

1 Photochemical evolution of continental air masses and their influence on ozone formation
2 over the South China Sea

3 Yu Wang¹, Hai Guo*,¹, Xiaopu Lyu¹, Luyao Zhang¹, Yangzong Zeren¹, Shichun Zou**,²,
4 Zhenhao Ling³

5 ¹ Air Quality Studies, Department of Civil and Environmental Engineering, The Hong Kong
6 Polytechnic University, Hong Kong, China

7 ² School of Marine Sciences, Sun Yat-sen University, China

8 ³ School of Atmospheric Sciences, Sun Yat-sen University, China

9 * Corresponding author: ceguohai@polyu.edu.hk

10 ** Second corresponding author: ceszsc@mail.sysu.edu.cn

11 **Abstract**

12 To investigate photochemical ozone (O₃) pollution over the South China Sea (SCS), an
13 intensive sampling campaign was conducted from August to November simultaneously at a
14 continental site (Tung Chung, TC) and a marine site (Wan Shan Island, WSI). It was found
15 that when continental air masses intruded the SCS, O₃ episodes often occurred subsequently.
16 To discover the causes, a photochemical trajectory model (PTM) coupled with the near-
17 explicit Master Chemical Mechanism (MCM) was adopted, and the photochemical processes
18 of air masses during the transport from TC to WSI were investigated. The simulated O₃ and
19 its precursors (*i.e.* NO_x and VOCs) showed a reasonably good agreement with the
20 observations at both TC and WSI, indicating that the PTM was capable of simulating O₃
21 formation for air masses traveling from TC to WSI. The modeling results revealed that during
22 the transport of air masses from TC to WSI, both VOC and NO_x decreased in the morning
23 while O₃ increased significantly, mainly due to rapid chemical reactions with elevated
24 radicals over the SCS. The elevated radicals over the SCS were attributable to the fact that
25 higher NO_x at TC consumed more radicals, whereas the concentration of radicals increased

from TC to WSI because of NO_x dilution and destruction. Subsequently, the photochemical cycling of radicals accelerated, leading to high O₃ mixing ratios over the SCS.

Furthermore, based on the source profiles of the emission inventory used, the contributions of six sources, *i.e.* gasoline vehicle exhaust, diesel vehicle exhaust, gasoline evaporation and LPG usage, solvent usage, biomass and coal burning, and biogenic emissions, to maritime O₃ formation were evaluated. The results suggested that gasoline vehicles exhaust and solvent usage largely contributed the O₃ formation over the SCS (about 5.2 and 3.8 ppbv, respectively). This is the first time that the contribution of continental VOC sources to the maritime O₃ formation was quantified.

Keywords: Ozone; Continental air masses; VOC sources; Photochemical Trajectory Model; South China Sea

1 Introduction

Surface ozone (O₃) is a secondary pollutant which has large detrimental effect on human health, visibility and climate change (NRC, 1991; Seinfeld and Pandis, 2016). Ozone pollution involves many aspects, such as emissions of precursors (*i.e.* volatile organic compounds (VOCs) and nitrogen oxides (NO_x = NO + NO₂)), photochemical formation and dynamic transport on different scales, making it a complex phenomenon (Jacob, 1999; Guo *et al.*, 2017). O₃ precursors emitted in urban areas can not only cause photochemical smog in the source region, but also travel to broad remote areas, resulting in increased O₃ in these downwind receptor regions (Finlayson-Pitts and Pitts, 1993; Solberg *et al.*, 2004). In addition, the emission sources of these primary precursors are variable, making O₃ pollution more complicated (Guo *et al.*, 2017). Thus, understanding the emission sources of precursors and their impact on O₃ formation, and subsequently formulating and implementing control measures are critical to mitigate O₃ pollution.

50 With high emissions of VOCs and NO_x, the fast developing Pearl River Delta (PRD) region,
51 including Hong Kong and Macau, has suffered from high O₃ pollution during the last three
52 decades (Wang *et al.*, 1998; Guo *et al.*, 2009; Wang *et al.*, 2009; Xue *et al.*, 2014; Li *et al.*,
53 2014; Wang *et al.*, 2017a, b). Numerous studies have focused on the VOC source
54 apportionments in this region, which well elaborated VOC sources and their impact on O₃
55 formation (Zhang *et al.*, 2007; Guo *et al.*, 2017). For example, Guo *et al.* (2004) first probed
56 the sources of VOCs in urban Hong Kong and found that vehicle exhaust was the dominant
57 sources of VOCs, followed by solvent usage, natural gas/liquefied petroleum gas (LPG)
58 leakage and industrial emissions. Based on VOCs observations at seven sites in the PRD, Liu
59 *et al.* (2008) identified 12 sources and confirmed that vehicular exhaust made the highest
60 contribution (~50% of ambient VOCs) at the urban sites while solvent usage was remarkable
61 (~30%) at an industrial and a non-urban sites. A long-term study during 2005–2014
62 conducted at a suburban site in Hong Kong concluded that the two major sources of VOCs,
63 *i.e.* solvent-related (~55%) and traffic-related sources (~45%), had no significant changes
64 over the period (Ou *et al.*, 2015). Moreover, Ling *et al.* (2011) assessed the relative
65 contribution of VOC sources to O₃ formation at a suburban site of PRD and demonstrated
66 that solvent usage, diesel vehicular emissions and biomass/biofuel burning were the key
67 contributors to the O₃ formation. Based on the measurements at an urban site in autumn 2010,
68 Ling and Guo (2014) found that paint and sealant solvents, diesel exhaust and LPG usage
69 contributed most to O₃ formation in urban Hong Kong.

70 Compared to urban locations, O₃ levels are often greater in remote areas such as mountainous
71 (Guo *et al.*, 2013; Lam *et al.*, 2013) and oceanic areas (Wang *et al.*, 2018a, b). Guo *et al.*
72 (2013) revealed the association of high O₃ concentrations observed at a mountainous site of
73 Hong Kong with the urban emissions from the foot of the mountain where urban centre is
74 located. Lam *et al.* (2013) further applied a moving box model to evaluate the impact of

VOCs emitted from urban sources (*i.e.* mountain foot) on the O₃ formation at the receptor site (*i.e.* mountain summit) under the influence of mountain-valley breezes. They found that vehicular emissions from the foot of the mountain were the dominant VOC sources contributing to O₃ formation at the mountain summit. Different from the causes of O₃ formation in mountainous areas, the interaction between continental and maritime air such as sea-land breezes sometimes forms weak circulation cells in which pollutants become entrained, enhancing the O₃ formation in coastal areas of the PRD region (Wang *et al.*, 2018a). Furthermore, Wang *et al.* (2018b) confirmed that O₃ precursors effused to the maritime atmosphere elevated OH radicals, and accelerated the photochemical reaction cycling processes, resulting in enhanced photochemical O₃ production. In this study, although the high O₃ pollution over the SCS is recognized, how the emissions of O₃ precursors in the continental PRD affect the O₃ formation over the maritime atmosphere remains unclear.

To tackle the air pollution problems in the study region, a Photochemical Trajectory Model coupled with near-explicit Master Chemical Mechanism (PTM-MCM) was developed to simulate O₃ mixing ratios and the processes of O₃ photochemistry during the transport of continental O₃ precursors from coastal PRD region to the SCS. Furthermore, the contributions of individual continental VOC sources to O₃ formation during their transport to the SCS were evaluated, and some control policy suggestions were provided accordingly.

