

Sintering temperature dependence of thermoelectric performance in CuCrSe_2 prepared via mechanical alloying

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Abstract:

CuCrSe_2 compounds were synthesized by mechanical alloying followed by annealing and spark plasma sintering at temperatures of 873K, 923K, 973K and 1023K. Our investigation revealed that high spark plasma sintering temperature (above 973K) is the key to obtaining CuCrSe_2 single phase completely free of a metallic secondary phase CuCr_2Se_4 . The ultralow thermal conductivity caused by the Cu ions disorder, complex layered structure and dislocation leads to the favorable thermoelectric performance in CuCrSe_2 compound. A maximum dimensionless figure of merit ZT value over 0.75@873 K is achieved in the absolutely pure sample sintered at 1023K.

Keywords: CuCrSe_2 , single phase, layered structure, PLEC

With the rapid development of society, growing energy consumption has led to the emission of more greenhouse gas and pollutants [1-2]. To resolve the environmental issues, the Chinese government has issued new green energy policies to encourage more efficient utilization of energy. Thermoelectricity, which can directly convert heat into electricity, opens a promising path for recovering the lost energy in industrial and automobile waste heat [3-5]. The efficiency of thermoelectric conversion is defined by a dimensionless figure of merit, $ZT = S^2 \sigma T / \kappa$, where S is the Seebeck coefficient, σ is the electrical conductivity, T is the absolute temperature, and κ is the total thermal conductivity. The total thermal conductivity can be further decomposed to contributions from the lattice part (κ_L) and the electronic part (κ_e), where for metals and degenerate semiconductors $\kappa_e = L \sigma T$ (L is Lorenz number). A good thermoelectric material should have both a high power factor (PF) σS^2 and low thermal conductivity.

Although the expression of ZT is simple, S , σ and κ_e are all functions of the carrier concentration, and hence are strongly coupled with each other [6-7]. Virtually none of these properties can be independently tuned for achieving a high ZT , but the lattice thermal conductivity κ_L is the only term that is governed by a material's structural features. It has been reported that many state-of-the-art thermoelectric materials feature complex structures, such as Cu-based compounds [8-9], SnSe [10-12], In₄Se₃ [13], where strong intrinsic phonon scattering gives rise to ultralow thermal conductivity and hence high ZT values.

These materials inspired a novel concept in the thermoelectric known as phonon glass and electron crystal (PGEC), which means good thermoelectric materials should

possess low κ like glassy materials but high σ (low ρ) like crystalline semiconductors [14]. Since then, many PGEC have been extensively studied by researchers such as: CoSbS [15], Cu₂Se [16], AgCrSe₂ [17], and CuCrSe₂ [18-19] etc. Recently, CuCrSe₂, an ionic-electronic conductor, has regained popularity for its phonon-liquid electron-crystal (PLEC) transport behaviors. CuCrSe₂ features a layered structure composed by CdI₂ type CrSe₂⁻, in which the Cr³⁺ cations are surrounded by the Se²⁻ anions that form a distorted octahedral coordination sphere. Between the CrSe₂⁻ layers, the interlayer space is occupied by Cu ions. The Cu ions are disordered in this structure, occupying two different tetrahedral sites. At low temperatures Cu ions only occupy one of the tetrahedral sites while Cu ions begin to migrate to other tetrahedral site at a transition temperature round 365 K [18-19]. The strong disorder of Cu ions in the CuCrSe₂ is expected to result in low thermal conductivity like that of glassy materials. At room temperature, CuCrSe₂ has a band gap $E_g = 0.157\text{eV}$ [19], which conforms to the selection rule of having a narrow band gap for good thermoelectric materials. Recently, S. Bhattacharya *et al* synthesized CuCrSe₂ via solid state reaction and reported a peak ZT of 0.25 at 773K [18]; Meanwhile, it is worth to mention that there is no sign of anisotropy in CuCrSe₂ compound with layered structure [19].

