered:7.0%) are assembled into a 4-T tandem device, giving a higher PCE of 13.2% than the single-junction PSCs with typical Au contacts (10.2%).

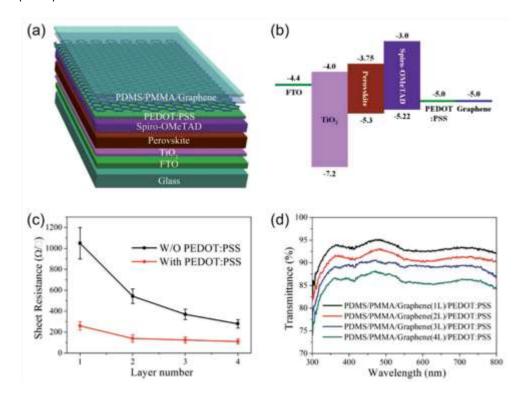


Figure 4. (a) Device architecture of a semi-transparent PSC. (b) Energy level distribution of layers of semi-transparent PSCs. (c) The sheet resistance of 1-4 layers of stacked graphene films with and without PEDOT:PSS. (d) UV–vis transmittance spectra of PEDOT: PSS doped transparent graphene electrodes.²⁰

In the perovskite-silicon 4-T TSCs, the top PSCs are commonly stacked onto the bottom c-Si devices with an air gap. Due to the significant refractive indices mismatch between the air and functional layers, a part of NIR light would be reflected to the PSCs. In 2018, Jaysankar and coworkers proposed a refractive index matching layer (IML) to replace the unfavorable air gap.²¹ The introduction of the optical spacer increased the transmission of NIR light through the top PSC and boosted the photocurrent of the c-Si bottom solar cell.

Although mainstream 4T TSCs adopt the "planar structure" aforementioned, in principle they can be constructed using other configurations. For example, an optical splitter is a multi-layer beam splitter with high reflection in the shorter-wavelength range and high transmission in the longer-wavelength range. By splitting the incident solar spectrum and distributing them to each subcell, solar energy can be utilized more effectively. Via optical splitter, subcells in 4-T structure can be set spatially separated yet optically coupled (Figure 5). The optical splitter is fabricated by sputtering multi-layer dielectric oxides with high (n~1.9-2.2) and low (n~1.5) refractive indices on both sides of the glass. The cut-off wavelength can be controlled by altering the number and thickness of the stacking layers. The Yamamoto group designed a series of optical splitters with cut-off wavelengths of 550, 600, and 640 nm, respectively. They achieved 28%-efficiency 4-T devices based on 7.5%-efficiency MAPbl₃ PSCs and SHJ solar cell (unfiltered: 25.2%, filtered: 20.5%) with 550-nm cut-off wavelength.²²

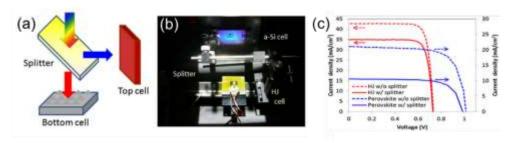


Figure 5. (a) Scheme of the optical splitting system, (b) measurement setup of an optical splitting system, and (c) *J-V* curves of SHJ and perovskite solar cells with and without a splitter.²²

2.2 Perovskite-Silicon 2-terminal TSCs

The 2-T configuration, where subcells are constructed as a single integrated unit, allows simple electrical integration and bypasses additional transparent electrodes. Nonetheless, the monolithic approach requires a tunneling/recombination interlayer that can effectively facilitate the flow of photogenerated carriers between two subcells. A reasonable tunneling/recombination layer should introduce small electrical and optical losses and low processing costs.

Si is an indirect- E_g semiconductor. It is common to design tunneling junctions consisting of two heavily doped p++ and n++ regions for charge recombination. McGehee and coworkers demonstrated the first 2-T monolithic perovskite-silicon TSC with an inter-band tunneling junction. To form n++/p++ tunneling junction, they deposited heavily doped n++ hydrogenated amorphous silicon (a-Si: H) on the p++ emitter via plasma-enhanced chemical vapor deposition (PECVD) and subsequently thermally annealed the a-Si: H layer to activate the dopants. Besides, a 2-3 nm-thick intrinsic a-Si layer was inserted between the P++ emitter and n++ amorphous Si layer to mitigate the dopant interdiffusion and prevent the degrading of the tunneling-junction conductivity. They obtained a PCE of 13.7% and a $V_{\rm OC}$ of 1.58V for the 1cm², 2-T MAPbl₃/Si tandem device. The slow current-dynamics and corresponding hysteresis of the tandem device resembled the sluggish dynamics of the MAPbl₃ PSC, which indicated that the perovskite subcell limited the $J_{\rm SC}$ of the TSCs. The n++/p++ tunneling junction demonstrated high shunt resistances and potentially facilitated the fabrication of large-area tandems. However, this Si-based tunneling junction requires an additional PECVD step and introduces parasitic optical absorption.

The transparent conductive oxide (TCO) recombination layer is an alternative for constructing 2-T devices, like indium-doped tin oxide (ITO) and indium-doped zinc oxide (IZO). Before 2015, the reported c-Si/perovskite tandems usually required high-temperature (\sim 500 °C) processing to sinter mesoporous TiO₂. However, this processing is not compatible with the best-performing SHJ technology, as a-Si: H requires the temperature to remain below 200 °C. To tackle this problem, Grätzel and coworkers sputtered the ITO recombination layer and used the atomic layer deposition (ALD) SnO₂ layer as the electron selective contact for PSCs to construct a 2-T perovskite-SHJ tandem architecture.²⁴ The 2-T devices demonstrate an efficiency of 19.9%, with a J_{SC} of 14 mA cm⁻², a V_{OC} of 1.78V, and FF of 79.5%. In their work, the SHJ structure's potential is not completely fulfilled due to the absence of a pyramid texture on the Si wafer, as the solution processing of PSCs requires a smooth substrate underneath. Ballif group chose IZO as the recombination layer in the 2-T monolithic perovskite-silicon tandem cell, which is inspired by the successful application of the IZO recombination layer in tandem organic solar cells.²⁵ They presented low-temperature-processing MAPbl₃/SHJ monolithic TSCs with efficiencies of 19.2% with an aperture area of 1.22 cm² and 21.2% with 0.17 cm² aperture area, respectively. According to transfer matrix simulations, the PSCs can achieve a maximum current density of top cell with a ~40-nm thick IZO recombination layer and a 150-nm thick spiro-OMeTAD layer, and the SHJ bottom cell needs to be textured to harvest more infrared photons.

Impediments to more advances in TCO interlayer include substantial parasitic absorption due to free-carrier absorption at long wavelengths and the proneness of carriers to be funneled to shunt paths along the recombination layer, which limits large-area fabrication.

In general, an interface (tunneling or recombination) layer makes electrical contact between the top and the bottom subcells but incurs higher fabrication costs and parasitic absorption. The "interlayer-free" concept offers the advantages of more straightforward processing and reduced optical losses. For the first time, Baillie group demonstrated a monolithic planar- MAPbl₃ perovskite-homo-junction silicon tandem without the use of an additional interface layer.²⁶ The mechanism is that the solution-processed SnO_2 layer serves as an electron transport layer for the perovskite top cell and as a recombination layer between the perovskite top cell and the Si bottom cell. The champion 2-T device with a 4 cm² area achieves a steady-state efficiency of 20.5% when the PDMS foil is applied. Besides, upscaling to large-area devices becomes straightforward as the monolithic tandem device relies on the p++ emitter instead of SnO_2 for lateral conduction. Thereby, a steady-state PCE of 17.1% can be achieved in a large area of 16 cm². They further optimize the device structure in some aspects. First, a new metal grid design is used for demonstrating 16 cm² large-area devices. Second, the planar homojunction silicon subcell is replaced with a textured homojunction silicon bottom cell. Third, a mixed perovskite absorber (FAPbl₃)_{0.83}(MAPbBr₃)_{0.17} is used instead of MAPbl₃, which can improve the V_{OC} of the tandem device and reduce hysteresis, owing to the larger E_g and better quality of mixed perovskite. Correspondingly, they demonstrated a 16 cm²-area perovskite- homojunction silicon tandem device with a high PCE of 21.9% under reverse scan and a high FF of 78%.²⁷

Parasitic UV absorption and UV-induced degradation by spiro-OMeTAD in PSCs hinder device performance and stability when the perovskite cell is a top cell of perovskite-Si TSCs. Hence, Baillie group applied textured polydimethylsiloxane (PDMS) films that incorporate a downshifting material, (Ba, Sr)₂SiO₄: Eu²⁺ micron phosphor, in front of the monolithic 2-T perovskite- homojunction silicon tandem device (Figure 6a).²⁸ (Ba, Sr)₂SiO₄: Eu²⁺ phosphor had high luminescent efficiency with broad UV absorption and short decay time, which is suitable for a down-shifting material in photovoltaics. This film served multiple purposes: antireflective control for the top cell, light trapping in the Si cell, and absorbing UV and reemitting green light with high quantum yield. Figure 6b shows the PL spectra (green curve, λ_{ex} =370 nm) and photoluminescence excitation spectra (black curve, $\lambda_{em} = 517$ nm) of (Ba, Sr)₂SiO₄: Eu²⁺. The green emission ranges from 460 to 580 nm, and is peaked at 517 nm, corresponding to a Commission Internationale de l'Eclairage (CIE) chromaticity coordinate of (0.1891, 0.6441). The emission of Eu²⁺ ions is due to the transition from the relaxed 4f⁶5d¹ state to 4f⁷ ground state. The PL decay curve of (Ba, Sr)₂SiO₄: Eu²⁺ phosphor (λ_{ex} =405 nm), is detected at 520 nm at room temperature (Figure 6c). When applied onto a 4 cm² monolithic perovskite-silicon tandem solar cell, PCE was improved from 20.1% (control device without PDMS layer) to 22.3% (control device with pure PDMS layer) to 23.1% (experimental device with 0.5 wt% (Ba, Sr)₂SiO₄: Eu²⁺ doped PDMS). Moreover, the champion device maintained 90% of the initial performance after 648 h UV exposure at 5 mW cm⁻². Catchpole and coworkers also demonstrated a monolithic 2-T perovskite-Si tandem by straightforward integration of PSCs and Si solar cells without an additional layer (Figure 7a). This interlayer-free device is enabled by highly conductive contact between ALD n-type TiO₂ and P-Si.²⁹ A possible reason is the substantial density of localized midgap states at the interface between p-Si and TiO₂. These interfacial states can facilitate band to band tunneling at reverse bias and work as generation-recombination centers at all bias voltages. Electrons move into/out of the defect states via local capture/emission and tunneling in this circumstance. Figure 7b shows the photovoltaic J-V performance exhibited by the highest-performing poly-Si tandem cell and the corresponding photovoltaic metrics. The corresponding monolithic 2-T perovskite (Cs_{0.05}Rb_{0.05}FA_{0.765}MA_{0.135}Pbl_{2.55}Br_{0.45})/SHJ tandems demonstrate an efficiency of 24.1% at reverse scan, with a $V_{\rm OC}$ of 1.763V, a $J_{\rm SC}$ of 17.8 mA cm⁻², and a FF of 78.1%.