2 Methodology

2.1 Sampling site

In order to understand the chemical evolution of air masses during the transport from continental source region to maritime receptor region, concurrent field measurements were conducted at a suburban site of Hong Kong and a remote maritime site over the SCS in autumn 2013 (Figure 1). The suburban Tung Chung (TC, 22.29° N, 113.94° E) site, located

99 in southwestern Hong Kong, was selected as the continental source site, where highly emitted
100 O₃ precursors and O₃ episodes were often observed, particularly in autumn (Cheng *et al.*,
101 2010; Jiang *et al.*, 2010; Ling *et al.*, 2013; Xue *et al.*, 2014; Wang *et al.*, 2017b). The distance
102 from TC to the Hong Kong International Airport, urban center of Hong Kong and Macau is
103 about 3 km, 20 km and 38 km, respectively. Moreover, TC site belongs to the Hong Kong
104 Environmental Protection Department (HKEPD) air quality monitoring network, in which
105 high quality online data are measured. The measurement instruments were installed in a room
106 on the rooftop of a building (~27.5 m above ground level). Additional details for TC site have
107 been described elsewhere (Cheng *et al.*, 2010; Jiang *et al.*, 2010; Wang *et al.*, 2017b).

108 In contrast, the maritime Wanshan Island (WSI, 21.93° N, 113.73° E) site, which is bounded
109 to the north by the Pearl River Estuary, was chosen as the receptor site. The island is right in
110 the downwind direction of the PRD region in autumn and winter, when the northeast
111 monsoon prevails (Ling *et al.*, 2013; Wang *et al.*, 2017b). During this period, WSI is
112 regarded as a receptor site of TC, which provides a unique opportunity to evaluate O₃
113 formation during the transport of air masses from the continental region to the maritime
114 atmosphere. The straight distance is about 44 km between WSI and TC. The island has an
115 area of 8.1 km² and a population of about 3,000 with sparse anthropogenic emissions. The
116 sampling instruments were installed at the top floor of the National Marine Environmental
117 Monitoring Station. The height was about 65 m above sea level and ~10 m above ground
118 level.

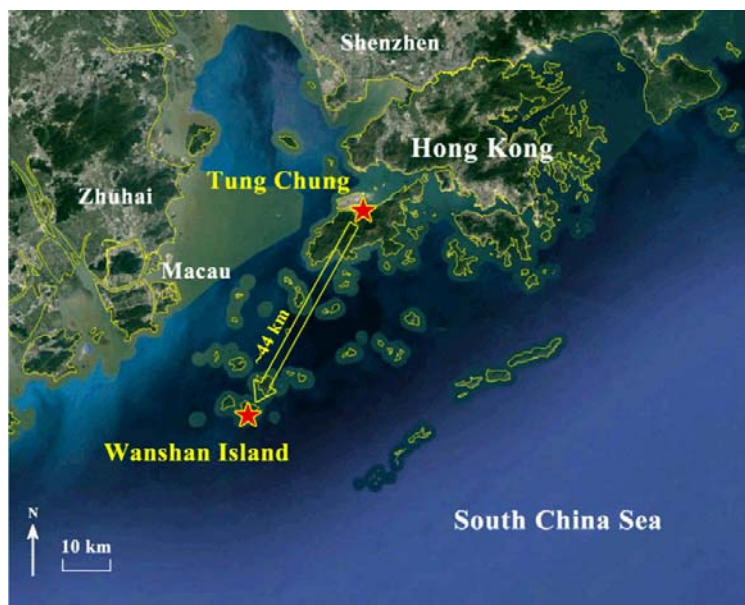


Figure 1. Location of the sampling sites (Tung Chung and Wanshan Island).

2.2 Measurement techniques

2.2.1 Measurements of trace gases and meteorological parameters

At the WSI site, trace gases (*i.e.* O₃, NO-NO₂, CO and SO₂) were continuously monitored with commercial analyzers developed by Advanced Pollution Instrumentation (API) Inc. Briefly, O₃, NO-NO₂, CO and SO₂ were monitored with UV photometric O₃ analyzer, chemiluminescence NO/NO₂/NO_x analyzer, gas filter correlation trace level CO analyzer and UV fluorescence SO₂ analyzer, respectively. The time resolution of all instruments was 1 minute. The instrument mode and detection limit are listed in Table 1. More information of quality assurance and control protocols for these devices can be found in our previous studies (Guo *et al.*, 2009; Guo *et al.*, 2013; Wang *et al.*, 2018a). In addition, meteorological parameters (*i.e.* temperature, relative humidity, solar radiation, wind speed and wind direction) were routinely monitored by a weather station (Vantage Pro 2 plus, Davis Instruments) with a time resolution of 5 minutes. At the TC site, hourly data of the aforementioned trace gases and meteorological parameters were obtained from the HKEPD

(<http://epic.epd.gov.hk/ca/uid/airdata>). The descriptions of the quality assurance and control protocols are provided in the HKEPD report (HKEPD, 2015).

Table 1 Descriptions of O₃, NO-NO₂, CO and SO₂ analyzers

trace gas	Equipment	Model	Detection limit
O ₃	UV absorption O ₃ analyzer	<i>API, Model 400E</i>	0.6 ppbv
NO-NO ₂	Chemiluminescence NO/NO ₂ /NO _x analyzer	<i>API, Model 200E</i>	0.4 ppbv
CO	Gas filter correlation trace level CO analyzer	<i>API, Model 300E</i>	0.02 ppmv
SO ₂	UV fluorescence SO ₂ analyzer	<i>API, Model 100E</i>	0.4 ppbv

2.2.2 Sampling and analysis of VOCs

During the sampling period, non-methane hydrocarbons (NMHCs) and carbonyls were collected simultaneously at both sites for 21 days, including both non-O₃ episode and O₃ episode days. An O₃ episode day is defined as the day with the peak one-hour averaged O₃ mixing ratio exceeding 100 ppbv (Grade II, China National Ambient Air Quality Standard). 1-hr NMHCs canister samples and 2-hr carbonyl cartridge samples were simultaneously collected every two hours during 7:00 – 19:00 at each site. Intensive NMHCs sampling was also carried out at WSI on selected seven days (*i.e.*, 3, 4, 9 and 22-25 October) with eleven one-hour samples per day (every two hours during 1:00 – 22:00 inclusive). In total, 311 valid VOC samples (144 at TC and 167 at WSI) and 227 carbonyl samples (124 at TC and 103 at WSI) were collected. More descriptions of the VOC sampling can be found in Wang *et al.* (2018a).

The whole-air samples of NMHCs were collected using 2-L electro-polished stainless steel canisters. The canisters were cleaned, conditioned and evacuated before being used for

sampling. A metal bellows pump was used to fill up the canisters with sample air over one-hour integration (with a flow restrictor) to a pressure of 40 psi. The speciation and abundance of 59 C₂-C₁₁ NMHCs in the canisters were determined by a Model 7100 preconcentrator (Entech Instruments Inc., California, USA) coupled with an Agilent 5973N gas chromatography-mass selective detector/flame ionization detector (GC-MSD/FID, Agilent Technologies, USA). The detection limit of NMHCs was 3 pptv with a measurement precision of 2-5%, and a measurement accuracy of 5%. Detailed information of the analysis system, and quality control and quality assurance for VOC samples can be found elsewhere (Simpson *et al.*, 2010).

Carbonyl samples were collected using silica gel filled cartridges impregnated with acidified 2,4-dinitrophenylhydrazine (DNPH). An O₃ scrubber was connected to the inlet of the DNPH-silica gel cartridge to prevent interference from O₃. Air samples were drawn through the O₃ scrubber and cartridge for two hours. Before and after each sampling, the flow rate through the cartridges was measured with a rotameter to keep the range of 0.4–0.6 L min⁻¹. All cartridges were stored in a refrigerator at 4 °C after sampling. The sampled carbonyl cartridges were eluted slowly with <5 ml of acetonitrile in the direction opposite to sampling flow into a 5-ml brown volumetric flask, followed by adding acetonitrile to a constant volume of 5 ml. A 20-μl aliquot was injected into the high performance liquid chromatography (HPLC) system through an auto-sampler. Typically, C₁–C₉ carbonyl compounds were measured efficiently with a detection limit of ~0.2 ppbv.