In spite of its promising attributes for TE applications, it is extremely difficult to obtain a pure phase of CuCrSe₂ due to its complex phase transitions during solidification. As shown in the phase diagram of Fig.S1, the CuCrSe₂ prepared by conventional vacuum melting and solidification contains the second phase of CuCr₂Se₄. To overcome the synthesis challenge, we resorted to mechanical alloying (MA) [20-22] followed by

vacuum annealing. After that, CuCrSe₂ ingot was pulverized and sintered into bulk samples by SPS. By properly controlling the SPS sintering temperature, we have successfully obtained totally single phase CuCrSe₂ that possesses thermal conductivity lower than that reported in previous works [18-19]. We believe the ultralow thermal conductivity of these SPS samples originates from the intrinsic layered structure combined with Cu ions disorder and dislocations within a crystalline solid.

High-purity elemental powders of Cu(99.99%), Cr(99.9%) and Se(99.99%) were weighed according to the stoichiometric ratio. All constituents were loaded into a planetary ball mill (*Pulverisette 7*) placed in the glove box. Tungsten carbide vessel and corresponding balls were used, and the weight ratio of ball to powder was kept at 10:1. The powders were mechanically alloyed for 8h at 500 rpm, which was found to be the optimum MA condition for the synthesis and thermoelectric performance of CuCrSe₂. The as-milled powders were sealed in quartz ampoules under vacuum of $\sim 7.8 \times 10^{-4}$ mbar, slowly heated to 1173 K in 20 h and kept at such temperature for 72h before cooling down to room temperature. The ingots obtained were re-grounded into fine powder and then sintered into dense pellets by SPS (*SPS-625*) under 50 MPa at varying temperatures (873–1023 K) for 5 min in a graphite die. The densities of final products are greater than 95% of the theoretical density.

The obtained pellets were cut into about 9 mm×3 mm ×3 mm rectangular parallelepipeds for electrical property measurements and round disks of 10 mm in diameter and about 1 mm in thickness for thermal conductivity measurement. Seebeck coefficient and electrical conductivity were measured using LSR-3 (*Linseis, Germany*)

instrument measuring system under a static helium atmosphere. The thermal conductivity is determined by $\kappa = \rho D C_p$, where ρ is the density measured using the mass and geometric volume of the pellet, the thermal diffusivity (D) measured by a Netzsch 457 (Netzsch, Germany) microflash LFA system, and specific heat C_p measured using a Netzsch, 404 F3. As mentioned in the introduction, the lattice thermal conductivity was obtained by $\kappa_L = \kappa - \kappa_e$, where κ_e is estimated by Wiedemann-Franz law $\kappa_e = \sigma L T$. The errors in the measurements of electrical conductivity, Seebeck coefficient, thermal diffusivity, and C_p are about 5%, 5%, 3%, and 3%, respectively. All thermoelectric properties were measured from 300K to 873K. Powder x-ray diffraction (XRD) patterns were collected from powders using a PANalytical X'Pert apparatus with Cu K α radiation. A scanning electron microscope (JSM-7800F, JEOL) was used to image the morphology. The verification of the crystal structure was carried out by transmission electron microscopy (TEM, JEOL 2100F).

Fig .1 shows the XRD patterns of all bulk CuCrSe₂ samples after SPS at different temperatures. All major reflections of the polycrystalline samples are indexed to the CuCrSe₂ phase. A small amount of secondary phase CuCr₂Se₄ was found in samples sintered at 873K and 923K. When the sintering temperature reached 973K and above, the secondary phase disappeared and single phase of CuCrSe₂ was successfully obtained.

Figs. 2a-b display the scanning electron microscope (SEM) images of the cross section of samples sintered at 873K and 1023K. It is the evident that all samples are of intrinsic layered structures, but the bulk polycrystalline samples show no sign of anisotropy [19].

As shown in Figs. 2a-b, it is apparent that the grains are closely packed, which is in agreement with the measured high density of our bulk samples. Figs. 2c-d are transmission electron microscope (TEM) images of CuCrSe_2 compound sintered at 1023K and Figs. 2e-f are for the sample sintered at 873K. In Fig. 2c, one can see that there are many dislocations present in the grains. To illustrate their crystal structures, high-resolution TEM images and selected area electron diffraction (SAED) patterns were taken. Fig. 2d is a typical high-resolution TEM image, indicating the hexagonal crystal structure and lattice spacing of CuCrSe_2 , which is further confirmed by the SAED pattern in the inset. Similar TEM investigations were performed on the sample sintered at 873K, and Fig. 2e shows the bright-field TEM image with clear grain boundaries. Our detailed high-resolution TEM images prove the hexagonal crystal structure and lattice spacing of CuCrSe_2 as well. In addition, a secondary phase of CuCrSe_4 with cubic structure was also founded in this sample, as demonstrated in the Fig. 2f and its inset. This is in excellent consistence with the XRD results.