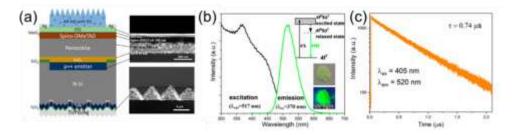


Figure 6. (a) Schematic illustration of perovskite-silicon homojunction solar cells with a down-shifting AR PDMS layer. Cross-sectional SEM image of the perovskite top cell and the rear textured silicon bottom cell. (b) Excitation and emission spectra of (Ba, Sr)₂SiO₄:Eu²⁺ phosphor (inset: schematic diagram of the Eu²⁺ 4f–5d transition and photos of phosphor powder under ambient light and UV light (365 nm). (c) Decay curve of the Eu²⁺ emission in the (Ba, Sr)₂SiO₄: Eu²⁺ phosphor.²⁸

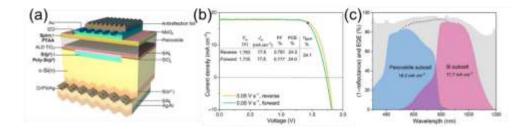


Figure 7. (a) Scheme of the interlayer-free monolithic perovskite-c-Si tandem solar cell. (b) J-V curve of the tandem device with both reverse and forward scanning at 0.05 V s⁻¹ based on heterojunction poly-Si subcell. (c) Absorbance (1–reflectance) of the tandem device (gray), EQE of the perovskite top cell (blue), and the c-Si bottom subcell (pink).²⁹

2.2.1 Light management of perovskite-Silicon TSCs

Light management (LM) is critical for achieving high-performance monolithic 2-T perovskite-Silicon TSCs. Si presents a larger optical refractive index than interlayer and perovskite layers, which results in substantial parasitic reflection losses at the internal junction between two subcells. An ideal tandem solar cell usually utilizes surface texture to remove reflection losses by improving light in-coupling and enhance the NIR response at the band edge.

It is possible to position an electrically inert but optically textured layer with an appropriate refractive index directly atop the top contact of TSC to improve the light harvest. Albrecht and coworkers employed textured LM foils on the front side of a TSC. The LM foil significantly improved the PCE of the 2-T perovskite-SHJ tandems from 23.4% to 25.5%. 30 Besides, through simulations, they predicted a slightly narrower optimal E_g of the perovskite top sub-cell is needed to match a textured Si device compared with a flat device, and the strong interdependency between E_g and the texture position in the monolithic 2-T device. Their simulations showed that a 32.5% efficiency is realistically achievable with a 1000-nm 1.66 eV- E_g perovskite atop a both-side silicon device. Snaith and coworkers incorporated an optical interlayer consisting of nanocrystalline silicon oxide (nc-SiO₂:H) to mitigate the optical drawback of a flat silicon substrate and increase bottom-cell current density. The 110-nm nc-SiO2:H interlayer with a refractive index of 2.6 resulted in 1.4 mA cm⁻² current density gain in the silicon bottom cell. The anisotropic nature of the nc-SiO₂:H interlayer with high transversal and low lateral conductivity supports the further reduction of top-cell shunting via shunting quenching. The champion 1 cm² monolithic 2-T perovskite-Silicon tandem cell exhibits a total current density of 38.7 mA cm⁻² under AM 1.5G irradiation, and a certified stabilized PCE of 25.2%.

Surface texture can also be realized by using c-Si wafers with um-sized pyramids. Such texture improves light in-coupling by minimizing reflections off the front surface while scattering the light within the solar cell to enhance light trapping. However, most monolithic 2-T perovskite-silicon tandems use Si wafers featuring a smooth front side and textured backside to meet the requirement of PSCs wet processing. Difficulties of depositing perovskite via solution techniques atop micrometer-sized Si pyramids include uncovered Si pyramids, large shunt paths, and inefficient charge collection. Besides, conventional surface-passivation techniques are incompatible with the rough perovskite surfaces. This configuration, nonetheless, limits light-trapping functions and requires additional anti-reflection foils, the effectiveness of which may be compromised by encapsulation. Moreover, perovskite-Silicon tandems with a polished wafer surface have a cost issue, as industrial Si wafers are typically etched and textured on both sides, and chemical-mechanical polishing is cost-prohibitive.

Sargent and coworkers constructed high-performance 2-T perovskite tandems based on fully textured SHJ bottom cells, which is sketched in Figure 8a.³² They increased the depletion width by three times at the bases of silicon pyramids to facilitate charge collection in micrometerthick perovskites. The J_{SC} of the PSC increases with the thickness of the perovskite layer. The experimental J_{SC} value (certified ~19.3 mA cm 2) of the ~1.1-µm-thick perovskite is thereby only within 10% deviation of the theoretical limit (Figure 8b). As figure 8c shows, spectrally weighted reflectances are calculated from AM 1.5G spectrum at 350 to 900 nm. They observed that the benefit to reflection began to saturate when the pyramid size is reduced to 2 μ m, suggesting the solution-processing techniques of perovskite can be integrated with textured silicon. They also increase carrier diffusion length and suppressed phase segregation by anchoring 1-butanethiol as a self-limiting passivator (SLP) on the perovskite surfaces. The SLP method increases FF from 72% to 77% (Figure 8D). EQE characterization demonstrates the advantage of switching from a polished front side to a double-side textured architecture (Figure 8G and 8H). The introduction of a front texture reduced reflection losses and exhibited efficient light trapping. The device demonstrated a certified PCE of 25.7% and negligible performance loss after the thermal stability test (400 hours, 85°C) and after MPP tracking (400 hours, 40°C) (Figure 8E and 8F). Huang and coworkers use a blade-coating technique to deposit PSCs' layers on the double-side-textured silicon cells with sub-micrometer pyramids (Figure 9). 33 These pyramids with shrunken size are also rough enough to scatter light within the silicon relative to large-size pyramids. A N_2 knife helps remove solvent vapor, facilitating a quick transition of solution into solid during the blading process (Figure 9a). When PTAA is spin-coated on a silicon bottom cell, the resulting layer is non-uniform with a thickness of 45 nm in the valleys but a thickness of 8 nm on the pyramids (Figure 9b). Even with N₂-assisted drying, blade coating PTAA at room temperature results in a non-uniform layer (Figure 9c). When the substrate temperature was elevated to 70 °C, the solvent's accelerated evaporation makes PTAA more uniform on the textured silicon surface (Figure 9d). The high-throughput blade-coating process can deposit both a conformal hole transport layer and a planarizing perovskite layer that thoroughly covers the textured silicon with elevated substrate temperature. This work elongates the path length of longwavelength light in the silicon and reduced reflection loss to afford 2-T tandems with a J_{SC} >19 mA cm⁻² and an efficiency of 26.2%.

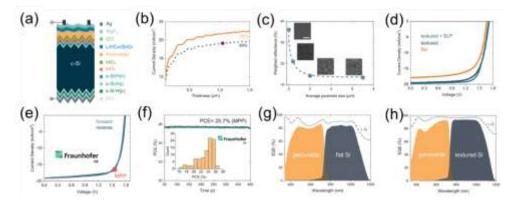


Figure 8. (a) Structures of solution-processed perovskite—textured silicon tandem. (b) Calculated J_{SC} values of the perovskite cells as a function of perovskite layer thickness. The red dot represents the EQE-integrated J_{SC} value in the textured tandem top cell. (c) Weighted reflectance as a function of c-Si pyramid sizes. The SEM images of different texturing sizes are shown in the insets (scale bar, 2 mm). (d) J-V characteristics of flat, textured, and SLP-treated textured tandems. (e) J-V curves of certified SLP-treated textured tandems. (f) MPP tracking of certified SLP-treated textured tandems and PCE distributions of 88 individual tandem devices. (g) EQE of the flat devices. (h) EQE of the textured devices.³²

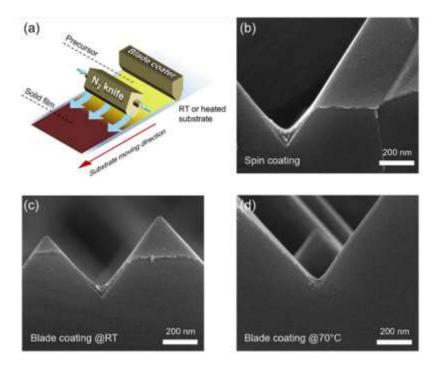


Figure 9. (a) Scheme of N_2 -assisted blade coating. (b–d) SEM images of PTAA on textured silicon bottom cells deposited by (b) spin coating, (c) N_2 -assisted blade coating with the substrate at room temperature, and (d) N_2 -assisted blade coating with the substrate at 70 °C. The three visible layers are the silicon wafer, ITO front electrode of the silicon cell, and PTAA hole-transport layer.³³

Note that the bifacial c-Si solar cell and module technology have received considerable attention, which collects light from both the front and rear sides. The bifacial power gain can be up to 50%,³⁴ and is highly dependent on the albedo of the surroundings and installation geometry.³⁵ The concept of bifacial cell and module can be naturally transferred to perovskite-c-Si tandems for both 2-T and 4-T configurations.³⁶⁻³⁸ In 2021, Sargent, Wolf, and co-workers systematically studied the performance of bifacial 2-T perovskite-c-Si tandems. Because of the additional incident light from the rear side, the optimized band gap of perovskite is reduced to 1.59-1.62 eV. Thus the bromide content is reduced compared with that in the wide band gap perovskite, which would relax the stability issue coming from the halide segregation.³⁹

2.2.2 Bandgap Engineering of perovskite-Silicon TSCs

The origin of designing tandems is to reduce the thermalization loss. As predicted by the detailed-balanced model, the ideal $E_{\rm g}$ for the tandem configuration is ~1.67 to 1.75 eV for the top cell and 1.12 eV for the bottom cell.⁴⁰ The 1.12-eV $E_{\rm g}$ of silicon renders itself a nearly ideal bottom light absorber. Bandgaps from 1.6 to 3.06 eV can be achieved through I/Br and Br/Cl alloying in Perovskites. However, wide-bandgap alloys in the optimum range for TSCs suffer from short diffusion lengths and photo-induced phase segregation, which lead to poor device performance and large $V_{\rm OC}$ loss. For instance, the replacement of I with Br by >20%, which is necessary to enlarge the $E_{\rm g}$ to 1.7 eV, leads to the formation of I-rich and Br-rich structures under illumination (a process termed the Hoke effect⁴¹) and limits the Fermi level splitting to that of the lower-bandgap I-rich domains, which effectively act as trap sites. Besides, Br-containing high $E_{\rm g}$ perovskites exhibited shorter carrier lifetime than pure I-based perovskite. Compositional tuning of the A and X site in perovskite structures and surface passivation can improve the device performance of high- $V_{\rm OC}$ PSCs.

Cation substitution with Cs or DMA can widen the E_g of perovskite and prevent the large fraction of Br. For instance, Zhou group reported high-efficiency monolithic 2-T perovskite-Si tandems by tuning the E_g and the optical density of perovskite absorbers. ⁴² Four different perovskite absorbers, (FA_{0.48}MA_{0.37}Cs_{0.15}Pbl_{2.23}Br_{0.77}, FA_{0.57}MA_{0.43}Pbl_{2.04}Br_{0.96}, FA_{0.5}MA_{0.38}Cs_{0.12}Pbl_{2.04}Br_{0.96}, and FA_{0.51}MA_{0.38}Cs_{0.11}Pbl_{1.85}Br_{1.15}) were adopted to adjust the E_g of perovskite absorbers from 1.65 to 1.72 eV. The FA_{0.5}MA_{0.38}Cs_{0.12}Pbl_{2.04}Br_{0.96} film presented a much lower trap density of 1.13×10¹⁶ cm⁻³ than FA_{0.57}MA_{0.43}Pbl_{2.04}Br_{0.96} (5.7 ×10¹⁶ cm⁻³), presumably because the small amount of Cs acts as nucleation sites, effectively promotes crystal growth and suppresses defect formation. To maximize the potential of (FA, MA, Cs) PbX₃, they optimized the residual Pbl₂ content in the film, the thickness of the perovskite layer, and the SnO₂ electron transport layer. Semi-transparent PSCs with 410-nm FA_{0.5}MA_{0.38}Cs_{0.12}Pbl_{2.04}Br_{0.96} layer and transparent MoO_x/ITO transparent electrode exhibited an efficiency of 13.09%. The 2-T -TSCs based on FA_{0.5}MA_{0.38}Cs_{0.12}Pbl_{2.04}Br_{0.96} and SHJ delivered a PCE of 22.22% and retained over 85% of their original efficiency after the long-term stability test (500 h).