2.3 Photochemical trajectory model (PTM)

2.3.1 General description

In this study, the photochemical O₃ formation processes during the transport of air masses from TC to WSI site were simulated by running a photochemical trajectory model (PTM)

incorporating the MCM. Briefly, the PTM calculates chemical reactions within a well-mixed boundary layer air parcel. The air parcel moves through the pre-defined trajectory and sequentially picks up emissions from each located grid cell of anthropogenic and biogenic VOCs, CO, NO_x and SO₂, and experiences photochemical and depositional processes. In addition, the observed values of meteorological parameters (temperature and relative humidity) were used to constrain the model (Guo *et al.*, 2013; Ling *et al.*, 2014; Lyu *et al.*, 2015a, b). For each trajectory simulation, the model was run for a 13-hr period between 0700 and 1900LT (local time, LT) to calculate the daytime photochemical O₃ formation according to the MCM mechanism.

2.3.2 Master chemical mechanism

The MCM is a near-explicit chemical mechanism, involving 5,900 chemical species and around 16,500 reactions. Previous studies demonstrate that the MCM performs well in calculating the O₃ production (Jenkin *et al.*, 1997, 2003; Saunders *et al.*, 2003). Facsimile software (Curtis and Sweetenham, 1987) is used to integrate the model system of differential equations with integration output set each hour. The model output simulates mixing ratios of O₃, O₃ precursors, radicals (*i.e.* OH, HO₂, RO and RO₂), and intermediates. The initials of O₃ and its major precursors were set up as follows according to the observations during the sampling period and previous studies: O₃ (40 ppbv), NO (10 ppbv), NO₂ (20 ppbv), CO (600 ppbv), methane (1.79 ppmv), ethane (1.5 ppbv), propane (2.0 ppbv), ethene (2.0 ppbv), propene (0.6 ppbv), ethyne (4.5 ppbv), toluene (3.0 ppbv), *m,p*-xylene (1.5 ppbv), formaldehyde (5.0 ppbv), and acetaldehyde (2.0 ppbv) (Guo *et al.*, 2004, 2007; Barletta *et al.*, 2005; Wang *et al.*, 2008; Cheng *et al.*, 2010).

2.3.3 Emission inventory

The model simulation was based on the combination of the 2006-based PRD emission inventory (3×3 km) (Zheng *et al.*, 2009) and an emission inventory for Hong Kong compiled by Hong Kong Environmental Protection Department (HKEPD). The inventories included six sources, namely gasoline vehicle exhaust, diesel vehicle exhaust, solvent usage, gasoline evaporation and LPG usage, biomass and coal burning, and biogenic emissions, and all the sources had speciated VOC profiles containing 92 VOCs (Liu *et al.*, 2008; Zheng *et al.*, 2009). According to Zheng *et al.* (2009), the speciated VOC source profiles in the emission inventories were mainly established on the basis of measurements of 158 samples collected from the emission sources by Liu *et al.* (2008). The temporal and diurnal variations of emission profiles for on-road mobile sources can be found in Zheng *et al.* (2009), while for biogenic source they were given by Zheng *et al.* (2010) and Situ *et al.* (2010), respectively. Since only 59 VOC species were measured in this study, the summed emission of each of the 59 VOC species from the six sources was individually input into the PTM-MCM for model simulations.

2.3.4 Settings of physical modules

Apart from the module of chemical reactions, photolysis rate, dry deposition and the boundary layer height are all considered in the PTM. The photolysis rates of different chemicals in the model are parameterized using the photon flux determined from the Tropospheric Ultraviolet and Visible Radiation (TUVv5) model (Lam *et al.*, 2013), while the dry deposition module includes the dry deposition of air pollutants, which are parameterized as an average deposition rate within the mixed layer height (MLH). The MLH is assumed to vary from 300 m at night to 1400 m during the daytime (Lam *et al.*, 2013). More detailed descriptions of the model are provided in Cheng *et al.* (2010), Guo *et al.* (2013) and Ling *et al.* (2014). The transport process within the PTM is illustrated using backward trajectory

analysis. Respective backward trajectories arriving at WSI were calculated using NOAA-HYSPLIT4.8 (Draxler and Rolph, 2012) for 1-hr intervals to the end point of 200 m above sea level (Guo *et al.*, 2009; Cheng *et al.*, 2010). The horizontal dimensions of the air parcel were 3×3 km, which is consistent with the resolution of emission inventory.

2.3.5 Model performance

Similar to previous modeling studies (Wang *et al.*, 2015; Lyu *et al.*, 2015a, b, 2016; Wang *et al.*, 2017b), the index of agreement (IOA) was applied to evaluate the model performance. The IOA is defined as follows (Equation 1) (Huang *et al.*, 2005; Wang *et al.*, 2015; Lyu *et al.*, 2015a, b).

$$IOA = 1 - \frac{\sum_{i=1}^n (O_i - S_i)^2}{\sum_{i=1}^n ((O_i - \bar{O})^2 + (S_i - \bar{O})^2)} \quad (\text{Eq. 1})$$

where S_i and O_i represent simulated and observed values, respectively; \bar{O} represents the mean of observed values, and n is the number of samples. The IOA value lies between 0 and 1. The better agreement between simulated results and observed data is, the higher the IOA value is (Huang *et al.*, 2005).

3 Results and discussion

3.1 General features of O₃ episodes over the SCS

An overview of this sampling campaign (August–November 2013) was given by Wang *et al.* (2018a). This study mainly focused on the O₃ episodes at both TC and WSI, when continental high pressure dominated and north winds transported polluted continental air masses to maritime atmosphere (Wang *et al.*, 2018a, b). Figure 2(a) displays daily average backward trajectories of air masses on 8 O₃ episode days (red lines) and 13 non-episodes (blue lines) at WSI during the sampling campaign when VOCs samples were measured. Clearly, on all O₃ episode days at WSI, the air masses originated from the inland. Moreover, the hourly

backward trajectory in Figure 2 (b) shows that the air masses passed through the TC site and then arrived at WSI, resulting in the concurrent O₃ episodes at both sites. Since 17 of 21 O₃ episode days (at WSI) intensively occurred in October during the sampling period, mainly due to the impact of two continental anticyclones (Wang *et al.*, 2018a), the days when trace gases and VOCs were both monitored in this month were selected for further investigation. Table 2 shows the daytime (0700-1900 LT) average levels of pollutants during the O₃ episodes (2-4, 22-25 October) and non-episodes (9-10, 17, 30-31 October) at TC and WSI. The TVOCs was defined as the sum of 73 measured VOC species (Table S1). Compared to non-O₃ episodes, higher levels of O₃ and its precursors (*i.e.* NO₂, CO, SO₂ and TVOC) were found during O₃ episodes. The maximum O₃ was around 110 ppbv and 130 ppbv at TC and WSI, respectively, indicating severe O₃ pollution in October. Moreover, the elevated anthropogenic precursors at WSI during episode events also implied that the invaded continental precursors profoundly changed the atmospheric composition and influenced the secondary O₃ formation over the coastal marine areas (Wang *et al.*, 2018a). As such, the PTM-MCM was applied to simulate O₃ formation during the transport of air masses from TC to WSI in the next section.

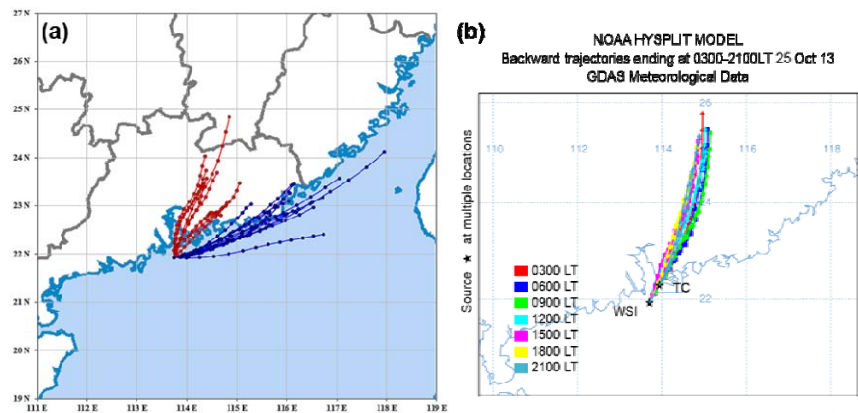


Figure 2. (a) Daily average backward trajectories of air masses on 8 O₃ episode days (red lines) and 13 non-episodes (blue lines) at WSI during the VOC sampling days; (b) Hourly

backward trajectories of air masses arriving at WSI on 25 October 2013. The backward hour of each trajectory in (a) and (b) are 12 hr and 24 hr, respectively. The hourly trajectories of other VOC sampling days in October are given in Figure S1.