The electrical conductivity, Seebeck coefficient and the calculated power factors are shown in Figs. 3a-c. Abnormalities around 365 K in the electrical and thermal transport data indicate the onset of the disorder of Cu ions. Below 365K, the sharp decrease of electrical conductivity with increasing temperature is indicative of a typical degenerate semiconductor. As temperature rises above 365K, the electrical conductivity shows a very weak temperature dependence, which resembles the high temperature behavior of an ionic semiconductor such as AgCrSe_2 [17]. The electrical conductivity of samples sintered at 873K and 923K are in agreement with the previously reported values [18-

[19] and much higher than those sintered at 973K and 1023K due to the presence of the metallic secondary phase CuCr_2Se_4 . The reason why samples sintered at 873K and 923K has higher electrical conductivity is attributed to the carrier concentration enhancement. As shown in Table I, the carrier concentration for samples sintered at 873K and 923K are much higher than samples sintered at 973K and 1023K. Although the presence of second phase CuCr_2Se_4 in CuCrSe_2 samples sintered at 873K and 923K may result in the reduction of carrier mobility, the increase of carrier concentration far exceeds the reduction of carrier mobility, yielding the improvement of electrical conductivity in samples sintered at 873K and 923K.

Fig. 3b shows the temperature dependence of Seebeck coefficient. The positive sign of S implies all samples are intrinsic p-type semiconductors. For samples sintered at 873K and 923K, the Seebeck coefficient decreases from room temperature to about 365K and then steadily increases from $80\mu\text{V/K}$ at 365K to $140\mu\text{V/K}$ at 873K. The Seebeck coefficients of samples sintered at 973K and 1023K exhibit a similar trend: decreasing from room temperature and then becoming almost temperature independent. In addition, the value of Seebeck coefficient increases with the decrease of carrier concentration, however, it decreases with the decreases of effective mass for similar carrier concentration as shown in Table I. It is noted that the band gap of CuCrSe_2 is 0.157eV [19], the activation of minority carriers play negative effects on Seebeck coefficient above 600K. This behavior and values of Seebeck coefficient have not been observed in previously published work on CuCrSe_2 [18-19]. Fig. 3c displays the calculated power factors. Because of the reduced electrical conductivity, the peak power factors of pure

single phase CuCrSe₂ samples are lower than those of samples containing the metallic secondary phase.

Fig. 4a displays the temperature dependence of total thermal conductivity κ of bulk CuCrSe₂ samples. From 300 K to 365 K, the reduction of κ is attributed to the disorder of Cu ions. Above 365 K, thermal conductivity becomes nearly temperature independent as amorphous materials. The total thermal conductivity, especially that of samples sintered at 973K and 1023K, is extremely low even compared with other well-known thermoelectric materials with intrinsically low thermal conductivity [8-13]. This could be contributed to the following synergistic phonon scattering mechanisms: (a) dislocations proved by TEM attribute to the extremely low total thermal conductivity; (b) the phonons with a larger mean free path of several mms are scattered by micron-size layered grain structure; (c) the property of a superionic conductor with liquid like substructures (i.e. diffusion of ions within the crystalline sublattice) also attributes to ultralow thermal conductivity by reducing the phonon mean free path and eliminating some of the phonon vibration modes [23]. Besides, the lower values of total thermal conductivity of samples sintered at 973K and 1023K is caused by reduced electrical thermal conductivity (inset). It should be noted that the total thermal conductivity for samples sintered at 873K and 923K shows good consistency with other independently reported values [18-19]. The temperature dependence of lattice thermal conductivity are shown in Fig. 4b which is calculated by subtracting the electronic thermal conductivity from the total thermal conductivity. The Lorenz number L , which depends on the degree of elasticity in carrier scattering, can be calculated by Equation (1):