MAPbCl₃ has a wider E_g of 2.88 eV and higher stability than its Br and I counterpart. However, incorporating Cl into perovskite precursor solutions typically cannot increase E_g but increase the perovskite domain size, because the formation and outgassing of MACl alter the crystal nucleation and growth dynamics. Hence, the Cl concentrations in perovskite film were below the detection limit of energy-dispersive X-ray spectroscopy (EDX) and X-ray photoelectron spectroscopy (XPS), even when a large portion of Cl was incorporated in the precursor solution. McGehee and coworkers alloyed 2-5 mol% MAPbCl₃ into the lattice of FA_{0.75}Cs_{0.25}Pb($I_{0.85}Br_{0.15}$)₃, $I_{0.15}^{43}$ and thereby increased the solid solubility of Cl and the PVSK I_g from 1.63 to > 1.67 eV (Figure 10a). In their work, the Cl to Pb ratio in solid films is increased with the increasing fraction of Cl in the precursor solution, with grain domain size barely enlarged. Specifically, $I_{0.85}Br_{0.15}Pb(I_{0.85}Br_{0.15})_3$; $I_{0.85}Br_{0.15}$ has a lattice constant of ~6.24 Å, like that of $I_{0.85}Br_{0.15}Pb(I_{0.85}Br_{0.15})_3$ (Figure 10b and 10f). It is because that the lattice constant of the host perovskite was reduced by Cs and Br, making Cl closer to the ideal size for the X-site in the lattice. Consequently, Cl is located in the perovskite lattice and increased I_g , instead of being sacrificed in a MACl volatile phase. The Cl element in the triple-halide films was uniformly distributed through the entire film thickness (Figure 10c). The bandgap decreases at high Cl content due to double-phase segregation into high Br/Cl phase and high I/Br phase (Figure 10d, 10e and 10f). The extension from double-halide (I, Br) to triple-halide (I, Cl, Br) significantly increases photocarrier lifetime and charge-carrier mobility and effectively suppresses light-induced phase segregation. This strategy reduced the $I_{0.05}$ hope and $I_{0.05}$ hope an

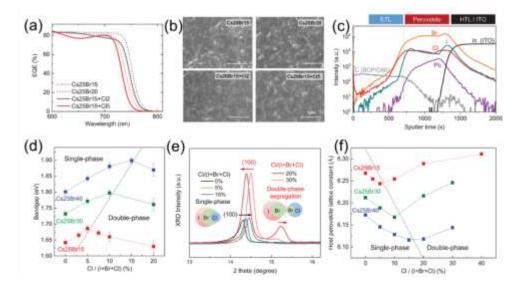


Figure 10. (a) EQE curves of perovskite with and without Cl. Cs25Br15 and Cs25Br20 refer to Cs_{0.25}FA_{0.75}Pb(I_{0.85}Br_{0.15})₃ and Cs_{0.25}FA_{0.75}Pb(I_{0.8}Br_{0.2})₃, respectively. Cs25Br15+Cl2 and Cs25Br15+Cl5 refer to the triple-halide perovskites with 2 or 5 mol % MAPbCl₃ in the Cs25Br15, respectively. (b) Top-view SEM images of double-halide and triple-halide perovskites. (c) The uniform distribution of halides throughout the entire film thickness of the triple-halide perovskite (Cs25Br15+Cl5). (d) Evolution curves of band gap for triple-halide perovskite films with increased ratio of Cl/(I+Br+Cl), based on Cs25Br15 (red), Cs25Br30 (green), and Cs25Br40 (blue), respectively. The

dashed line delineates the single-phase and double-phase zones corresponding to bandgap raising and reducing, respectively. (e) Shifting and splitting of XRD (100) peaks shows the transition from single-phase triple-halide alloy to double-phase segregation in Cs25Br40 with increased Cl content. (f) Curves of host perovskite lattice constant with increasing ratios of Cl/(I+Br+Cl). The dashed line delineates the single-phase and double-phase zones corresponding to band gap raising and reducing, respectively.⁴³

The two-dimensional (2D) phase was used to stabilize wide- E_g perovskite, where sheets of [PbX₆]-² octahedra are separated by long-chain or aromatic molecules that act as a passivation agent. For example, n-butylammonium iodide (n-BAI) and phenethylammonium iodide (PEAI) are common 2D additives. Snaith and coworkers introduced n-butylammonium (BA) cations into (FA_{0.83}Cs_{0.17}Pb(I_yBr_{1-y})₃) perovskite. The 2D perovskite platelets were formed and interspersed among highly orientated 3D perovskite grains, suppressing non-radiative charge recombination. The device with an optimal concentration of BA cations exhibited an average stabilized PCE of 17.5 \pm 1.3% with a 1.61 eV- E_g perovskite and 15.8 \pm 0.8% with a 1.72eV- E_g perovskite. Unencapsulated devices retained 80% of their post burn-in efficiency after 1000 h in air, while encapsulated device retained 80% after nearly 4000 h.⁴⁴ Shin and coworkers incorporated binary additives, Pb(SCN)₂ and PEA(I_{0.25}SCN_{0.75}), into 1.68 eV- E_g (FA_{0.65}MA_{0.2}Cs_{0.15})Pb(I_{0.8}Br_{0.2})₃ perovskite and thereby delivered a 26.7%-efficiency, stable monolithic 2-T perovskite-Si TSCs.⁴⁵ The 2-T devices retained >80% of their initial PCE (20.7%) after 1000 hours of continuous illumination (underneath a sulfur plasma lamp at ~0.8 sun). Anion engineering of phenethylammonium-based 2D additives was critical for controlling the structural and electrical properties of the 2D passivation layers bearing a PbI₂ framework. Pb(SCN)₂ accelerates the growth of 3D perovskite grain. The 2D additive PEA(I_{0.25}SCN_{0.75}) enables the formation of 2D-phase perovskite both on the surface and at grain boundaries, compared with the sample with Pb(SCN)₂ forming only surface 2D phases.

Al-Ashouri and coworkers recently demonstrated a monolithic perovskite-Silicon tandem with a certified record-high PCE of 29.15%, surpassing the best silicon single-junction cell (26.7%). The 1.68 eV-bandgap perovskite absorber remained phase-stable under illumination through the synergy of fast hole extraction and minimized nonradiative recombination at the hole-selective interface. These features were afforded by a self-assembled, methyl-substituted carbazole monolayer (Me-4PACz) as the hole-selective layer in the PSC. The accelerated hole extraction went along with a low ideal factor of 1.26, a high single-junction FF of 84%, and a high tandem $V_{\rm OC}$ of 1.92V. In air, the tandem device without encapsulation retained 95% of its initial efficiency after 300 hours of operation.

Because of the series connection feature and the spectral mismatch of solar simulators, it is difficult to measure the photovoltaic properties of 2-T tandem devices. Kim and coworkers designed three-terminal (3-T), monolithic perovskite-Si tandem cells, where the ITO recombination layer was also used as an additional electrode for characterization⁴⁶ (Figure 11a and 11b). The top subcell was a semi-transparent perovskite solar cell with MAPbl₃ as the high-energy absorber, deposited on a flat silicon bottom cell with a textured backside. With the 3-T architecture, the J-V characteristics of each subcell and the tandem cell can be measured independently; the EQE of each subcell can be accurately measured without any light/potential bias or filtration of the probing light. Through optical engineering and bandgap tuning, the best monolithic 3-T perovskite (MAPb($I_{0.95}Br_{0.05}$)₃) /SHJ tandem cell exhibits a high PCE of 23.5% and retains 97% of the initial efficiency stored in the dark after 100 days.

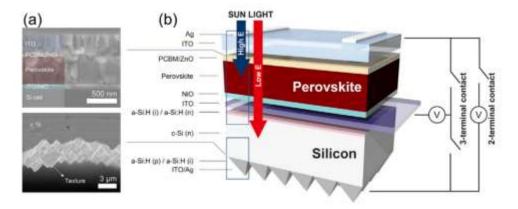


Figure 11. (a) Cross-sectional FESEM images of perovskite top cell and silicon bottom cell, and (b) schematic illustrations of the monolithic perovskite-Si tandem cell and circuits of 2,3-T architecture. 46

The monolithic 2-T perovskite-perovskite-silicon triple-junction solar cells can be a low-cost alternative of III-V semiconductors based triple-junction solar cells, as the latter device requires complicated and costly manufacturing processes. Snaith and coworkers predicted that triple-junctions based on 1.95 and 1.44 eV perovskite subcells and a 1.1 eV silicon bottom cell have an efficiency potential of 38.8%.⁴⁷ Ballif group realized a proof-of-concept 2-T perovskite-perovskite-silicon triple-junction device by monolithically growing two perovskite cells on double-side textured silicon wafers.⁴⁸ They employed an evaporation-spin coating sequential process to deposit the perovskite absorbers, enabled the conformal growth of all the layer stacks directly on the micrometer-sized pyramids of Si bottom cell. The triple-junction devices exhibited a $V_{\rm OC}$ of ~2.7 V, an equivalent cumulative current density of 38.8 mA cm⁻², and MPP tracked efficiency of 13.2%.

3. Perovskite-Perovskite TSCs

Perovskite-perovskite tandems offer a route to high efficiency and reserve the strengths of low-cost, low-temperature, and low environmental impact fabrication and the possibility of lightweight and flexible form of both subcells. Perovskite has the compositional versatility and thus highly adjustable E_g , which offers the potential of realizing high-performance perovskite-perovskite tandems. For instance, the E_g of Pb-based perovskite can be elevated from 1.58 (pure I) to 2.2 eV (pure Br) by substituting I with Br in MAPbI₃. Zhou group pioneered in constructing bottom-up solution-processed 2-T all-perovskite TSCs, where two MAPbI₃ subcells are bridged by spiro-OMeTAD/PEDOT: PSS/PEI/PCBM: PEI charge recombination layer. Despite the low PCE of 7%, this device yields a high $V_{\rm OC}$ of 1.89 V close to the sum of the two MAPbI₃ subcells.⁴⁹ Im group fabricated 2-T MAPbBr₃-MAPbI₃ TSCs by laminating single-junction MAPbBr₃ and MAPbI₃ planar solar cells.⁵⁰ In their work, P3HT and poly[bis(4-phenyl)(2,4,6-trimethylphenyl)amine] (PTAA) hole conductors with Li-TFSI and t-BP additives improved conductivity and reduced charge-transport barriers due to the assistance of Li/Li⁺ redox shuttle. Despite the high $V_{\rm OC}$ of 2.25V, the monolithic 2-T MAPbBr₃-MAPbI₃ TSCs gave a low $J_{\rm SC}$ of 8.3 mA cm⁻² and a frustrating sum PCE of 10.8%, suggesting the combination of only wide- E_g Pb perovskite absorbers fundamentally limits device performance.