Table 2. Descriptive statistics of air pollutants on O₃ episode and non-O₃ episode days in October 2013 at TC and WSI sites (Average \pm 95% C.I.; unit: ppbv; C.I.: Confidence Interval). Max. O₃ refers to the average of daily maximum O₃. The average values were calculated during daytime hours (0700-1900LT).

	TC		WSI	
	O ₃ episode	Non-O ₃ episode	O ₃ episode	Non-O ₃ episode
Max. O ₃	112.8 \pm 9.7	61.8 \pm 6.8	128.6 \pm 15.7	75.7 \pm 5.5
O ₃	62.5 \pm 4.3	44.1 \pm 2.9	88.1 \pm 4.3	63.4 \pm 2.4
CO	691 \pm 14.2	500 \pm 20	408 \pm 12.3	210 \pm 10
NO	9.5 \pm 1.4	7.6 \pm 1.2	0.6 \pm 0.1	0.5 \pm 0.2
NO ₂	37.1 \pm 1.7	24.2 \pm 1.5	7.1 \pm 0.6	1.3 \pm 0.4
SO ₂	9.7 \pm 0.4	6.6 \pm 0.2	4.7 \pm 0.2	2.2 \pm 0.1
TVOCs	19.4 \pm 1.7	11.7 \pm 1.9	17.6 \pm 1.4	8.1 \pm 2

3.2 Model construction and comparison with observations

Prior to the study of O₃ formation and its photochemistry over the SCS, the PTM-MCM model performance was evaluated. Figure 3 compares day-to-day patterns of simulated and observed O₃ on the 12 days at TC and WSI. In general, the model well presented the diurnal variations of O₃ at TC and WSI with the same trends as the observations, and the discrepancy between the simulated and observed values was in the range of \pm 30%. The simulated values were generally lower than the observations. The IOA of the 12-day simulation at TC and WSI was 0.75 and 0.83, respectively, suggesting that the PTM-MCM model, which considered boundary layer trajectories, precursor emissions and chemical processes, provided a

reasonable simulation of O₃ formation in the coastal area. Moreover, the simulated precursors were compared with the observations at TC and WSI on the seven O₃ episode days in October 2013. Figure 4 displays the diurnal variations of simulated and observed NO, NO_x and TVOC at TC and WSI. The simulated NO, NO_x and TVOC values presented the same magnitudes as the corresponding observations, and the diurnal variations were consistent. In addition, the simulated VOC species (*i.e.* ethene, ethane, 1-butene, benzene, toluene, formaldehyde and acetaldehyde) had good agreement with the observed values, which are shown in Figure S2.

The above differences between the observations and simulation results were likely caused by the following reasons: (1) High vertical gradients of NO_x whereas the PTM assumes that pollutants are completely mixed in the vertical direction; (2) The uncertainties in the 2006-based PRD emission inventory, which showed medium to high uncertainty for the NO_x emission and high uncertainty for VOC and CO emissions (Zheng *et al.*, 2009); (3) The same profiles and fractions of VOC emission sources used for both sites, which in fact were different; (4) The uncertainties of speciated VOC source profiles and the initial settings of precursors used in the model, which influenced the O₃ formation; (5) The emission cell extracted by backward trajectory which might vary; and (6) The physical processes, such as horizontal and vertical dispersion, which were not considered in the PTM-MCM.

In summary, the above results indicated that the PTM-MCM was acceptable for the simulations of O₃ and its precursors during the movement of air from TC to WSI. Hence, the process analysis of O₃ photochemistry and the influences of its precursors were further investigated.

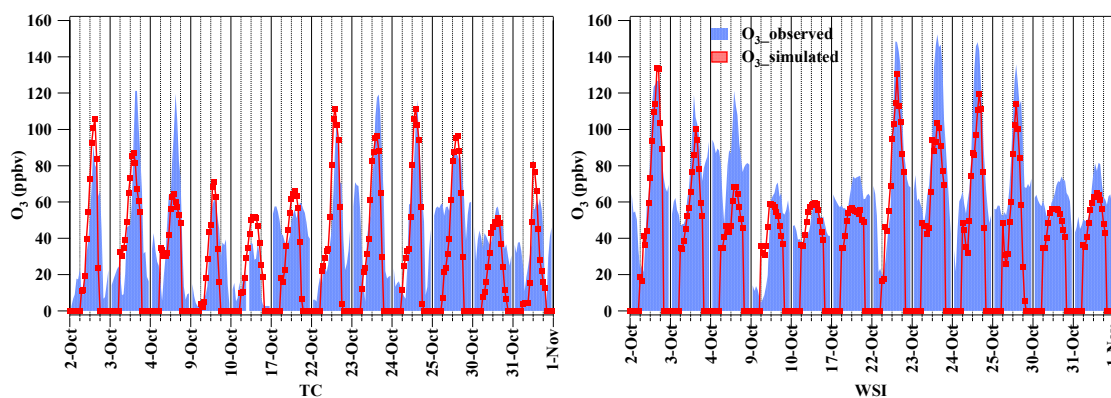


Figure 3. Comparison of observed O_3 (blue shaded areas) with simulated O_3 (red lines) at TC and WSI.

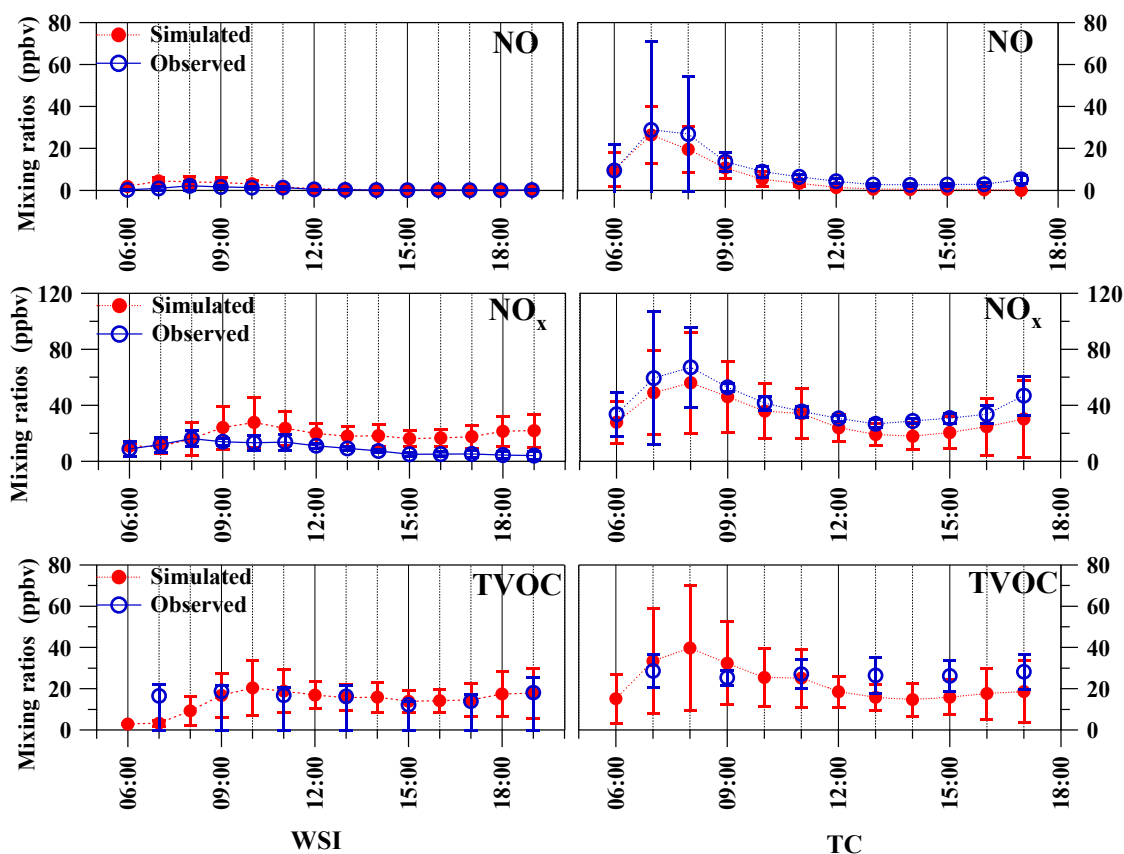


Figure 4. Comparison of observed precursors (blue lines and markers) with simulated precursors (red lines and markers) at TC and WSI.