$$L = \left(\frac{k_B}{e}\right)^2 \left[\frac{\left(r+\frac{7}{2}\right)F_{r+\frac{5}{2}}(\xi)}{\left(r+\frac{3}{2}\right)F_{r+\frac{1}{2}}(\xi)} - \left(\frac{\left(r+\frac{5}{2}\right)F_{r+\frac{3}{2}}(\xi)}{\left(r+\frac{3}{2}\right)F_{r+\frac{1}{2}}(\xi)} \right)^2 \right] \quad (1)$$

where the scattering factor r is the exponent of the energy dependence on the charge carrier mean free path. For acoustic phonon scattering in the CuCrSe₂ system, r takes a value of $-1/2$. $F_n(\xi)$ is the Fermi integral defined as:

$$F_n(\xi) = \int_0^\infty \frac{x^n}{1 + e^{x-\xi}} dx \quad (2)$$

where ξ is the reduced Fermi energy that can be deduced from the Seebeck coefficient and the scattering factor based on the following formula assuming a single parabolic band:

$$S = \pm \frac{k_B}{e} \left(\frac{\left(r+\frac{5}{2}\right)F_{r+\frac{3}{2}}(\xi)}{\left(r+\frac{3}{2}\right)F_{r+\frac{1}{2}}(\xi)} - \xi \right) \quad (3)$$

The lattice component κ_L as a function of temperature is plotted in Fig. 4b and the inset in Fig. 4b shows κ_e of all samples. Overall, the lattice component κ_L for all samples are extremely low due to the complex layered structure combined with PLEC mechanism. In the end, it is worth mentioning that the observed high temperature κ of CuCrSe₂ is comparable to various recently reported ionic semiconductors such as AgCrSe₂ [17], Zn₄Sb₃ [24-26] and Cu₂Se [16], showing advantages of layered structure combined with PLEC mechanism.

Fig. 4d shows the temperature dependence of ZT values of SPS samples. Although the maximum ZT values of sample sintered at 873K and 1023K show little difference, the average ZT value of sample sintered at 1023K is higher. Since the efficiency of thermoelectric device is determined by the average ZT of constituent materials, the SPS

sample prepared at 1023K has an advantage over the one prepared at lower temperatures for waste heat recovery applications. In addition, compared with other ionic semiconductors possessing intrinsic low thermal conductivity [17], CuCrSe₂ shows excellent thermoelectric performance regardless of SPS temperature.

In this study, via mechanical alloying followed by annealing and spark plasma sintering, we have successfully synthesized pure single phase CuCrSe₂ polycrystalline compounds. Our investigation focused on the temperatures dependence of SPS on the thermoelectric performance in CuCrSe₂ based compounds. It is found that high SPS temperature (above 973K) plays a vital role for obtaining pure single phase CuCrSe₂. In addition, disorder of Cu ions and complex layered structure combined with dislocations contribute to the ultralow thermal conductivity while maintaining a reasonable power factor. A peak ZT up to 0.75 @873K was obtained in the completely pure sample sintered at 1023K.

Acknowledgements

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Captions for Figures

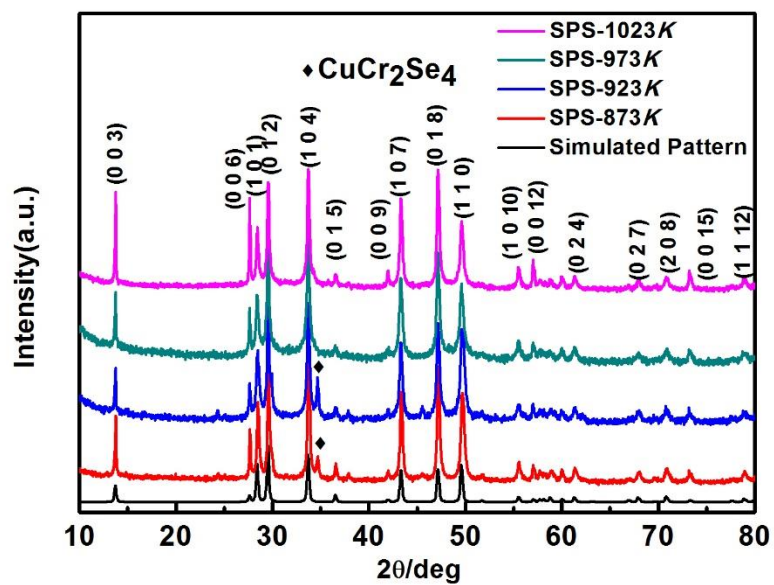


Fig. 1 XRD patterns of CuCrSe_2 after SPS at different temperatures.