The mixed Sn-Pb perovskite presents the bandgap bowing effect, which means the mixed perovskite can present a lower bandgap than its Sn-free and Pb-free counterparts. Alloying Pb and Sn in metal halide perovskites can narrow E_g s down to 1.17 eV for MASn_{1-x}Pb_xl₃(X=0.25 and 0.5). In principle, the low- E_g Sn-Pb perovskite could be an ideal candidate as the bottom subcell, opening a pathway for high-performance perovskite-perovskite tandems. Recent optical and device modeling shows that all-perovskite tandems based on 1.22 and 1.80 eV PSCs have a potential to reach 33.4% efficiency with EQE over 90% and V_{OC} of 2.17 eV, rendering this technology potentially competitive with perovskite-silicon tandems.⁴⁷ To achieve high-performance all-perovskite tandems, low- E_g Sn-Pb perovskite should have a large thickness to harvest sufficient photons, and long carrier diffusion lengths to ensure efficient charge transport and extraction. However, narrow E_g Sn or Sn-Pb PSCs lag far behind wide E_g PSCs currently, which is the major obstacle hindering the advances of all-perovskite tandems. Recently, scientists have carried out composition engineering, additive engineering, solvent engineering, and interface engineering to improve Sn-Pb perovskite based single-junction and tandem devices.

In the early time, devices based on Sn-containing perovskites performed much worse than their pure-Pb counterparts, even though Sn-containing perovskite absorbs a larger fraction of the solar spectrum. Early in 2012, Shum group invented the first Sn (CsSnl₃) PSCs, which delivered a PCE of 0.9%. In 2014, Kanatzidis group reported that PSCs based on $1.30 \text{ eV-}E_g \text{ CH}_3\text{NH}_3\text{Snl}_3$ and $1.17 \text{ eV-}E_g \text{ MASn}_{0.75}\text{Pb}_{0.25}\text{I}_3$ afforded efficiencies of 5.44% and 7.37%, respectively. The disappointing performance of Sn or Sn-Pb based PSCs is associated with both material and device issues. First, Sn and Sn-Pb perovskites have smaller absorption cross-sections than Pb perovskites, which requires a thick-film Sn or Sn-Pb perovskite absorber to achieve high EQE and in turn long carrier diffusion lengths for efficient charge extraction. Second, the Sn-substituted perovskite typically crystallizes more quickly than Pb-based perovskite, resulting in inferior-quality films. Last but not least, the facile oxidation of Sn²⁺ to Sn⁴⁺ reduces the device performance and stability.

To improve the film quality of Sn-based perovskite, Huang and coworkers reported a simple Pb-assisted two-step spin-coating method. In the control group without Pbl₂, a sequential deposition of pure Snl₂ and MAI afforded MASnl₃ film with cracks and discontinuous grains.⁵³ By contrast, the perovskite film is uniform and crack-free with the addition of Pbl₂. Therefore, the MASn_{0.5}Pb_{0.5}l₃ PSCs achieved a PCE of 13.6% with high reproducibility. Jen and coworkers further improved the performance of low- E_g Sn-Pb PSCs via integrated composition, process, and interface engineering.⁵⁴ The solvent-washing methodology with a DMSO-based co-solvent system effectively retards the growth of Sn-Pb perovskites and helps form homogeneous and smooth films by forming the Snl₂·DMSO intermediate phase. MAPb_{0.75}Sn_{0.25}l₃ with ~1.35 eV E_g was chosen after the investigation of surface morphology and optoelectronic properties of MAPb_{1-x} Sn_xl₃ perovskites with different x values. As the partial substitution of MA with FA stabilizes the Sn-Pb perovskite, the final composition they chose is 1.33 eV E_g FA_{0.5}MA_{0.5}Pb_{0.75}Sn_{0.25}l₃. The corresponding devices achieved a PCE of 14.19% and retained >94% of its initial PCE after 30 days in inert atmosphere. The coupling of the Sn-Pb perovskite bottom subcell (unfiltered: 14.19%, filtered: 5.56%) and a semi-transparent MAPbl₃ (13.52%) top subcell endowed the 4-T device with an encouraging PCE of 19.08%.

Via cation and device engineering, Snaith and coworkers demonstrated a 14.8% efficiency for Sn-Pb PSCs. They implemented a technique of precursor-phase antisolvent immersion to deposit uniform FASn_{0.5}Pb_{0.5}I₃ film.⁵⁵ The mixed solvents of DMSO and DMF with low-vapor pressure retard crystallization via the formation of precursor complexes, followed by an immediate antisolvent bath and gentle heating to crystallize the perovskite. The partial substitution of FA with Cs effectively improved PCE from 10.9% to 14.1%. The combination of the 1.2-eV Sn-Pb perovskite (FA_{0.75}Cs_{0.25}Sn_{0.5}Pb_{0.5}I₃) cell and 1.8-eV E_g perovskite cell (FA_{0.83}Cs_{0.17}Pbl_{0.15}Br_{0.15}) demonstrates a 17.0%-PCE monolithic 2-T TSCs. They also fabricated 20.3% efficiency small-area all-perovskite 4-T tandems based on a semitransparent 1.6-eV E_g FA_{0.83}Cs_{0.17}Pb(I_{0.83}Br_{0.17})₃ front cell (filtered: 15.8%) and the FA_{0.75}Cs_{0.25}Sn_{0.5}Pb_{0.5}I₃ bottom cell (unfiltered: 14.8%, filtered: 4.5%).

The development of mixed Sn-Pb PSCs can draw lessons from the recent progress of the Pb-free PSCs.⁵⁶ First, FASnI₃ is more stable than MASnI₃ to work as the Sn source. Second, SnF₂ additives are useful for increasing carrier lifetime (by reducing Sn vacancy and background hole density), changing the energetics of the material, improve the film morphology and stability. Third, hole transporting layers using Li or Co salts as dopants may pose damage to Sn-containing Perovskite. Accordingly, Yan and coworkers employed 1.25 eV Eg (FASnI₃)_{0.6}(MAPbI₃)_{0.4} absorber, SnF₂ additives, and PEDOT: PSS hole transporting layer to fabricate Sn-Pb inverted PSCs. Figure 12a sketches the steps of forming the precursors for one-step deposition of mixed Sn-Pb perovskite thin fims, consisting of the FASnI₃ precursor with the MAPbI₃ precursor at different molar ratios. MAPbI₃ and FASnI₃ films both contain many large grains, yet with some small ones (figure 12b and 12c). As comparison, the (FASnI₃)_{0.6}(MAPbI₃)_{0.4} film contains relatively smaller but more uniformly distributed grains (figure 12d). The device based on (FASnI₃)_{0.6}(MAPbI₃)_{0.4} exhibited a PCE of 15.08% with good reproducibility and negligible hysteresis.⁵⁷ The Sn-Pb perovskite exhibits a carrier lifetime of 0.7 ns, much longer than FASnI₃ (0.2 ns), but much shorter than MAPbl₃ (109 ns). The low-E_g PSCs show relatively low spectral response in the NIR region, which is resulted from the short carrier lifetime (<20 ns) and the relatively thin (~400 nm) Sn-Pb perovskite film. By optimizing the precursor solutions, Yan and coworkers simultaneously increased grain size, crystallinity, and carrier lifetime to 255 ns at a thickness of 620 nm for the perovskite layer. The low- E_R PSCs achieved a PCE of 17.5%, and a J_{SC} of 28.7 mA cm⁻² with EQEs of >70% in the NIR range.⁵⁸ When it was mechanically stacked with a FA_{0.3}MA_{0.7}PbI₃ (~1.58 eV) front cell, a maximum PCE of 21.2% at reverse scan was demonstrated for the 4-T all-perovskite tandem cell. Yan and coworkers further incorporated 2.5% Cl into (FASnI₃)_{0.6}(MAPbI₃)_{0.4}, which simultaneously enlarged grain, increased crystallinity, and functionalized grain boundaries. Via the photothermal deflection spectroscopy (PDS) characterization, they found the incorporation of 2.5% Cl reduced Eu from 27.9 meV (the undoped perovskite device) to 23.9meV (the experimental device), illustrating that CI reduced electronic disorder.⁵⁹ Transient photocurrent measurement illustrated that CI incorporation effectively increased minority carrier lifetime. The inverted planar Sn-Pb PSCs with ~750-nm active layer demonstrated a PCE of 18.40%, a $V_{\rm OC}$ of 0.850V, and a high $J_{\rm SC}$ of 29.31 mA cm⁻² with negligible $\text{J-V hysteresis. By combining this narrow-} \textit{E}_{g} \text{ PSC, wide-} \textit{E}_{g} \text{ FA}_{0.8} \text{Cs}_{0.2} \text{Pb} (I_{0.7} \text{Br}_{0.3})_{3} \text{ PSC, and ultra-flat and smooth Ag/MoOx/ITO ICLs, } \\ \text{PSC, wide-} \textit{E}_{g} \text{ FA}_{0.8} \text{Cs}_{0.2} \text{Pb} (I_{0.7} \text{Br}_{0.3})_{3} \text{ PSC, and ultra-flat and smooth Ag/MoOx/ITO ICLs, } \\ \text{PSC, wide-} \textit{E}_{g} \text{ FA}_{0.8} \text{Cs}_{0.2} \text{Pb} (I_{0.7} \text{Br}_{0.3})_{3} \text{ PSC, and ultra-flat and smooth Ag/MoOx/ITO ICLs, } \\ \text{PSC, wide-} \textit{E}_{g} \text{ PSC, wide-} \textit{E}_{g} \text{ PSC, wide-} \text{PSC, wide-} \text{PSC,$ the monolithic 2-T tandem device achieved PCEs of 21.0% and promising operational stability. Likewise, the incorporation of Br endowed the 1.27 eV Sn-Pb perovskite with a high PCE of 19.1% and a small $V_{\rm OC}$ loss of 0.384 V. 60

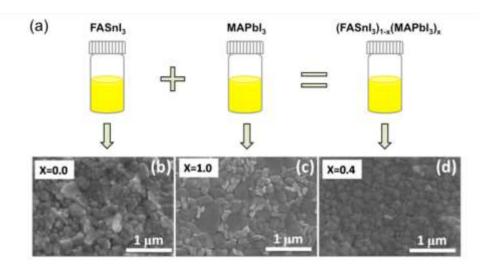


Figure 12. (a) Scheme of the formation of $(FASnI_3)_{1-x}(MAPbI_3)_x$ precursor. (b, c, d) SEM images of $(FASnI_3)_{1-x}(MAPbI_3)_x$ perovskite films (x = 0.0, 1.0, and 0.4).⁵⁷

Commonly used Sn-based narrow- E_g perovskite has shorter carrier diffusion lengths and lower absorption coefficient than Pb-based perovskite. In Sn-containing PSCs, electrons exhibited much shorter diffusion length than holes. Huang and coworkers incorporated 0.03 mol% cadmium ions (Cd^{2+}) into $FA_{0.5}MA_{0.45}Cs_{0.05}Pb_{0.5}Sn_{0.5}I_3$ perovskite to de-dope Sn-Pb perovskite and reduce the electron trap density, which effectively enhances recombination lifetime, electron mobility, and electron diffusion length. With this strategy, the optimized perovskite thickness is increased from 620 to 1000 nm to yield stabilized efficiencies of 20.2% for single-junction 1.22 eV Sn-Pb PSCs. When 1.80 eV $FA_{0.6}Cs_{0.4}Pb(I_{0.65}Br_{0.35})_3$ PSCs and the 1.22 eV Sn-Pb PSCs are combined, the best 2-T tandem devices achieve a high PCE of 23.0% under reverse scan, with a V_{OC} of 1.99 V, a J_{SC} of 15.1mA cm⁻², and FF of 77%.