3.3 Process analysis of O₃ formation from TC to WSI

3.3.1. Temporarily evolutionary variations of O₃ and its precursors

According to the hourly backward trajectories of air masses on each VOC sampling day (Figure S1), 5 days (*i.e.* 2, 22-25 October) when air parcels passed through TC and arrived at WSI were selected for process analysis. Given the 44 km distance between TC and WSI and the daytime average wind speed of 3.9 ± 0.2 m/s on the selected five O₃ episode days, it would take approximately three hours for air parcel at TC to reach WSI. This was consistent with the backward trajectory results, which found that the air parcel from TC to WSI took about three hours (Figure S1).

Figure 5 displays temporarily evolutionary variations of simulated and observed O₃ and its precursors from TC to WSI. Obviously, after the 3-hr transport of air masses from TC to WSI, the simulated precursors (*i.e.* NO, NO₂ and TVOC) decreased intensively in the morning (1000-1200LT), while simulated O₃ increased steadily and the increment sustained high till 1500LT. The maximum decrements of the three simulated precursors occurred simultaneously in the morning, caused by the large variations of mixing ratios at the two sites due to the 3-hr time lag. At TC, the early morning peaks of simulated precursors (0700-0900LT) were consistent with their observations, such as NO (Figure 4), ethene, ethane and other VOCs shown in Figure S2, mainly due to the morning rush-hour emissions and the variation of boundary layers in the morning (So and Wang, 2003; Guo *et al.*, 2013). During the transport processes from TC to WSI, these precursors emitted in early morning at TC experienced the most intensive sunlight and were consumed by the most efficient photochemical reactions during daytime period (Figure 6a, b), resulting in the largest decrements before arriving at WSI at 1000-1200LT. The maximum $\Delta_{\text{precursors}}$ values (differences of NO/NO₂/TVOCs between WSI and TC, with a 3-hr time lag) on these 5 days appeared at 1200LT, 1100LT, 1100LT, 1000LT and 1000LT, respectively. Accordingly, O₃

presented the maximum increments at 1100-1400LT, indicating daytime strongest photochemical O₃ formation. The calculated largest Δ O₃ value (differences of O₃ between WSI and TC, with a 3-hr time lag) was 45.7 ppbv on 22 October, followed by 45.0, 37.5, 36.2 and 34.7 ppbv on 2, 24, 25 and 23 October, respectively. The Δ O₃ began to decrease in early afternoon and turned to be negative around 1600-1800LT. The averaged negative Δ O₃ value (-19.7 ± 8.9 ppbv) was much less than the averaged positive Δ O₃ (39.8 ± 5.6 ppbv) before 1600LT ($p < 0.05$), implying the net O₃ production in these five days. Indeed, the daytime averaged Δ O₃ (0900-1800LT) was 22.7 ± 8.1 ppbv, which was comparable to the difference of observed O₃ between TC and WSI (26.5 ± 3.1 ppbv, $p > 0.05$), suggesting the contribution of continental precursors to O₃ formation (more than 20 ppbv on average in these 5 days) over the SCS.

To further reveal the changes of O₃ and its precursors from TC to WSI, the temporarily evolutionary variations of radicals and photochemical pathways during the 3-hr transport were illustrated in the next section.

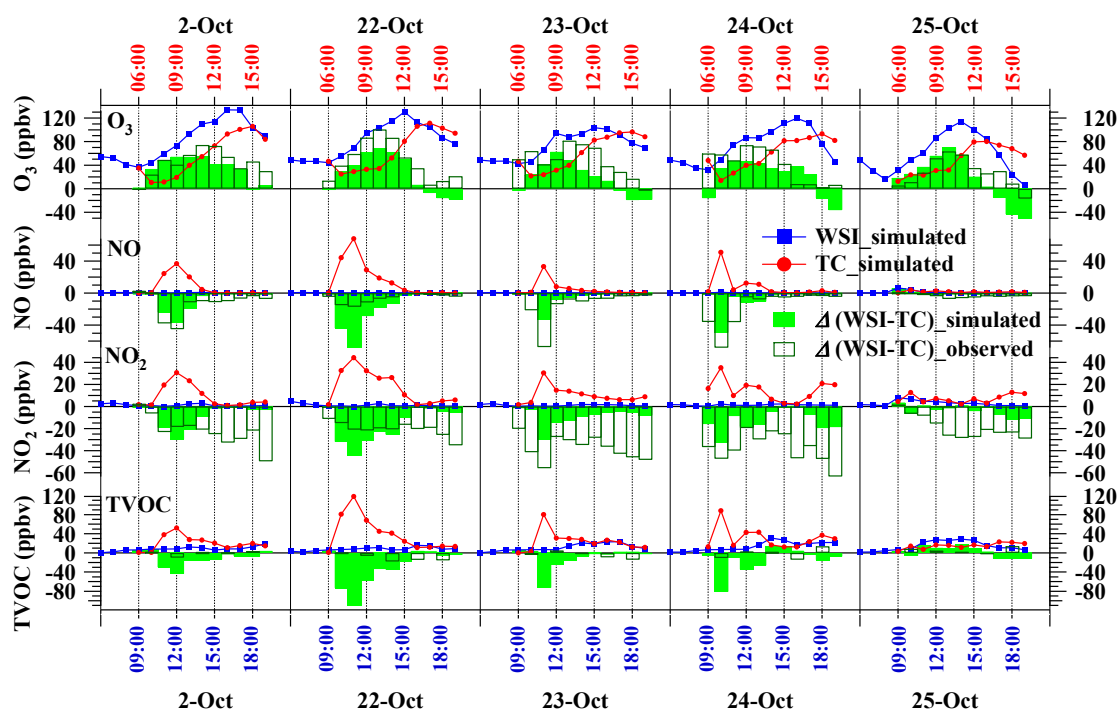


Figure 5. Temporally evolutionary variations of simulated O₃ and its precursors (*i.e.* NO,

NO₂ and TVOC) from TC (red lines and markers) to WSI (blue lines and markers) on 2, 22-25 October 2013. The solid green bars represent the differences between mixing ratios of simulated air pollutants at TC and those after 3 hours at WSI, while the unfilled green bars are the observed differences. Note that the simulated values at TC and WSI correspond to the top (red) and bottom (blue) axis, respectively, and the time lag is 3 hours.

3.3.2. Temporarily evolutionary variations of radicals and photochemical pathways

Figure 6(a) presents the temporarily evolutionary variations of simulated OH and HO₂ radicals from TC to WSI on the five episode days. Unlike the process analysis of O₃ and its precursors from TC to WSI, radicals cannot be transported due to their short lifetime (<1 minute). During the process from TC to WSI, both Δ_{OH} and Δ_{HO_2} (differences of radicals between WSI and TC, with a 3-hr time lag) were positive before 1500LT on these five days, with an average increment of $7.6 \pm 1.7 \times 10^6$ molecules cm⁻³ and $3.7 \pm 9.1 \times 10^7$ molecules cm⁻³, respectively, whereas Δ_{OH} and Δ_{HO_2} became negative after 1500LT, with values of $-4.8 \pm 1.3 \times 10^6$ molecules cm⁻³ and $-5.9 \pm 4.4 \times 10^7$ molecules cm⁻³, respectively. Clearly, both Δ_{OH} and Δ_{HO_2} were overall positive, indicating that the radicals over the SCS were higher than those on the continent. In addition, the time when maximum OH radicals and maximum photochemical reaction rates appeared was consistent with the maximum decrements of the three precursors (Figure 5).