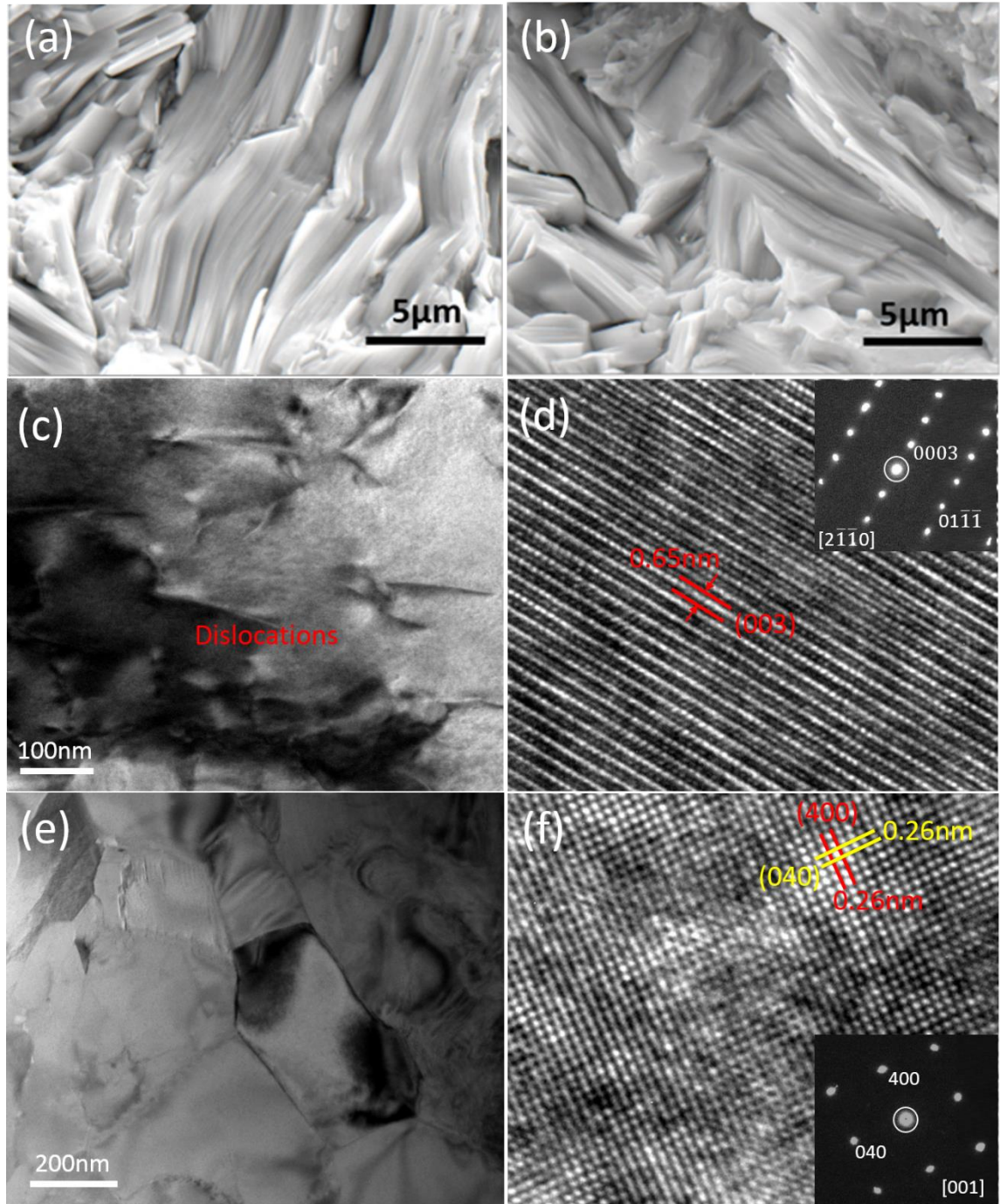


Fig. 2 SEM images of the fracture surface of samples sintered at (a) 873K, (b) 1023K. Transmission electron microscope (TEM) images of samples sintered at 873K and 1023K for bulk CuCrSe₂, images (c) and (d) are for sample SPS at 1023K showing the hexagonal phase, images (e) and (f) are for sample SPS at 873K showing the extra cubic phase in addition to the hexagonal phase.