Via a comproportionation reaction, Tan and coworkers utilized metallic Sn to reduce Sn^{4+} to Sn^{2+} in mixed Sn-Pb perovskite solution.⁶² When Sn^{4+} free precursor solution is processed, the formation of Sn vacancies caused by Sn^{4+} in the precursor solution can be suppressed in perovskite films (Figure 13a). Thereby, the Sn vacancies in the grains were reduced, and a 3- μ m long carrier diffusion length was achieved in mixed Sn-Pb perovskite. They obtained a PCE of 21.1%, along with FF > 80% and J_{SC} >32 mAcm⁻²

for 1.22-eV Sn-Pb perovskite. They fabricate monolithic all-perovskite TSCs using a 1.77 eV-Eg front subcell and a 1.22-eV back subcell (Figure 13b). The 2-T TSCs delivered a certain PCE of 24.8% for 0.049 cm² device, with a high V_{0c} of 1.965V, a J_{sc} of 15.6 mA cm² and a high FF of 81%, and retained 90% of their initial performance after 463-hour MPP operation under simulated 1-sun illumination (Figure 13c and 13d). Tan group recently used the surface-anchoring zwitterionic antioxidant to inhibit Sn²+ oxidation and passivate defects at the surfaces and grain boundaries. Thereby they fabricated 21.7%-efficiency narrow- E_g single-junction solar cells, and obtained certified 24.2%-efficiency 1-cm²-area all-perovskite tandem cells. The encapsulated tandem devices retained 88% of their initial PCE after 500 hours' operation under 1-sun illumination in ambient conditions.

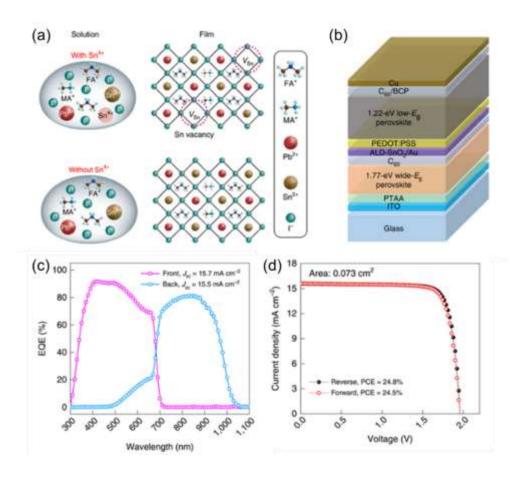


Figure 13. (a) Scheme of the formation of Sn vacancies in mixed Sn-Pb perovskite due to the presence of Sn⁴⁺ in the precursor solution and the suppression of Sn vacancy formation because of the absence of Sn⁴⁺. (b) The device architecture of the perovskite-perovskite tandem solar cell. (c) EQE curves of the front (15.7 mA cm⁻²) and back (15.5 mA cm⁻²) subcells. (d) *J-V* curves of the tandem solar cell (0.049 cm²).⁶²

Zhu and coworkers incorporated 7% guanidinium thiocyanate (GuaSCN) to improve the structural and optoelectronic properties of 1.25-eV Sn-Pb perovskite absorber ((FASnI₃)_{0.6}(MAPbI₃)_{0.4}).⁶⁴ On the one hand, SCN ions with a reasonable amount enlarged grain size and reduced pinholes. On the other hand, Gua with a large ionic radius separated the original 3D ABX₃ perovskite and implemented the 2D-3D heterojunction. The 2D structures passivate surface and grain boundaries through its wider bandgap, block the diffusion of oxygen into grains; and prevent Sn from diffusion out of grains. Consequently, GuaSCN reduced defect densities by a factor of 10; extended carrier lifetimes to 1232 ns; elongated carrier diffusion length to 2.5 μ m; and reduced surface recombination velocity to 1.0 ×10² cm s⁻¹. These improvements endowed (FASnI₃)_{0.6}(MAPbI₃)_{0.4} with >20% efficiency. The combination of low- E_g and wide- E_g PSCs makes 25.0%-efficiency 4-T and 23.1%-efficiency 2-T all-perovskite TSCs.

4. Perovskite-CIGS TSCs

Apart from perovskite-c-Si and perovskite-perovskite tandem cells discussed above, a perovskite cell can also be integrated with a CIGS cell, which has the optimal E_g of 1.08-1.15 eV for the high-performance single-junction device.⁶⁵ CIGs have a large absorption coefficient (10⁵ cm⁻¹) and highly tunable E_g from 1.00 to 1.70 eV by tuning the ratio of Ga/In. Besides, the CIGS technology is rapidly growing with mature commercial thin-film products. These demonstrated the great potential of perovskite-CIGS TSCs, considering

easier processing, flexible form factor, and light weight, etc. The requirement of perovskite bandgap in this TSC is generally similar to perovskite-Si tandems.

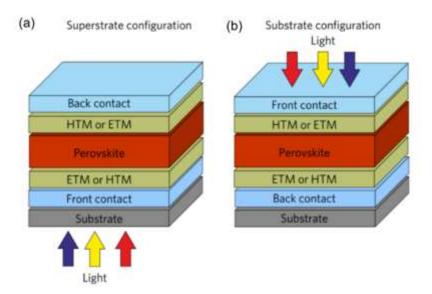


Figure 14. Scheme of a perovskite solar cell in the (a) conventional superstrate configuration and (b) the substrate configuration.66

4.1 Perovskite-CIGS 4-T TSCs

Perovskite-CIGS 4-T tandems do not need to match current density, and both subcells can easily work at the MPP point. In the 4-T configuration, ST-PSCs can freely choose the n-i-p or p-i-n structure. Transparent electrodes for semi-transparent PSCs are carefully summarized in the section of perovskite-Si 4-T tandems and do not need emphases in this section. PSCs can have superstrate (light enters the devices from the substrate) and substrate configuration (light enters from the film side) (Figure 14). The superstrate configuration limits the choice of the substrate to transparent materials. In the substrate configuration, perovskite can be grown on substrates like flexible polymer films and other metal foils. The substrate configuration can facilitate processing directly on CIGS solar cells to realize high-performance polycrystalline all-thin-film tandem devices. However, the substrate architecture constrains device processing and the electronic quality of the perovskite absorber and interlayer. Tiwari and coworkers reported 16.1%-efficiency substrate-configuration ST-PSCs with an architecture of substrate/In₂O₃:H/PTAA/MAPbl₃/PCBM/ZnO NPs/ZnO: Al/Ni-Al grid.⁵⁶ F4-TCNQ-doped PTAA and PCBM ensure efficient charge extraction and transport, and the ZnO NPs can reduce the sputtering damage and provide suitable band alignment with sputtered ZnO: Al. The semitransparent PSC exhibits a high AVT of 80.4% between 800 and 1200 nm, allowing the realization of tandem devices with sum PCE of 22.1% and 20.9% for CIGS (unfiltered:19.2%, filtered:6.0%) and CulnSe₂ (unfiltered:13.0%, filtered:4.8%) bottom cells, respectively.

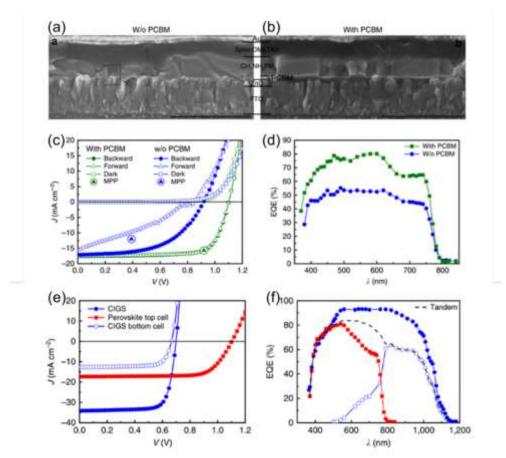


Figure 15. (a,b) The cross-sectional SEM images of devices without (a) and with (b) PCBM, respectively. (c,d) The current density–voltage (*J*–*V*) curves (c) and EQE spectra (d) of the planar perovskite solar cells. (e, f) The current density–voltage (*J*–*V*) curves (e) and EQE spectra (f) of the four-terminal perovskite-CIGS tandem device.⁶⁷

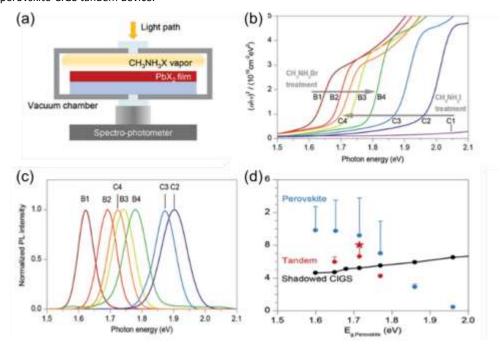


Figure 16. (a) Scheme of in-situ bandgap tuning apparatus. (b) Tauc plots of sample series B and C, showing a blueshift progression with CH₃NH₃Br vapor treatment and a redshift progression with CH₃NH₃I vapor treatment, respectively. (c) PL spectra of series B and C. (d) Efficiency plotted against the bandgap of perovskite.⁶⁸

Mathews and coworkers used triple cation perovskite to demonstrate a 16% efficiency ST-PSCs, which had an average transmittance of 12% in the visible region and > 50% in the NIR region, respectively. They observed that the Ag buffer layer improved the adhesion to the underlayer and FF, outcompeting the MoO_x buffer layer. They demonstrated a 20.7% efficiency for 4-T tandem based on the ST-PSCs and CIGS (unfiltered:12.3%, filtered:4.7%).⁶⁹ Catchpole and coworkers engineered the perovskite composition: FA is the primary cation (75%) due to its stability towards moisture and heat; 15% MA can reduce the size discrepancy of FA and Cs, reduce defects and facilitate the formation of α-phase perovskite. Cs can inhibit humidity-induced decomposition and strain-induced phase segregation, remove the yellow phase, and improve device performance and stability. Rb can balance the tolerance factor, enhance efficiency and stability. So they utilized the 1.62 eV E_g Cs_{0.05}Rb_{0.05}FA_{0.75}MA_{0.135}Pbl_{2.55}Br_{0.45} and developed a 18.1%-efficiency S-T PSC. The combination of S-T PSC and CIGS subcell (unfiltered: 16.5%, filtered: 5.8%) gave a 4-T tandem efficiency of 23.9%. They also delivered a 16.0% PCE ST-PSCs based on 1.75 eV Cs_{0.1}Rb_{0.05}FA_{0.75}MA_{0.15}Pbl_{1.8}Br_{1.2}, and 23.4% 4-T perovskite-CIGS tandem based on 16.0% efficiency ST-PSCs and CIGS. Besides, the four-cation perovskite absorbers exhibit less oxygen sensitivity than MAPbl₃. They deduced that the interaction between MA and oxygen molecules increased surface recombination, resulting in oxygen-induced degradation.