Figure 6(b) illustrates the evolutionary variations of photochemical pathways during the 3-hr transport from TC to WSI. The pathways of VOC+OH initialized the photochemical reactions. Subsequently, HO₂ and RO₂ radicals were generated. The reactions of HO₂+NO and RO₂+NO competed with O₃+NO, resulting in the preservation of O₃. Overall, the temporarily evolutionary variations of photochemical pathways from TC to WSI were similar

377 to those of radicals; Namely, photochemical O₃ production was accelerated in the morning
 378 and at noon, with positive values of averaged $\Delta_{\text{VOC}+\text{OH}}$, $\Delta_{\text{HO}_2+\text{NO}}$ and $\Delta_{\text{RO}_2+\text{NO}}$ (6.0 ± 1.5 ,
 379 5.9 ± 1.5 and $7.8\pm2.0 \times 10^7$ molecules cm⁻³, respectively), but decelerated after 1500LT with
 380 negative values of -2.7 ± 0.8 , -3.0 ± 0.7 and $-4.2\pm1.0 \times 10^7$ molecules cm⁻³, respectively. Since
 381 the acceleration rates were much higher than the decelerate rates ($p<0.05$), the overall
 382 photochemical O₃ production during the daytime hours increased from TC to WSI, causing
 383 higher O₃ levels at WSI.

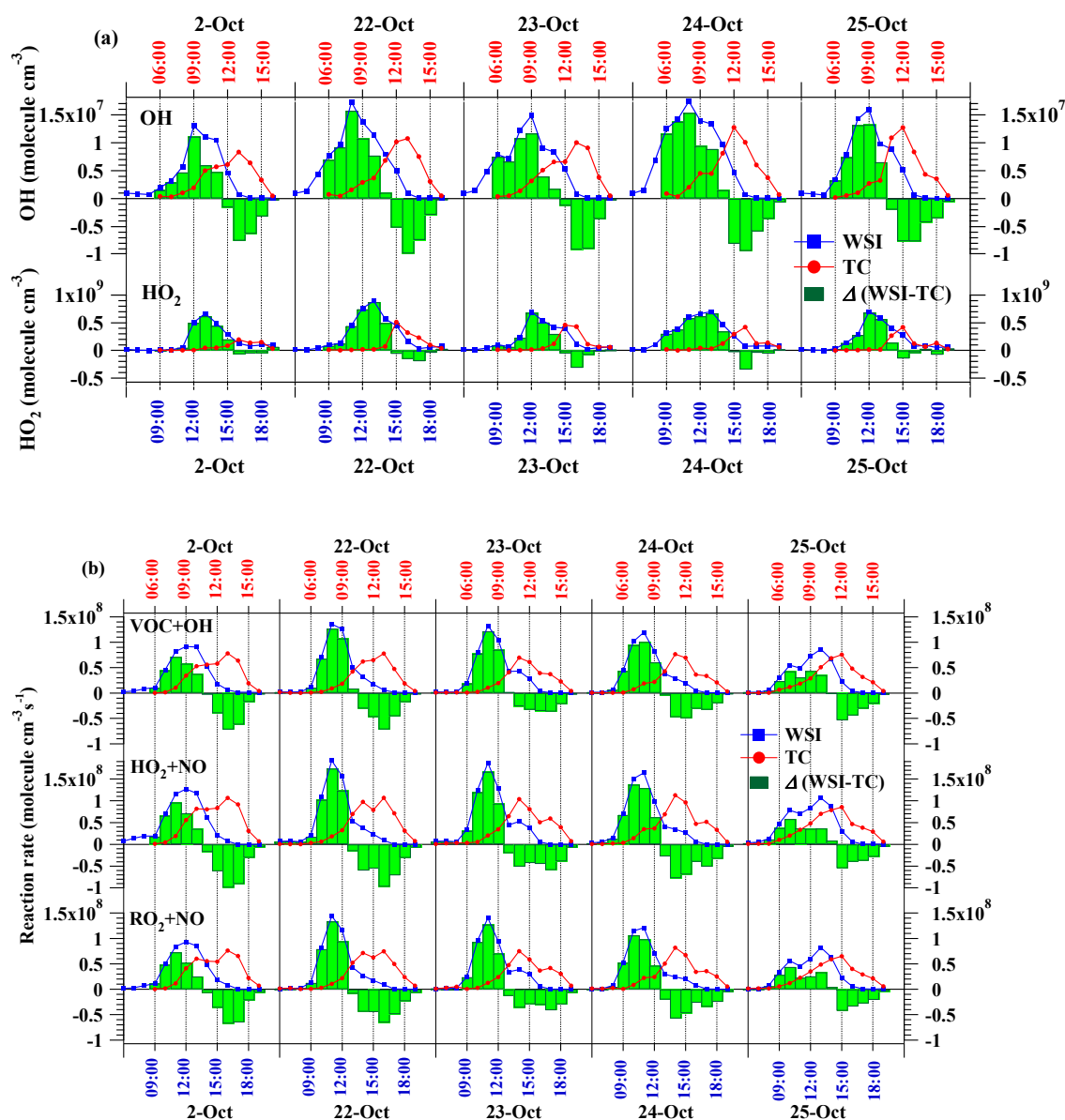


Figure 6. Temporarily evolutionary variations of simulated (a) OH and HO₂ radicals, and (b) major photochemical pathways (*i.e.* VOC+OH, HO₂+NO and RO₂+NO) from TC (red lines and markers) to WSI (blue lines and markers) on 2, 22-25 October 2013. Note that the simulated values at TC and WSI correspond to the top (red) and bottom (blue) hours. The green bars represent the differences between levels of radicals and photochemical formation pathways at TC and those after 3 hours at WSI.

3.3.3. Spatial evolutions of radicals and O₃ photochemistry

Figure 7 shows the spatial evolutions of the average diurnal variations of the simulated OH and HO₂ from TC to WSI. To display the evolutionary processes between the two sites, the diurnal patterns of radicals at 15 km and 30 km (from the TC site) are presented as well. The highest concentration of OH was $9.7 \pm 2.5 \times 10^6$ molecules cm⁻³ at TC. After the air mass left the TC site towards the WSI site, the highest OH concentration changed to $1.2 \pm 0.3 \times 10^7$ at the location of 15 km and $1.3 \pm 0.2 \times 10^7$ at the location of 30 km to TC. Eventually the highest OH level reached $1.4 \pm 0.1 \times 10^7$ molecules cm⁻³ at WSI, about 1.5 times that at TC ($p < 0.05$). Similarly, the highest HO₂ increased from $3.5 \pm 1.3 \times 10^8$ molecules cm⁻³ at TC to $6.8 \pm 1.1 \times 10^8$ molecules cm⁻³ at WSI ($p < 0.05$). The increase of OH and HO₂ levels from TC to WSI was mainly due to the consumption (*i.e.* the reaction of HO₂+NO) and dilution of NO_x mixing ratios from the continental site to the SCS, which regulated the temporal variations of radicals at different locations. The increased gradients of radicals from TC to WSI reflected that the atmospheric oxidative capacity over the SCS was greater than that on the continent, leading to more intensive O₃ production in the maritime environment.