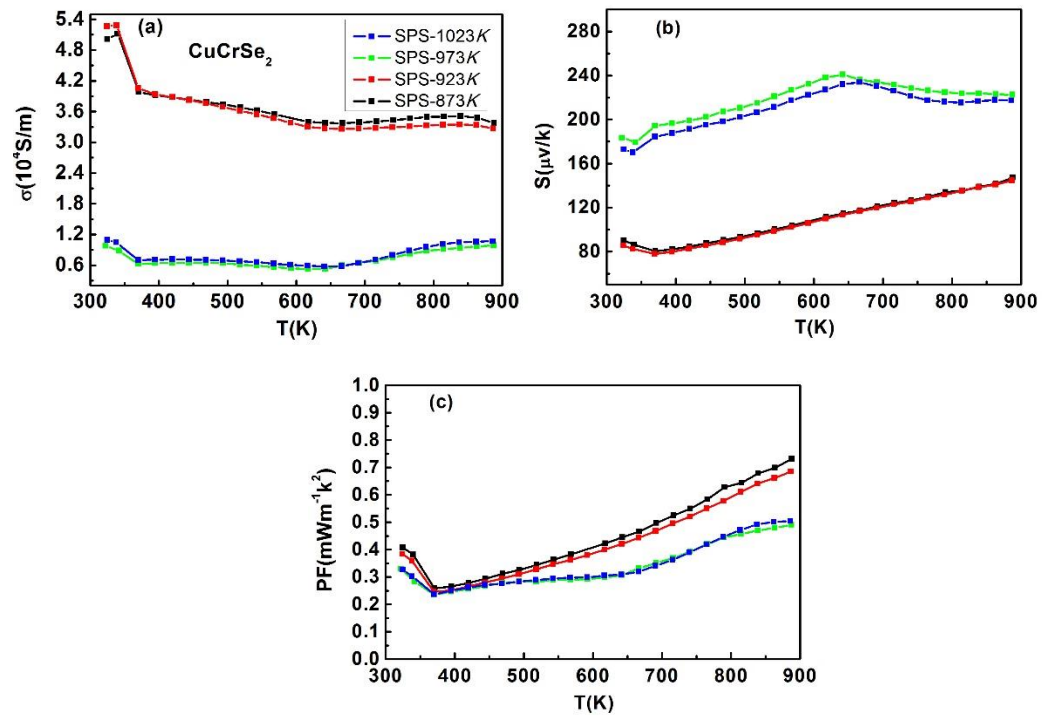


Fig 3. Temperature dependences of the electronic transport properties of all samples sintered at 873K, 923K, 973K and 1023K: (a) electrical conductivity. (b) Seebeck coefficient. (c) Power factor. The observed abnormalities at ~ 365 K indicate the onset of the disorder of Cu^+ in CuCrSe_2 .