For Perovskite-CIGS 4-T tandems, it is essential to reduce the defect density and increase the long lifetime of wide- E_g perovskite film. In 2015, Tiwari and coworkers updated the performance of perovskite-CIGS 4-T tandems through concerted efforts.⁶⁷ On the one hand, PCBM was incorporated into the regular device structure, enabled by a hybrid thermal evaporation-spin coating technique. PCBM facilitated the growth of high-quality perovskite absorbers and inhibited hysteresis. Figure 15a and 15b presents the cross-sectional SEM images of the planar PSCs with and without the PCBM layer, respectively. The perovskite layer on ZnO exhibits considerable surface roughness and thickness non-uniformity. When grown on PCBM, a uniform and compact perovskite layer with a large grain size can be obtained. As figure 15c and 15d shows, the device with PCBM layer exhibits a PCE of 14.5%, together with a V_{OC} of 1.101 V, J_{SC} of 17.6 mA cm⁻², and FF of 74.9%. On the other hand, sputtered In_2O_3 :H is employed as a transparent rear electrode, yielding 14.2% steady-state efficiency, and 72% average transmittance between 800 and 1150 nm (Figure 15e and 15f). Furthermore, they demonstrated 20.5% efficiency for 4-T perovskite-CIGS (unfiltered: 18.3%; filtered: 6.3%) tandem configuration. Via the 2D-3D heterostructure strategy, Zhu and coworkers from the NERL demonstrated ~20%-efficiency opaque PSCs by incorporating PEAI and Pb(SCN)₂ binary additives in the FA_{0.65}MA_{0.20}Cs_{0.15}PbI_{2.4}I_{0.6} precursor⁷⁰. These additives, enhanced perovskite morphology, inhibited PbI₂ formation (compared with the sole addition of Pb(SCN)₂), and reduced defect density and energetic disorder. The coupling of a 1.68-eV ST-PSC (17.1% efficiency) and a 1.12-eV CIGS bottom cell (unfiltered:20.7%; filtered:8.8%) afforded a record-high 25.9%-efficiency 4-T polycrystalline perovskite-CIGS 4-T TSCs.

Segawa and coworkers achieved a high PCE of 28% in a perovskite-CIGS 4-T tandem comprising a perovskite top cell and a CIGS bottom cell with $E_{\rm g}$ of 1.59 and 1.02 eV, based on a spectrum splitting system equipped with a dichroic mirror with splitting wavelengths of 775 nm. Tandem devices using perovskite and CIGS can potentially exhibit high flexibilities, which cannot be achieved with Si cells. Spectrum splitting tandem cells may not be ideal for practical applications. Nonetheless, this study shows experimentally that perovskite-CIGS TSCs is not only mechanically superior but also exhibits a higher efficiency than Si-based TSCs. 71,72

4.2 Perovskite-CIGS 2-T TSCs

It is noteworthy that, different from other tandems, only n-i-p PSCs can work as top cells for perovskite-CIGS 2-T tandems. In 2015, Guha and coworkers realized monolithic 2-T perovskite-CIGS tandems, where both subcells are customized (Figure 16).⁶⁸ The E_g of CIGS was narrowed from 1.1 eV to 1.04 eV. The optical properties of perovskite were controlled by vapor-based halide exchange reactions. They designed a reactor for continuous in-situ monitoring and precise control of perovskite's E_g via vapor-based halide exchange reactions (figure 16a). The sample of spin-coated PbI2 was annealed by the excess solid MAI vapor source until the layer is converted to 1.58-eV perovskite. MAI can be replaced with MABr. With continued annealing, the E_g will increase when I is replaced by Br, and samples exhibited different Eg values: 1.65 eV (B2), 1.72 eV (B3), and 1.75 eV (B4). In a reverse-sequence process, the E_g of a pure bromide perovskite (C1 with 2.29 eV Eg) is reduced to the following values of 1.95 eV (C2), 1.86 eV(C3), and 1.7 eV (C4) with the MAI vapor (figure 16b and 16c). Consequently, a PCE of 10.9% was obtained for a monolithic 2-T perovskite-CIGS tandem cell (figure 16d).

Besides, CIGS has a very rough surface due to the common vacuum fabrication (sputtering and co-evaporation), which is problematic for building a smooth ICL atop it and may cause the failure of PSCs. Albrecht and coworkers introduced two new hole-selective contacts (MeO-2PACz and 2PACz) for p-i-n PSCs. ⁷³ These two molecules feature carbazole block with phosphonic acid anchoring groups, which enabled the formation of self-assembled monolayers (SAMS) on various oxides, which is ordered and stabilized by π - π interactions between adjacent carbazole fragments. These two SAMS exhibit weaker absorption in the visible wavelength compared with the PTAA interlayer and have an energy alignment to the perovskite absorber. The MeO-2PACz SAM layer enabled dopant-, additive- and interlayer-free PSCs with stabilized PCEs of up to 21.1%, and allowed the realization of a monolithic 2-T CIGSe/perovskite TSC with a certified PCE of 23.26%. The simplicity and diverse substrate compatibility of SAMs help ease the fabrication of 2-T CIGSe/perovskite TSC. Yang and coworkers designed a 22.43% efficiency 2-T perovskite-CIGS tandem with an architecture of Glass/Mo/CIGS/CdS/i-ZnO/boron-doped ZnO (BZO) /ITO/PTAA/Perovskite-PCBM/ZnO NPs/ITO/MgF2. They carried out nanoscale interface engineering of the CIGS surface and used a heavily doped PTAA HTL between two subcells that preserved V_{OC} and enhanced FF and J_{SC} . ⁷⁴ They first deposited an ITO layer and then used chemical

mechanical polishing to smooth out the ITO surface. The polishing process reduced the vertical distance of ITO to 40 nm. Note, the BZO layer under the ITO was not polished, and the original CIGS solar cell can be retained. As the BZO layer displayed a much lower work function (4.0 eV) than PTAA (5.1 eV), the ITO between BZO and PTAA created better ohmic contact. A monolithic 2-T perovskite-CIGS TSC achieved a 22.43% efficiency, and the unencapsulated devices in air retained 88% of their initial efficiency after 500 hours of aging under continuous 1-sun illumination.

4.3 Perovskite-CZTSSe TSCs

Compared with CIGS, Cu2ZnSn(S, Se)4 (CZTSSe) does not use rare metal elements, In and Ga, and thus have a low cost. CZTSSe has a high absorption coefficient of > 10^4 cm⁻¹ in the visible wavelengths and a high theoretical PCE of 28%. The E_g of CZTSSe can vary from 1.00 to 1.50 eV, which is mainly tuned by the ratio of S/Se.

Guha and coworkers demonstrated a monolithic 2-T tandem cell with solution-processed kesterite CZTSSe and MAPbl₃. The device exhibited a high $V_{\rm OC}$ of 1.353 V, close to the sum $V_{\rm OC}$ of the single-junction subcells. The 2-T tandem exhibited a very disappointing PCE of 4.6%, compared with 11.6% for CZTSSe and 12.3% for MAPbl₃.⁷⁵ Although other factors to the high Rs (15.7 Ω cm²) cannot be ruled out, the drastically reduced illumination in all layers and high Rs of the Al contact are major contributors to the low $J_{\rm SC}$ and FF. Ding and coworkers present a tandem combining solution-processed perovskite and CZTSSe subcells in a reflective configuration.⁷⁶ The 1.70 eV- $E_{\rm g}$ perovskite subcell acts as a spectral filter reflecting the sub-bandgap photons to a low bandgap CZTSSe cell. A 4-T measurement afforded a sum efficiency of 16.1%, showing the potential of perovskite-CZTSSe 4-T tandems in realizing high performance and low fabrication cost simultaneously.

5 Perovskite-organic TSCs

Organic photovoltaics (OPVs) are another emerging PV technology.^{77, 78} OSCs can be fabricated through low-cost solution processes using environmentally friendly materials. OSCs can be made into lightweight, flexible and semi-transparent devices. Currently, OSCs have achieved up to 18.2% certified PCE⁷⁹. Perovskite and organic semiconductors have composition and molecular tunability to enable a large pool of photoactive material with different optoelectronic properties, offering the feasibility to construct high-performance perovskite- organic TSCs.

5.1 Perovskite-organic 4-T TSCs

Hygroscopicity risk and easy decomposition are the obvious drawbacks of organic-inorganic hybrid perovskites. Compared with the hybrid organic-inorganic perovskites, all-inorganic perovskite solar cells display better stability toward moisture, light soaking, and thermal stressing. 80, 81 In particular, CsPbI₃ perovskite with E_g of 1.73 eV can serve as top cells in tandem devices with silicon solar cells. CsPbI₃ also exhibits good thermal stability. However, it is difficult to maintain the bandgap and phase stability of CsPbl3, due to their low tolerance factor, the large crystal distortion, and moisture. The all-inorganic perovskite CsPbBr₃ has demonstrated the most promising stability against heat and humidity. For instance, it has been reported that CsPbBr₃-based mesoporous PSCs displayed minimal performance degradation when exposed to highly humid air and extreme temperatures for longer than 3 months. CsPbBr₃ presents a wide E_g of \approx 2.3 eV, efficiently harvest short-wavelength sunlight in 300-530 nm wavelength range, and can achieve ≈ 90% EQE at the wavelength ≈400 nm. A high V_{OC} of ≈1.25 V and high transmittance in the visible region at wavelengths >530 nm can be afforded by mesoporous CsPbBr₃ PSCs, indicating the potential of CsPbBr₃ in ST device and TSCs. Li and coworkers addressed OSCs' photostability issue without efficiency loss by constructing a CsPbBr₃/organic 4-T TSCs. They prepared high-quality CsPbBr₃ all-inorganic perovskite films with a low defect concentration using the dualsource vacuum thermal evaporation method. The corresponding CsPbBr₃ film displays a high AVT of 90.9% in the 530-800 nm wavelength range. The opaque planar PSC exhibited a Voc f 1.44 V, a PCE of 7.78%, and good UV stability. The semitransparent PSC with the transferlaminated PH1000 anode showed a PCE of 5.98% with a J_{SC} of 6.15 mA cm⁻², a V_{OC} of 1.38 V, and a FF of 70.51%. The perovskite ST devices were paired with PTB7-Th: PC71BM (unfiltered: 9.06%, filtered:4.63%), PBDB-T:ITIC (unfiltered: 10.10%, filtered:5.59%) and PBDB-T-SF:IT-4F(unfiltered: 13.72%, filtered:8.05%) systems, to deliver sum PCE of 10.61%, 11.57%, and 14.03%, respectively. These tandems exhibit good UV stability, which is reflected in that they reserve 97.3%-99.4% of the initial PCEs after 100 mW cm⁻² for 120 h.⁸²

5.2 Perovskite-organic 2-T TSCs

The study of perovskite-organic TSCs was pioneered by Yang group in 2015. In their work, MAPbI3 was processed via a one-step deposition with a small-molecule additive named BmPyPhB, which provided heterogeneous nucleation sites and improved the uniformity of the initial nucleation process of perovskite crystals. The organic system (PBSeDTEG8:PCBM) exhibited photosensitivity up to 950 nm and good thermal stability to endure PSC's thermal annealing process. The monolithic 2-T perovskite-organic tandem cells exhibit an efficiency of 10.2%.⁸³

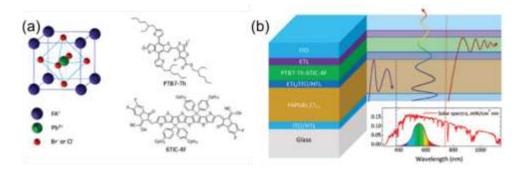


Figure 17. (a) Scheme of FAPbBr $_{x}$ Cl $_{3-x}$ perovskite crystal lattice, Chemical structures of PTB7-Th and 6TIC-4F. (b) Device architectures of selective absorbing tandem ST-PVs. ⁸⁴