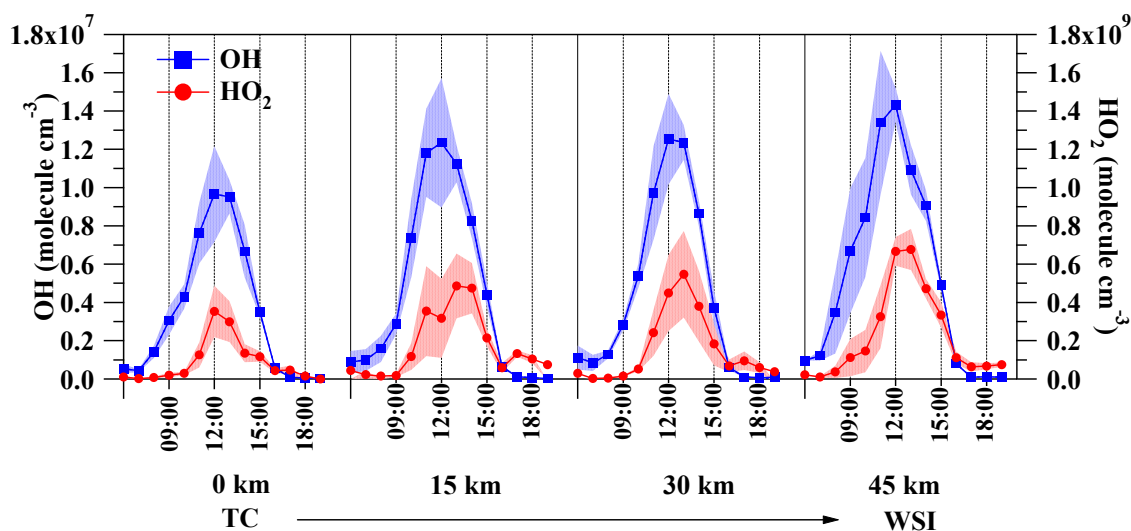
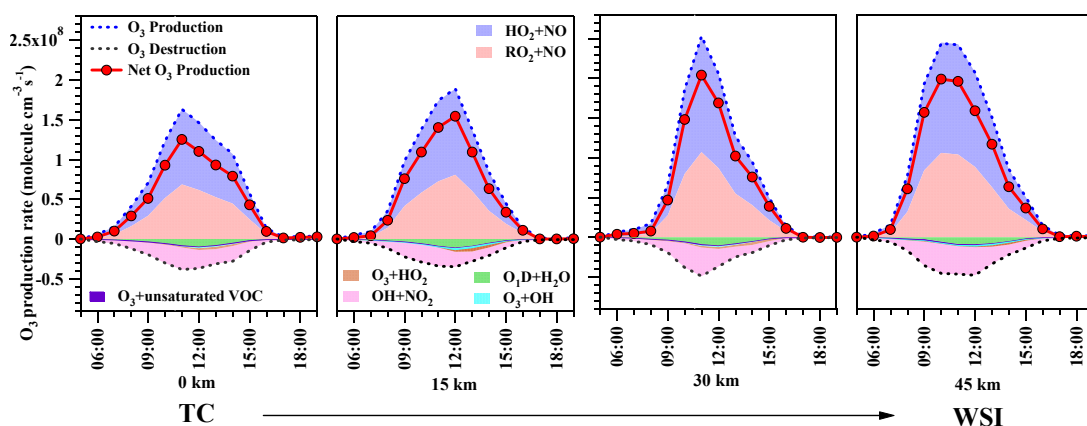


Figure 7. Spatial evolutions of the average diurnal variations of the simulated OH and HO₂ radicals from TC to WSI. The shaded area indicates the 95% Confidence Interval of the averages.

Figure 8 depicts the spatial evolutions of the simulated diurnal profiles of O₃ production and destruction rates from TC to WSI. Overall, with the distance farther away from the TC site, the reaction rates of O₃ production/destruction were gradually faster. Consequently, the averaged maximum net O₃ production rate at WSI was $2.0 \pm 0.5 \times 10^8$ molecules cm⁻³, much larger than that at TC ($1.3 \pm 0.2 \times 10^8$ molecules cm⁻³, $p < 0.05$). The increase of net O₃ production rate from TC to WSI was mainly due to the enhancement of RO₂/HO₂ radicals, which accelerated the photochemical cyclings and resulted in more O₃ production. Specifically, O₃ production was dominated by the reaction of HO₂ with NO, with an average contribution of 58.1% and 57.5% at TC and WSI, respectively, while the NO oxidation by RO₂ accounted for 41.9% and 42.5%, respectively. In contrast, the reactions of OH+NO₂ and O₃ photolysis were the major contributors to O₃ destruction at TC with an average contribution of 69.9% and 18.2%, respectively, while their contributions at WSI were 66.1% and 20.1%, respectively. The contribution changes of the O₃ destruction pathways were

424 mainly attributed to higher NO₂ level at TC, which facilitated the production of HNO₃ via the
 425 terminal reaction of OH and NO₂.



426
 427
 428 Figure 8. Spatial evolutions of simulated O₃ production and destruction rates from TC to WSI.

429 3.4 Contributions of continental VOC sources to maritime O₃ formation

430 Figure 9 shows the speciated profiles of six major VOC sources used in the 2006 emission
 431 inventory (Liu *et al.* 2008). As described in section 2.3.3, the six sources were gasoline
 432 vehicle exhaust, diesel vehicle exhaust, solvent usage, gasoline evaporation and LPG usage,
 433 biomass and coal burning, and biogenic emissions, which contributed 38.4%, 3.2%, 26.8%,
 434 3.3%, 2.5% and 25.8% to total VOC emissions, respectively (Liu *et al.*, 2008; Zheng *et al.*,
 435 2009).

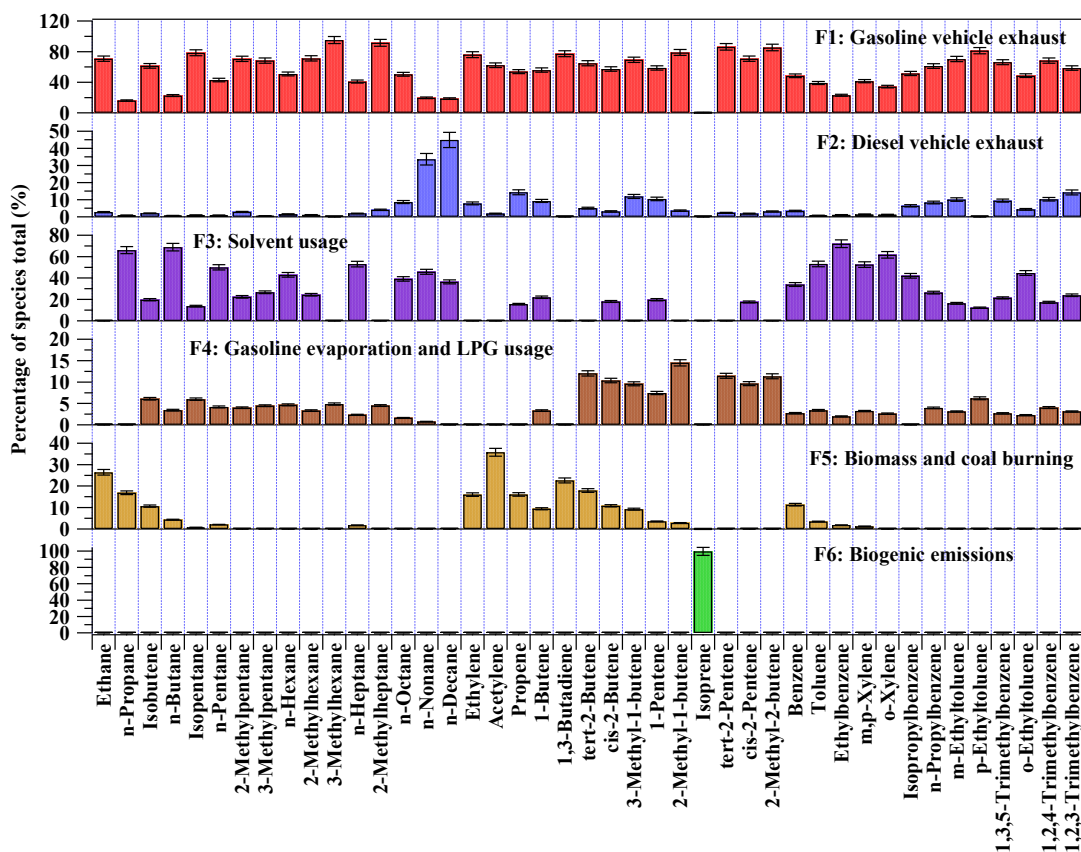


Figure 9. Speciated profiles of the six major VOC sources in the emission inventory, shown in the percentage of species total. Error bars indicate the 95% Confidence Interval of the averaged percentages.