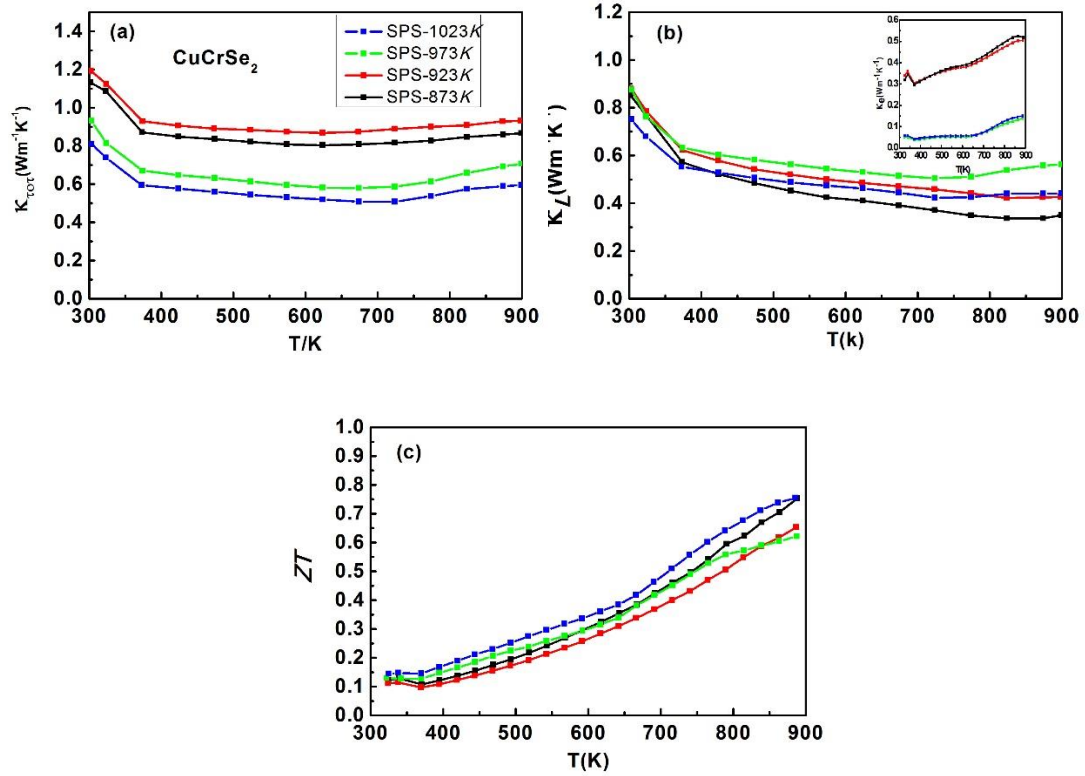


Fig. 4 Temperature dependence of all samples: (a) total thermal conductivity. (b) lattice thermal conductivity. (c) dimensionless figure of merit ZT for samples sintered at 873K, 923K, 973K and 1023K.

Supplementary Material

Sintering temperature dependence of thermoelectric performance in CuCrSe_2 prepared via mechanical alloying

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1. Phase diagram of Cu-Cr-Se system

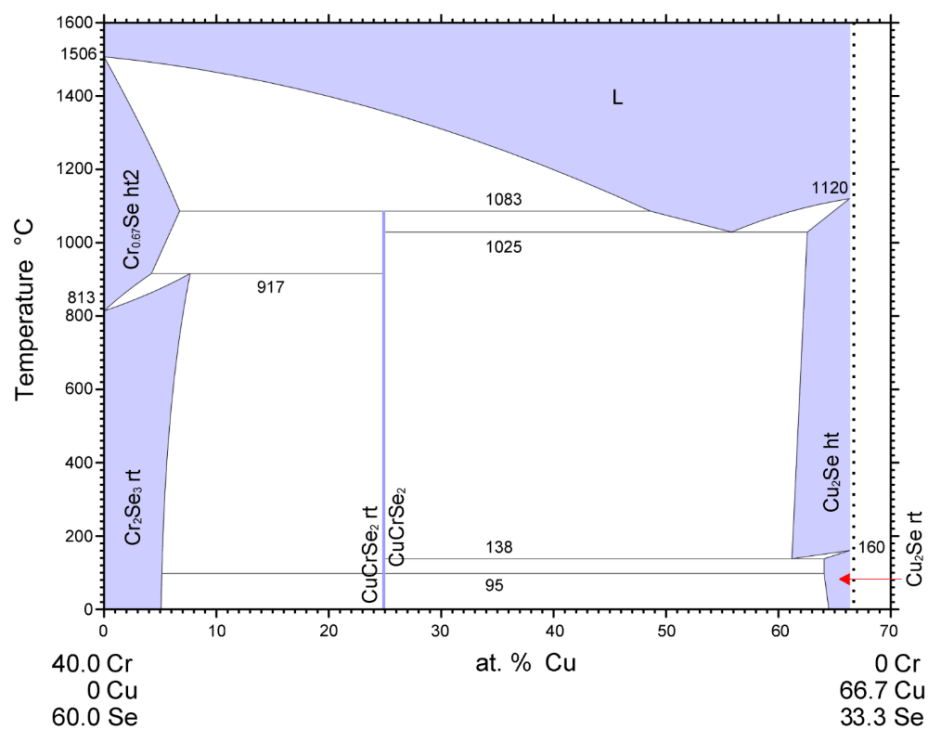


Fig. S1 Phase diagram of Cu-Cr-Se system.