Jen group demonstrates unique semi-transparent monolithic 2-T TSCs by rationally choosing selective absorbing photoactive materials.84 They developed 2.36 eV-E_R FAPbBr_{2.43}Cl_{0.57} perovskite with efficient selective absorption in the UV region and meanwhile utilized an organic BHJ blend of PTB7-Th:6TIC-4F with an absorption onset at ≈980 nm (Figure 17a). Therefore, the balance of energy loss between thermalization and transmission is improved, and the resulting 2-T S-T TSC delivered a PCE of 10.7%, an averaged transmittance of 52.91%, and a light utilization efficiency of 5.66% (Figure 17b). Jen group further employed phenmethylammonium bromide (PMABr) to passivate the wide- E_g perovskite $Cs_{0.1}(FA_{0.6}MA_{0.4})_{0.9}Pb(I_{0.6}Br_{0.4})_3$ and obtain a high V_{OC} of 1.22 V for the PSC.⁸⁵ The assembled rigid perovskite-organic 2-T TSCs yielded a remarkable PCE of 15.13%, with $V_{\rm OC}$ of 1.85 V, $J_{\rm SC}$ of 11.52 mA cm⁻², and FF of 70.98%. Moreover, they realized flexible perovskite-organic TSCs with a PCE of 13.61% and a $V_{\rm OC}$ of 1.80 V. The rigid and flexible perovskite-organic TSC generated a solar-to-hydrogen efficiency of 12.30% and 11.21%, illustrating the potential application of perovskite-organic TSC in a photovoltaics-driven water splitting system. Zeng et al. demonstrated all-inorganic perovskite-organic TSCs with CsPbl₂Br cell as the front cell and PTB7-Th:CO_/8DFIC: PC₇₁BM as the rear cell. The wide-Eg CsPbl₂Br can absorb short-wavelength photons before 650 nm, while the low-bandgap NFA CO_/8DFIC can absorb long-wavelength photons until 1,050 nm. They use a device structure of ITO/SnO2/CsPbl2Br/PTAA/ICL/ZnO/PTB7-Th: CO_i8DFIC: PC₇₁BM /MoO3/Ag, where the ICL consisting of MoO3/Au was made by thermal deposition. The corresponding best tandem exhibits a PCE of 15.04%, with a V_{OC} of 1.71 V, a J_{SC} of 11.98 mA cm⁻², and a FF of 73.4%.⁸⁶ Very recently, Chen et al. developed a semi-empirical model to calculate the practical PCE limits of perovskite-organic TSCs. Based on their calculation, an efficiency >30% is approachable as the rapid advances of low-Eg organic semiconductors and wide-Eg perovskite materials. They prepared monolithic perovskite-organic TSCs by employing a 1.77 eV-Eg FA_{0.8}MA_{0.02}Cs_{0.18}Pbl_{1.8}Br_{1.2} for the front cell and a 1.41 eV -E_g organic materials for the rear subcell (figure 18a).⁸⁷ As shown in figure 18b, the absorbed energy distribution is simulated by the transfer matrix method. To effectively match the J_{SC}, they established the relationship between the photo-current and the thickness of each subcell. The highlighted blue curve indicates that the J_{SC} of both subcells is matched theoretically (figure 18d). Featuring all-thermally evaporated BCP/Ag/MoO₃ ICLs, the 2-T monolithic perovskite-organic tandems delivered good reproducibility and landmark PCE of 20.6% (certified as 19.54%). The UV sensitivity of OPVs is significantly reduced with the UV filtering function of the perovskite layer. Recently, the integrated perovskite-organic solar cells without any recombination layers have generated wide attention. This integrated structure takes the advantages of tandem cells using both perovskite solar cells and near-infrared (NIR) BHJ organic solar cells for wide-range absorption and the simple fabrication of a single-junction device.88-90 The high Voc and wide-range absorption can be maintained at the same time. However, as this device does not belong to the tandem device in the strict sense, it will not be discussed here.

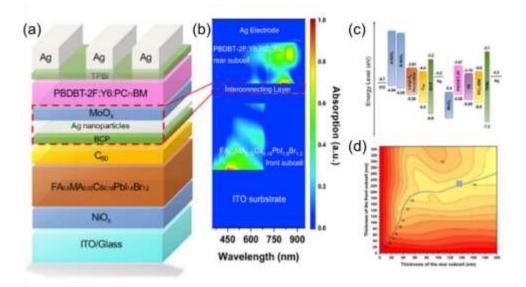


Figure 18. (a) Perovskite-organic tandem structure with a ICL of MoO_x/Ag nanoparticles/BCP. (b) The simulation of the photon absorptions distribution in tandem. (c) Energy levels distribution of components in the tandem device. (d) Stimulated current density as a function of the thicknesses of the front and rear subcells.⁸⁷

Conclusions and outlook

Perovskite solar cells are an ideal candidate to construct tandem cells, due to their high $V_{\rm OC}$, highly tunable $E_{\rm g}$, and low-temperature solution processability. We have presented the key advances of perovskite-silicon, all-perovskite, perovskite-CIGS, and perovskite-organic TSCs. Currently, the 1-cm² perovskite-silicon monolithic two-junction, the two-terminal device has achieved the highest results of 29.15%, which pushes further beyond the highest efficiency for a single-junction silicon cell with a 9% relative advantage. Perovskite TSCs are stepping towards commercialization, which requires high device efficiencies, long-term stability, and low-cost and reproducible fabrication. The following challenges and strategies should be considered for future developments.

The high-performing tandems required the reduced additional optical losses (reflective losses and parasitic absorption losses) and electrical losses in tandem structure. The optical losses of PSCs are greatly reduced by changing the configuration of PSCs from n-i-p to p-i-n, which may account for the phenomenon that the maximum efficiency of TSCs are obtained with p-i-n PSCs. The parasitic absorption in p-i-n PSCs can be minimized by reducing the thickness of electron transport layer like C_{60} and PCBM, as the wide-bandgap BCP/Bphen can work as the hole-blocking layer and the buffer layer. Take perovskite-Si TSCs as an example, PSCs with the p-i-n structure are the most promising for high PCEs due to the less optical losses, low cost, compatible fabrication process with the SHJ structure, and the highest PCE of 22.3%.

LM layers like MgF_2 , SiO_x on the substrate surface should be utilized to reduce the reflective optical loss. The transparent electrode should have high conductivity and high transparency. ICLs should have small resistance and lateral conductivity and high transparency. The high-performance tandems also require efficient subcells.

- 1) wide-bandgap PSCs
 - Compositional engineering, the partial substitution of I by Br for instance, can enlarge the bandgap of perovskite. The quality of wide-bandgap perovskite film should be improved, with low defect concentration, weak charge recombination, and long carrier lifetime. As discussed above, many strategies are reported, such as the incorporation of Gua/PEA-based additives for constructing the 2D-3D hybrid structure, Lewis bases/acids, PbI₂, Br and Cl, ligand passivation, and metal ions. To fabricate PSCs on the textured or rough surfaces of Si and ClG solar cells, some special fabrication methods should be utilized, like blade-coating, thermal co-evaporation, vacuum-solution binary step.
- 2) narrow-bandgap bottom cells
 - The Sn-Pb PSCs should be improved by tackling the material and device issues. First, Sn and Sn-Pb perovskites require thick film perovskite absorbers and in turn long carrier diffusion lengths. Second, the quick crystallization of Sn-substituted perovskite results in inferior-quality films. Last, the facile oxidation of Sn^{2+} to Sn^{4+} degrades its device performance. Several strategies are reported and summarized in the manuscript, which is partially similar to the strategies for wide- E_g perovskites, such as the addition of Cl, excess metal ions, Snl_2 , SnF_2 , $SnCl_2$, FA, Cd, insulating polymers, and alkali metal ions. For organic bottom cells, high-performance non-fullerene acceptors, especially fused-ring electron acceptors, should be developed. The

match of donors and acceptors should be guaranteed. The ternary/multiple strategies can be applied. The bulk-heterojunction active layer should be optimized by device engineering, including solvent additive, thermal annealing, and solvent vapor annealing. As Si and CIG bottom cells already have very mature products, their improvement is not the highlight to some extent. The bandgap and thickness of the top and bottom cells should be prudently optimized together.

To reduce the cell-to-module PCE gap, the main step is to obtain high-performance large-scale perovskite top cells. Our group performed systematic investigations on fabricating PSCs via blade coating in ambient conditions with high humidity and room temperature. The morphological and photoluminescence characterization indicate that air-knife-assisted drying significantly contributed to nucleation of the perovskite, determination of the thin-film morphology and device performance, and protection of the perovskite and its intermediate phase against moisture during solidification. We fabricated hysteresis-free PSCs with PCE of 21.1% and 18.0% for small and large-area blade-coated devices, comparable to that of the glovebox-processed blade-coated device. These PSCs also exhibit comparable stability to those fabricated in a glovebox. Our group also induce a multifunctional sulfobetaine-based zwitterionic surfactant to facilitate room-temperature meniscus coating of high-quality perovskite films. The systematic in-situ study revealed the crystallization pathway of perovskite and emphasized the synergistic effects of surfactants in film formation, crystallization kinetics adjustment, defect passivation, and moisture barrier protection. The coated PSCs have negligible deterioration when the device area is upscaled to 0.8 cm², which is among the highest records for the upscaling coated PSCs. Besides, this method significantly improves the stability of perovskite films and devices under different aging conditions. 92- 94 It is noteworthy that it is still challenging to deposit sub- μ m thick perovskite layers from solution onto textured silicon with typical pyramids of 3-10 µm heights. Huang group demonstrated a new perovskite/silicon tandem solar cell architecture involving silicon wafers textured with pyramids less than 1 µm in height, which are rough enough to scatter light within the silicon but smooth enough to solution process a perovskite film. The silicon bottom cell was fully textured with scalable chemistries. The subμm pyramids and precursor additives enabled solution-processed dense perovskite layers that fully cover the silicon texture. This tandem architecture can be adopted by silicon solar cell manufacturers, which hunt for tandem technologies with low required capital expenditures, the potential for rapid scaling, and minimum modification of their silicon cells.

For perovskite-organic tandems, large-scale fabrication also imposes requirements of large-scale OSCs. Orthogonal solvents should be used to protect the layers underneath. Otherwise, barrier layers should be designed, or a vacuum-deposition method should be utilized. Optimization of interconnection design to achieve a right balance between leakage currents and resistive losses is also indispensable for large-scale fabrication.

Stability is another essential factor for the commercialization of perovskite TSCs. For instance, the stability of perovskite-Si tandems should match the 25-year lifetime of silicon solar cells. The polycrystalline perovskite films contain a high density of defects, including point and extended imperfections, which heavily influences the devices' stability. In terms of perovskite composition (ABX₃), the X site poses photo-instability concerns; the A-site cation has a relationship with thermal, moisture, and oxygen instability. The volatile MA needs encapsulation to prevent volatilization. Besides, the narrow $E_{\rm g}$ perovskite with ${\rm Sn}^{2+}$ on the B-site has a strong tendency from the Sn²⁺ to Sn⁴⁺ state when exposed to oxygen or moisture. The A-site and X-site cations of perovskite are very mobile, which can also degrade stability. Photo-induced phase segregation in wide- $E_{\rm g}$ perovskite and the oxidation of Sn²⁺ to Sn⁴⁺ in low-bandgap are the main factors to limit the stability of PSC tandems. Admittedly, some strategies can use coordinate and ionic bonding to target the defects of the surface and grain boundaries, the conversion of those extended defects to wide Eg materials, and ion migration. 95-99 For example, our group introduce NFA molecule (IT-M) as lewis base to improve perovskite interfaces, which effectively enhances the electronic properties of Cs_{0.05}(MA_{0.17}FA_{0.83})_{0.95}Pb(I_{0.83}Br_{0.17})₃ thin films, passivates the perovskite surface defects, and further boosts the performance of PSCs. The introduction of IT-M significantly enhances the charge transport properties and boosts the photogenerated photoluminescence lifetime from 1.46 to 2.20 ms. Our group also found that the π -conjugated polymer PBDB-T trigger heterogeneous nucleation over the perovskite precursor film and passivate the trap states of the mixed perovskite through the formation of lewis adducts between lead and oxygen atom in PBDB-T. The p-type hydrophobic PBDB-T fills in the perovskite grain boundaries to improve charge transfer for better conductivity and prevent moisture invasion into the perovskite active layers, thus simultaneously improving efficiency and stability. However, it is still unexplored whether the phase segregation and the oxidation issues will reappear in the long term and how long the effects of the abovementioned strategies can last. Besides, the weak layer to layer adhesivity and different thermal expansion coefficients also deteriorate perovskite tandems' long-term stability. Currently, there are no standards of stability measurements for scientific research. The literature involving the stability of PSC tandems has different test conditions, increasing the difficulty for scientists to compare the effectiveness of different stability strategies.