Since O_3 formation was VOC-limited over the SCS on episode days (Wang *et al.*, 2018b), to evaluate the contribution of VOC sources to O_3 formation, two scenarios were designed for data analysis. The first scenario was “base case”, which used original emission inventory as input. The second scenario was “constraint case”, which excluded each of the six sources in turns from the emission inventory in the “base case”. The excluded proportions of each VOC source in the six sources were based on the source profiles in Figure 9. Hence, the contribution of each VOC source was obtained from the difference of simulated O_3 between the “base case” and the corresponding “constraint case”. Figure 10 shows the contributions of the six continental VOC sources to photochemical O_3 production at TC and WSI. It was

found that the gasoline vehicle exhaust, solvent usage, and biogenic emissions made large contributions to O₃ formation at TC and WSI, which contributed 8.7±1.7, 6.4±1.2 and 6.1±0.8 ppbv at TC, and 13.9±2.7, 10.2±2.0 and 6.2±0.9 ppbv at WSI, respectively. The contributions of diesel vehicular exhaust, biomass and coal burning, and gasoline evaporation and LPG usage were insignificant at TC and WSI (<2 ppbv).

Since the simulated maritime O₃ formation during the process from TC to WSI was the accumulatively formed O₃ along the upwind locations including TC and the in-situ formed O₃ from the residuals of continental precursors arriving at WSI, the difference of O₃ between WSI and TC was the net contribution of each continental VOC source to maritime O₃ formation. The estimated contributions of gasoline vehicle exhaust and solvent usage emitted at TC to O₃ at WSI were about 5.2 ppbv and 3.8 ppbv, respectively, which dominated the contributions of continental sources to maritime O₃ formation. The large contributions of the aforementioned two sources were mostly attributed to the high reactivity and large abundance of aromatics (above 70%, shown in Figure 9). Moreover, the contributions of alkenes in vehicle exhausts to O₃ formation were not insignificant due to their high reactivity (Atkinson and Arey, 2003). Previous studies already revealed the significant contributions of VOC emissions from gasoline vehicle exhaust and solvent-related sources to O₃ formation in the PRD region (Liu *et al.*, 2008; Guo *et al.*, 2011; Ou *et al.*, 2015; Lyu *et al.*, 2017; Wang *et al.*, 2017b). This study for the first time demonstrated the significant influence of these two sources on the maritime O₃ formation over the SCS. In contrast, because isoprene, the major component of biogenic emissions, has the highest reactivity among VOCs with short lifetime (<2 hours), it was consumed rapidly to form O₃. As such, the continentally-emitted isoprene at TC and upwind locations would be run out before arriving at the maritime WSI site, leading to negligible contribution of continental biogenic emissions to maritime O₃ formation. It is noteworthy that the PTM-MCM model simulation might overestimate the contribution of

VOC sources to O₃ formation due to the use of the same profiles and fractions of VOC emission sources (extracted from the 2006 emission inventory) at the two sites with very different land-use functions (Figure S3).

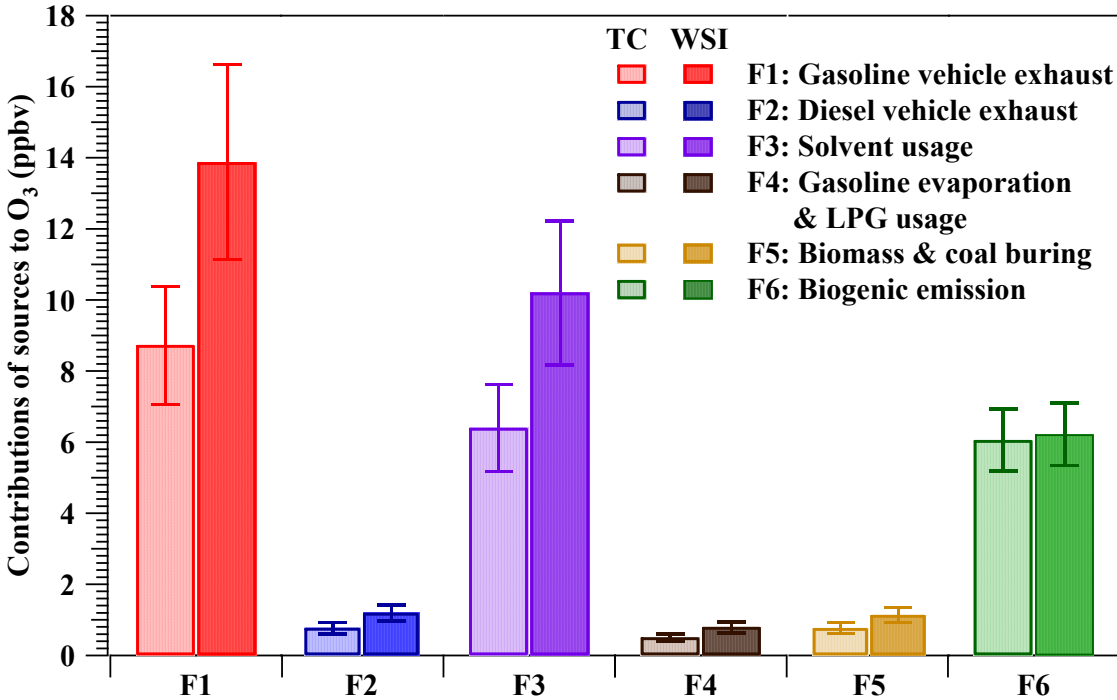


Figure 10. Contributions of six VOC sources to O₃ formation at TC (light-colored bar) and WSI (dark-colored bar). Error bars are the 95% Confidence Intervals of the overall averaged contributions.

4 Conclusions

In this study, a PTM model coupled with a near-explicit chemical mechanism was applied to simulate the O₃ formation during the process of transport from a continental site to a maritime site, and to assess the impact of continental VOC sources on O₃ formation over the SCS. According to the backward trajectory analysis, five O₃ episode days at WSI (*i.e.* 2, 22-25 October, 2013), when air masses passed by both TC site of Hong Kong and marine receptor WSI site, were selected to simulate the photochemical processes with the application of PTM. The simulated O₃ and its precursors (*i.e.* NO, NO₂ and VOCs) showed a reasonable

agreement with the observed ones at TC and WSI, indicating that simulated O₃ formation processes from TC to WSI were acceptable. The process analysis of simulated O₃ formation from TC to WSI showed that (1) the precursors decreased significantly in the morning through photochemical consumption and atmospheric dilution; (2) radicals (*i.e.* OH and HO₂) increased and photochemical reactions were accelerated from TC to WSI; and (3) net O₃ production increased from TC to WSI and the daytime average O₃ increment was about 22.7 ppbv from TC to WSI. In addition, based on the source profiles of the emission inventory, the contributions of six sources, namely gasoline vehicle exhaust, diesel vehicle exhaust, gasoline evaporation and LPG usage, solvent usage, biomass and coal burning, and biogenic emissions, to maritime O₃ formation were evaluated. The results for the first time proved that gasoline vehicle exhaust and solvent usage, which contained relatively high reactive aromatic VOC species, largely contributed to O₃ formation over the SCS (~5.2 and 3.8 ppbv, respectively). This study provides insightful method to assess the photochemical evolution of air masses from the continent to the maritime environment, and the impact of continental VOC sources on maritime O₃ formation over the SCS. The findings are helpful to formulate and implement appropriate O₃ reduction strategies.

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