Overall, through concerted efforts of materials design, device engineering, interface engineering, device physics, and improved fabrication technology and encapsulation, perovskite tandems can overcome the abovementioned challenge. Ideal perovskite-based tandem cells, especially the perovskite-Silicon tandem cell, can surpass 30% efficiency and achieve good stability, to meet commercialization requirements in the near future.

Conflicts of interest

There are no conflicts to declare.

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Table 1. Summary of the structure and performance of perovskite-based tandem solar cells

	Eg	Transparent		Voc	J _{SC}	FF	PCE	total	
Absorber	[eV]	electrode	ICL	[mV]	[mA cm ⁻²]	[%]	[%]	PCE [%]	Ref.
MAPbl ₃	~1.6	MoO _x /ITO		821	14.5	51.9	6.2	13.4	9
SHJ	~1.1		_	689	13.7	76.7	7.2		
MAPbI ₃	~1.6	AZO/ITO/MgF ₂		952	16.5	77.4	12.3	18.0	10
mono-Si	~1.1		_	562	13.3	76.2	5.7		
MAPbI ₃	~1.6	MoO _x /IZO		870	17.5	68	10.36	18.18	
SHJ	~1.1		_	690	14.6	77.6	7.82		11
MAPbI ₃	~1.6	IZO		938	17.36	59.6	9.71	18.19	
SHJ	~1.1			693	15.81	77.44	8.48		
$Cs_{0.05}FA_{0.81}MA_{0.14}PbI_{2.55}Br_{0.45}$	1.63	MoO ₃ /IZO/MgF ₂	_	1120	22.3	77.7	19.4	28.6	12
SHJ	~1.1			700	17.2	76.6	9.2		
MAPbl₃	~1.6	Cu/Au		1080	20.6	74.1	16.5	23.0	16
SHJ	~1.1			679	12.3	77.9	6.5		
MAPbI₃		AgNW	_	1025	17.5	71	12.7	17.0	17
mc-Si				547	11.1	70.4	4.3		
MAPbl ₃		MoO _x /AuNW/	-	1156	19.8	79.9	18.3	27.0	18
SHJ				698	15.6	80.1	8.7	•	
MAPbI ₃		graphene		900	12.6	55	6.2	13.2	19
SHJ			_	670	14	73.8	7	15.2	
MAPbI ₃		optical splitter		987	10.6	71.5	7.5	28	22
SHJ			_	728	34.9	80.9	20.5		
MAPbI ₃	~1.6		n++/p++ Si						
c-Si	~1.1		tunnelling junction	1580	11.5	75		13.7	23
MAPbl₃			ALD SnO ₂	1780	14	79.5		19.9	24
SHJ			3.102	1,30		. 5.5		19.9	
MAPbI ₃			IZO	1692	15.8	79.9		21.4	

		1	T	T	T	ı	ı	Т	ı
SHJ									25
MAPbI ₃			SnO₂	1676	16.1	78		21	
homo-junction Si									26
MAPbl ₃			SnO ₂	1660	15.6	68		17.6	
homo-junction Si			31102			08		17.0	
(FAPbI ₃) _{0.83} (MAPbBr ₃) _{0.17}			SnO ₂	1740	16.2	78		21.9	27
homo-junction Si			552						
(FAPbI3) _{0.83} (MAPbBr3) _{0.17}			SnO ₂	1730	16.5	81		23.1	28
homo-junction Si			552	2700	20.0	02		20.2	
Cs _{0.05} Rb _{0.05} FA _{0.765} MA _{0.135} PbI _{2.5}	1.63								
₅ Br _{0.45}	1.03		TiO ₂	1763	17.8	78.1		24.5	29
SHI									
Cs _{0.05} (MA _{0.17} FA _{0.83})Pb _{1.1} (I _{0.83} Br									
0.17)3			ITO	1760	18.5	78.5		25.5	30
SHI									
Cs _{0.05} (FA _{0.83} MA _{0.17}) _{0.95} Pb(I _{0.8} Br									
0.2)3			nc-SiO₂:H/ITO	1791.9	19.02	74.6		25.43	31
SHJ									
Cs _{0.05} MA _{0.15} FA _{0.8} Pbl _{2.25} Br _{0.75}			NiOx/InOX	1780.5	19.07	75.36		25.7	32
SHJ				2700.0					
Cs _{0.1} MA _{0.9} Pb(I _{0.9} Br _{0.1}) ₃			ITO	1820	19.2	75.3		26.2	33
SHJ									
FA _{0.5} MA _{0.38} Cs _{0.12} Pbl _{2.04} Br _{0.96}	1.69		ITO	1655	16.5	81.1		22.22	42
SHJ									
(FA _{0.65} MA _{0.2} Cs _{0.15})Pb(I _{0.8} Br _{0.2}) ₃	1.68		ITO	1756	19.2	79.2		26.7	45
SHJ	1.12		110	1730	13.2	73.2		20.7	
FA _{0.75} Cs _{0.25} Pb(I _{0.85} Br _{0.15}):2 to 5	1.67		_	1886	19.12	75.3		27.13	
mol% MAPbCl₃	1.07								43
SHJ									
1		L	i		I .	l	1		l

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$MAPb(I_{0.95}Br_{0.05})_3$	~1.6		ITO	1670	18.29	77		23.5	46
SHJ	1.1								
CsFAPbIBr	~1.77		IZO						
CsFAPbIBr	~1.55		nc-Si:H	2700				13.2	48
SHJ									
MAPbBr ₃	2.2		DCDM DQUT	4050	0.4	66		10.0	50
MAPbI ₃	1.58		PCBM:P3HT	1950	8.4	66		10.8	
MAPbI ₃	1.58	ITO		1080	16.69	75	13.52	19.08	54
MA _{0.5} FA _{0.5} Pb _{0.75} Sn _{0.25} I ₃	1.33		-	760	9.14	80	5.56	19.08	
FA _{0.83} Cs _{0.17} Pb(I _{0.5} Br _{0.5}) ₃	1.8		0.0/570/570	1000				100	
FA _{0.75} Cs _{0.25} Sn _{0.5} Pb _{0.5} I ₃	1.2		SnO ₂ /ZTO/ITO	1660	14.5	70		16.9	55
FA _{0.83} Cs _{0.17} Pb(I _{0.83} Br _{0.17}) ₃	1.6	ITO		970	20.3	79	15.7		_
FA _{0.75} Cs _{0.25} Sn _{0.5} Pb _{0.5} I ₃	1.2		_	740	7.9	73	4.4	20.1	
FA _{0.3} MA _{0.7} PbI ₃	~1.58	MoO _x /Au/MoO _x		1141	20.1	80.0	18.3		58
(FASnI ₃) _{0.6} (MAPbI ₃) _{0.4}	1.25		-	808	4.8	74.4	2.9	21.2	
FA _{0.8} Cs _{0.2} Pb(I _{0.7} Br _{0.3}) ₃	1.75		Ag/MoO _x /ITO	1922	14	78.1		21	59
(FASnI ₃) _{0.6} (MAPbI ₃) _{0.4} :Cl	1.25		Ag/MOO _x /110	1922	14	76.1		21	
FA _{0.6} Cs _{0.4} Pb(I _{0.65} Br _{0.35}) ₃	1.8		SnO₂/ITO/PEDO						
$FA_{0.5}MA_{0.45}Cs_{0.05}Pb_{0.5}Sn_{0.5}I_{3}:C$ d	1.22		T:PSS	1990	15.1	77		23	61
Cs _{0.2} FA _{0.8} PbI _{1.8} Br _{1.2}	1.77		SnO ₂ /Au/PEDOT						62
MA _{0.3} FA _{0.7} Pb _{0.5} Sn _{0.5} I ₃	1.22		:PSS	1965	15.6	81		24.8	02
CS _{0.05} FA _{0.8} MA _{0.15} PbI _{2.55} Br _{0.45}	1.63	IZO		1124	20.8	79.3	18.5	25.4	
(FASnI ₃) _{0.6} (MAPbI ₃) _{0.4}	~1.25		-	806	10.5	80.6	6.8	25.4	64
Cs _{0.05} FA _{0.8} MA _{0.15} PbI _{2.55} Br _{0.45}	1.75			1042	45.04	00.24		23.4	-
(FASnI ₃) _{0.6} (MAPbI ₃) _{0.4}	~1.25		-	1942	15.01	80.31			
MAPbI ₃			ITO	4252	F. C	66.4		4.5	75
Cu ₂ ZnSn(S, Se) ₄	1.1		ITO	1353	5.6	60.4		4.6	
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MAPbl ₃		AgNW		1025	17.5	71	12.7	18.6	17

MAPbl ₃		In ₂ O ₃ :H		1104	17.4	73.6	14.1	20.4	67
CIGS			-	667.4	12.7	74.9	6.3	20.4	
Perovskite				1450	12.7	56.6		10.9	68
CIGS			-	1450	12.7	30.0		10.9	
(FA _{0.65} MA _{0.20} Cs _{0.15})Pb(I _{0.8} Br _{0.2})	1.68		IZO	1137	19.6	76.8	17.1	25.9	70
CIGS	1.12		-	715	15.6	79.2	8.8		
				, 13	13.0	75.2	0.0		
Cs ₅ (MA ₁₇ FA ₈₃) ₉₅ Pb(I ₈₃ Br ₁₇) ₃	1		AZO/SAM	168	19.17	71.9		23.16	73
CIGS									
Cs _{0.09} FA _{0.77} MA _{0.14} Pb(I _{0.86} Br _{0.14})	1.59								
3	1.59		ITO/PTAA	1773.9	17.3	73.1		22.43	74
CIGS	1		-						
MAPbI ₃	1.52		PFN/TiO ₂ /PEDO	4520	10.05	67		40.22	83
PBSeDTEG8:PCBM			T:PSS	1520	10.05	67		10.23	
CsPbBr ₃	2.3	PH1000		1380	6.15	70.51	5.98	14.03	82
PBDB-T-SF:IT-4F			-	880	12.18	75.16	8.05	- 14.05	
FAPbBr _{2.43} Cl _{0.57}	2.36		ZnO-NP/ITO	2230	6.61	74		10.83	84
PTB7-Th:6TIC-4F				2230	0.01	'.		10.03	
Cs _{0.1} (FA _{0.6} MA _{0.4}) _{0.9} Pb(I _{0.6} Br _{0.4})	1.74								
3	2., 1		PEDOT/Ag/BCP	1850	11.52	70.98		15.13	
PBDB-T:SN6IC-4F	1.30		1						85
Cs _{0.1} (FA _{0.6} MA _{0.4}) _{0.9} Pb(I _{0.6} Br _{0.4})	1.74								1
3	1.74		PEDOT/Ag/BCP	1800	11.07	68.31		13.61	
PBDB-T:SN6IC-4F	1.30		1						
FA _{0.8} MA _{0.02} Cs _{0.18} PbI _{1.8} Br _{1.2}	1.77		BCP/Ag/MoO ₃	1902	13.05	83.1		20.6	87
PM6:PC ₇₁ BM:Y6	1.41		BCF/Ag/IVIOU3	1902	13.03	65.1		20.0	